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AGRICULTURE

OF

NEW-YORK:

COMPRISING

AN ACCOUNT OF THE CLASSIFICATION, COMPOSITION AND DISTRIBUTION OF THE SOILS AND ROCKS,

AND THE NATURAL WATERS OF THE DIFFERENT GEOLOGICAL FORMATIONS;

TOGETHER WITH A CONDENSED VIEW OF THE

CLIMATE AND THE AGRICULTURAL PRODUCTIONS OF THE STATE.

BY EBENEZER EMMONS, M. D.

VOLUME II.

ALBANY:

PRINTED BY C. VAN BENTHUYSEN. 1849.

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TO HIS EXCELLENCY JOHN YOUNG,

Governor of the State of New-York.

SIR,

The volume which I now place before your Excellency, is devoted mainly to the composition of the inorganic parts of vegetables. It is designed to aid the farmer in his immediate avocation. I hope, Sir, that the counsel which you have imparted in its preparation may not have been lost. It will be a source of satisfaction to me if I have succeeded in fulfilling your wishes, and have at the same time performed my duty acceptably to the farmers of New-York.

I am, Sir, your most obedient servant,

E. EMMONS.

ALBANY, November 1, 1848.

•

PREFACE.

The preparation of this volume has occupied two years. To some, this may appear more than sufficient to have accomplished what is here recorded. Whatever view may be taken of it, I believe I am justified in saying that I know not how it would have been possible to have increased very materially the amount of labor in the time specified above. The law, whether right or wrong, required its completion on the first day of October last. The work would have been improved, and more valuable, had it been possible to have given more time to some subjects; inasmuch as the disproportion of labor would have been diminished, and each subject have received its due examination. But this, I repeat, has been impossible, notwithstanding the fact that I have employed an assistant for the whole period, and have continued the work about four months beyond the time specified in the act authorizing the continuance of the Survey.

Mr. J. H. Salisbury, who commenced the study of analytic chemistry three years since in my laboratory, and whose services were noticed in my last volume, has continued them, and his work is distinguished by the initial of his name. He has also kept for me the observations on temperature, as recorded in the appendix, for this place. L. C. Ball, esquire, of Hoosic-falls, has also contributed observations and analyses for this volume. Mr. C. B. Salisbury, of Scott, Cortland county, is deserving of the thanks of the community for his observations on the temperature of the soil of that place, which is supposed to be elevated about 1200 feet above tide at Albany.

It will be observed, that in some of the analyses of the ash of wood, the footing is too great. This arises from calculating an amount of carbonic acid sufficient to saturate the lime. There are cases where the carbonic

Vi PREFACE.

acid was not actually obtained; and this result, showing an excess in the amount of the elements over that employed in the analyses, is due to the caustic or subcaustic state of the ash. The error is in the carbonic acid, and not in the essential elements. Carbonic acid is sometimes referred to as if it were one of the original elements of the wood, while in reality it usually results from the conversion in combustion of an organic into carbonic acid. So in the case of sulphuric acid, it is not oil of vitriol in grain or straw; but when burned, the sulphur of the proteine compounds acquires oxygen, and hence it is often a secondary result.

The irregularity in the numbering of the plates has arisen from the fact, that when I began the volume, it was intended to embrace the descriptions of the fruits; but it was ascertained that the volume would be too large if the original intention was carried out, and hence it was deemed advisable to give the matter treating of the fruits in a separate volume. This part of the work is considerably advanced, and will be ready for publication in the course of the year.

ALBANY, February 1, 1849.

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REPORT

ON THE

AGRICULTURE OF THE STATE OF NEW-YORK.

PART HI.

CHAPTER I.

PRELIMINARY OBSERVATIONS.

THORGANIC ELEMENTS OF VEGETABLES: THEIR NUMBER FEW; THEIR CHARACTER, ORIGIN, AND SOME OF THEIR COMBINATIONS. SILEX, ITS RELATIONS AND USES IN PLANTS. ALUMINA, LIME, MAGNESIA; PHOSPHORUS, SULPHUR; IRON, MANGANESE; POTASH, SODA AND AMMONIA; CHLORINE, CARBON, OXYGEN, NITROGEN AND HYDROGEN. CARBONIC ACID AND ORGANIC MATTER; THEIR UTILITY IN SOILS. VEGETABLE SUBSTANCES: STARCH, LIGNIN, GUM, DEXTRINE; EXTRACTIVE AND COLORING MATTERS. ANIMAL SUBSTANCES: PROTEIN; ALBUMEN, FIBRIN, CASEIN, GLUTEN, GELATIN; BLOOD; MILK.—RECAPITULATION.

In the preceding volume of the Agriculture of New-York, I attempted to give the results of many analyses of the soils peculiar to the State, the object of which was the determination of the amount of nutritive matter contained in them. This seemed to be required, as one of the first steps which should be taken to elucidate, in the order most natural, the principles which lie at the foundation of all improvements in husbandry. That the composition of soils requires a full and perfect determination, is now admitted. It is only by possessing this knowledge, that the farmer can cultivate his lands understandingly, or cease to work empirically or by rote. A knowledge of the composition of soils is not all that is requisite to good and profitable farming: there is still remaining an entire field of facts, to which the husbandman should by no means remain a stranger. These facts relate

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to the food of plants. There is first that general store-house of inorganic matter, from which all kinds of plants select the peculiar elements which they respectively require for the nutrition of their bodies. We may note, in the first place in this branch of inquiry, that different families of vegetables require different sorts of food. That which nourishes the wheat plant, is not so essential to flax and hemp, or to the potato. It is true, that in a majority of cases, the same substances may be found in different families of plants; but there still remains the fact, that the proportions required of a given element differ greatly. It is then quite important to know what is especially wanted in a given vegetable, as wheat, rye or indian corn. In addition to the study here briefly alluded to, we, unconsciously, as it were, extend our inquiries to the conditions on which life and organization depend. The distribution or destination of nutritive substances is governed by law. We are not left for a moment to suppose that nutriment is borne to each organ in the same degree, and in the same amount. This, as will be seen by reference to analyses to be detailed hereafter, is beautifully illustrated in the cereals: we find only traces of the carbonates of lime and magnesia in their seeds or kernels, while the ash of the stems and leaves furnishes both in considerable quantities. So carbonate of lime is in great abundance in the bark of most trees, while it is less in quantity in the wood, and still less in leaves, flowers and seeds. Such results could never have been secured, except under conditions that have been imposed by laws which regulate the life and growth of all living beings.

In the execution of the plan of this work, I shall proceed to detail the researches I have made on the composition of the inorganic parts of plants which have grown upon the soils of this State, the general characters of which have been already given in a former volume. I desire to connect together, as far as possible, the soil and its products; or, in other words, to trace the relations which exist between the soil, and the living beings which vegetate upon it. That this desire might be carried out as far as practicable under existing circumstances, repeated additional analyses have been made of the same soil when its immediate products could be obtained. I deemed it highly necessary to investigate minutely the products of the soil of each of the several districts, for the purpose of ascertaining the adaptations of each. The work, however, is too great to be executed in the time allotted to the survey, but it is hoped that enough has been done to aid the farmer to a certain extent. This will not appear improbable, when it is known that the agricultural districts possess points of great similarity, and that the most essential properties of soils are common to each district respectively. So great indeed is the uniformity, that we may calculate the quantity of certain elements which exist in their respective areas, and hence determine in each instance how long it will answer to cultivate exhausting crops.

It is scarcely necessary to say, that we arrive at a knowledge of the most essential part of the food of plants by an analysis of the ash. Although the carbonaceous matter which is burned off is also important, still, so far as inquiries of the capabilities of the soil are concerned, the carbonaceous are of less importance than the inorganic and fixed parts; for so essential are these fixed matters, that the very existence of vegetables depends upon them. They constitute the frame work or skeleton, and can not be dispensed with; and hence

where lands have been cropped for a long time, these matters become so far diminished in quantity that a profitable cultivation ceases. The object, then, of the analysis of plants is to ascertain what elements enter into their composition, and how much is requisite to give that perfection which is most profitable to the cultivator. And it is upon this kind of knowledge that the most profitable rotation of crops is based; for, as has been stated already, plants differ among themselves as it regards the special elements which they require, and also in the relative amount of those elements; and hence it is possible to devise an expenditure of a store of food in the soil, which shall ultimately end in a great saving of labor, as well as an increase of direct profits.

To carry out the objects here alluded to, numerous analyses of plants have been undertaken, the results of which it is proposed to give in this volume. I should perhaps proceed at once to this part of the work; but it seems best, in the first place, to state somewhat in detail the characters of those elements which constitute so much that is essential to the plant. I shall first recount all that seems to be essentially necessary for the farmer to know concerning the functions or offices of the elements of plants; and I do not, therefore, here take it for granted that my readers are already familiar with these subjects; for experience in these matters not unfrequently proves that our readers or hearers know less than we expected, and, in fact, are ignorant of many things of which we by no means had supposed it possible. I would wish certainly to pursue a plan in which plainness and simplicity shall appear prominent, and which shall be as free as possible from subjects that have not been explained to ordinary readers.

We may then regard it as an essential part of the attainments of a farmer, to possess a knowledge of those bodies which enter into the composition of the plants which he cultivates, and even of all plants, inasmuch as they minister directly or indirectly to his interests. Regarded in a general point of view, it is not difficult to see in what way they affect his interests; even in vegetables, though they do not contribute to his nourishment, or that of his cattle, still their office must be taken into account. It is better that a field be covered with vegetation, than to lie naked. In the former case, the growing plants bring up from beneath, and to the surface, elementary substances which, if properly managed, can not fail of contributing to the growth of a succeeding crop. That which it is necessary to know upon this subject, is what particular nutritive elements are brought to the surface, which, by decay, will remain for the improvement of the crop we propose to cultivate. What elements, for example, does the growing buckwheat bring up to the surface, which, when ploughed in, shall become food for wheat or maize? This question can be answered only by an analysis of the ash of the plant itself. This being correctly performed, we shall then know not only what elements are mixed with the surface soil, but we possess facts by which we can determine the amount we have in the crop of buckwheat also. So too we may know how rapidly we exhaust the store of food by the continual raising of a given crop. By such researches, we connect the character of the elements with that of the vegetable of which they form a constituent portion, and we connect the elements of the vegetable with vegetable and animal life.

It is a remarkable fact that those elements which are of the highest importance in the animal and vegetable economy, form but an inconsiderable part of the soil. The tissues of the animal frame are composed of the salts of phosphorus, which, in order to be detected in the earth, require the utmost nicety in chemical manipulation; while the compounds of lime, magnesia and silica, are far more abundant.

It can not be said that the elements forming the essential part of the soil are numerous; and hence it may be inferred, that the elements which constitute the frame work of vegetables and animals are also comparatively few in number. Of those which exist in the soil, the following enumeration comprehends the entire list, so far at least as can be regarded as essential to organization: Silex, alumina, lime, magnesia, iron, manganese, potash, soda, phosphoric acid, sulphur, chlorine, carbonic acid, ammonia, the elements of water and air, and nitrogen, oxygen, hydrogen and carbon in an oxidized state. They may be regarded as forming very naturally two classes of bodies: those which form the frame work of the tissues; and those which fill up, as it were, the interstices. In the former, or in the frame work of the tissues, we find silex, lime, magnesia in combination with phosphoric and carbonic acids; in the latter, carbon, hydrogen and nitrogen in various combinations are found. It is an important fact, established by the analyses which I am about to give, that each tissue has its own distinct organization; that they respectively possess their elementary bodies in proportions peculiar to themselves. For example, the constituents of the leaf of a currant differ from those of the stem, bark, flower or wood; the kernel of wheat, from its envelopes, the husk or stalk. But what is practically still more interesting and important, these component parts may be modified by culture and by soil. The law, however, that certain elements or bodies are determined towards specific parts, is unaffected: that the phosphates of magnesia and lime are determined to the essentially nutritive parts of the plants, is a universal fact; the modifications produced by a rich soil and high culture are mainly seen in their accumulated quantity, and it is not very unlike that process called the laying on of fat in certain parts by some breeds of domestic animals. We shall have occasion to treat of this interesting subject more at large, being now more immediately concerned with the elementary bodies alluded to in the preceding paragraphs.

SILEX, SILICA, SILICIC ACID.

These three names are synonymous. The substance is familiar to all, in the forms and conditions of white sand, flint, and crystals of quartz. In those forms it is commonly regarded as pure silex, and is so called; but when it is combined with other bodies, or is in a soluble state, it is called *silicic acid*, because it performs the functions of that class of bodies which are known as acids, and not because it has the taste which acids are commonly supposed to possess; for it is really tasteless, and produces no change upon litmus.

Silex is scarcely soluble in water, or in the ordinary acids; but it is perfectly soluble in fluoric acid, and in potash when aided by heat. The alkalies give it a solubility which it

retains as long as it is moist; but when dried, and especially when ignited, it becomes insoluble again: hence it is often spoken of as soluble and insoluble silica. In silicated plants, as the cereals, it retains its solubility if they undergo in the ground a slow decay, and we may dissolve it in our analysis along with the phosphates. If, however, these vegetables are burned, their silica becomes insoluble mostly, although it is minutely divided.

Silex is white when pure, and harsh and gritty to the feel, but fuses easily with soda into a transparent bead. It is also dissolved in the hydrated vapor of fluoric acid, for which substance it is regarded a test.

In this connection, it is proper to speak of silica as a constituent of soils, and of its uses and functions, as a part of the vegetable tissues. It has its mechanical importance in both relations; and so abundant is it both in its separate and combined states, that it must be regarded as one of the most essential bodies in nature.

- 1. Silica as an element of soil. In quantity it forms more than 60 per centum, and sometimes its percentage is as high as 95. Taking its average range as about 78.79 per centum, we find it entering more largely into the composition of soil than any other element. In this fact, we discern that its function as an earth in the midst of earths is important: it must impart its own characters to the compound. In itself, silica is a dry, white, harsh-feeling powder, destitute nearly of affinity for water, and admitting the free passage of fluids through it without affecting them in the least. It is then an element which is designed to give porosity to soil, in order that water and air may be admitted into its texture. If soils contain too little of it, they are close and impervious; if too much, water percolates too rapidly through them. Soil is not tempered with an excessive dose of silex when it amounts to 85 per centum: above that proportion, the soil becomes rapidly porous and loose, and can not be cultivated without annual additions of manure.
- 2. Silica as an element of organized bodies. Silica or sand is not taken up by the roots of plants as such, in consequence of its insolubility; but it requires to be in combination with other substances, in order to give it this property. What these substances are, has been stated already, namely, potash and the alkalies. Silica then enters into the composition of vegetables, though not in equal proportions in different parts of the same plant. Its presence can not be regarded as accidental; for instance, in 100 parts of the ash of the straw of the creeping wheat, it amounts to 69.66; in the grain, to only 2.56. In the ash of forest trees, it is never half as much. In the stalks of all the cereals, however, it exists in large percentages. It gives strength to the straw, and may be regarded as the element which supports and protects it. In order that the grains and grasses may take up silica, it is necessary that the fluids of the soil should be able to dissolve it.

Although an immense quantity of silica exists in a free and insoluble condition, and it would seem necessarily so, yet, in the constitution of the globe, provision is made for its solubility. This provision may be seen in its numerous combinations with the alkalies and alkaline earths. With these it has been united by fusion; and, as we have already stated, this is one of the modes resorted to for giving it solubility. In the natural progress of the decomposition of the silicated earths and alkalies in a moist soil, and by the at-

mospheric influence, a large portion of this combined silica, when set free, retains its solubility. New or virgin soils are particularly rich in soluble silica and the elements of carbonic acid: hence the cereals find in them their best location, and yield abundant returns. When tillage has exhausted them of silica in its soluble state, the straw is weak, and the harvest fails, unless artificially supplied. All plants take up this earth, but it is in the cereals and grasses that it abounds. In some the quantity is so small, that it may be regarded as accidentally present.

But I may yet remark, that many bodies which contain potash are easily decomposed by the ordinary atmospheric agents to which they are exposed; thus, granite, which is in part felspar, is a remarkable instance where decomposition furnishes soluble silica. The potash with which it is combined in the hard rock, yields to the action of carbonic acid and water. The same may be said of hornblende and basalt, or of the pyrogenic rocks as a class. The clay slate, or the slates of the Taconic system, as well as those of the higher classes in the New-York system, furnish both silica and potash by decomposition: hence a glass may be formed which is very soluble, and may be used as a manure.

A remarkably striking instance of decomposing action is that of carbonic acid upon the hardest of substances, such as that which is constantly exhibited upon the tumblers used for dipping the carbonated waters of Saratoga. These, in the course of a few days' use, lose their transparency, and look as if they really required washing before they would be fit for dipping. It is the alkaline matter of the glass which is attacked in the first instance. The silica, however, is all of it soluble, and is slowly washed away. This action will be increased in proportion to the amount of alkali used in the composition of the glass.

The formation of silica, as described in the foregoing paragraphs, is an illustration of the mode by which soils have been produced. Only a small portion, however, of the debris of rocks is in a condition to become the food of plants at any given period. The process is slow, but, as must be seen, it is the one best adapted to the wants of vegetation.

The most important and interesting fact in regard to silica, is the two distinct chemical characters which it possesses; that of being soluble in water and weak acids at one time, and almost insoluble at another. In one state, it is the sustaining and protecting agent in the tissues of a vegetable; in the other, it sustains the root and whole plant, and is the medium through which nutritive matters are introduced, that is to say, it forms the soil in which plants are destined to grow, and at the same time affords them an outward mechanical support. It is in the most delicate and useful vegetables, the grasses and grains, that silica forms a large proportion of the tissues, or in those plants where it seems lime would be insufficient to perform the same office.

ALUMINE, ALUMINA, SILICATE OF ALUMINA, OR CLAY.

It will not be far from the truth, to assert that alumine or clay possesses characters opposed to those of silica. Whether true or not in its full extent, it is certainly less soluble

than silica, and very rarely if ever enters into the constitution of vegetables, so as to form an essential part of their tissues. If its functions as a part of the soil are also taken into account, it is found to differ essentially from the former. Silica, as already observed, preserves the soil in an open or free state: alumine, on the contrary, makes it consistent, compact and impervious. The functions of this body are confined to the soil. It is necessary to observe that it is not the pure alumina of chemists which forms so large a portion of what is termed clay soils, but a silicate of alumina. Alumina, however, exists in two conditions even here: one of which is easily soluble, and may be called free alumina; the other is comparatively insoluble and fixed, and before it can be fully dissolved even in the strong acids, requires fusion with the alkalies or alkaline earths.

The facts most important to be known in the history of alumina, are the following: First, its insolubility in water; secondly, its neutral action on plants; thirdly, the impervious condition it imparts to soil; fourthly, its uniform state of fineness; and, fifthly, its high affinity for water. It never forms a gravel, nor is ever found in coarse particles like the gravel of silica. This great degree of fineness is seen when it is diffused in water: weeks are sometimes required for the water in which it is diffused to become clear.

It is unnecessary to dwell longer upon the properties of alumina: the most important facts, though briefly alluded to, are sufficient to answer the objects in view.

The two foregoing bodies, silica and alumina, form the basis of all soils, and to their presence the soils owe their main and permanent characters. The presence of other elements produce but trifling modifications. Organic matter exerts a greater influence than any of the earths. The body or basis being thus constituted, the remaining elements, along with soluble silica, may be regarded as food, or as matter out of which the frame work of the various tissues is formed. Although this is without doubt true, yet it is proper, before leaving the subject, to say of soil composed entirely of those bodies termed nutritive, that if composed of either singly, it would form but a barren waste. A pure limestone soil, or one mixed with silica and subcarbonate of lime, would be infertile. It is not a matter of indifference what constitutes the basis or substratum of soils. This must be a peculiar medium having certain relations to water and other bodies, which, if not secured by and in their physical or mechanical properties, would render nugatory the labors of the farmer.

Most of the earths possess low absorbent properties. Alumina, however, when pure, or as a silicate, ranks high in this important particular. The rapid absorption of ammonia by clay, renders it a receptacle or reservoir of this element. Odors, too, may be expelled from clothes and other articles, by burying the articles in aluminous soil.

LIME.

Caustic lime, or the subcaustic hydrate of lime, is never an element of soil. Its affinity for carbonic acid permits it to exist only as a carbonate, or some neutral salt of lime.

Carbonate of lime, if obtained from pulverized limestone, would have nearly the same mechanical effect in soil as sand. It has but little affinity for water; and hence so far as it exerts any mechanical agency, it operates like sand. In the soils of New-York, and more especially in those of the New-England States, the quantity of carbonate of lime is so small, that it has no perceptible mechanical influence. But there is another form of carbonate of lime, which exerts a decided influence: this is marl, a carbonate of lime, which is in a state of fine subdivision, and is combined with 4 or 5 to 10 per centum of organic matter. This substance, which is esteemed highly as a fertilizer, is a powerful retainer of water: it even ranks higher than alumina. It truly deserves the reputation it has acquired as a fertilizer, though we doubt whether its real action in the soil has been understood.

Carbonate of lime, then, in the form we should obtain it by grinding rocks of limestone, would operate only mechanically like fine sand, by giving more porosity to the soil. In the form of marl, however, it gives tenacity to soil, by increasing its retentiveness.

Carbonate of lime, as a salt of the earth, owes its importance to the relations it sustains with organized bodies, as is shown by their analyses. It is almost always an element, and a most essential one, of the animal tissues. Thus in the lower orders of the animal kingdom, lime is necessary to form the shell, covering or habitation of the species; in the higher, the bones are composed of salts of lime, all of which are derived from the soil. So in the vegetable kingdom, salts of lime are abundant in the ashes of all kinds of woods, and especially so in that of bark. We may therefore regard this element as one of the most important, and one which must be present in all good soils.

Carbonate of lime is insoluble in pure water, but the farmer is under no necessity to provide the means for its solution. Rain water carries down to the earth carbonic acid, the presence of which enables water to dissolve it. So probably the development of carbonic acid in the soil itself may aid in giving solubility to carbonate of lime. I need not, however, dwell upon this subject, as, under carbonic acid and the organic matters of the soil, I shall have occasion to call the attention of the reader again to it.

The salts of lime, found in soils, are the phosphate, crenate, apocrenate, carbonate, and perhaps the humate and silicate. In weak acids, lime is quite soluble, but they form salts with different degrees of solubility in water. These salts are never in excess in any of the New-York soils: indeed they are really deficient, and, in good husbandry, have to be added in forms which are considered as a manure. In fact, one of the great efforts of the farmer is to supply lime in sufficient quantity to meet the wants of his crops.

Few subjects have enlisted the attention of agriculturists, so much as the use and effect of lime in and upon soils. The facts very generally go to prove its great value: its action, however, has not been so generally understood. A subject which involves many intricate questions can scarcely be expected to obtain for itself an uniform opinion, or a theory which all will readily adopt. Analysis proves the constant presence of lime in vegetables: hence there is no doubt that it should be present in all soils, to supply the wants of vegetation. But its use and functions do not terminate in supplying a material for nutriment: there are certain reactions of lime upon other elements in the soil, which equal in utility the one

just referred to. This reaction is upon the silicates of potash, and other alkalies. It has been shown by Prof. Fuchs of Munich, that the silicates of alumina and the alkalies are decomposed by mixture with milk of lime in a subcaustic state. The same action takes place when lime is mixed with the soil. There is, therefore, a liberation of silica and of potash, or the alkali with which the silica is combined. Besides this chemical reaction, the texture of clay soils is loosened, and the whole mass becomes porous and friable.

The functions of lime, then, it will be seen, consist, first, in supplying an element necessary to the plant; secondly, in liberating the alkalies in combination with silica; thirdly, in rendering the freed silica soluble; and fourthly, in giving porosity to argillaceous soils. Too much importance, therefore, can scarcely be given to lime as an agricultural agent. There are probably other actions and uses of a minor importance, which may be passed over without particular notice. Some of the uses attributed to lime seem rather problematical, or not well sustained by facts.

I have already intimated that all plants require lime. I do not know that this intimation requires modification or reserve. It is important, however, to state that the different parts and organs of plants require it in different proportions. The wants of a tree for lime can not be determined by an analysis of its wood or leaves, nor by that of its bark. In the outer coverings it is always greater than in the internal parts. Its presence is therefore functional and necessary.

In the foregoing remarks, I have had in view one variety of lime only, the subcaustic or subcarbonate in a hydrous condition. It is in this condition only that it is usually employed. I except here of course the phosphate of lime, as contained in bones, which exerts more of a physical effect upon the clay soils than the hydrous subcarbonate does.

Many farmers have used lime without benefit. In view of this fact, it may be stated that at the time the lime was used, it was not required, there being already sufficient in the soil for the purposes of vegetation, or for the use of the crop to be raised. Whenever there is a deficiency of the silicated alkalies and of organic matter, the effects of lime are not so apparent.

MAGNESIA.

The uses of magnesia, either as an element of soil or as a constituent of plants, are not so well known as the use of the salts of lime. It can exert but little influence mechanically on the soil, its proportion being very much smaller than that of lime; and though magnesian rocks are not unfrequent, still it never occurs in quantities sufficient to produce any perceptible effect in the texture of the soil. Magnesia, however, appears to be an essential constituent of the grains, as indian corn, wheat, etc., in which it even exceeds lime in quantity; thus, in the ash of rye it amounts to 12 per centum, while the lime is only 2.61. The carbonate requires 2000 parts of water for its solution, and is more insoluble than carbonate of lime. All the rocks furnish magnesia by their decomposition.

The most common form in which magnesia enters into the composition of plants, is that [Agricultural Report — Vol. II.] 2

of a phosphate. Whether the phosphate is formed in the soil, or in the plant, is a point difficult to determine. It is an element of food, and not of mechanical support. It is required as a constituent part of the grain; and the grain itself, in consequence of its composition, becomes an important nourishment for animals. It is necessary, however, to say that magnesia enters into the composition of the straw of the cereals, but in small proportion to the quantity in the grain. So it is found in the ash of fruit and forest trees, but the bark is almost destitute of it. In animals, it enters into the composition of the harder parts, as the internal and external skeletons, but in less proportion than lime. It forms salts with the same acids as lime. We have no facts which go to show that lime may replace the salts of magnesia, or the contrary.

The state in which magnesia is required by the cereals, is that of a phosphate. No grains contain the carbonate or other salts of magnesia, except in combination with phosphoric acid. I have so arranged the analyses of the cereals, that this fact may assume its proper importance. It will be seen, on consulting these analyses, that both magnesia and lime exist in the straw and chaff, while in the kernels they appear only in traces as carbonates.

PHOSPHORUS, PHOSPHORIC ACID, PHOSPHATE OF LIME, &c.

Phosphorus, which is one of the most remarkable of the simple substances or elements, exists in all organized bodies. It is never found simple in nature, and never enters as such into organized bodies. We have therefore nothing to say of it as such; but on its compound forms, or combinations with oxygen, lime and magnesia, and some other bodies, it is necessary to dwell for a moment. In plants, it is found combined in the four principal alkalies and alkaline earths, viz. lime, magnesia, potash and soda. It is also in combination with iron.

If the general presence of a substance is an evidence of its importance, the phosphoric compounds rank high in this respect. It is, however, in the fact that grains are its main receptacles, that its real importance is indicated. In respect to the quantity present in any particular part or organ, its distribution follows the same law as that which governs the distribution of other bodies; thus it is present in the wood, bark, leaves and fruit, but not in equal quantities. In the bark of trees, straw and chaff of grain, it is comparatively trifling in amount. In the leaves of many plants it is abundant. In all edible parts, however, it forms the largest proportion of the whole inorganic matter. These facts indicate very clearly the utility of its presence in soils; and not only this, but its amount; for as it is comparatively only in small quantities that it is always found, it is probable that it is one of the first which becomes exhausted by culture. In most soils, the phosphates are only appreciable, or just susceptible of being weighed in one hundred grains. A total absence of phosphates in a soil will render it barren and unproductive. It is not improbable that phosphate of lime may replace and become a substitute for that of magnesia, when the latter is very deficient in quantity.

On the general distribution of the phosphates, it is quite interesting to know that the phosphate of lime is distributed to the outer envelopes of a plant, the bark, cuticle, rind of fruits, etc.; while the phosphates of magnesia and iron pass to the interior, or to the seed, kernel or grain. These seem therefore especially designed for the animal tissues. Membrane and bone, blood and muscle, must contain a definite quantity of the phosphates in a healthy state; and when they are deficient in quantity, the bones are soft and flexible, and refuse to support the weight of the body.

Much has been said of the importance of supplying nitrogenous matters to soils. The phosphates can never be derived from the atmosphere, or from rain water, nor are they furnished by any of the ordinary processes of nature: hence, if they become exhausted, direct application must be made of some substance which contains them. They must be supplied by labor, and by somewhat expensive materials. Nitrogenous matters, however, are constantly being added to soils, by unceasing operations or agencies independent of man. The rain brings down ammonia to the earth, so that nitrogen can never be totally absent from any soil, though it may be deficient in quantity, or insufficient to supply the wants of artificial crops; but the phosphates can not be obtained when the soil has been exhausted of them, except by waiting the slow decomposition of rocks, or by a direct application of some material which contains them. It is towards this one thing, the supply of a sufficient quantity of substances abounding in the phosphates, that the farmer should direct all his measures. Other substances, though they may be important, yet are frequently so abundant that there is really no difficulty in obtaining them. This is the case with sulphuric acid, abundance of which exists in gypsum, and other cheap salts of the alkalies and alkaline earths.

Not intending, however, to diminish the weight of opinion in regard to the importance of adding nitrogenous matters to soils, it is notwithstanding proper that correct views should be entertained upon this subject; and I think that the statement of Liebig, in his work entitled "Chemistry in its applications to agriculture and physiology," puts the question in its proper light. He states, p. 75, that "the most decisive proof of the use of strong manure [strong in nitrogen] was obtained in Bingen (a town on the Rhine), where the produce and development of vines were highly increased by manuring them with such nitrogenous manure as the shavings of horn, etc.; but after some years, the formation of wood and leaves decreased, to the great loss of the proprietor, to such a degree that he has long had cause to regret his departure from the usual methods, ascertained by long experience to be the best. By the manure employed by him, the vines had been too much hastened in their growth: in two or three years they had exhausted the potash in the formation of fruit, leaves and wood, so that none remained for future crops, his manure not having contained any potash. There are vineyards," he remarks, "upon the Rhine, the plants of which are above one hundred years old, and all of them have been manured with cowdung, a manure rich in alkaline ingredients, but poor in nitrogen." The same system of manuring would end in the same disaster in all cultivated crops.

SULPHUR, SULPHURIC ACID.

It is now well known that sulphur and its combinations play an important part in the economy of life: hence it is another product to which the farmer must direct his attention.

No better criterion exists relative to the importance of a product, than that it should be found constantly in the blood and animal tissues. In the case of sulphur, it is always present in albumen and fibrin; and these are the proximate elements which supply the waste of the body, and by which its growth is promoted and secured. Sulphur, as it exists in albumen, fibrin or casein, is not in an oxidized state, as is proved by the spontaneous decomposition of these substances. When undergoing this change, it is sulphuretted hydrogen which is emitted.

Sulphur exists in many of the useful vegetables, as peas and beans, and indeed in all leguminous vegetables, both in their juices and mature seeds. In another family of plants, also, sulphur is an essential ingredient, namely, the *Cruciferæ*, as horse-radish, mustard and scurvy-grass.

Sulphur, as in the case of phosphorus, must be supplied to the soil, by the farmer, whenever it becomes deficient in quantity. That plants invariably derive it from the soil, is proved by the fact that no compound exists in the atmosphere which can furnish it. Sulphuretted hydrogen may sometimes be detected in particular locations, but it is not a substance universally present, or at all adapted in that state to supply the wants of vegetation.

Sulphur is without doubt derived from the sulphates, of which there are abundant sources in the earth and rocks. Sulphate of lime, or gypsum, is one of the most common sources for supplying this substance to vegetables. The sulphurets of the metals may also furnish it by decomposition. There is, therefore, no want of materials from which sulphur may be obtained.

Sulphate of ammonia is regarded by Liebic as the substance best adapted for assimilation in the vegetable tissue. This opinion is founded upon the composition of this body. It contains nitrogen; and as nitrogen is also a constituent of fibrin and albumen, it furnishes both elements by decomposition. The simple removal of the elements of water, hydrogen and oxygen, enables the nitrogen and sulphur to pass over into the composition of the vegetable juices. In the case of gypsum, inasmuch as it is soluble in water, it may also be taken up by the roots of plants, and in their juices undergo decomposition in the presence of carbonate of ammonia.

The sulphurous, or, as they are usually termed, nitrogenous compounds, albumen and fibrin, are insoluble substances, out of the animal body, or after coagulation by heat. It is supposed that, in plants and animals, their solubility is maintained by the presence of the alkalies. The white of egg furnishes invariably free soda.

In all the changes and facts respecting the state, condition and growth of vegetables, it is not difficult to see the mutual adaptations of bodies to each other. In the case of sulphur,

these adaptations are of the most interesting kind. Sulphur is one of those bodies which exists in its simple and elementary condition in the vegetable and animal fluids. It is a rare instance, as most of the solid or fluid bodies are oxygenated, or otherwise combined with other elements. Sulphur, whenever it is procured by the burning of vegetable substances, is obtained in the form of sulphuric acid, oxidation having taken place during the ignition of the vegetable.

IRON, OXIDES OF IRON.

Iron is an essential constituent of the blood of all vertebrated animals; but whether it is equally essential to the invertebrata, has never been determined. It is sufficient for my purpose to know that it is found in all animals with red blood. Of its source or origin there can be no doubt: all soils contain it, and all vegetables have the power to take it up. From the vegetable kingdom, it finds its way into the animal. What special function does it perform in the soil?

As in many other instances, so in the case of the oxide of iron, its function is not to be considered as confined to the production of one single result. In the vegetable economy, its office must be regarded as the same; but in the soil, it undoubtedly aids or promotes the formation of ammonia. To understand the mode by which this result is brought about, we must consider that iron exists in two states, viz. in that of a protoxide and that of a peroxide. This fact has been fully established by many analyses; but there is no constancy in the relative proportion of the two oxides: these are found to vary. The two oxides are made to play conflicting parts. When the iron is at its maximum point of oxidation, organic matter in the soil robs it of its oxygen, and the formation of an organic acid is the result. When, however, it is in its lowest state of oxidation, its affinity for oxygen is so strong that it robs water of that element; and the hydrogen, being set free, combines, while in its nascent state, with the nitrogen of the air in the soil, and forms ammonia. This will be dissolved in water, or may combine with any free acid, as the carbonic or sulphuric, when it is fitted for the uses of vegetation, or is ready to enter the vegetable tissues. Such changes may be carried on so long as the soil is furnished with organic matter. The presence of iron, then, aids in furnishing ammonia; and were it of no use itself in the vegetable and animal economy, its function would still be highly important.

The proto-salts of iron are usually regarded as injurious to vegetation. This is certainly true when they exist in considerable quantities, yet in small doses they do not destroy vegetation: hence the injurious effect of a proto-salt may be owing rather to the quantity, than to its poisonous properties. These salts are, however, easily neutralized by the application of lime. A barrenness arising from an excess in quantity of these astringent salts of iron, may be immediately remedied by an application of the hydrous subcarbonate of lime, by which gypsum is at once formed, and the iron remains a simple protoxide with the powers and functions ascribed to it in the preceding paragraph.

MANGANESE, OXIDE OF MANGANESE.

It is not as yet determined that this oxide is an essential constituent of any class of plants. So far as this question has been investigated, it appears to be an accidental substance in the ash of plants. Like iron, it is soluble in organic acids, and forms an earthy black substance, the particles of which cohere but slightly. It is analogous to bog ore, being formed in the same way and in similar locations. The organic salt of manganese is undoubtedly the substance which is taken up by the roots of plants, and is occasionally detected in small quantities by appropriate tests.

POTASH.

It is now well known that soils destitute of potash are nearly barren; that, at least for the cultivation of some plants, they are totally worthless. Liebig has proposed to divide plants into groups or classes, according to the predominance of a particular earth or alkali. This has some show of a systematic arrangement; but when it is attempted to carry it out, it fails, as generally in the composition of the seed we find potash; in the leaves and stalks, lime, etc. If the whole plant is taken into consideration, it is difficult to determine whether it should be called a lime, potash, soda, magnesia or silica-plant.

The original source of potash, and indeed of all the alkalies, is found in the rocks, particularly clay slate, and those containing felspar. In the Hoosic roofing slate 3.52 per centum of potash exists; in a slate of the same age in Washington county, 0.60. Marls and clays contain 0.50 to 1 and 2 per centum of potash and soda; but one of the shales from the Chemung series contains 5.47 per centum.* I have reason to believe, however, that this large amount of potash is only local; for it not unfrequently happens that a single specimen yields a large amount of some valuable fertilizer, as phosphate of lime; but a few yards distant, it is only in diminished quantities.

For analyses of the rocks, clays, marls, limestones, etc., see Vol. I. of the Agriculture of New-York. Numerous analyses are also given in the American Journal quoted below.

Potash, besides its direct use as a constituent itself of vegetables and animals, is equally useful, if not more so, in rendering silica and other bodies soluble. Its absence, or its presence in insufficient quantities, is an evil which must be removed by a direct application of those substances which contain it. Ashes of plants furnish it more economically than other substances. Even leached ashes contain potash, and are competent to impart fertility for many years; and in this condition, too, they give tenacity to loose and sandy soils. Our neighbors of Long Island understand well the value of leached ashes; while the farmers in the valley of the Mohawk have not yet discovered their use, and the fact that they would do much to restore to fertility the worn out fields in this beautiful valley.

^{*}American Journal of Agriculture and Science, Dec. 1847, p. 343.

SODA.

Both potash and soda are bases for organic and mineral acids, in which combination they are connected with the growth and development of plants. The tubers of potatoes require both potash and soda, and, when grown in a suitable soil, they form a valuable food: if, however, they vegetate in the open air, a poisonous alkali is formed from the elements of the tuber, in which there exist mere traces of potash and soda.

The great source of soda is sea water and saline springs, where it is in combination with chlorine. It is also found in mineral bodies, in the same relations as those of potash: hence soils may be supplied with soda by the decomposition of slates, shales and clays.

It will be observed, on consulting the analyses, that soda is in less proportion in soils, rocks, and in the ash of plants, than potash. Plants growing in sea water and near brine springs, as at Salina and Syracuse, contain soda. The Salsola kali is common about the salt-pans and fields moist with chloride of sodium. It here finds its proper food, and is as flourishing as upon the shores of the Atlantic.

In rocks the percentage of soda is sometimes as high as 11.48, as in albite, a common variety of felspar; in mica, it is only from 3 to 5. Notwithstanding the apparent small percentage of soda in rocks and soils, we see that it has accumulated in immense quantities in some locations, as in the rock salt of Cheshire in England, Cracow in Poland, &c. The sea, however, forms the great reservoir.

It is maintained by many that potash and soda may replace each other, in case of an absence of either; that marine plants which naturally require soda, if cultivated far inland, take the vegetable alkali in place of the mineral. This, however, is a forced state; and the probability is, that in these cases the plant in a few years would cease to vegetate.

AMMONIA, VOLATILE ALKALI, HARTSHORN.

Ammonia, like pctash and soda, is an original constituent of the globe; but unlike those alkalies, it is constantly produced and destroyed by the affinity of the elements which compose it. Nitrogen and hydrogen being its elements, may unite whenever they exist in a complex substance, when that substance is decomposed. It is exhaled from animal and some vegetable matters in the process of decay, during which it is probably formed by the union of its elements, but in which it did not exist as ammonia when the decay began.

Ammonia is also exhaled from the deep interior of the earth. Its salts condense in and upon the fissures of the rocks near volcanic vents. Its vapor rises from the lagoons of Tuscany, in company with boracic acid. This fact, however, does not prove that it exists in masses and reservoirs in the interior of the earth: it may be formed in its bowels, by the decomposition of water and other bodies in which nitrogen is an element.

The importance which ammonia takes in the processes of agriculture, arises from the

presence of nitrogen, a substance essential to the composition of the nutritive elements, as albumen and fibrin. It is not the ammonia in its entire constitution which enters the tissues of plants, and exists in them as such: it is necessary that it should undergo decomposition, and part with its nitrogen, to combine with other elements for the purpose of forming the tissues. One of the salts of ammonia exists ready formed in the cererls, viz. phosphate of ammonia and magnesia. The bran is richer in this salt than the flour; and it is stated by chemists, that when horses have been fed upon bran for a considerable time, balls of phosphate of ammonia and magnesia form and accumulate in the large intestines. This circumstance, however, I believe is a rare occurrence.

Ammonia exists in the soil either as a carbonate or sulphate, according to circumstances: it exists also in a free state. Clay and oxide of iron both attract this substance strongly: they serve therefore to fix it, and prevent its speedy evaporation.

Ammonia exists also in the atmosphere, and is brought down in rain water and snow, from which it may be obtained by evaporation. This fact probably explains the adage, that "Spring snows are the poor man's manure."

An important means of fixing ammonia is furnished by the use of plaster or gypsum. Privies and stables are in a measure freed of ammoniacal odor by sprinkling plaster upon the floor, or about the place: this plaster becomes then doubly valuable as a manure. The ammonia decomposes the gypsum, by combining with its sulphuric acid. The farmer will always find it for his interest to employ plaster abundantly about his premises, or in all places where decompositions are going on.

CHLORINE.

Chlorine is united to sodium or sodium and water, or potassium, when it exists in organized bodies. It is one of the elements of common salt. It combines with many other bodies, as potash, lime, magnesia, ammonia, iron, etc.

Common salt has often been extolled as a manure or fertilizer: its effects, however, are not uniform, and hence conflicting opinions exist as to its true value in agriculture. It has also been highly spoken of as a remedy for worms and insects, but here opinions do not agree. It is probably more important to animals than vegetables. Chlorine is not abundant in grains and seeds: it is more so in the stalks, leaves, etc. Clover contains more than wheat. It is not improbable that much chlorine is lost in the process of burning. We know that salt volatilizes by heat, or is carried off in the vapor of boiling water; hence, in this combination, it is undoubtedly lost in part.

Chlorine exists in the soil in combination with sodium. In the vicinity of the sea, its presence is readily accounted for by the spray and vapor carried inland by winds, which, when very strong, have been known to carry a quantity sufficient to impart a saline taste to leaves, grass, etc. to the distance of many miles. So it may be transported in a state of more minute division from sea to sea, though it is only over or near the ocean that the atmosphere is sufficiently charged to denote its presence by nitrate of silver.

Hardness in well and spring waters is often due to the presence of the compounds of chlorine. Chloride of potassium forms an important element of tobacco; and indeed it is somewhat remarkable that both chlorides, that of sodium and that of potassium, exist together in such large proportions as they are found in this deleterious weed. One variety of tobacco was found by Will and Fresenius to contain 8.53 of chloride of potassium. The stalks of hops contain 9.64 per centum. The Saccharum officinarum contains a much larger quantity; amounting, according to Stenhouse, to 30 per centum.

CARBON, OXYGEN, HYDROGEN AND NITROGEN.

In various states of combination, these bodies constitute those forms of matter which are called organic. The most important, or those which are most generally distributed, are carbon and oxygen forming carbonic acid, oxygen and hydrogen forming water, hydrogen and nitrogen forming ammonia. In the present constitution of bodies, not one of these compounds could be dispensed with: they are universally diffused and present in some form or other, in the vegetable, animal and mineral kingdoms.

Carbonic acid is the source of carbon in plants. It is also the great solvent in nature for the hardest materials, such as felspar in granite. Its constant, though slow action, compensates for the rapid and powerful action of mineral acids. I have already alluded to this property of carbonic acid.

Water in itself, and as water, must be furnished to all living bodies, and there are but few substances in the mineral kingdom which do not require it; but when it is considered in its constitution and the decompositions which it is susceptible of, and the changes it can effect in other bodies, or in its actions and reactions, its influence and importance are exceedingly magnified and extended. The same may be said of ammonia, the great source of nitrogen in organized bodies.

I need not here dwell longer either upon these elements themselves, or upon their compounds*. Of carbonic acid, I would remark in this place, that I have some doubt as to the absorption of it by the leaves of vegetables; and even admitting that it is absorbed, I can not but maintain the position that the roots are the principal organs which convey it to the plant. Leaves may condense carbonic acid on their surfaces, without absorbing it. It is, however, a point upon which I do not propose to insist. The fact that it is necessary that it be supplied by the roots, I have no doubt will be readily admitted; and hence practically the materials which are capable of furnishing it, must be supplied where they are required.

^{*} See Vol. I, pp. 223 - 227 of the Agriculture of New-York.

ORGANIC MATTERS.

Fertile soils always contain organic matters. By this term, however, is not intended to be understood a definite compound, but a complex substance derived from the decay of organized bodies, and existing in various states from the recently dead vegetable leaf or stem, to the perfectly disorganized product or products which have been formed in the process of decay. These products become finally converted into organic acids, which are capable of uniting with bases, as potash, soda, lime, magnesia, iron, manganese, ammonia, etc. In the character of salts, these compounds are absorbed by the roots of vegetables; and they constitute, at least in part, their food, and minister to their growth.

Some theoretical chemists look upon the organic matter of soils as of little consequence. If, however, the subject is considered in all its bearings, I believe its importance will be conceded. There are important functions which organic matter performs in soil. In the first stages of decay, and indeed through all its changes, it is an absorbent of water and ammonia. Indeed, the absorbent power of a soil is in a direct proportion to its organic matter in a minute state of division. In this respect it ranks higher than clay: it preserves the porosity of soil; it is a source of carbonic acid; it aids in the decomposition of the peroxide of iron, by which ammonia is furnished. We therefore find it an active agent at all times, performing some of the most essential offices to the growing vegetable; offices, which, though they may not be regarded as vital, yet unprejudiced minds must admit are of the highest consequence.

SUBSTANCES PECULIAR TO THE VEGETABLE KINGDOM.

Starch. This well known substance is a product of many plants, in some of which it is quite abundant: this is the case with the potato, wheat, barley and oats. It consists of rounded grains imbedded in the cellular tissue, and entirely destitute of a crystalline structure. Starch is insoluble in cold water; and, hence, after the cells in which it is contained are ruptured, it may be washed freely, and obtained in a pure state. The starch which is deposited in the tuber of the potato, and in grains of the cereals, is changed by germination into sugar: this transformation is effected by the saccharine fermentation. This susceptibility of starch being converted into sugar, is turned to advantage by brewers, in the formation of alcoholic liquors. A seed, when it begins to germinate, absorbs water and swells up, and its temperature rises: it then absorbs oxygen, and evolves water and carbonic acid, and the starch gradually diminishes or changes into sugar. Soon the sugar itself disappears, by ministering to the growth of a sort of stem. The office of the starch in the seed, tuber and root, is to furnish nutriment, until the plant can obtain it by its radicle from the soil. So long, however, as the seed is kept in a dry place, its starch remains unchanged.

LIGNIN, WOOD OF PLANTS. There are various forms of this substance; thus, the cells of

the tissues, the fibres of flax, cotton, hemp, etc. are nearly pure lignin. To this substance the coloring matter adheres when employed with a mordant, as the acetate of alumina. It is in consequence of this affinity that colors are fixed, and continue unchanged for a long period.

Gum. Gum arabic is one of the most important gums. It is soluble in water; is in brittle transparent pieces, but never exhibits a tendency to crystallization. Gum may be changed into sugar by means of sulphuric acid, and also by the vital action of the plant which produces it.

There is a great deal of interest in the question of the changes effected on starch in the vegetable tissues. The origin of lignin is traced back to starch. It is known that the grains of starch possess a structure quite different from those substances which are formed by affinity. They have a structure analogous to certain animal organs, as the crystalline lens of the eye; and starch, when its changes are carefully noted, suffers a gradual transition into lignin, the arrangement of the granules being such that they form fibrous tubes in which considerable unaltered starch still remains adhering to the walls of the fibre or tube*. This peculiar change may be better understood, when the composition of starch is stated: it is composed of twelve atoms of carbon, ten of hydrogen, and ten of oxygen. Starch also being formed of concentric layers, the outer one continually increases in density by the absorption of water from the inner layers: this forms a space within it, which may only constitute a cell; but, in contiguous granules, they may form continuous cells, or, in other words, a tube.

Starch is regarded as the first stage in the organization of tissues. It may also perform other offices: it may be changed into sugar or gum, acids, oils, coloring matter. Changes of an analogous kind continually take place in the ripening of fruit. Let any one note the changes in an apple, or a nut, from its earliest period up to the perfect fruit, and he will witness numerous distinct products at different periods, which are changed or metamorphosed into each other.

Dextrine. This is regarded as a kind of gum, which is formed from starch. It is formed by digesting starch in sulphuric acid. The proportions employed, are, one of sulphuric acid, fifteen of water and five parts of starch; the mixture being heated to 200° Fahr., and maintained at that temperature for some time. The starch is perfectly dissolved, loses its peculiar character, and, instead of giving a deep blue violet color to iodine, imparts to it only a wine red.

Extract, or Extractive matter. Extract is a watery solution of several substances, dried down to the consistence of syrup. It is bitter and astringent usually, or sweetish bitter; but it can not be regarded as a proximate principle. It is an impure mixture of several substances. Many medicinal substances are obtained in this state, but they can never be regarded as chemical substances. The bitter principle of aloes, of hops, and of various roots, are complex bodies. Their composition always depends upon the mode in which they are prepared.

^{*} Kane's Chemistry, p. 652.

20 ALBUMEN.

Coloring matters. Coloring matter, as obtained, exists in two states: it is, first, a pre-existing substance of a particular vegetable; and, secondly, it is one which is formed by the chemical action of another substance upon a given vegetable product. Madder is an instance of the former; and perhaps indigo, of the latter. This substance, it is true, preëxists in the leaf of the *Indigofera* in a white condition, and so remains until the leaf begins to decay, when the white indigo absorbs oxygen from the atmosphere, which is known by the appearance of many blue points in the texture of the decaying leaf.

ALBUMEN, FIBRIN, CASEIN.

It is not my purpose to eulogize the age, or the men of the age; still it is a just tribute to science, and to men devoted to science, to say that the discovery that the three organic productions under consideration are indentical in composition, is really one of the greatest in modern times, especially when coupled with the fact that they are also identical with bodies of the same name derived from the animal kingdom. Vegetable albumen is the same as animal albumen, in composition. This discovery points out the source of these important bodies, and establishes clearly the offices delegated to the vegetable kingdom, namely, the elaboration of the fit elements of food, or nutrient matters, leaving the formative functions only to the animal; the power of shaping or moulding them, to the necessities imposed by nature. The animal consumes or eats what is essentially his own flesh, which is duly prepared in the grass of the fields. This subordination of the vegetable to the animal kingdom, is the greatest proof of design: it supports fully the doctrine maintained by Paley.

ALBUMEN. The animal and vegetable fluids abound in this substance. In the white of the egg, and indeed in all eggs and in all animals, it exists in its greatest purity. Without undergoing chemical changes, albumen may be said to exist in two states; one soluble, and the other insoluble. The first state is the natural one: it passes into the other at the temperature of 167° Fahr. If albumen is carefully dried at a temperature not exceeding 120°, it still retains its solubility, and may even be exposed, when thus dried, to a temperature of 212° without losing its solubility. When it is exposed to a temperature of 167°, it coagulates, and then becomes nearly insoluble. Acids also possess the power of coagulating it, except the acetic, tartaric and phosphoric acids, in each of which it is perfectly soluble. So also the alkalies and their carbonates form soluble compounds with albumen.

Albumen is usually detected in fluids by the application of heat. The addition of nitric acid also coagulates the albumen.

The four organic elements, carbon, hydrogen, nitrogen and oxygen, together with sulphur and phosphorus, constitute this body. The proportions which are given are

$$C_{400}+H_{310}+N_{50}+O_{120}+SP$$
.

The albumen of the blood differs from that of eggs, in containing one atom more of

sulphur. The salts contained in albumen amount to from 4 to 8 per centum, consisting of phosphate and sulphate of lime, and chloride of sodium.

The importance of albumen is directly indicated by the phenomena which occur in the incubation of the egg. It is a familiar fact, that in this process, all parts of the animal are developed from the albumen, inasmuch as no other nitrogenous body is present. Bones, muscles, feathers, nails, claws, the brain, the membranes and tissues of the body, and the blood, are all generated from this substance in a few days. The yolk is albumen intermixed largely with large yellow oil globules.

Fibrin. Fibrin is so closely allied to albumen, that it is regarded as chemically the same, although it presents some physical properties not found in albumen; thus, fibrin coagulates spontaneously from the blood when it ceases to circulate. By fibrin, then, it will be understood that I mean that clot or coagulum which forms when blood is drawn from a blood-vessel. As it exists in this state, it is mixed largely with the coloring matter of the blood. When it is wished to obtain it pure, blood may be shaken in a bottle with bits of lead or tin, when the fibrin will adhere in a fibrous mass, which may be washed with cold water till all foreign matter is removed. The fat which still adheres to it, may be dissolved out by ether. It is then a substance of a pale yellow, and devoid of taste or smell. Its chemical composition is the same as that of albumen. The fat associated with fibrin varies from 2 to 4 per centum.

Fibrin, like albumen, is always associated with a certain amount of salts. It is readily distinguished from other bodies, by its spontaneous coagulation.

CASEIN. Casein is an important constituent of milk. Curd of milk is casein combined with some foreign matters. To obtain it pure, milk is evaporated to dryness, and its butter dissolved out with hot ether. Dissolve the residue in water, and filter, and then throw down the casein with alcohol. When a solution contains casein, a pellicle forms upon the surface when heated or boiled: this is regarded as the effect of oxygen upon it.

The composition of casein, according to Mulder, is

$$C_{400}+H_{310}+N_{50}+O_{120}+S.$$

Casein is converted into albumen by digestion; and so, by the same action, albumen is converted into casein. It is precipitated from its natural solution by all acids, but is redissolved by the same. The most familiar case of precipitation, or rather coagulation of casein is that produced in milk by the mucous membrane of a calf's stomach: this is the method employed in the manufacture of cheese. The addition of an alkali to the milk would prevent coagulation. When milk is allowed to stand for some time, a naturally coagulated casein surrounds the butter vesicles, which are broken by agitation in stirring or churning, and collect in the form of butter.

Casein contains from 3 to 8 per centum of ash after incineration, consisting of phosphoric, carbonic and hydrochloric acids in combination with lime, magnesia and iron. To distinguish casein from albumen, it may be heated to a little over 167°, when it does not coagulate, but a pellicle will be formed upon the surface.

The three preceding substances, viz. albumen, fibrin and casein, are called proteine bodies; protein itself being regarded as a mere modification of them. They exist ready formed in the juices of vegetables, and may each be separated by suitable methods. Thus, juices which are newly expressed, and allowed to stand a short time, will separate into two or more parts: one is a green gelatinous precipitate, which, when the coloring matter is removed, is a grayish white substance, and has been named vegetable fibrin. It separates from the juices, precisely as does the fibrin of the blood. So the clarified juices of all nutritive vegetables, among which we may enumerate asparagus, cauliflower, cabbage, turnip, oats and the various other kinds of grain, when boiled, produce a coagulum which is identical in composition with the serum of the blood, or the white of an egg: this is called vegetable albumen. In the leguminous vegetables, as peas, beans, etc., the other proximate element, casein, more particularly abounds. This, as has been already observed, forms a pellicle upon the top of the heated juices, and does not coagulate.

These substances, too, I may add, are the chief constituents of the blood, and hence must be regarded as the proximate bodies which build up and form the basis of the animal frame.

Protein is obtained from either albumen, fibrin or casein. In order to effect this, the substance must be well washed in succession with water, alcohol and ether, by which means the extractive matter, soluble salts and fats are removed. The phosphates of lime, and other salts insoluble in water, are removed by hydrochloric acid. Potash in solution in water, and moderately strong, dissolves the remaining earths, as well as the sulphur and phosphorus which is usually present. The protein is then ready to be thrown down by acetic acid, which must be added in slight excess only. It is a gray gelatinous flocky substance, which must be washed from the acetate of potash upon a filter. When dried, it becomes hard and yellow. It is insoluble in water, alcohol or ether, and devoid of taste or smell. Its composition is represented by the formula

$$C_{40}+H_{31}+N_5+O_{12}$$
.

Its symbol is Pr. It burns without leaving a residue.

Protein may be obtained from most of the tissues or organs and fluids of organized bodies. Thus besides pure albumen, fibrin and casein, it may be obtained from hair, horn, and the crystalline lens of the eye. It is regarded by Mulder as the first product of organization: hence the term I am first; and it is regarded, of course, as the starting point of the tissues in the animal kingdom. The relations of protein to the acids need not be stated here: it is sufficient for my purpose to observe that it dissolves in dilute acids, especially the acetic and phosphoric. It is precipitated from them by tannin, absolute alcohol, ferrocyanide of potassium, etc.

In regard to protein, I would, with great deference to the opinion of chemists, question the real existence of this body in the fluids or tissues in an insulated state, or one which exists independently of all other bodies. I do not question the product itself, or that such a product is readily obtained; but that it is ever formed or found in the fluids or tissues as

a product of organization, I do not believe. I may illustrate my views in this way: Bone is a homogeneous product of organization: so is fibrin, casein, etc. When these bodies, however, are washed with water, alcohol and ether, the extractive matter, fat and oils are removed. If we now subject them to the action of hydrochloric acid, we remove the solid matters, and we have remaining a flexible cartilaginous body of the original shape of the bone. So if we subject bone to incineration, we remove all but the solid parts: we have the earthy matters remaining. In this case, neither the cartilage nor the solid phosphates could say 'I am first;' for the fact is, the fluids, which form bone, contain simultaneously the special elements which are destined to form it. So it may be said of protein, that it never forms a tissue in its independent capacity: the elements of protein, as given in the formula, must be in combination with other bodies in order to form the tissues. These elements are all removed by the treatment to which the juices, tissues, etc. have been subjected. Cartilage of bone is far more likely to be formed in a state free from phosphate of lime; this, however, would be an abnormal state. So bone not unfrequently contains too much bony matter; and both bone and cartilage may be insulated by proper treatment, yet no one would have a right to call either normal bone. The insulation of protein from albumen and fibrin or casein, is an extraction by chemical affinity analogous to cartilage in bone; a matter, which never exists in an independent state. If, from proteine bodies, the fat and extractive matter is removed, leaving the inorganic substances, I believe it is in the condition in which it forms tissues. Fat, starch, oil, sugar, and analogous bodies, are the only ones which are destitute of inorganic matter. Even starch leaves a residue on burning, but I am unable to determine whether it is accidentally present or not.

The foregoing bodies exist in the animal and vegetable kingdoms. There are other bodies, however, which are found only in one: thus, gum and starch belong to the vegetable kingdom exclusively; while gelatin, or rather gluten, and chondrin, are products of the animal kingdom only. So pus or pyin, pepsin, and ptyalin, are also exclusively of animal origin. Only two of these bodies require a notice in this place, viz. gluten and chondrin.

GLUTEN. In its ordinary state and condition, it is gelatin or glue. Two distinct substances, closely related to each other, are obtained from skin, cartilage and bone, namely, gluten and chondrin. The former is obtained by boiling serous membranes, skin, etc. in water: when cold, it forms a tremulous jelly. Chondrin is obtained by boiling the cartilage of the ribs, or larynx: when cold and dried, it is hard and brittle. Both bodies behave alike when their solutions are treated with acetates of lead, sulphate of iron, chlorine and iodine: they form precipitates, which are not soluble in an excess of the precipitating substance. Alcohol also precipitates gelatin from its solutions. Tannin (tannic acid) is the proper test for gluten or chondrin.

Glycicoll is a species of sugar produced from gelatin, by boiling it with potash: it is the sugar of gelatin, and crystallizes in colorless rhombs from a spirituous solution.

The origin of gelatin is unknown, except that it is an animal product. It leaves a residue on being incinerated. It is supposed, however, that it is formed in the organism by the

decomposition of protein by the alkalies. Neither chondrin nor gelatin yield protein when treated with potash, as they do not become purple with hydrochloric acid. Gelatin does not contain fibrin, albumen or casein. Blood, therefore, can not be formed from gelatin; and hence an animal, fed exclusively upon it, must die from starvation. Though proteine compounds can not be evolved from gelatin, yet gelatin is formed from the proteine bodies. We have an example of such a production in the chick in the egg.

BLOOD.

It is a compound substance, though homogeneous when flowing either in its appropriate vessel, or in the moments during which it is issuing from a wounded vessel. When it has stood a short period, however, it separates spontaneously into three parts: serum, which is a yellowish somewhat viscid fluid; fibrin, a fibrous white coagulated mass; and blood globules of a red color, which, when circulating, impart a scarlet red to the whole mass. Blood, however, is not necessarily red: the white fluid of mollusca, and other invertebrate animals, is a true blood. It is also of different colors in insects and worms, as green, yellow, orange.

The temperature of blood varies in different classes of animals: in the ox, it is 103°; in the hog, 99.5°; horses, 96.8°; sheep, 101.3°; and the duck, 105.8°, being uniformly higher in birds than in mammalia. Arterial blood is 1.8° higher than venous. Its specific gravity varies from 1.041 to 1.082. In robust men, it is high; in very young infants, thin, and of a low specific gravity.

The function of the blood is to transmit nutritive matter to the various parts of the system. It must be nutritive itself; and as there is a continual waste in the system, it must necessarily receive continual additions: these are furnished by digestion, or, in vegetables, from matter taken up by the roots. The nutritive matter of blood consists of plasma, as it is sometimes called, and which is itself composed of albumen, fibrin, casein, fat, salts, iron, extractive matter, and a peculiar coloring substance called hamaphain*. These constituents, however, are not all nutritive. As certain bodies are carried out of the system in a healthy state, we are to regard them as in a condition unsuited for nourishment: thus saline, coloring and extractive matter are excreted; while albumen, fibrin, and fat, never are, except in a morbid state. They neither are found in sweat, urine or mucus.

The nutritive power of the blood depends upon the food, and the health or normal functions of the system. We may form some opinion upon this subject, from the composition of blood from a horse.

^{*} Simon's Chemistry, Vol. I, p. 101, 102.

					A	NA	LYS.	s e	Y S	SIMON.	
										Arterial blood.	Venous blood.
Water	-	-	-	•	-		-	-	-	760.084	757 · 35 1
Solid residue	-	-	-	-	-	-	-	-	-	$239 \cdot 952$	$242 \cdot 649$
Fibrin	-		•		-	-	-	-		11.200	11.350
Fat	-		-	-	-	-	-	-	-	1.856	$2 \cdot 290$
Albumen	-		-	-	-		40	-		78.880	$85 \cdot 875$
Globulin	-	-	-		•	-	-	-	-	$136 \cdot 148$	$128 \cdot 648$
Hæmatin	-		-	-	-		-	-		4.872	$5 \cdot 176$
Extractive mate	er	and	sa	lts	-	-				6.960	9 · 160*

The testimony and opinions of many able chemists and physiologists go to establish the doctrine that there are important differences in venous and arterial blood. The following paragraph is a summing up of the differences, as established by careful experiments and analyses: "That arterial contains less solid residue generally than venous blood: it contains less fat, less albumen, hæmatin, extractive matters and salts, than venous blood; and also that the blood corpuscles of arterial blood contain less coloring matter than venous":

It must be evident that blood can possess no definite composition; that the blood of two individuals must be somewhat dissimilar; and probably that the period of life, food, etc. must particularly modify its composition. It is well established that the condition of the organs materially affects it. Food which is intended to impart strength, must be rich in vegetable fibrin and albumen. Animals which are worked, or which afford milk, must be supplied also with the same materials, with food which contains elements of the composition of their own flesh.

The composition of different parts and organs of animals may be stated here, and in this connection, from the bearing which the subject has upon the growth and renewal of tissues.

Composition of bones, cartilage, teeth, muscles, liver, brain.

BONE.		
Phosphate of lime, with a little fluoride of	Femur.	Decipital bone.
calcium	$54 \cdot 07$	$52 \cdot 51$
Carbonate of lime	12.71	11.14
Phosphate of magnesia	1.42	1.05
Salts		0.50
Cartilage	29.09	$32 \cdot 80$
Fat	1.91	2.00 Von Bibra.

^{*} Simon's Chemistry, Vol. I, p. 194.

[†] Ditto, p. 195.

Water

Phosphorized fat - -

Extractive matter and salts - - - - -

.0		CONST	TTUE	INTS	8 0	F	BLOOD.	•
		C'A	RTILA	GE C)F A	. CI	HILD.	
	Phosphate of lime			-	-	_	20.86	
	Sulphate of lime			-	-	-	50.68	
	Phosphate of magnesia	ı	es es	-	ď	~	9.88	
	Phosphate and carbona				•	-	traces.	
	Chloride of sodium		-	ď	eř.	œſ	9.37	HERZ and GUGERT.
	FROM THE COS	TAL CART	ILAGE	or	A M	IAN	TWENTY	YEARS OF AGE.
	Per cent of ash in the	cartilag	e •	e ⁱ		•	$2 \cdot 24$,
Dried	muscle of the ox has	the sam	e coi	mpc	siti	on	as blood	l. It is composed of
							Beef.	Blood.
	Carbon	~	- o∵	-	es"	-	51.83	51·95
	Hydrogen			-	-	-	7.57	7.17
	Nitrogen	.		-	-	-	15.01	15.07
	Oxygen	·	eo.	•	-	-	21.37	21.39
	Ashes	9 5 -	• •	-	e	æ	4.23	4.42 PLAYFAIR
The co	onstitution of teeth is	rather	differ	ent	fro	m	that of c	ordinary bone.
							Enamel.	Osseous portion.
	Phosphate of lime wit				ium	i -	89.82	66 • 72
	Carbonate of lime -			-	-	-	$4 \cdot 37$	$3 \cdot 36$
	Phosphate of magnesi	a		-	-	-	$1 \cdot 34$	1.08
	Salts		o, o	•	-	•	0.88	0.83
	Cartilage			-	-	-	$3 \cdot 39$	27 ·61
	Fat		• •	-	æ	-	0.50	0.40 Von Bibra
	сом	POSITION (of "GL:	ANDU	JLA	R I	1ATTER —	LIVER.
	¥¥7 ,						Healthy liver	•
	Water	= a d		• •	4	-	76.39	<i>55</i> · 1 <i>5</i>
	Solid matter				-	-	23.61	44.85
	Animal matter dried a	it 212°		-	-	-	21.00	13.32
	Saponifiable fat -		9 9	-	•	-	1.60	$30\cdot 20$
	Cholesterin	• • a		•	9	9	0.17	1.33 Bondet
The b	rain is composed of						•	

The quantity of water in the brain varies, according to the statement of several eminent chemists. According to Fremy, it amounts to 88 per centum.

78·0 7·3

 $12 \cdot 4$

1.4 Denis.

It will be observed that from the foregoing analyses, the composition of the organs and parts vary essentially from each other; and though it may not be precisely the same in individuals at different times, still the departure from a certain standard is never very wide. The law of the vital economy secures a composition constant within narrow limits.

MILK:

A secretion of the mammary glands, of a white color, and, like the blood, is of a complex constitution.

Milk, when pure and healthy, has an alkaline reaction. This alkaline reaction continues for periods varying according to the state of the weather, and the meteorological condition of the atmosphere. The kind of milk, also, varies in this respect; human milk remaining longer in its alkaline condition than cow's milk.

Milk consists of casein, a coagulable fluid in which fat vesicles may be observed under the microscope. In addition to the casein and butter, certain salts and sugar are invariably dissolved in a large quantity of water. When milk is allowed to stand, the fat vesicles or butter rise to the surface. When boiled, casein rises to the surface and forms a pellicle. The same substance is coagulated by rennet, sulphuric and other acids. The solid matter of milk is composed of the following substances:

						I.	II.
Phosphate of lime	-	-	-	-	-	47.1	50.7
Phosphate of magnesia	-	-	-	-	-	8.6	9.5
Phosphate of peroxide of iron	-	-	-	-	-	1.4	1.0
Chloride of potassium	-	-	-	-	-	$29 \cdot 4$	27 · 1
Chloride of sodium	-	-	-	-	-	$4 \cdot 9$	5.0
Soda	-	-	-	-	-	8.6	6.7

The fluid matter of cow's milk has the following constitution in 1000 parts:

							I.	II.	
Water	-	-	-	-	-	-	$857 \cdot 0$	853.0	
Solid constituents	-	-	-	-	-	-	$143 \cdot 0$	177.0	
Butter	-	-	-	-	-	-	40.0	38.9	
Casein	-	-	-	-	-	-	72.0	69.8	
Sugar and extractive matter	-	-	-	-	-	-	28.0	⁴ 31·3	•
Fixed salts									HERBERGER.

The milk may be diseased, by the general disease of the animal. When the cow has vaccinia, or vaccine disease, the butter is diminished, and pus is found mixed with casein. The milk from a healthy teat contains no pus, and has the appearance of healthy milk. Cows affected by the grease, do not yield healthy milk.

Milk is also liable to certain changes which can not be explained very satisfactorily. It sometimes becomes blue and yellow, from the presence of two species of animalculæ: the

first is the Vibrio cyanogenus, and the other Vibrio xanthogenus. I have also seen a bright red, from the presence of another species probably of the same genus. One or two pans of milk in a dairy will be affected, and all the rest escape.

Milk becomes sour, by the change of the sugar into lactic acid.

RECAPITULATION.

A brief notice of the foregoing substances was necessary for various reasons. The agriculturist should know the destination and use of those elements of the soil, the channels through which they pass, and the preparation they undergo before they are converted into products suitable for the consumption of man and animals. He should have in his mind a condensed scheme of the physiology of the natural products, and of the changes which the elements undergo in passing from their comparative inert condition in the soil, to their first semi-organized state in the tissues, and finally to their perfect organization in the bone, cartilage, membrane, muscle and brain.

In order to reproduce the foregoing matter in a distinct and more intelligible form, and in conclusion of the chapter, I shall here recapitulate the most important facts and principles.

- 1. Soil is composed of a few essential elements only; the most important, in one sense, the sense in which they are most necessary for animals, are in the smallest proportions.
- 2. If plants have no power of selecting the elements essential to their growth, they have the power of distributing it to certain parts and organs. Phosphate of magnesia and lime exist only in small quantities in the chaff of wheat, but both are quite abundant in the kernel which the husk envelopes. In no case do we find the reverse of this fact.
- 3. Every organ of a vegetable, and we may extend the remark to animals, has a reticulated frame work of inorganic matter, the base of which is either silex or lime. Monocoty-ledonous plants, particularly the cereals and grasses, have a silicious skeleton; the dicotyledonous have usually a lime skeleton, or a predominance of lime. This statement refers to the stalks, stems and leaves; while if their seeds or tubers are edible, their composition bears a resemblance to that of the grains.
- 4. The fluids of plants and animals contain all the nutritive bodies in solution, and become vitalized by contact with vitalized matter. A very large proportion of the fluids circulating in plants is water, the solvent powers of which are increased by the presence of alkalies and carbonic acid. It is by the presence of these bodies in the soil, too, that some substances quite insoluble out of the soil, are quite soluble in it: the organic matter of soils is soluble in most earths, and associated with other elements, and hence becomes an essential class of matters forming the growth and perfection of certain cultivated plants.

- 5. The importance of organic matter in the soil, is sustained by many well established facts:
 - a. In the removal of crops of beans, wheat, indian corn, etc., the soil is exhausted of not only inorganic but organic matter; and in order to restore fertility, experience proves the necessity of adding nitrogenous matters. The most striking results flow from those manures which contain the most organic matter already prepared for the uses of the plant, such as guano and night soil; and in the application of these substances, we become aware of the value of their presence in the soil. Ashes, which is usually regarded as a manure wholly inorganic, is really complex, and contains much organic matter. Organic matter in fact adheres so obstinately with phosphoric salts, as well as the alkalies, that no form of matter which is applied to land as a fertilizer is entirely free from it. If there are any exceptions, it is in the use of lime of the oldest rock, and in pure gypsum.
 - b. It is well established that one of the conditions necessary to secure the favorable action of lime, is the presence of organic matter. It is not sufficient that there be carbonic acid in the atmosphere, or ammonia: it is necessary that it should exist there in the condition of a product undergoing decay, by which the oxygenized products may be acted upon, and by which the peculiar organic acids may be produced.
 - c. In regard to the entrance of nutriment into a plant, I can not but regard the root as its channel. Experience upholds the idea at least; and though the leaf has the power of absorbing carbonic acid and ammonia, yet it is really analogous to the power of the skin also to absorb matters: still it is not the function of the skin to supply food to the system. Vicarious functions are quite different from natural ones. Hence the argument that organic matter in the soil could not furnish enough for the wood produced in a forest, does not prove that the forest received its increase through the channels of the leaves. The ammonia and carbonic acid falling with the rains to the earth, supply additions of nutriment to the soil. Again, it is not enough that the inorganic manures be employed. Experience proves that their good effect on crops fails in due time: indeed, perfect seed can not be produced in their absence. All essential and perceptible increase of products comes from manuring, and in proportion to the manure; and trees or shrubs whose branches are cut off from the supply below, die, not from the absence of water alone, but from starvation: they can maintain but a precarious existence under the most favorable circumstances.
 - d. The quantity of carbonic acid is about one-thousandth of the weight of the atmosphere. This is sufficient, no doubt, to preserve, so far as this is concerned, the balance of nature. Being produced by the respiration of animals and by combustion, and diffused through the atmosphere, it is again brought to the soil. So it is produced in vast quantities in the soil by slow combustion, and manures must yield the same product in the very place where it is particularly wanted.

- Must we suppose that the carbonic acid thus derived from the decay of vegetables, and from the manures added, is evolved in a gaseous state, and ascends to the leaves for absorption? This will not gain credence.
- e. So of ammonia: this must be continually supplied. The original store of it, which may be supposed, for argument, to have been thrown into the atmosphere, would long ere this have been exhausted. It must be reproduced, and it is well known to be reproduced in the decay of vegetables, and also that a part is resolved into nitrogen and hydrogen. The decompositions in the soil become a source of ammonia: this need not escape into the atmosphere for the absorption of leaves. Manures, too, furnish it during their changes in the soil; and it is not probable that it must leave the soil in order to reach the vegetable tissue.
- f. If, however, manure is left uncovered, will the crop get the benefit of it? Here it goes into the atmosphere; but who will maintain that the field will yield the crop it would, had it been covered and well mixed with the soil? When ammonia is fixed by ground gypsum, is not the conclusion evident that it reaches the leaves of plants through the root?
- g. The soil must possess all the inorganic substances, as well as organic, which are essential to the perfection of vegetables: if any one is wanting, it must be supplied. But to secure its benefits in the highest degree, the soil must be put into a state which shall make all those matters accessible to the roots of plants. This calls for the attention of the husbandman to its mechanical condition.
- h. It is maintained that certain crops, as clover, take nitrogen from the air. May it not be doubted, and may not the advantages to be derived from its cultivation arise from its large and widely branching roots, whereby a rapid growth is secured by an absorption of nitrogenous matters from comparatively large areas.
- 6. Modern chemists and physiologists have established the doctrine, that the vegetable kingdom is the great source of nourishment to the animal kingdom. In the vegetable kingdom, albumen, fibrin and casein, substances essential to the maintenance of animal life, are elaborated. So both kingdoms yield back the inorganic matters to the mineral in their decay, and combustion and respiration are other means by which the food of plants is in part prepared. It will suggest itself to the reader, probably, that the actions of the vegetable end in results totally different from those of combustion and respiration. In the leaf, oxygen is set free; in combustion, it is fixed. Oxygen and carbon being united in animal life, it becomes the part of the vegetable to separate them; to appropriate the carbon in the growth of wood, while at the same time oxygen is once more in a condition to meet the wants of animals.
- 7. It is impossible to overlook, in these changes, the balance which is preserved by the controlling agencies of nature. Mutual adaptations prevail: harmony is secured. To the vegetable is allotted the task of elaborating the fluids most essential to the growth of animals. The vegetable has time and leisure to do this. To say that the plant vegetates, is to express the whole of its life and doings.

- 8. A large proportion of the weight of all living bodies is water: four-fifths of an animal is lost by drying. So also in some of the most important products of vegetable life, a very large amount has to be set down to water. These facts prove the importance of water; and it surprises every one, when he is told for the first time that highly organized animals, weighing twenty and thirty pounds, when dried, have only a few grains of saline ash: they are vesicles of animated sea water.
- 9. Many persons have been mistaken in their notions of scientific husbandry; or they have failed to seize upon the higher idea embraced in the investigations of the philosophic farmer. We can regard scientific husbandry as an investigation of the mutual adaptations of the three kingdoms of nature to each other, and of the methods of applying fixed principles to practice. No other method of farming, but that founded on adaptations fixed in nature by its Great Author, can be successful, and repay the efforts of the laborer. Some of these methods are easily discovered, and have ever been used: others are not so accessible, and require other sources of knowledge for their discovery. Chemistry has in this way become the handmaid of agriculture, and has already unfolded new and most important principles in the employment of this first and chiefest of arts.

CHAPTER II.

ANALYSIS OF PLANTS.

GENERAL REMARKS ON THE IMPORTANCE OF AN ANALYSIS OF THE ASH OF PLANTS, AND ON THE DISTRIBUTION OF THE ELEMENTS OF PLANTS. PREPARATION OF THE ASH FOR ANALYSIS. MODE OF ANALYSIS: INORGANIC AND ORGANIC. ANALYSIS OF SEVERAL KINDS OF POTATOES IN COMMON USE; TOMATO. ROOT CROPS: CARROT, BEET, RUTA BAGA, SWEET POTATO. CONCLUDING REMARKS ON THE POTATO.

The analysis of the cultivated, as well as of those vegetables which grow without cultivation, is a necessary work, and is especially promotive of agriculture in its present state. Modes of culture, preparation of the soil, and treatment of the growing crop, have reached a very perfect stage, and but little more can be expected from methods which are usually called improved. It has now become interesting to inquire how the produce may be increased, by improving the quality and excellence of the crop, by a systematic application of matters which the crop requires, by administering them in new modes and forms, and at times more in accordance with the period when peculiar elements are deposited in the seeds, grains or straw.

The possibility of bringing about improved results from the modes here alluded to, rests upon experimental knowledge of the constitution of the bodies we have under culture, and also upon the progressive changes and progressive accumulation of nutritive matter during the periods of growth. We obtain the knowledge necessary to secure the ends in view, by the analysis of the ash of the perfect and mature plant, and by successive analyses during its progress to maturity.

That the inorganic matters vary at the different stages of growth, is now well established; that organs differ in composition, and that the same may be said of parts, is no longer to be questioned. I shall maintain that these variations are not accidentally produced, but are results founded upon a law which regulates the distribution, and directs the final destination, of every particle of nutritive matter received into the tissues of a vegetable. If the distribution of nutritive matter had been left to chance, we might as frequently find the gluten and casein in the straw and chaff, as in the grains; the phosphates of magnesia, lime, etc. in the chaff, rather than in the kernel. Such a result has never been met with; the kernel being known as the principal storehouse of food, from the experience of the whole cycle of ages which has elapsed since their cultivation and use by man.

This law is one which may be expressed, so far as direction is concerned, by an upward

and outward movement: a peripheral force is given to the nutritive matters in the early stages of growth. It is perhaps premature to attempt to speculate upon the ends which are secured by the supposed law; but one or two remarks may be offered upon the subject. First, the products rich in elements required by all plants, are speedily sent back to the soil, for an early and renewed use: the leaf, the rind and husk, the tender stem, are all annual growths, and return annually to the soil to undergo decay. The fruit, with its envelopes, is situated at the extremity of a floral branch or bud: it is at the end of the channel of the coursing fluids; and its influence robs the base or bottom of the straw of a portion of its inorganic matter, that an accumulation may be secured in parts which are specially appropriated to the use of animals. The bottom of the straw has less nutritive matter than the top: it has moved upwards; but the usual supply is diminishing in consequence of age, and hence its deficiency remains to the end. Pea vines die upwards: their main stalk ceases to elaborate or arrest the nutritive matter; and it often presents the appearance of death, while the upper leaves and fruit are yet fresh: it is robbed by the activity of the superior organs. So numerous are the instances of this kind, that we can scarcely refuse to admit the law of an upward and outward force in the distribution of nutritive matters.

Another important result, is the perfection of the seed and fruit. This seems to be the great end to be secured; and so rare is a failure of this end, that we scarcely consider the law which secures the result: it is regarded as a matter of course. A stone projected into the air, surely returns to the ground, but the law of gravitation is rarely thought of; so the seed ripens by an accumulation of nutritive matters scarcely less surely, and, as in the case of gravitation, we forget there is any law in operation. Though a vast amount of nutritive matter is locked up in a forest, this law still operates. There is no centralization of important substances in the interior of the wood: in the growth even of the trunk of an oak of a thousand summers, the outward and upward forces, are still recognizable in the percentage of ash in the wood of the outside, the leaves, and extreme branches.

The action of vegetables upon the nutritive matters in the soil is in accordance with the same law: we can not fail to recognize the upward movement. The roots bring up, from the deepest parts of the soil to which they can penetrate, the acids and bases which are required to sustain life: the leaves and annual stalks receive a large share of it; these fall upon the surface, where they undergo decay; and hence upon the surface, the matter which has been drawn from the deepest soil, is left where it is required. We often see roots shooting upward into the richer stratum at the top; still, when not thus invited by a richer storehouse at the surface, they penetrate deeply and widely.

In the constitution of the soil, no depth has been reached, which has been on that account deficient in the elements of nutrition. Organic matter exists in the Albany clay, at the depth of at least fifty feet, and at hundreds of feet in the calcareous shale of the Salt group.

In confirmation of what has been stated in the foregoing paragraphs, the reader may [AGRICULTURAL REPORT — Vol. II.] 5

consult the analyses, many of which were undertaken with a view to determine the law of distribution of the inorganic matter*.

I now propose to give in detail the results of the analyses which have been made in furtherance of the objects of the State Survey. In doing this, I find it impossible to state these details in that complete and perfect form which is desirable. This arises from the fact that the analyses are still going on, will continue through this season, and can not be completed till its close. What then remains, will be given in an appendix to this volume.

The analysis of the ash of plants presents some difficulties which are not easily overcome. The first step is to procure an ash in a proper condition: if it is highly alkaline, it is very liable to fuse at low heat, and the fused particles embrace particles of coal, and the ash is black. So it is liable to be caustic, or partly in a caustic state; and hence there will arise, in summing up the results, an apparent error: the figure will be too high, if the carbonic acid is reckoned or calculated, instead of being obtained and weighed; it will be too low, if the carbonic acid is omitted. It is indeed difficult to obtain an ash entirely free from this uncertainty. The ash of indian corn is the most difficult to obtain; and, besides this, I take it upon me to say, that up to this time, no one has yet made a correct analysis of this grain, or without having too much loss.

The method which I have pursued for the analysis of the ash of plants, has been the one approved of by distinguished chemists. It has, however, been modified during the progress of the work; and if it has differed essentially from that of some distinguished chemists, it has been for special reasons.

I will now state the method which has been followed in the laboratory, in order that the results may be appreciated. When the method of obtaining a result is defective, I have no desire that it should pass for more than it is worth; and as the results given have been those which were actually obtained, it certainly is proper that the method also should be known. It has not been my practice, nor that of my assistants, so far as I have known, to distribute losses among the several results. The loss is a result as important as any, and should be known, and never concealed, however great it may be. That a small loss should occur, is inevitable; this will vary with the care bestowed, and the nature of the substance analyzed.

I. The preparation of the ash. The vegetable is burned at as low a temperature as possible, and sufficient time is given to consume the coal perfectly. Sometimes, however,

^{*}It is not intended here to claim the entire merit of determining the fact that the organs and parts of plants differ in chemical constitution, and differ also in different periods of growth. That the kernel differed in composition from the straw, has been known from the earliest time; but the differences which are very constant in the inside and outside wood of forest and fruit trees, the differences between the leaf and bark, the law of upward and outward movements of the nutritive fluids, were, so far as I know, first observed in my laboratory. Our results, however, were not published as early as some others bearing upon and proving the existence of the same general laws; but I do not, therefore, feel bound to credit discoveries in this field to others. By these remarks, it is not designed to claim more than is my due, nor to withhold praise from others who have labored in the same field. The present period has been remarkable for action and unremitting labors in the chemistry of plants, and many discoveries have been made almost simultaneously by different individuals.

the impure ash is weighed, and all the then soluble matters dissolved out, when the residue is again burned. This method is adopted only when the ash fuses at a very low temperature, or when the proportion of mixed alkalies is comparatively large. Generally the ash has been used soon after it has been prepared: if it stand it is dried previous to analysis.

II. The analysis begins with the weighing two or three parcels of 20 grs. each: sometimes a less quantity is used. One parcel is used for the earth, alkalies and phosphates; another for chlorine, sulphuric acid and organic matter. I obtain the chlorine from one half, and the sulphuric acid and organic matter from the other half. The other parcel is dissolved in hot hydrochloric acid. In many cases, this is a speedy operation: time, however, is given in the straw and chaff of the cereals, when silicates may be expected. We usually obtain in these cases a gelatinous solution, which indicates the perfect action of the acid. Silica is obtained by filtration: it is ignited and weighed. The filtrate contains the phosphates, lime, magnesia, potash and soda. The phosphates are thrown down by excess of ammonia; the lime, afterwards, by oxalate of ammonia. The filtrate, after the lime has been separated, is divided into two equal parts: one is used for magnesia; the other, for the potash and soda. To obtain the magnesia, I prefer the phosphate of soda with ammonia. We have used the peroxide of mercury, and it saves time and some labor; but we have been better satisfied with the phosphate of soda. The half reserved for potash is evaporated, and the salt exposed in a Berlin capsule to low ignition. It is then weighed, and the result is set down as chlorides. These are dissolved out with water, and filtered; the filtrate is evaporated again, and the dry residue submitted to the action of absolute alcohol. The insoluble matter remaining in the filter, which is washed with alcohol, is dried, and heated rather strongly, and weighed and set down as chloride of potassium, from which the amount of potash is calculated. The chloride of potassium and magnesia being subtracted from the chlorides, gives the chloride of sodium, from which the soda is calculated. The insoluble matter in water is also noted. This method, though not recommended in Fresenius's work, I can not but regard as quite accurate. Fresenius says that chloride of potassium is nearly insoluble in absolute alcohol; and we have found, on testing the alcohol made in the laboratory, that it does not dissolve a perceptible quantity of pure chloride of potassium, while chloride of sodium dissolves in it rather freely, and chloride of magnesia at once.

III. After going through with the preceding work, the phosphates which have been preserved are taken up for analysis. They are redissolved in hydrochloric acid, and filtered to free them from soluble silica. The filtrate from the soluble silica is mixed with ammonia in slight excess; and while the mass is still wet, acetic acid is added, which dissolves all but the phosphate of peroxide of iron: this is separated by filtering, and dried, ignited and weighed. From the filtrate, oxalic acid throws down the lime, which is also filtered, ignited and weighed. The magnesia is now thrown down by phosphate of soda and ammonia, the latter in slight excess, or else the acetic acid will still hold the magnesia in solution.

I can not but regard this method of treating the phosphate, as preferable to that where

the whole of the phosphoric acid is obtained at once and separately. If it is suspected, however, that we do not obtain all the phosphates or phosphoric acid, we adopt the separate method with another quantity of ash, using for this purpose a weighed portion of iron dissolved in nitric acid. By this method, all the phosphoric acid is obtained in combination with peroxide of iron.

By the method I have pursued with the phosphates, a more satisfactory result is obtained. We know not only what parts and organs contain the most phosphates, but also with what bases they are in combination.

The ash always contains organic matter. In almost every result we find it: it is, however, the most troublesome in the phosphates, and it has been suspected that it decomposes phosphoric acid, and that a loss ensues when strongly ignited. Whether there is a loss of potash in burning the vegetable, is not as yet well determined. The organic matter is supposed to exist as an organic salt of lime, magnesia, and perhaps potash. It seems impracticable to remove it, and it is quite abundant in the best or whitest ash which can be obtained. For a manure, ashes are so much the better for the organic matter: it is quite soluble in water, and its amount is obtained from the watery solution; the whole product being perfectly dried and weighed, and afterwards ignited till it is entirely consumed, it is again weighed, and the loss is organic matter.

It is hardly necessary to say that the chlorine is calculated from the chloride of silver, and the sulphuric acid from the chloride of barium. The proportions were obtained by taking 100 grs. (usually) of the substance, drying in a water bath, and then burning it in a platina or porcelain capsule.

Organic analysis, as detailed in the following paragraphs, consists in the separation of the proximate elements, as starch, albumen, casein, dextrine, etc. The mode is sufficiently simple, but seems to be liable to some variations in consequence of the easy decomposition of some of these bodies on exposure to the atmosphere.

The potato is first washed, and then dried, or freed from the outside water. It is then sliced longitudinally, and laid upon blotting paper a moment in order to absorb the exuded moisture. Two or three hundred grains are then weighed, when it is immediately grated carefully, and so as to be free from unbroken pieces of the potato. The grated portion is allowed to subside in pure water, after passing through fine mushin. The starch collects at the bottom; and the fibre, after thorough washing, remains upon the muslin filter. The supernatant liquor is drawn off from the starch by a syphon. The small quantity of fluid remaining upon the starch, and the starch itself, is thrown upon a filter and washed again, and the filtrate is added to the liquid drawn off by the syphon. The liquid is often quite bulky: it may, however, be divided into two equal parts; one for albumen, and the other for casein. The first is boiled or heated to above 160° Fahr., when the albumen separates in thin coagulated masses, and subsides in the course of twelve hours. This is filtered upon a prepared filter, and dried in a water bath preparatory to weighing. The casein is precipitated from the other half by acetic acid. The liquid, after complete subdivision, is drawn off with a syphon, and filtered, dried and weighed. The dextrine may be obtained

from either half. The acetic solution, however, must be first neutralized by carbonate of ammonia, when the albumen goes down, and from which the liquid must be freed. It may then be evaporated to a small bulk, and alcohol added in a large quantity, when the dextrine appears in flocculi which must be left to subside. It may then be filtered upon a prepared filter, and weighed. The remaining liquid is now evaporated in water, in a cnp whose weight is accurately determined, until it ceases to lose weight: this matter is called sugar. The fibre which had been obtained in an uniform state, is freed from fatty matter and gluten by means of alcohol in a retort. Ether is afterwards employed to free the gluten from fat, which is cautiously distilled off or evaporated in a weighed cup.

My readers will now be prepared to set a proper value upon the labors which have been performed in the laboratory. The foregoing statements, I hope, have also prepared the way for entering upon a detailed account of the results which have been obtained. I shall, in the first place, give the analyses of the potato plant; and I may again say that the order of arrangements is governed by necessity, being obliged to give in the first place the analyses of those substances which are in the greatest state of forwardness; and as analyses are still in progress, and will be for three months to come, those which are made hereafter will be placed in an appendix.

The analysis of the potato plant is extended to the stalks, leaves and tubers; they have not been confined to the tuber. All the varieties which I could obtain have been analyzed.

I. MERCER POTATO.

Color white or grayish white; flesh white; form elongated, much longer than wide.

The potatoes analyzed were raised in Cortland county, by N. Salisbury, Esq. Length, 5 inches; thickness, $2\frac{1}{2}$ inches.

1. Analysis of the ash of the whole potato.

Silica	•		۵						$4 \cdot 400$
Phosphat	es	•		•	•	۵			38.500
Lime	•					•	•	•	0.150
Magnesia	ı	•	•	4	•	-	-	-	0.800
Potash	•	•	•	4	4	•	•	•	13.263
Soda	-	•		-	•	•	•	-	$24 \cdot 925$
Sulphuri		l	-	-	-	*	•	-	$6 \cdot 254$
Carbonic	acid		-	•	-		•	•	trace.
Organic	matte	er	•		•			-	2.536

102:434 S.

2. Analysis of the phosphates.

Рнозриа	TES	•	•	•	•	-	٠	38.500
Phosphate of	pero	xide	of iro	n -	8		-	7.900
Lime -	•	•	•	•		•		2.933
Magnesia	-		-	•			-	0.055
Silicic acid	-		*				٠	trace.
Phosphoric ac	$^{\mathrm{id}}$		•		-	٠	۰	27.612

For an analysis of the soil upon which this potato grew, see Vol. I, p. 342.

3. Organic analysis of the Mercer potato.

Size the same as that from which the ash was obtained.

. 1	PROPO	RTIONS	5.			
Water of the ends -	•		4	ě		79 509
Dry matter in the ends		à	•	-	-	20.492
Ash of the ends	4	•	÷	6		0.579
Ash calculated dry •	•	-	•		4	3.746

ANALYSIS.

							rom which albumen was separated.	Half from which albumer was not separated.
Starch -	is.	۵	•	-	-	-	9.710	9.710
Fibre -	ė.	-	-	•			5.779	5.779
Gluten -	-	-	-	-			0.205	0.205
Fatty matter		-					0.084	0.084
Albumen	-	-			-		0.249	$0 \cdot 249$
Casein -	-		•	٠	•		0.506	0.468
Dextrine	-	-	-	-	-		0.721	1.265
Sugar and ex	tract	•		b	-	-	$3 \cdot 931$	$2 \cdot 638$
							${21\cdot 185}$	$\frac{-}{20 \cdot 398}$

It is found that the amount of water, ash, etc. differ in the ends of the potato. The following results exhibit this fact:

				 Seed or rose end. 	2. Heel end.
Per cent of water -	•		è	• 83·839	75 · 177
Dry matter		•	•	- 16 · 161	$24 \cdot 823$
Ash		6		0.725	0.431
Ash calculated dry -	•	•	•	5 · 197	2·296 S.

These two results with the ends may be compared with the mean, or both ends used together. It becomes necessary, from these differences in the composition of the ends, to examine them separately when time will permit. The result is interesting, and agrees with other facts in the same class.

II. EARLY SHAW, MOUNTAIN JUNE, EARLY JUNE POTATO.

PL. 3 B. Fig. 2.

Color white; flesh white; form round; eyes in fives; skin slightly rough.

This is one of the most valuable of the varieties of this vegetable. It is not predisposed to rot, or to be impaired by the disease. It is early, and may be used as an early or late potato; and my own experience is that it keeps well, and is really one of the best kinds in the spring, never becoming watery or strong. If it is planted early, it reaches that point of maturity which enables it to escape disease. The cause which occasions the rot acts only upon the immature crop. A ripe potato is no more affected by the rot, than an apple, or any other fruit.

1. Organic analysis of the tuber.

							I.	II.
Water -	-	•	-	-	•	-	$74 \cdot 902$	75.06
Starch -	-	-			-	-	13.378	$10 \cdot 45$
Casein -			-		-		$2 \cdot 053$	0.73
Albumen -	-	-					0.085	0.27
Sugar and extr	act		-	4	-	•	$1 \cdot 364$	$2 \cdot 04$
Dextrine -	-				-	-	0.912	0.44
Gluten and fat	-		•		-		0.008	
Fibre -				•	-	-	6.829	10.70
							99.531	99.69

PROPORTIONS.

Water -	-	-	•	-	•	-	-	74.3598	
Ash -	-	-	•	-	-	-	•	0.5492	
Dry matter	-	-	•	•	•	-	-	25.0910	
Ash calculate	d dry	•	•	-	-	-	-	$3 \cdot 4470$	B

The second analysis was made of a specimen obtained from Lansingburgh, from the garden of Mr. A. Walsh; and I have reason to suspect that the fibre was not thoroughly washed from starch.

The Early Shaw is a potato of English origin, and I believe has not been cultivated many years in this country. It never attains a large size; neither is it as productive as some other kinds, yet it is very excellent in its product. Its superior qualities render it a very desirable kind for general cultivation for the table.

2. Potato vine — Early Shaw.

Cut August 6, when it had just passed its flowering.

S.
S.
S.

III. FLESH-COLOR OR BURR POTATO.

PL. 2 B. Fig. 2.

Color purple; flesh white; form wider than long; eyes elongated, and in fives.

In this analysis the whole potato was taken. Grown in the garden of Mr. Walsh, Lansingburgh.

_			_		_
1.	Ane	alusi	s of	the	tuber.

			-					
Silica -		-	-	-		-	-	0.300
Phosphates	-	**	-	-		-	-	$32\cdot 000$
Lime -	•	-	-	-	-		-	0.582
Magnesia	•	-	-	-	-	-	-	0.400
Potash -	-	-	100	-	-	-	-	20.798
Soda -	-	ю	-	-	-		-	$43 \cdot 044$
Chloride of	sodiun	n -	-	-	-	-	-	$3 \cdot 924$
Sulphuric ac	id	-	•	-	-	-	-	0.859
Carbonic aci	d -	•	-	-	-	-	-	trace.
Organic mat	ter		-	-	-	-	-	$2 \cdot 700$
								104 · 247

2. Analysis of the phosphates.

Pноsрн	ATES	-	-		-		77	32.000
Phosphate o	f perc	xide	of iro	n -	-	-	-	4.900
Lime -	•	-	*	-	1	•	•	0.733
Magnesia	ъ	-	-	-	-	-:	•	$0 \cdot 155$
Silicic acid	-	-	-	-	-	-	-	0.155
Phosphoric a	acid	-	-	-	-		-	$26 \cdot 057$

3. Organic analysis.

							I.	11.
Water -	-	-	-	•		-	$72 \cdot 52$	$76 \cdot 3260$
Ash -			is	-		-	0.68	
Starch			-	-			7.77	$9 \cdot 7850$
Albumen	-	-		-	'n	ter	} 0.88 {	$0 \cdot 1203$
Casein		-	-	-	-		} 0.88 {	0.9724
Dextrine	-		-	-	-	-	0.46	0.5015
Sugar -	-	~	bn .	-		-	$3 \cdot 45$	$5 \cdot 7630$
Gluten	*		-	-				0.0407
Fat -	-	-	-	-	-	-	0.18	
Fibre -		-		in	~		9-74	$8 \cdot 1504$

It will be observed that this potato contains less starch than many others in common use.

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For the analysis in another specimen, a slice was cut through the middle of the potator from end to end. The following three proportions were made of the ends and middle.

Water -		w.	ď	1. Rose end. 78 977	2 Middle.76 328	3. Heel end. 74.918
Dry matter			_	21.023	23 672	25-082
Ash -			•	0.942	0.674	0.577
Ash calculated	₫ry -	-	*	4-479	2.847	2 300
Mean per centum do Mean per centum do Mean per centum do Mean per centum do	of ash	whole in two in the matter er in two	e pota ends whol in th v	e potato ne two ends whole tuber	 - 3·389 - 3·208 - 0·759 - 0·731 - 23·052 - 23·259 - 76·947 - 76·741	95 10 } Moist potato. 25 90 75

IV. MERINO POTATO.

LA PLATA RED, LONG RED, LONG JOHNS, SPANISH RED.

Pr. 5 B.

Color red; flesh white, with reddish circular patches in the direction of the reddish or purple axis; form greatly elongated; eyes numerous, and in 14.

This is one of the largest of potatoes: it has been known to attain a foot in length. It is also productive. Though not highly esteemed for the table, yet I believe it is a better potato than many others in common use, when served in its season. It ought not to be used till late in winter and early spring. Cattle are fond of them, as its flesh is tender and juicy. Like many other fruits, as apples and pears, this has its period when it is ripened for the table. It has not, however, been customary with the great body of farmers to select their potatoes in their proper periods, or those periods when they become ripened by age. A potato is obliged to be cooked and eaten at any time after harvest. I can not, however, but believe that the time is not far distant, when the different kinds will be used as they attain maturity; and as in the case of fruits, there will be distinguished the early harvest, the fall, winter, spring and summer potatoes. Accurate experiments upon their keeping qualities are needed, and the time when, like greening and russet apples, they become fit for use. I have subjected this variety to a careful analysis; but the ash which was subjected to examination was obtained from potatoes growing on a different soil from those from which the starch, albumen, etc. were obtained.

1. Analysis of the ash of the Merino potato in three parts.

				1. Rose end.	2. Heel end.	3. Middle.
Silex	-	•	-	1.586	4.019	1.658
Coal	•	-	-	0.180	0.612	
Carbonate of lime		*	-	0.070	0.060	0.984
Phosphates -	-	-	-	14-533	20.842	23-714
Potash - ·	-	-	-	50.213	48.855	38.708
Soda	-	-	•	13.713	15.066	18.518
Sulphuric acid	•	-		8-125	8-424	8.471
Chlorine	-	-		1.110	$1 \cdot 242$	0.897
Insoluble -	-	-	-	$3 \cdot 466$		4.210
Organic acids -	•	-	-	6.000	0.813	
Organic matter	-	-	-			2.830
Magnesia -	-	•		trace.	trace.	trace.
				00.000	00.000	
				98 - 996	99 - 933	99.990

These analyses, it will be observed, are calculated without carbonic acid. The ash of the ends effervesced strongly; the middle, less so: in each instance showing the former existence of organic acids, which, by burning, had been converted into carbonate. The ash, however, in each instance, was converted in part into a caustic state. The potato, it is proper to say, was not peeled; but the outside was well cleansed from dirt by a brush, without abrading the skin.

2. Organic analysis of the Merino potato.

Fı	rom th	e gard	en of I	Mr. W.	ALSH,	Lansin	gburgh.	Middle.
Starch -		•	•			-	-	13.755
Fibre -	-	-	-	-	-	•	-	6.766
Albumen	-	-	-	-	-	-	-	0.112
Gluten and	l fat	-	-	-	-	-	-	0.284
Casein	-	-	-	-	-	•	•	0.605
Dextrine	-	•	-	-	-	-	•	0.556
Sugar and	extra	ct	-	-	-	-	-	2.773
MEA	NT NOTE	CENT.	AGE O	f WAT	ER IN	mire a	rwo en	70
MEA	N PE	CENT	AGE OF	WAT	ER IN	THE T	WO EN	
Water	•		-	•	-	•	-	74.3865
Dry matter			-	•	-	-	•	25.6135
Ash -	-	-	-	-	-	p	-	0.7165
Calculated	dry	•	-	-	-	-	-	3.7685
			mr. 0 br. 4		. D			
		KOPOR	TIONS	OF TH	E TWO) ENDS		

			PROP	OKITOR	S OF	THE T	WU ENDS.	
							Rose end.	Heel end.
Water	-	-			-	-	81.301	$67 \cdot 472$
Dry matter	-	-	-	-	-	-	18.699	$32 \cdot 528$
Ash -	•	-	•	-	-	-	0.973	$0\cdot 460$
Ash calcula	ted	dry		_	_		5.401	2 · 136 S.

Another specimen of the rose end gave,

Water	-			٠.	-	•	$74 \cdot 04$
$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-		-	-	0.94

As this variety is one of the best for illustrating the difference in composition of the different parts, I have divided it into three portions, viz. rose end, heel end, and middle. These divisions, however, are not exact by measure; three parts, nearly equal lengths, only being taken, without regard to exact lines of demarkation denoted by the position of the eyes.

It is evident from an inspection of the results of the analyses, that the ends differ from each other, and also from the middle. I had, however, too little ash in each instance, and not as much as is required for the most trustworthy analysis; still, on comparing them with the results of others, I find there is a sufficient coincidence to bear out the expectation that they approximate very closely to their true composition. I fear, however, that there is too great a difference between the middle and ends, or it may be that it is greater than will be found by subsequent trials; yet there is clearly a foundation for suspecting that there are real and permanent differences in their composition. The amount of water in the ends, and of solid matter, and the proportion of ash, are considerations which lead to the conclusion that these differences are not due to errors of analysis. The most remarkable differences are found in the amount of potash and of the phosphates.

It may be interesting to copy two analyses of P. F. H. Fromberg, assistant in the Laboratory of the Agricultural Chemistry Association of Scotland:

						Rose end.	Heel end.
Potash	-	-	-	-	-	$38 \cdot 15$	29.53
Soda	-	-	•	-		$5 \cdot 40$	11.26
Chloride of sodiu	ım	-		-		$6 \cdot 25$	9.66
Lime		-	-	-	-	$1 \cdot 28$	1.55
Magnesia -	-			•	-	5.90	$3 \cdot 96$
Oxide of iron	-	-	-		-	$1 \cdot 03$	$1\cdot34$
Sulphuric acid	-	-	-	-		$24 \cdot 32$	20.63
Phosphoric acid	-	-		-	_	12.81	20.73
Silica	-	-	-	-		4.86	$1 \cdot 34$
							
						$100 \cdot 000$	$100 \cdot 000$

In order to compare these analyses with my own, it is necessary to state that the lime and magnesia here given is in combination with phosphoric acid; and that in my own, those bodies are merged in the general composition of the phosphates.

V. COW-HORN POTATO.

Pr. 3 B. Fig. 1.

This is a small elongated potato of purple color, and usually slightly bent or curved. It is not in common use, but is esteemed as a very good baking potato. I have not been able to analyze it, but I have given a figure which represents the variety perfectly. It is a very good spring potato; and those who are fond of cultivating singular kinds, will add this to their number.

VI. CARTER POTATO.

PL. 2 B. Fig. 1.

Color gray and yellowish gray, often tinged slightly with purple; flesh white; form elongated; eyes few, and in 5's.

This is esteemed as a rich variety, and is extensively cultivated for market, and usually bears a high price.

			0	rgani	c anal	ysis.			
Water		•			-		-	-	$74 \cdot 00$
Starch	-					-	-	-	11.92
Casein	-	-		-	-		-	-	0.55
Albume	n								lost.
Dextrin	е	-	-			-	_		0.41
Sugar a	nd e	extract	-	-	-	-	-	-	3.78
Gluten	-	_	•	-	-	•	-	•	
Fat	-	-	-	-		-	-	-	0.21
\mathbf{Fibre}	-	•	•	•	-	-	-	-	$7 \cdot 02$
Ash per	cen	tum	-	•	-	-	-		0.88

VII. PEACH-BLOW POTATO.

WESTERN RED, LAKE ERIE, SAND LAKE, BUFFALO, KENTUCKY RED.

Pr. 2 B, fig. 1.

Color red, and marked with numerous dots; form elongated; eyes few, and in 5's.

It has some resemblance to the Merino, but is of a brighter red and rarely as long. The quality of the Western Red ranks only with the medium kinds. It is a good bearer, and is common in the winter in the Albany market. It is not much affected with the potato

disease, especially upon the sandy lands between Albany and Schenectady. From the numerous names under which it is known, it seems to be cultivated in a large extent of territory, being common in New-England and in the Western States.

Organic analysis.

			, -		-			
Water -	•	-	-	•	-			73.78
Starch -	-	•	•	-	•	-	•	12.60
Albumen	•	•	-					0.36
Casein -	-			-	•	-	-	0.45
Dextrine	-	•	•		-			0.25
Sugar -	-	-	-	-		•	-	2.22
Gluten and	fat	-	•	•	-	-	-	0.18
Fibre -	•	•	•	•	•	•	-	8.35
								98 · 19
Ash per cer	ntum	-		•	-	-		1.07

It yields more than an average quantity of starch.

VIII. HAVANA POTATO.

This variety is characterized by its elongated and tapering form. Two or three are often united at the smaller ends. It is not generally cultivated. Although it is of a medium quality, its shape is against it, and it will probably never become a favorite kind with farmers. From the garden of A. Walsh, Esq. Lansingburgh.

Organic analysis.

Water	•	-							75 · 15
Starch	-	•	•	-	-	-	-		8.93
Albume	n	•		-			-	-	0.30
Casein	-	•	•	-		-	•	-	0.05
Dextrin	e	-	•	-	•	•	-	-	0.48
Sugar	•	-	-	-	-	•	•	•	0.60
Gluten	•	•	•	•	•	•	-		0.77
Fat	-	•	•	-	•	•	•	-	0.10
Fibre	•			-	-	-			8.41
Ash per	cent	um	-	•		-			1.41

IX. LADY-FINGER POTATO.

PL. 6 B, fig. 1.

The Lady-finger is eteemed for baking: it is an elegant variety, and might be considered a fancy potato. It frequently grows quite large and long, especially when planted in highly manured land. System of eyes in 14.

Organic analysis.

Water	-	-	-	•	•	-	•		78.07
Starch	•	•	-	-	-	•	•	-	8.25
Albume	n	-	•	-	-	•	-	-	0.43
Casein	•	-	-	-	۰	•	-	•	0.43
Dextrine	е	•	-	-	-	-	-	-	0.98
Sugar	-		-	-	•	-	-	-	$2 \cdot 31$
Gluten	-	-	-	-	- 35	•	•	-	
Fat	-	•	-	-	-		-	-	0.09
\mathbf{F} ibre	•	4	-		-	-	•	•	14.37
Ash per	cen	tum	-	•	•		•	-	1.03

X. EARLY KIDNEY POTATO.

Pr. 4 B, fig. 1.

Color grayish and yellowish gray, smooth; flesh white; form elongated; eyes few, small, and in 5's.

It is a delicate early potato, but not very productive.

Organic analysis.

Water	٠	-		-			-	•	73 · 13
Starch	4	-	-	•	-	•	-	-	12.91
Albume	n	•	-	-	•	-	•	•	0.95
Casein	-	-	-	-	•	-	-	-	0.10
Sugar	-	•	-	-	-	•	•	-	3.29
Gluten	-	•	•	-	-	-	-	-	
Fat	-	•	•	-	-	-	-	•	0.05
\mathbf{Fibre}	•	•	-	-	-	•	-	•	7.13
Ash per	cen	tum	•	•	•	•			1.47

XI. ORANGE POTATO.

Color yellowish gray, eyes slightly margined with pink; flesh yellowish; form roundish; eyes in 5's.

				Or	ganie	anal	ysis.			
	Water	6	ė	-	•		-	è	-	78.44
	Starch	-	-	•	-			-	-	8.05
	Albume	n	•	•	ě -	•	•			
	Casein	-	-		-	-	•	•	•	0.35
	Dextrine	е	•	•	•	-	-	-		0.17
	Sugar	-	-	**	-	-	è	-		$2 \cdot 38$
	Gluten	and f	at	-	-	-	•	-	-	0.36
	\mathbf{Fibre}	-	-	٠	é	•	-	٠	•	8.05
A diseased specimen gave										
	Water	-		•	ė		-	•	•	81.58
	Starch	-	-	-	-	-		-	•	2.80
	Fat			-	-	•				$0 \cdot 15$
	\mathbf{F} ibre	-	6	•	•	•	•	-	•	9.65
	Ash per	cent	um	de						0.76

XII. SCOTCH GRAYS POTATO.

EARLY BLUES, STAFFORD HALL.

PL. 4 B, fig. 2.

Color bluish purple with yellowish patches, rose end darker than the heel end; flesh white; form elongated; eyes in 5's.

Organic analysis.

Water -		-	-	-	-		-	71.63
Starch -	-	• ~	•	•	•		•	$9 \cdot 28$
Albumen	•	â	•	•	-	• "	-	0.92
Casein -	•	-	-	-		-	-	0.20
Dextrine	•					ù		0.40
Sugar -	-	-			-		-	3.64
Gluten and	fat	*	ė			٠		0.40
Fibre -	-	-	•				-	11.39
Ash per cer	ntum	-	•	•	-	•	•	1.12

The Scotch Gray gives a good result in its analysis. It is, however, complained of as disposed to become watery and rather strong.

XIII. PINKEYE POTATO.

Pr. 1 B, fig. 1.

Color orange gray; eyes purple, with a waved line; flesh yellow; form round.

Organic analysis.

Water	-	.0	-	-	-	-	-	-	75.06
Starch	**	-	-	-	~	-	-	-	8.81
Albume	n	•	-	-	-	-	-	-	0.17
Casein	**	۰	•	_		-		-	0.60
Dextrine	е			-	•	-00	•	-	0.68
Sugar	-	•	-	-	-	-	-	-	2.29
Gluten	-	•	-	-	-	•	-	-	
Fat	**	-	-	-	-	-	-	-	0.12
Fibre	-	.00	-	-	-	-	-	•	8.77
Ash per	cen	tum	-	-	•	-	49	-	0.96

XIV. BLUE PINKEYE POTATO.

Bluish or purplish red, uniform; broader than long; flesh white; eyes in 5's.

The young tubers, obtained from the Albany Market, June 12, 1848; raised in New-Jersey.

-	\sim		7 .		000	
1.	Urg	anic	analysis	of	300	ors.

									rer cemum.
Water	-	•	-	-		•	•	$221 \cdot 15$	$73 \cdot 71$
Starch	-	-	•	-	-		-	$63 \cdot 43$	21.14
Albumen	•	-	-	-	-	-	•	$2 \cdot 00$	0.66
Casein	-	-	•	-	•	_	-	1.64	0.54
Dextrine	-	-	-	-	•	-	-	0.76	0.25
Fibre, gl	utei	n and	fat	-	-	-	-	11.70	3.50
								300.68	$100 \cdot 20$

Ash of the fibre 1.	5	é	5
---------------------	---	---	---

2. Analysis of the ash.

Silex										0.000
SHEX	•		•	-	•	-	-	-	-	2.000
Phospha	ate	of	lime	, mag	nesia	and	iron	-	-	$19 \cdot 083$
Magnes	ia		-		-	-	-	-	-	trace.
Carbona	ate	of	lime	•	-	-	-	-	-	0.200
Potash	-		-	-	-	-	-	-	-	40.767
Soda	-		-	-	-	-	-	-		
Chlorid	e of	f s	odiun	n.	-	-	-	-	-	$23 \cdot 452$
Sulphu	ric :	aci	d	-	-	-	-	-	-	7.001
										$100 \cdot 503$

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XV. ROUND PINKEYE (Young).

Pr. 1 B, fig. 1.

Yellowish white; roundish but flattened, somewhat shorter than wide; eyes purplish, in 5's; flesh yellowish, uniform.

Obtained in the Albany Market, June 12, and was about two-thirds its natural size.

			1.	Orga	nic an	alysi	s of 3	800 grs.		
337								238.6	or.	Per centum. 79.540
Water		•	•	•	•	•	•	40.4		13.486
Starch	•	•	•	-	•	•	•			
Albume	en	•	•	•	٠	-	•	1.5		0.526
Casein	-	-	•	-	•	-	•	1 · 7	-	0.583
Dextrin	e	•		-	•	•	-	0.5	36	0.120
Fibre,	fat and	l glut	en	-	-	•	•	12 · 8	50	$4 \cdot 133$
								295 · 27	1	98.388
Dry ma	.tter							20 · 4	16	
Ash		-			•		•	0.7	78	
Per cer	ntage o	f ash	of d	lry m	atter	•	•	$3 \cdot 8$	31	
Ash of	the fil	ore		•	-	-	*	2.0	05	
				2	Analy.	sis of	the a	ish.		
	Silex	-		-	-	-			-	1.100
	Coal	•		-	-	-	-	-	-	0.500
	$\mathbf{P}_{\mathbf{hosph}}$	nates	of m	agne:	sia, li	me a	nd irc	n -	-	27.761
	Carbor						-	-	•	0.410
	Magne					-	-		-	trace.
	Potash								•	19.941
	Soda									20.116
	Chlori	do of	eod:	11111	_	_	_			25.447
				ши	•	_	Ī	_	_	5.938
	Sulphu	iric a	cia	•	•	•	-	•	-	0 000

This analysis is calculated without carbonic acid or organic matter.

Soda, in this potato, greatly exceeds the potash, and the amount of phosphates is greater than is usually obtained. Probably the early potatoes are forced forward by the aid of rich manures, and hence the excess of phosphates and soda. This, however, is only a conjecture, which I have not the means of verifying.

XVI. ANALYSIS OF THE ASH OF THE YOUNG POTATO VINES (EARLY SHAWS).

I separated the leaves from the stem, in conformity with the method I have usually pursued in the analysis of similar substances.

1. Analysis of the leaves.

		Quai	ntity of	ash u	sed, 20	9 grs.	Per centum.
Silica -				ъ	•	5.500	27.500
Phosphates	• •		•	-	-	4.090	$20 \cdot 450$
Carbonate of l	ime •	•	-	-	-	2.599	$12 \cdot 995$
Magnesia		•	-	-	-	0.230	$1 \cdot 150$
Potash -		•	•	-	-	1.570	7.850
Chloride of po	tassium		-	•	-	2.061	10.305
Chloride of so	dium	-	•	19		1.850	9.250
Sulphuric acid	19	•	-	•	•	0.448	$2 \cdot 240$
Carbonic acid	80	ъ.	•	73	-	0.731	3.655
Soluble silica	•	•		•	-	0.260	1.300
Organic matte	r -		9	ta	•	0.231	1 155
_							
							97.850

2. Composition of the phosphates.

PHOSPHATES	-	-	ь	16	-	6	4.09
Phosphate of perc	oxide o	f iron	-	ъ		•	0.90
Lime	-	-	-	-		ъ	0.73
Magnesia -	-	*	-	-	•	-	2.08
Phosphoric acid	•	· (5, 2)	•	•	*	*	1.28
							4.09

XVII. ANALYSIS OF THE ASH OF THE POTATO STEM (EARLY SHAW). Effervescent.

								Per centum.
Silica -		•	*	-	-	-	-	5.000
Phosphates	•	6	•	•	-		-	$13 \cdot 000$
Carbonate of	lime	:	-	-	-		-	11.700
Magnesia	•	-	•	-	-	•	-	0.560
Potash -	•	•	•	•	I		•	20.740
Soda -	•	•	-	-	100	•		$2 \cdot 000$
Chloride of p	otass	ium	•	-	•	-		$34 \cdot 960$
Sulphuric aci	d	•	-					4.000
Carbonic acid	ĺ		•			R		$9 \cdot 032$
								$100\cdot 992$

XVIII. GILKEY POTATO.

MERCER, CHENANGOES, MESHONNOC, NERCHANNECK.

The young tuber, obtained in the Albany market the 7th or 8th of June. They were only about half grown, or considerably less than when in market in their season.

			O	rgani	c ana	lysis (of 30	0 grs.	
Water	us.	_	_	a ²	ď		•	243.00	Per centum.
Starch	ul'	3 ′	-		-	-	-	$39 \cdot 99$	13-33
Albumen	-	-	۰,	ο,	۰,	•	œ'	3.38	$1 \cdot 12$
Casein	-	-	-	-	-	<u>~</u> ;	4	3.64	1.21
Dextrine		-	-	-	•	-	æ	1.81	0.60
Fibre, oi	l, s	ugar a	nd gl	uten	e,	W ₁₀	-	11.85	3.95 S.

From this analysis it appears that this variety, when young, is as rich in nutritive matter as when mature.

The fibre gave ash 1.8143, or 0.6037 per centum; and consists of sulphate of lime, magnesia, and but a small proportion of phosphate of lime and magnesia.

Proportions of water and ash of young and old potatoes.

	You	ing po	tatoes,	100	grs. ga	ive		
Dry	-	•	-	-	•	-	•	17.55
Ash	es [†]	-		-	es.	•	ď	1.07
Calculated dry	o T		e.	-	•	-	-	6.08
	Olo	ī set,	100 gr	s. insi	ide ga	ve		
Dry matter	-	ei	- 6		æ			$3 \cdot 54$
Ash -	•	er"		-	o ⁱ	•	•	0.51
	Old s	et out	side, i	n 100	grs. g	ave		
Dry matter	-	ď	-	•	~	•	œ	5.72
Ash -	cot		5 ^	d	_	e.	-	1.64

It will be observed, that in the old set, the inside is mostly water; that the water in fact increases, while the inorganic matter diminishes, and it probably goes into the young set in the condition of nutriment.

Proportion of water and ash in different parts of the vines.

1. Leaves of one stalk or vine, collected on the 12th of June.

Quantity	•	-	•	-		-	781·50 grs.	Per centum. 100.00
Dry matter	ai .	•		-	•	ď	93 · 10	11.88
Ash -	•	es ⁻		ď	4	ď	$16 \cdot 03$	$2 \cdot 05$

2.	Stalk	divided	in	length	into	3	egual	parts	:	bottom.	middle,	top	4
----	-------	---------	----	--------	------	---	-------	-------	---	---------	---------	-----	---

		Вотто	m.							Per centum.
Quantity	-	•		-	4	•		210,00	grs.	$100 \cdot 00$
Dry matter	-	-	-	-	-	-		1.28		7.09
Ash -	•	-	•	-	•	•		2.84		1 · 17
		Middi	Æ.							Per centum.
Quantity	•	•	•	-	*	-		186.70		100.00
Dry matter	-	-	-	-	•	•		8.73		4.68
Ash -	-	-	•	-	•	-		1.68		0.91
		Top.								Per centum.
Quantity	•	•	•	-	•	-		$77 \cdot 200$	grs.	$100 \cdot 00$
Dry matter	•	•	-	•	•	•		3.440		2.65
Ash -	-	-	•	ė	*	à		0.408		0.23
				Ash	calcu	lated	dr	/ •		
							•	Ash.		Water.
Bottom -		٥	•	•	•	-	•	16.5		$92 \cdot 91$
Middle -		4	•	•	4	•	-	$19 \cdot 4$		$96 \cdot 32$
Top -		-	-	-	•	•	•	20.0		97.35

XIX. HARPER'S POTATO.

1. Analysis of the ash.

Silex -	-	-	-	-	-	-	•	1.118
Coal -	•	٠	•	•	-	-	•	$1 \cdot 132$
Phosphates	•	-	-	•	-	•	-	11.111
Potash -	•	-	•	-	•	4	٠	50.094
Soda -	•	٠	•	•	4	•	•	7.607
Carbonate o	f lim	ıe		-	-	-	•	6.666
Magnesia	-	-	-	-	•	•	•	0.369
Phosphates	of so	da an	d pot	ash	•	-	٠	$4\cdot555$
Chlorine	-	-	•	٠	-	-	-	2.780
Organic mas	ter	-	•	•	•	٠	•	4.593
Sulphuric ac	id	•	•	•	-	-	-	$7 \cdot 321$

2. Analysis of the phosphates.

Р ноѕр н а	TES	-	•	•	•			$11 \cdot 111$
Phosphate of	perc	xide	of ir	on	•	•	-	0.308
Soluble silica		•	-	-	-	•	•	0.225
Lime -	•	•	•	-	•	•	-	1.050
Magnesia	ė	•	•	-	-	-	-	$2 \cdot 293$
Phosphoric ac	id	-	•	•	-		•	$7 \cdot 238$
<u>-</u>								
								$11 \cdot 194$

			PROPO	RTION	5.			
Water -		•	•				•	77.55
Dry matter	•	•		•	-	-	• >	$22 \cdot 45$
Ash -					-			0.79

This potato was introduced into this country by Mr. HARPER, from England, and is found to be an excellent variety, and very little subject to decay.

From the foregoing analyses, it will not be difficult to determine the exhausting powers of the potato crop. In doing this, it will be useful to consider the tuber only, as this is the part which is removed from the field. The stalks, I believe, almost invariably are left to decay upon the ground, and hence restore to the soil the substances they had taken from it during their growth. Sometimes, however, it may be an object with the farmer to put them into a compost heap, for the purpose of securing a more perfect decomposition than will usually take place upon the surface. In this case, it will be observed that the leaves and stem are rich in inorganic matter: potash, from 25 to 35 per centum. Chloride of potassium and sodium, as well as the phosphates of lime, magnesia and iron, form also important items in the ash, or inorganic matter. The tuber, however, being removed, and consumed often in a distant market, its elements are lost to the soil, and so far they exhaust it of important matters. But as the potato contains a large percentage of water, and a small amount only of ash, it would seem not to exhaust the soil rapidly; still when it is considered that a large amount is yielded, it will not escape our observation, that the loss to the soil is quite large. So also it will be observed that the ash is rich in those elements which it is expensive to restore, namely, the alkalies and phosphates, or bone earth as it is sometimes called. Nearly one half of the ash is potash. Hence it is evident that attention to the elements removed in the crop is of the highest importance.

The amount of elements removed in an ordinary crop of potatoes, may be stated as follows:

							In o	ne ton of tubers.	Removed from an acre.
Potash	•	•	•	•	•			8.40	42.00 lbs.
Soda	•	•	_	•	•		•	3.00	$15 \cdot 00$
Lime	•	-	•	-	•		•	. 1.00	5.00
Magnes	ia	-	-		•	•	-	1.61	8.05
Phospha	ites of	lim	e and	mag	nesia	•		3.57	17.85
Sulphur	ic aci	d	-		•	•	-	1.70	$8 \cdot 50$
Chlorine	-	•	•	-	-	-	•	0.21	$1\cdot05$
Silica	•		•			•	-	0.51	$2 \cdot 55$
									
								$20 \cdot 00$	$100 \cdot 00$

Thus one hundred pounds of inorganic matter is removed from an acre in the course of an ordinary crop of five tons; and this is a low estimate, as fourteen tons of potatoes are frequently obtained from the acre. More than one half is potash and soda, and a large item consists of the phosphates of lime and magnesia.

In the potato crop, as in the grain crop, magnesia and lime are in combination with phosphoric acid; and it is only in a very small proportion that these important bodies appear in the form of carbonates in the ash, a form which is derived from incineration, and consequent change of the organic acids present in the potato.

It is necessary to remark here in regard to the estimates which are made of the removal of the inorganic matter, that no one variety could exactly represent the amount removed: there are small variations in the percentage of varieties. These variations, however, are not much greater than those which are furnished by the two ends of the same potato; a difference which is not accidental, but one by which important results are accomplished in the economy of this valuable plant.

If we consult the analyses of the potato for the purpose of determining the best mode of culture, or what manures are the best adapted to it, we shall readily be able to satisfy our minds; for it is evident that it is strictly a potash plant, and requires the use of those substances which are rich in this element. There is no doubt, therefore, of the value of ashes, in some form or other, to this plant; and it is highly probable the potash in the soil is the element which is exhausted first of all. A plant, however, which every body feels competent to raise, scarcely requires comment in this place.

Closely allied to the potato, is the *Tomato*, a vegetable which is already extensively cultivated, and is rapidly becoming more so. The vine and leaves have been carefully examined in the laboratory, and are found to possess the following composition:

XX. TOMATO VINES (THE LARGE RED VARIETY).

They were cut August 6, 1847, and were only half grown.

1. Proportions of water, ash, etc.

Per centum	of w	ater	-	-	•	-	-	88.39
Dry matter	-	-	-	-	-	-	-	11.61
Ash -	-	-	-	-	•	-	-	1.68
Ash calculate	ed fr	om the	e dry	matte	er -	-	-	14.47

2. Analysis of the vine or stem.

Silica	-	•	-	-	•	•	•	-	4.720	
Phosphat	es of	lime,	etc.	-	-	-	-	-	$13 \cdot 200$	
Lime	-	-	-	-	•		-	-	8.151	
Magnesia	1	-	-	-	-	-	-	-	$0 \cdot 150$	
Potash	-	-	-	-	-	-	-	-	$45 \cdot 482$	
Soda	-	-	-	-	-	-	-	-	1.632	
Sulphuri			-	-	-	-	-	-	0.336	
Chloride	of so	dium	-	-	-	-	•	-	17 · 135	
Carbonic			-	-	•	-	-	-	trace.	
Organic	matte	er	-	•	-	-	-	•	$7 \cdot 275$	S.

3. Composition of the phosphates.

PHOSPHA	TES	•	-	•	-	-	•	$13\cdot 200$
TD1 1 0			۸.					
Phosphate of	perox	ide (ot iron	-	•	•	-	$1 \cdot 450$
Phosphate of	lime	-	-	-	-	-	•	$3 \cdot 243$
Magnesia	-	-	-	-	-	-	-	$0 \cdot 100$
Silicic acid	•	-	-	-	-	-	-	0.450
Phosphoric ac	cid	-	•	-	-	-	-	$7 \cdot 957$

Composition of tomato leaves obtained from the same vines.

1. Proportions of water, ash, etc.

Per cent	um	-	-	-	-	-	-	-	$80 \cdot 030$
Dry mat	ter	-	-	•	-	-	-	-	$19 \cdot 970$
Ash	-	-	-	-	-	-	-	-	$4 \cdot 250$
Ash calc	ulate	d fron	n the	dry m	atter	_	-	-	21.282

2. Analysis of the leaf (slightly effervescent).

•	, ,		•	•			•	
Silica	-	-	-	-	-	-	$33 \cdot 950$	
Phosphates -	•	-	•		-	-	13.825	
Lime	-	-	•	•	-	-	$22 \cdot 211$	
Magnesia -	-	•	-	-	-	-	$3 \cdot 710$	
Potash	-	-	-	-	-	-	13.378	
Soda	-	-	-	•	-		$2 \cdot 782$	
Chloride of po	tassium	-	-	•	-	-		
Chloride of so	dium	-	-	-	-	-	4.874	
Sulphuric acid	•	-	-	•	-	•	$2 \cdot 139$	
Carbonic acid	-	-	-	•	-	-	trace.	
Organic matter	r -	-	-	•	-	-	$1 \cdot 425$	
J								
							98·294 S	_
							50 201 0	•

3. Composition of the phosphates.

PHOSPHATES	-	-	•	•	•	•	13.825
Phosphate of perc	oxide	of ir	on	-	-		4.825
Lime	-	-	-	-	-	•	$3 \cdot 991$
Magnesia -	-	-	-	-	-	-	0.075
Silicic acid -	•	-	•	-	-	-	0.260
Phosphoric acid	-	-	-	-	-	-	4.674

It will be observed that the tomato vine differs in composition from that of the potato. The leaf of the tomato contains more silica, and no chloride of potassium.

This plant evidently requires a rich soil — a rich sandy loam. The varieties of it which may be produced by cultivation, are equally numerous with its congener the potato.

Composition of the leaves and stems, calculated without organic matter.

							1. STEMS.	2 LEAVES.
Silica -	-	-	-	-	-	•	$5 \cdot 152$	$35 \cdot 05$
Phosphates	-	-		-		-	14.410	14.27
Lime -	-	-	-	-	-	-	8.890	$22 \cdot 94$
Magnesia	-	-	-	-	-	-	0.163	3.82
Potash -	-	-	-	-	-	-	49.660	13.81
Soda -	-		-	-	-	-	0.770	2.87
Chloride of	sodiu	ım	-	-	-	-	18.706	5.03
Sulphuric ac	eid		-	-	-	-	0.366	$2 \cdot 20$

By this comparison, the actual differences between the leaves and stems are more distinctly seen. The organic matter is of no consequence, so far as the ash is concerned. It is remarkable that the leaves contain so large a percentage of silica, and so little potash in comparison with the stem. An analogous fact exists with respect to the potato vine.

XXI. ORANGE CARROT.

As only a few analyses have been as yet made of carrots, beets and other root crops, I deem it expedient to place them in this connection, rather than devote to them a separate chapter.

The specimen analyzed was a large fine root eight or nine inches long, and one and a half or two inches thick at the large end. It was of a bright salmon-color, and perfectly sound. It was raised in the Hudson-river district, upon a sandy loam based upon the Albany clay. Analyses of this soil have been given in the first volume, p. 260.

				1. 4	1nalys	sis of	20 g	rs.	
							Ŭ		Per centum.
Silica		-	-	-	-	-	-	1.580	7.900
Phospha	e of	lime,	iron	and r	nagn	esia	-	3.800	$19 \cdot 000$
Lime	-	-	-	-	-	-	-	0.010	0.050
Magnesia	ı	-	-	•	•	-	-	trace.	trace.
Potash	-	-	-	-	-	-	-	7.659	$38 \cdot 295$
Soda	-	-		-	-	-	-	1.467	$7 \cdot 335$
Chlorine	-	-			-	-	-	0.370	1.850
Sulphurio	acid	l	-	-	-	-	_	0.343	0.715
Carbonic	acid		-	-	-	-		4.591	$22 \cdot 955$
Organic 1	matte	r	-	-	-	-	-	0.036	0.180
								19.856	98.380

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2. Calculated without carbonic acid and organic matter.

Silex -	-	-	-	•	-	-	-	10.415
Phosphates,	etc.	-	-	-	•		-	25.051
Carbonate of	lime	9	-	-	•	-	-	0.060
Magnesia	-	-	-	-	-	-	-	trace.
Potash	-	•	-	-	-	-	-	50.391
Soda -	-	-	-	-	-	-	-	9.560
Chlorine	-	-	-	-	-	-	-	$2 \cdot 438$
Sulphuric ac	id	-	-	-	-	-	-	$2 \cdot 261$
								$100 \cdot 176$

PROPORTIONS.

One hundred grains of the inside of the root, in a water bath at 212° Fahr., gave

Water	•	-	-	•	-	-	89.520
Dry matter -	-			-	-	-	10.480
Ash	-	•	-	-	-		2.200
Calculated dry	-	-	-		-	_	20.992

The outside treated in the same way, gave

Water	-	-	-	-	-	-	-	83.78
Dry matter	-	•	-	-	-	-	-	16.22
Ash -	-	-	-	-	-	-	-	2.58
Calculated di	٠v	-	-	-	-		-	15.91

The carrot, of which the foregoing is an analysis, was preserved through the winter and part of the spring of 1848. Probably it had lost a portion of its water; and hence it would have furnished a larger percentage of dry matter and ash, had it been taken from the field.

3. Analysis of the young Orange carrot.

Silica			-	-	-	-	-	$4 \cdot 40$
Phosphates -			-	-	-		-	28.60
Carbonate of li	me ·		-	-	- ,	-	-	$5 \cdot 40$
Magnesia -		. ,	-	-	-	•	-	9.58
Potash		-	-	-	-	-	-	27.54
Soda			-	-	-	-	-	15.28
Sulphuric acid		•	-	-	-	-	-	$2 \cdot 42$
Chlorine -		-	-	-	-	-	•	2.60
Carbonic acid -		-	-	-	-	-	-	4.52
								100.34

From this single analysis, there appears more soda and less potash than in the old root. They were raised on different soils; and in order to establish the fact of material differences existing, it would require repeated analyses.

4. Analysis of the young carrot leaves, belonging to the preceding specimen.

Quantity	taken	20	grs.
----------	-------	----	------

Silica -		-	-			-	1.900	Per centum. 9:500
Phosphates	-	-		-	-	-	3.710	18.550
Carbonate of	lime	-	-	-	-	-	1.960	9.800
Magnesia	-	-			-	-	0.446	$2 \cdot 230$
Potash -	-	-	-	-	-	-	6.979	$29 \cdot 895$
Chloride of p	otass	ium	-	-	-	-	$1 \cdot 359$	$6 \cdot 795$
Chloride of s	odiun	n.	-	-	-	-	$1 \cdot 235$	$6 \cdot 175$
Sulphuric aci	d	-	-	-	-	-	1.750	8.750
Carbonic acid	l	•	-	-	-	•	$2 \cdot 101$	10.505
							${20 \cdot 431}$	${102 \cdot 200}$

Proportions of water and ash in the young root and stems.

Root	-	-	-	-	-	-	-	-	100.00
Dry m	atter	-	-	-	-	-	-	-	10.48
$\mathbf{A}\mathbf{s}\mathbf{h}$	•	-	-	-	•		-		0.94
Ash ca	lculate	ed di	у	•	-	•	•	•	8.99
plant, ex	clusiv	e of	root,	weig	hing	-	-	-	252·3, gav
Dry m	atter	-	-	•	•	-		-	35.5

One plant, exclusive of root, weighing - - 252·3, g

Dry matter - - - - - - 35·5

Ash - - - - - - - 5·2

Per centum of ash in the whole top - - 2·06

From another specimen, I obtained from 100 grs.

Dry ma	atter	-	-	-	-	-	-	-	14.00
Ash	-		-	•	-	-	-		1.30

The adherence of a small quantity of dirt may have aided in increasing the amount of dry matter, as well as of ash, though I am inclined to believe the last proportion nearer the truth than the first.

XXII. BLOOD BEET.

Soil a sandy loam, based upon the Albany clay.

1. Analysis of 18 grs.

						_				
				,		Quar	ntity obtain	ied.	Without car	bonic acid, entum.
Silex -	•	•	•	•	•	-	1.180			730
Phosphate	-	•	ar	•	-	•	2.615		19	324
Carbonate of	lime	-	-	-	-	•	0.140		1.	042
Magnesia	-	-	•	-	-	-	0.576		4.	146
Potash -	-	-	•	•	-	•	6.325		46 ·	684
Soda -	-	-	-	-	-	-	1·34 3		9 -	928
Chlorine	-	-	•	-	-	-	0.304		$2\cdot$	245
Sulphuric aci	d		•	•	-	•	1.032		7.	628
Soluble silica		-	-	-	•	•	0.012		0.	061
Carbonic acid	L	•	•	-	-	•	4.307			•
							$\overline{17.834}$		99.	788
				PRO	PORTIO	NS.				
100 gr	s. ga	ve,	Water		of.	æ	•		86.42	
	Ū		Dry ma	atter	-	-	•	-	$13 \cdot 58$	
			$\mathbf{A}\mathbf{s}\mathbf{h}$	-		-	-		1 · 10	
			Calcula	ated	dry	=	œ		8.10	

The specimen furnishing the foregoing results had probably lost a portion of its water, having been kept through the winter, and exposed a day or two upon a fruit stand in market.

The herbage	of one	plant	, we	ighing	-	•	-	••	648.5 grs. gave
Dry m	atter	•	-	•	-	•	•	-	47.6
$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	•	•	-	-	-	-	8.5
In one week,	the he	erbage	incre	eased	to -	•	-	-	3400 grs.
The re	not -								408

Another beet and its top, three weeks afterwards, weighed:

Root	-	-	-	-	-	15840 grs.	or 2 lbs. 1 oz.;
Top	-	-	-	-	-	7680 "	or 1 lb.;

the root having increased much faster than the top, during the time specified.

Those plants whose herbage is large, have large roots in the end.

2. Analysis of the young beet.

Diameter of the root, half an inch; length, 4 inches.

Silica -	•	-	-	-	•	•	-	$9 \cdot 96$
Phosphates	-		co.		•	-		$34 \cdot 32$
Carbonate of	lime		-		<u> </u>	-		7.96
Magnesia	-	•	-	-	à	ń	•	2.88
Potash .	6	-	6		-			11.74
Soda -	-	•	-	•	-	-	-	1.16
Chlorine	٠	•	•		-	•	•	$4 \cdot 00$
Sulphuric aci	d	•	•		•	-	-	$9 \cdot 90$
Carbonic acid		-	6	-	-	-	-	5.56
Soluble silica	•	-	-	۰	-	-	۰	1.60
Organic matt	er		6	•	•	-	•	8.82
Ü								
								$97 \cdot 90$

3. Analysis of the young beet top.

					Per centum.	Removed in a ton of dry tops.
Silica	-	•	-	-	$10 \cdot 550$	$236 \cdot 32$ lbs.
Phosphates -	٠	•	•	•	15.700	$351 \cdot 68$
Carbonate of lime	-	-	•	-	$6 \cdot 550$	$146 \cdot 72$
Magnesia	•	•		٠	2.900	$64 \cdot 96$
Potash	•	-	-	-	$30 \cdot 435$	$682 \cdot 08$
Soda	-	-	-	-	6.475	$144 \cdot 48$
Sulphuric acid -	۰	۰	•	•	none.	
Chloride of sodium	•	•	•	-	$9 \cdot 195$	$206 \cdot 08$
Carbonic acid .	•	•	•	•	$12 \cdot 190$	$273 \cdot 28$
					${93\cdot 995}$	$\frac{-}{2105 \cdot 60}$
					00 000	- 100 00

A ruta baga obtained at the same time, and in the same state, gave the following proportions in 100 grs.

Water .	•	•	٠	•		•		86.51
		•	•	•	•	•	•	$13 \cdot 49$
Ash -	-	•	-	•	•	-	•	1.25
Calculated d	lrv	•	•	-	-	•	•	9.26

These proportions, for one trial, gave a greater amount of ash and dry matter in the ruta baga than in the beet, and less water.

XXIII. PARSNIP.

This analysis, it will be observed, is incomplete; but it shows that the parsnip is rich in phosphates.

1. Analysis of 12 grains of the ash.

				Quai	ntity obtain	ned.	Calculated without carbonic acid.
Silica	-	-	•	-	1.600		Per centum. 13·705
Phosphates of lime, n	nagnesi	ia and	iron	-	4.750		$40 \cdot 689$
Carbonate of lime	•	•	•	-	0.010		0.085
Magnesia	-	•	•	•	0.080		0.680
Potash	-	-	-	-	3.744		31.560
Soda	-	•	-	-	1.478		12.666
Soluble silica -	•	-	•	-	0.012		0.102
Carbonic acid -	-	-	-	-	1.201		
					12.875		99.487
		PROI	PORTIO	NS.			
100 grs. gave,	Water			•	-		21 · 190
B B)	Dry m				•		78.810
	Ash	•	_	-	-		4 · 150

5.265

2. Analysis of 100 grains of the ash.

Calculated dry

Silica -	-				Per centum. 12:800	Removed in a ton. 286.72 lbs.	Removed from an acre. 1720:32 lbs.
Phosphates	-	•	•	•	$36 \cdot 110$	808.64	4851.84
Carbonate of	lime	-	-	•	0.890	$20 \cdot 16$	$120 \cdot 96$
Magnesia	-	-	•	-	0.650	14.56	$87 \cdot 36$
Potash -	•	•	•	•	35.052	$785 \cdot 12$	$4710 \cdot 72$
Soda -	-	-	-	•	12.954	$280 \cdot 08$	1680 • 48
					98.516		

An acre is estimated to yield at least six tons of parsnips.

RECAPITULATION AND GENERAL REMARKS ON THE FOREGOING ANALYSES.

- 1. It will be observed that the potato contains a large percentage of water, averaging seventy-one or two per centum. In this respect it exceeds the cereals, and hence is inferior to them in equal weights as a nutritious food. It contains, however, less water than the beet, carrot or turnip; and there is a closer resemblance in composition between the potato and the cereals, than between the latter and the other root The potato, for instance, contains a large amount of starch, an element which renders it a highly important article for food. The potato furnishes, in the chlorides which are formed in the analyses, phosphates possessing the same characters as that found in the grains, a phosphate which is insoluble in water, and fusible at a low red heat. This is not found, however, where the phosphoric acid is obtained in combination with a base, as iron. It is not thrown down by ammonia, with phosphate of iron, lime and magnesia, and hence appears among the last results of analysis in the chloride. I have given it as insoluble matter in the chlorides. It contains a trace of silex and phosphate of lime, but consists of phosphate of potash and soda. The grains and the potato always furnish this product, in which respect they resemble each other; but the beet and carrot, though they may give a small amount of insoluble matter, yet it consists principally of soluble silica and organic matter.
- 2. The nitrogenous matters, in an equal weight of the potato and the cereals, are much less in the former; yet the latter give a greater amount upon the acre. This fact, it appears to me, decides the question as to the real value of the crop. The percentage of the nitrogenous or proteine bodies, as they are usually termed, appear small when compared with the other elements, or with those of the grains: they rarely amount to more than 1.50 or 2 per centum. The albumen and casein, which represent the proteine bodies, seem liable to some considerable variation by circumstances. They are liable, for example, to spontaneous changes in the course of the analysis; for, in some instances, very little albumen is obtained, and the whole product for proteine matter may be coagulated and thrown down by acetic acid. This result, however, does not alter or change the fact in regard to their existence in the potato. It will be observed, also, that there are differences in the amount of albumen and casein in different varieties. For cooking, and suiting the palate, it is not perhaps necessary that the kind of potato should be that which is richest in proteine compounds; but for feeding stock, it is a matter of more consequence. I doubt much, indeed, whether the use and value of the starch in this esculent is well understood; that it is merely intended for the support of respiration and animal heat, may well be doubted. These purposes, it is true, are important, and highly so; still, animals fatten rapidly upon potatoes. We may, I think, without danger of falling into a gross error, admit that the nutritive properties of the potato are not due alone to the proteine bodies.

- 3. The ash of the potato is small, and in this respect again it resembles the cereals. Another peculiarity consists in the large percentage of potash, and almost total absence of lime and magnesia; which shows a still further resemblance to the cereals. In a few instances, soda has exceeded in amount the potash; but as a general rule, the proportion of potash amounts nearly to one half of the whole inorganic matter.
- 4. I may once more refer the reader to the differences which exist between the parts of the potato, as exhibited in the analyses of the Mercer and Merino. The differences are not confined to the water, but extend also to the alkalies and phosphates. These differences meet perhaps with a partial explanation in the fact that the phosphates and alkalies are not so much in combination with the solid parts (fibre), as in solution in the juices. This appears from the fact that the fibre contains but a small proportion of phosphates and alkalies.
- 5. The fibre, however, must be regarded as a highly nutritive substance, and as well worthy of attention when the potato is used for starch. Indeed, animals which have been fed upon it exclusively, have thriven and grown fat; and hence as a matter of economy, establishments for the manufacture of starch from the potato ought to undertake the fatting of stock with the fibre, which is now too often regarded as an entirely worthless article.
- 6. I have made but partial examination of the condition of the diseased potato. It has been found, however, to contain more water, and less starch, than the sound tuber. When placed in a posture favorable for the purpose, it dries rapidly.
- 7. A comparison of the relative composition of the tops of potatoes is given in the analysis of the Early Shaw. In the tops, lime, magnesia and potash in the form of carbonates, exist in their ash; though as carbonates, they are merely the products of combustion, being derived from organic acids, which, during ignition, are converted into carbonates. The leaves and stem, when compared, present remarkable differences. The leaves, for example, give a much larger per centum of ash. In the leaf I find a large per centum of silex; in the stem, much less. The leaf too is rich in phosphates, particularly phosphates of iron and lime. In the stem, the phosphate of iron is but little more than a sixth part. While a magnesian phosphate abounds in the tuber, as in grains, it forms only a trifling element in the stem and leaves.

There is an apparent anomaly in the proportion of water and ash in different parts of the stem. Thus the percentage of ash seems to diminish from the root upwards, while the water increases; but in the leaves, the solid residue and ash is much greater than in the bottom of the vine. In the stalks of ripe grain, the butt or base of the straw contains less ash than the top. It will be seen that the alkalies prevail in the stem of the potato, and the phosphates in the leaves. Soda, however, which is always present in the tuber, is only found in traces in the stem and leaves.

8. It can scarcely escape the observation of an attentive reader, that potato tops must form an excellent manure. They contain all the elements essential to the cereals; and hence it is that where the tops are left upon the soil to decay, the land does not im-

- mediately appear exhausted. Still it is evident that the removal of so much potash in a crop, must, after a few years, render the land less productive. The ash, it is true, exists as a small percentage, but a large quantity is removed in a ton of tubers.
- 9. The composition of the tubers and tops indicates the kind of manure potatoes require. Ashes, inasmuch as they contain much potash and lime, are particularly useful. The value of salt as a manure is not yet determined, and yet it seems highly probable that it ranks high as a fertilizer for this crop. Gypsum also is another manure which is required: at least from the constant presence of sulphuric acid, it is evident that some compound containing it is required. The same may be said of the phosphates, as bones, or bones dissolved in sulphuric acid. Some salt of iron will be found highly useful; or the application of cinders and refuse from the smith's forge, and from iron foundries. Very few soils, however, are met with, which are deficient in iron; but this does not render the use of these substances nugatory.
- 10. Some botanists regard the tuber as a subterranean stem; but this can not be regarded as an established position, inasmuch as the tuber is wanting in the principal characteristics of a stem. Its growth and structure are much more like those of a fruit, the apple for instance; and hence the propriety of the French name pomme de terre.
- 11. It is a subject which has often called for remark, that a better crop of potatoes is raised from cut or sliced tubers, than from those which are planted entire. Different opinions prevail on this subject, in different parts of the State. There is, however, a fact connected with this question, which is important; it is the difference in the time at which the ends send forth shoots. The rose end, for example, begins to grow a week or more before the heel end. This appears to have a practical bearing: the separation of the tuber in the middle, and planting each end by itself, would seem to be the proper method to be pursued. We may obtain from the rose end an earlier crop than from the heel end; and it is probable that the difference in the time of sprouting, is connected with the greater amount of water contained in the former. This view of the question seems to be supported by the fact that diseased potatoes send forth shoots preternaturally, or at least before they appear upon sound ones. It will be remembered that I have already stated that diseased tubers contain more water than sound ones, as a general rule.
- 12. The carrot, which is regarded by farmers as a valuable root, especially for feeding stock, must be placed among the exhausting crops. Fifteen tons have been obtained from an acre. But the course which has been recommended and adopted, viz. that of sowing it with flax, is still more ruinous; for, in this mode of culture, two exhausting crops are cultivated together.
- 13. The parsnip gives a larger per centum of ash than the carrot; and admitting an equal crop, the soil is more exhausted. It is not, however, an objection to a crop that it is exhausting; for in most cases its value will be in proportion to its exhaustion. The great desideratum is to keep the soil in a productive condition. This end can be secured only by a knowledge of the influence of crops on the soil.

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CHAPTER III.

OF THE FOOD OF CATTLE.

GRASSES, CLOVER, AND OTHER VEGETABLE MATTER USUALLY CALLED FODDER, etc.

THE food of cattle, of which I design to treat in this chapter, will be mostly limited to the grasses. This class of plants contains numerous species which are occasionally eaten; still the number on which cattle depend for support, is quite limited. They constitute a natural family of plants, possessing many characteristics in common. These are not confined to external appearance, but extend to their chemical constitution. In this latter point of view, the most remarkable character which they possess in common, is their silicious skeleton. Other plants which have been under consideration, were, properly speaking, lime plants, or lime united with an organic or phosphoric acid, or both. In the case of grasses, however, silica combined with an alkali constitutes the greatest part of their frame work, which is a hollow, light, fluted cylinder, possessing great strength for the amount of material which it contains. This is a necessary constitution, inasmuch as with almost any other they would possess stems too weak to brave the winds and storms. A sufficiency of silica, then, is the great requisite to their maturity; in which state only seed can be perfected, and their species propagated. They are, however, nutritious in their immature state, and are freely eaten in all stages of growth; but this fact should not lead us to adopt the opinion that they are equally nutritious at all stages of growth. They fulfil an important purpose, as a tender juicy food, in the early spring; but they soon reach a point where they are in the greatest perfection, after which they lose daily their best parts by transmission to the seed; while the stem becomes dry and ligneous, and is very poorly adapted even to sustenance, much less to improvement and progress.

For agriculturists, it is an important matter to ascertain when grasses are in their best condition for the food of stock. In determining this question, we may be guided by the nature of the case. It is evident that when a grass has put forth its blossoms, it has reached its natural size, its height and diameter; it ceases to grow: the energies of the plant are no longer directed to this end. Now another important function is to be fufilled: the perfection of seed. In order to fulfil this function, the nutritious matter in the stalk begins to concentrate in the embryo of the seed, or seed vessel; and in consequence of this, the stalk at least loses more than it gains; and after a short period, during which slow changes

take place, the stalk dries and becomes woody, hard, and less suitable both for mastication and nourishment. We arrive at the same results, the period of maturity, by analysis; for example, it is found that the nutrient matter does not increase in the stem of the grass after its blossoms are matured. These remarks of course apply only to those cases where the stalk, and not the seed, is to be used as food.

A character of the grasses, which is worthy of notice, is their mode of growth so as to form a turf. This, however, is not universal. The coarse grasses, as the sedges, form hummocks in marshes, but never a uniform surface. Rye, oats and wheat, never form turf. This property belongs to timothy, red-top, and several others of the better kinds of grasses. Besides giving to a landscape a superior beauty and elegance, an accumulation of nutriment is secured at the surface, which, when the turf is broken and turned over, is of the highest importance in the growth of many cultivated vegetables. It is a storehouse of nutriment, which may have been accumulating many years. This grassy surface again exerts an important influence upon the temperature, maintaining a more uniform state and condition, and preventing wide fluctuations. So we ought to remember that turf contributes largely to the stability of the earth's surface: rains and currents cease comparatively to wash and abrade or furrow the ground, and convenience is thereby secured; for what can a farmer do with a surface deeply channelled?

If we take a right view of the subject, we shall not probably form too high an estimate of the functions which the grasses perform in the economy of nature. We generally think of them as fodder for cattle, and in this light they are of the utmost importance; but we have seen that this is only a small item in the good they do, and in the services they perform. Though humble in their appearance and pretensions, they serve man's purpose in the turf, in the temperature, and in the stability and permanence of the earth's surface. To be impressed deeply with these facts, we have only to witness the moving sands of the seashore, or of a desert.

Grasses, when frequently cut or mown, become by the operation much finer than natural in their texture: it is in this way that a smooth velvety lawn is formed, which possesses so much beauty in parks. This fine grass, however, is not a very valuable fodder: it is fit only for the smaller and more tender herbivora, as lambs, calves, etc.

Grasses, though really as important as represented, are probably not the most profitable source of food for cattle. Many productions exceed the value of hay as fodder. Roots and grain, even though cultivated at an expense far greater than that which attends the cultivation and growth of timothy or red-top, still outrank them in profit and value. Still there is no substitute, in the long run, for grass and pasturage in the present arrangements: they will always form an important means of supporting our herds of cattle.

The valuable grasses belong to several genera, in each of which there are several species. Among the most important is the *Phleum*, of which there are two species in New-York; one of which, the *pratensis*, is the common timothy grass, known in New-England under the name of *foxtail*.

The common timothy may be regarded as the standard grass for all the States north of Virginia and Tennessee. South of these States, the temperature and dryness are not favorable to its cultivation. Even in the Northern States, it often suffers from drouth; or it seems to be more impatient of this state of the weather, than any of the other grasses. The drouth, or dry atmosphere, upon timothy, is often well marked after mowing it; when, if rains do not soon succeed, the sod becomes dry and parched, and the surface of the ground cracks. So a dry spring, especially if the field is fed, is seriously injurious to the future crop for the same cause. Timothy must first cover the ground, that its herbage may protect its roots; and then the farmer may be secure, the entire season, of a fine crop of this grass. It will continue to grow until late in antumn.

Timothy is regarded as a great exhauster of the soil, and on this account it is objected to, but without sufficient consideration; for this reason, it is the more valuable food, furnishing in itself an amount of nutriment equal to its exhausting powers. This view may be true only where the farmer consumes the product on his farm, and is in a condition to restore a large proportion of the consumed fodder to his meadows.

The soil most congenial to timothy, is one well charged with a light black mould and occupying a rather low position, as in the vicinity of river bottoms. The mould, from its absorbent and retentive powers, secures for the season an amount of moisture sufficient for the uses of the plant.

There are two other kinds of grass which resemble the timothy, viz. Alopecurus pratensis (Meadow foxtail grass), and A. geniculatus. The former is regarded as superior to timothy: it does not, however, attain maturity from the seed under four years, and hence is not well suited to a system of alternate husbandry. The latter is a native grass, and is more abundant in moist or wet places. It is smaller than the Phleum, and is bent at the joints.

Another kind is the *Dactylis glomerata* (Orchard grass), the flower heads being clustered into several smaller heads. It is an introduced grass, and is well adapted to be sown with clover, as it comes to maturity at the same time. It is better for pastures than meadows, and requires close grazing. It grows rapidly, and, perhaps for this reason, forms a delicate herbage for sheep.

An earlier kind of grass than timothy, is the Spear grass, Meadow grass, or Kentucky blue grass: botanically it is the *Poa pratensis*. It is a native of Europe, but is extensively cultivated in this country. It is perennial, and grows well in all parts of the Union, at least as far south as Charleston, S. Carolina. One of the merits of this grass consists in its being in a condition for use during a great part of the year. It is an early grass, but continues also to quite a late period in the autumn. It furnishes but one flower stalk, which, when matured, is succeeded by fine long blades or leaves.

The genns Poa furnishes also many other species worthy of notice. The P. trivialis (Rough meadow grass) resembles the former: it is not as early, but its yield is equally great, and its quality not inferior.

In addition to these, Mr. Buckley has described, in the American Journal of Science and Agriculture, about twenty kinds or species belonging to the genus *Poa*.

Manna grass, Glyceria fluitans. This grass is remarkable for its sweetish seeds. It grows from four to six feet high in wet places. As cattle are very fond of it, it may be profitable to cultivate it in those places where it is permanently wet and otherwise unproductive. Water fowls are all exceeding foud of the seeds; and hence the margins of ponds and streams may be very profitably stocked with it. Even fish are fond of the seed.

In addition to this very brief notice of the grasses, I may allude in passing to the family of *Clovers*, which, though not botanically grasses, are still important to cattle in many instances, and often occupy the place of grasses. The red and white clover seem to be distributed generally, and indeed as much so as any of the grasses proper. Their seeds spring up in every new field, and upon old fields under favorable circumstances, especially where gypsum has been strewed.

The habits of the two kinds of clover are different: the red has an upright, and somewhat robust stem; the white, a small wiry creeping stem, which becomes a pernicious weed in a garden, very difficult to eradicate, inasmuch as the remains of part of a stem with its root is sure to live and extend itself. It exhausts the soil, after a few years, of the peculiar elements which it requires, and gradually disappears from the fields. In new pastures it is thick and abundant, and fills the air with its perfume; but in the, old it has given place to the less nutritive but more hardy grasses.

The clovers, as is well known, furnish important means for increasing the fertility of soils. They improve the soils by their large and spreading roots. They also penetrate deeply, and bring up the phosphates and other fertilizing matters to the surface. They can not increase the mineral or inorganic matters, but they increase the organic.

According to Boussingault, one acre will produce 12\frac{3}{4} cwt. of perfectly dry clover roots, which is a large amount of useful matter in whatever light it is regarded; either as a production affecting the soil mechanically, or as one which acts chemically, first, in abstracting the deeper nutritive materials, and afterwards leaving them diffused in the soil as food for other more valuable plants for man. This is not to be regarded as new matter added to the soil, but as matter transferred through the agency of this plant from points inaccessible to others, and hence becoming available by a slight expense to the farmer. It is this view of the subject which makes the clover crop so valuable in a specific rotation, by furnishing an immense amount of nutritive matter to the wheat crop, which is in the most favorable condition to promote its growth and perfection.

Lucerne yields a much larger amount of roots, which, after the crop has remained a few years, are cut up with the turf and burned. This crop is well adapted to sandy soils of the Hudson river valley, and might, it is believed, profitably receive the farmer's attention. It is not, however, adapted to the use of milch cows, as it imparts an unpleasant taste to the milk and butter; but it is wholesome and nutritive to most kinds of stock.

Although the grasses and clover are spoken of as the food of cattle, either of which may be employed indiscriminately for the other, still it will be shown that they differ

materially in composition. Grasses are strictly silica plants; but the clovers, in place of silica, contain lime and magnesia. The other nutritive bedies, however, are not very dissimilar, consisting as they do in each class of albumen in conjunction with woody fibre, and phosphates of lime, magnesia and iron.

As in the potato, so in the grasses, it is essential that we should ascertain the proximate elements of which they are composed. I have been able as yet to make but a few analyses of these products, scarcely sufficient, it is supposed, to aid in estimating their practical use in feeding and fattening stock to the greatest advantage. Regarding the inorganic and organic parts as of equal importance, the work of analysis is but half completed when it stops with the ash. As my attention has been mostly directed to the ash analysis, I shall first give my results in this research; designing, if necessary, to select from the works of others such analyses of the organic parts as shall be deemed sufficient to answer the inquiries of agriculturists, or so much as I deem necessary to an entire view of the subject, or to a somewhat complete exhibition of the composition of these plants.

I. TIMOTHY GRASS (Phleum pratensis)*.

1. First specimen: collected May 20, 1848.

Stalk 24 inches long; head not visible.

		r	ROPOR	TIONS.				
Stalk	*	-	-		٠		65	30
Leaf	•	•	è	•	•	•	34.	70
							100.0	00
Water in the	stalk	-	-	•	•	•	-	81.00
Dry matter	•	•				-	•	17.80
Ash -	-	•	-	•	6	4	-	1.20
Ash calculate	d dry	-	•				-	6.74
Water in the	leaf	•					•	75.00
Dry matter	•	-	-	-	- "	-	-	23.00
Ash •	•	-		•	•		-	2.00
Ash calculate	d dry	٠	•	-		•		8.69
Water in the	whole	e plan	t	œ		•	6	78.00
Dry matter	•	-	-	•	•	-	-	20.46
Ash -	6	٠	•	-	•	-		1.60
Ash calculate	d dry	-	-	•	•	•	•	7.82

^{*} By Mr. Ball, Hoosic Falls, Rensselaer county.

2. Second specimen: gathered June 30, 1848.

Whole height 40 inches; head 5 inches long, in full blossom.

			PROPO	RTION	٤.			
Stalk	-	-	-	-	-	-	$52 \cdot$	03
Leaf	-	•	-	-	-	-	$27 \cdot$	64
Head	-	-	-	-	-	•	19 ·	7 3
							100	00
Water in the	stalk	-	-	•	-	-	-	$74 \cdot 75$
Dry matter	~	-	-	-	-	-	-	$23 \cdot 75$
Ash -	-	-	-	-	-	-	-	1.50
Ash calculate	d dry	-	-	-	-	-	-	6.31
Water in the	leaf	-	-		_	-	-	71.07
Dry matter	-	-	-	-	-	-	-	26.56
Ash -	•	-	-	-	-	-		$2 \cdot 37$
Ash calculate	d dry	-	-	-	-	-	-	$8 \cdot 92$
Water in the	head	-	-	-	-	-	-	$66 \cdot 25$
Dry matter	-	-	-	-	-	-	-	$32 \cdot 30$
Ash -	-	-	-	-	-	-	-	1.45
Ash calculate	d dry	-	-	-	-	-	-	$4 \cdot 45$
Water in the	whole	plaı	ıt	-	-	-	•	70.69
Dry matter	-	-	-	-	-	-	-	27.54
Ash -	•	-	-	-	-	-	-	1.77
Ash calculate	d dry	-	•	-	•	•	•	$6 \cdot 42$

3. Ash of timothy grass.

This specimen was made into hay, and was very bright and thoroughly sun-dried. Length 3 feet and 8 inches; head 7 inches, and cut before its blossoms were mature. 100 grains of the thoroughly sun-dried hay lost in the water bath, 10.30; ash, 3.00. Of carbonic acid, only a trace was found in the ash.

Quantity employed	-	-	_	-	-	20.000 grs.	
Silica	_	_	_	_	_	8:330	Per centum. 41.650
Phosphates	_	•		•	_	3.385	16.925
Carbonate of lime	-	-	-	•	-	9.040	0.500
Magnesia	-	-	-	-	-	0.100	0.500
Potash	-	-	-	-	•	$6 \cdot 152$	30.760
Soda	-	-		-	•	0.204	1.020
Soluble silica -	-	-	-	-	-	0.040	0.500
Chloride of sodium	-	-	-	-	•	0.498	$2 \cdot 490$
Sulphuric acid -	-	-	-	-	-	0.826	$4 \cdot 130$
						$\overline{19\cdot535}$	97.875

co	MPOSI	TION O	F THE	PHOSE	HATE	3.	
Phosphate:	s -	-	-	-	-	-	$3 \cdot 385$
Phosphate of pe	roxid	e of i	ron			-	0.400
Lime	-	-	-	-	-	-	1.700
Magnesia -	-	-	-	-	-	-	0.490
Phosphoric acid	-	-	-	-	-	-	1.537

The ash contained six per centum of organic matter, which is not noted in the work.

From the foregoing analyses, there will be removed from the soil, in timothy hay, the following amounts of the elements which the ash is found to have contained:

						In one ton.	From an acre.
Silica	-	-	-	-	-	26·292 lbs.	$65 \cdot 730$ lbs.
Phosphates -	-	-	-	-	-	$10 \cdot 155$	35.387
Carbonate of lime	- .	-	-	-	-	0.120	0.360
Magnesia	-	-	-	-	-	0.300	0.800
Potash	-	-	-	-	-	$18 \cdot 456$	$46 \cdot 160$
Soda	-	-	-	-	-	0.612	1.530
Chloride of sodium	-	-	-		-	1.492	3.730
Soluble silica -	-	-	-	-	-	0.120	0.300
Sulphuric acid -	-	-	-	-	-	$2 \cdot 476$	$6 \cdot 190$

II. MIXED HAY.

Fully ripe and bright: grown upon the clay loam of Albany county.

		AN	ALYSI	s.			
Silica	-	-	-	-	-	-	53.950
Phosphates -	-	-	-	-	-	-	$12 \cdot 050$
Carbonate of lim	ıe .	-	-	-	-	-	0.250
Magnesia -	-	-	-	-	-	-	0.045
Potash	-	-	-	-	-	-	$13 \cdot 955$
Soda	-	-	-	-	-	-	5.635
Sulphuric acid	-	-	-	-	-	-	5.500
Chlorine -	-	-	-	-	-	-	1.315
Soluble silica	-	-	-	-	-	-	0.400
Organic matter	-	-	-		-		5.350
							98.450

The soil upon which this and several of the preceding substances grew, has been analyzed, and found to be composed as follows:

								I. Surface.	II. 8 inches below surface.
Silica -	-	-	-	-	-	•	•	$86 \cdot 32$	$86 \cdot 25$
Water -	•	-	-	•	-	-	-	$3 \cdot 25$	$2 \cdot 00$
Organic matte	r	-		-	-	, -		$4 \cdot 25$	2 · 18
Carbonate of l		•	-	-	-			0.20	$0 \cdot 17$
Peroxide of ire	on ar	nd alu	m.ina	-	-	•	-	3.66	$8 \cdot 53$
Magnesia -		-		-	-	•	-	0.13	$0 \cdot 11$
Potash -		-	-	-	-	-		- 0.19	0.08
Soda -	-	-	-	-	-	-	-	0.37	$0 \cdot 05$
Soluble silica					-	-	-	$1 \cdot 00$	$0 \cdot 25$
								$99 \cdot 37$	$99 \cdot 62$

In addition to the mixed grasses growing upon the soil of this lot, I may add the Early Shaw potato, beet, carrot, and tomato vines. I have under cultivation upon the same lot, several varieties of indian corn which have been already analyzed, and more in a state of preparation for this process. The soil is sandy to the depth of two or three inches, and is then succeeded by a yellowish stiffish clay, known as the Albany clay. This soil, or some combination of it with more sand or clay, is the predominant soil of the towns in the valley of the Hudson river. It is an excellent soil for many productions, retaining admirably the manures when properly mixed with it.

III. RED-TOP (Agrostis vulgaris). Selected from dried hay.

			ANAI	YSIS.				
Silica -	-	-	-	•	-	-	-	$41 \cdot 90$
Phosphates	•	-	-	-	-	•	-	13.75
Potash -	-	-	-	-	-	-	-	$4 \cdot 92$
Chloride of s	odium		-	-	-	-	-	2.00
Carbonate of	lime	•	-	-	-	-	-	$10 \cdot 03$
Magnesia	•	•	-	-	•		•	6.64
Soda -	•	-		•	•	-	-	9.61
Organic matt	er	•	•	•	•	- ·	•	$2 \cdot 35$
Sulphuric aci	d	-	•	-	-	-	•	$7 \cdot 30$
								${98.50}$

It appears from the foregoing analysis, that this is a valuable grass, containing less silica than the mixed grasses consisting of timothy, speargrass, red-top, some clovers, etc. The amount of magnesia is large, and the soda nearly twice as large as in the mixed grasses.

IV. QUACK-GRASS (Triticum repens).

1. First specimen. Grown in the rear of the Old State House: cut May 1; height 11 inches. Five whole plants weighed 124 grs., and gave the following percentages:

Water -	-	-	-	-	-	-	-	$79 \cdot 355$	
Dry matter	-	-	-	-	•	-	-	$20 \cdot 645$	
Ash -	-	-	•	-		-	•	$2\cdot 528$	
Ash calculat	ed d	ry	-	•	-	•	-	$12 \cdot 245$	
Organic ma	tter	calcul	ated d	lry	-	-	-	87·755 S.	•

2. Second specimen. From the same place: cut June 8; height 46-48 inches, just beginning to head out. Five stalks weighed 361 grs., and gave the following percentages:

Water -	_	-	-	-	**	-	-	80.232	
Dry matter	-	-	-	-	-	-	-	19.768	
Ash -	-	-	-	-	-	-	-	2.081	
Ash calculat	ed a	lry	-	-	-	-	-	10.527	
Organic mat	ter	calcul	ated d	lry	-	-	-	$89 \cdot 473$	S.

3. Third specimen. From the same place: cut June 21; height 50 inches, in flower. Two stalks weighed 174:20 grs., and gave the percentages following:

Water -	-	-		-	-	-	-	71.618	
Dry matter	-	-	-	-	-	-	-	$28 \cdot 352$	
Ash -	-	-	-	-	-	-	-	$2 \cdot 612$	
Ash calculat	ted d	lry	-	_	-	-	-	$9 \cdot 203$	
Organic ma	lter (calcula	ated d	lry	-	-	-	90.797 S	١.

The above shows — excluding the water — a gradual decrease of inorganic, and a corresponding increase of organic matter as the plant advances in age.

4. Fourth specimen. From the same place: cut May 1; height 11 inches. 100 grains of ash gave

						Per centum.
-	-	-	-	-	-	$1 \cdot 455$
-	-	-	-	-	-	$27 \cdot 150$
-	-	-	-	•	-	$17 \cdot 250$
-	-	-	-	-	-	0.112
-	•	-	-	-	-	trace.
•	-	-	-	-	-	$7 \cdot 350$
-	-	-	•	-	•	$26 \cdot 785$
•	-	-	-	•	-	$3 \cdot 565$
-			-	•	-	$5 \cdot 425$
-	•	-		•	-	4.811
P			•	-	-	$5 \cdot 200$
						99·103 S,
	-					

The above analysis was made under more favorable circumstances, and with a better ash, and hence is more deserving of credit than the following: it is, however, in an earlier stage of growth, and would necessarily differ from the first analysis.

5. Analysis of the ash of Triticum repens, just previous to the appearance of the flower-spike or head.

					0,	, ceace	•			
										Per centum
	Silica -	•	ь	•	ь	ь		*	ь	40.500
	Coal -	•	•		•	•	-	-	-	$6\cdot300$
	Phosphat	tes	-	-	-	-	-		-	15.300
	Lime -		-	•	-	ь	•	•		0.959
	Magnesia	a	•	•	-	-	•	-	-	non ϵ .
	Potash -		-	•	•	-	-	-	-	$24 \cdot 417$
	Soda -		•	•	-		•	-	-	1.433
	Chloride	of s	odiun	n	ь		-	•	-	$3 \cdot 706$
	Sulphuri	c aci	id	-			-	-	-	
	Carbonic	acio	ł	-	-	-	•	-	-	trace.
	Organic	mat	ter	ь	~	-		ь	-	2.400
										98.015
			COM	Posit	ION O	FTHE	рноѕр	HATES		
	Рно	SPHA	TES	-	•	-	•	-	-	15.300
	Phosphat	e of	pero	xide	of ir	on		100	-	1.300
	Lime -		-	•	-	-	-	-	•	3.272
	Magnesia	a	-	-	•	-	•	-	-	2.600
	Silicic ac	id	-	•	•	•		-	-	trace.
	Phosphor	ic a	cid	•	-	•	••	-	ъ	8.128
					PROP	ortion	5.			
100 grs. of the	e green pl	ant	gave							
J	Water	_	_	_			_	_	_	75:39
	Dry matt	or	_	_	_	_	-	_	_	24.61
	Ash .	-	_	_	_	_	_	_	_	2.52
	Ash calcu	- ulate	d dry		-	19				10.24

V. SPEAR-GRASS (Poa pratensis).

1. First specimen. Grown in the rear of the State House: cut May 13, large growth 16 inches high, just heading out. Eight stalks weighed 98:12 grs., and gave

								Per centum.
Water -		•	٠	•	•	•	-	81.564
Dry matter	•	•	-	•	•	•	۰	18.436
Ash -	-	-	•	-	٠	•	•	$2 \cdot 267$
Ash calcula	ated d	$_{ m ry}$	-	•	•	•	•	$12 \cdot 296$
Organic m	atter	calcul	ated o	lry	•	•	•	87.704 S.

2. Second specimen. From the same place: cut June 8; height 30 inches, in flower. Eight whole stalks, with radical leaves, weighed 210.25 grs., and gave

Water -		a	•		•	•	•	Per centum. $77 \cdot 374$
Dry matter	•	•	•		•	•	-	22.626
Ash -	-	-	•	-	•		٠	2.073
Ash calculat	ed d	lry	-	•	-	-	-	9.162
Organic mai	tter	calcul	ated d	lry	-	•	•	90.838 S

This plant also shows the same comparative decrease of inorganic, and corresponding increase of organic matter as it advances in age, as the *Triticum repens*.

3. Third specimen. From Hoosic-falls, Rensselaer county: gathered May 20, 1848; whole height 22 inches.

			PROPO	RTIONS	3.			
Stalk	•	•	•	-	•	•	43	.00
\mathbf{Leaf}	•	•	•	-	-	•	57	.00
Water in the	stalk	-	•	•	•	-	-	80.75
Dry matter	•	•	-	-	•	•	•	17.91
Ash -	•	•	•	•	•	•	•	1.34
Ash calculate	d dry	-	•		-	•	•	7.48
Water in the	leaf		•	-	•	-	-	75.50
Dry matter	•	•	-	-	•	•	•	21.56
Ash -	-	•	•	-	-	•	•	2.84
Ash calculate	d dry	•	•	•	•		•	13.17
Water in the	whole	e pla	ant	•	w	•	•	7 8·12
Dry matter	•	•	•	-	•	•	•	$13 \cdot 79$
Ash •	•	•	-	•	•	•	•	2.09
Ash calculate	d dry	-	-	•	-	•	•	10.05

4. Fourth specimen. From the last mentioned locality: gathered June 15, 1848; stalk 32 inches, in full blossom.

			PROPOR	RTIONS	;.			
Stalk	۰	•	٠	•	å	•	71	.00
\mathbf{L} eaf	-	۰	-	•	•	-	6	$\cdot 45$
Blossom		۵	à	òn	۵	•	22	$\cdot 55$
			*				100	.00
Water in the	stalk	6		•		٠	•	68.50
Dry matter	•	•	۵	۵	6	•	-	30.50
Ash •	-		-	•	•	6		$1 \cdot 00$
Ash calculate	d dry	-	-	-	•	ě	•	3.28
Water in the	leaf	-	•		-	-		$66 \cdot 00$
Dry matter	4	60	*	-	•	-	•	$31 \cdot 90$
Ash -			•		ń	•	6	$2 \cdot 10$
Ash calculate	d dry	-	•	-	-	•	•	6.58
Water in the	blosso	m		~		-	-	58.00
Dry matter		•	a	•	â	•	à	$39 \cdot 97$
Ash -		۰	_	ė			٠	$2 \cdot 03$
Ash calculate	d dry	-	•	•	-	•	•	5.08
Water in the	whole	pla	ant	6	٠	å	۵	64.17
Dry matter				-		*	•	$34 \cdot 12$
Ash -			-			-	-	1.71
Ash calculate	d dry	•	۰	• 、	-	۰	۰	5.01

5. Analysis of the ash, selected from well made hay.

								Per centum.
Silica -	•				•	•	-	$56 \cdot 320$
Phosphates	-	•	-	٠	•	٠	•	14.980
Carbonate of	lime		-	-	•	•	•	$3 \cdot 540$
Potash -	•	•	•	•	•	-	•	15.624
Soda -		•	•	•	-		•	6.828
Magnesia	•	-	-	-	-			1.996
Sulphuric aci	id	-	•	-	-		-	0.200
Chlorine	-		•	•		•	•	0.863
								$100 \cdot 351$

It is highly probable that this grass had passed its mature state, and that its stem had become drier and more ligneous than when in blossom, and hence the per centum of silica is large when compared with that of the Red-top. This grass, in consequence of its early maturity, becomes dry and hard when left standing till timothy is ripened.

The following analysis of the ash of the plant in an earlier state, represents probably its true composition when mature.

6. Spear grass: cut May 13; height 16 inches, flower spikes just appearing. 100 grains of ash gave

Silicic acid •	•		٠		•	•	Per centum. 48:300
Phosphates •	-	-	-	-	-	-	11.650
Lime	•	-		۰	-	-	0.030
Magnesia -	•	•	•	•	•	•	trace.
Potash -	-	-	-	-	-	-	3.531
Soda	-	-	•	•	-	•	$12 \cdot 505$
Sodium -	-	•	•	-	•	•	$4 \cdot 180$
Chlorine -	•	•	•	-	-	-	6.365
Sulphuric acid	-	-	-	•	•	•	$5 \cdot 156$
Organic acids	•	-	-	-	•		$4 \cdot 400$
Carbonic acid r	not dete	ermin	ed.				
							

96·186 S.

VI. ROUGHISH MEADOW-GRASS (Poa trivialis).

Grown in the rear of the State House: cut June 21; height 40 inches, in flower. Two spears weighed 61.40 grs., and gave

								Per centum	١.
Water	•	•	•	•		-		$61 \cdot 407$	
Dry matter	-	-	-		-	-	-	38.593	
Ash -	•	-	•	•	•			3.071	
Ash calcula	ted d	lry	-	-	-	8	-	5.002	
Organic ma	itter	calcul	ated d	lry	-	•	-	94.998	S.
rns hard: asl	ı slig	htly s	lkali	ie.					

Burns hard; ash slightly alkaline.

VII. SWEET FESCUE-GRASS (Glyceria fluitans).

1. First specimen. Grew three miles south of Albany, in a swamp: cut June 22; height 40 - 48 inches, in flower (aquatic). One stalk weighed 123.25 grs., and gave

										Per centum.	
	Water .		•	•			-	•	-	58.369	
	Dry mat	ter	-	•	-	-			-	41.631	
	Ash	•	•		e				-	$2 \cdot 198$	
	Ash calc	ulate	ed d	lry		. •				$5 \cdot 279$	
	Organic	mat	ter	calcul	ated o	lry		n	•	94.721	\mathbb{S} .
Ash	saline.					•					

2. Second specimen. Taken from a swamp three miles south of Albany: plant in flower. Ash of a reddish tint, and slightly effervescent. 100 grs. gave as follows:

							Per centum.
-	-			•		•	$35 \cdot 250$
-			-	-	-	-	$19 \cdot 350$
-	•	•		•	•	-	0.055
	-	•	•	-	-	-	$0 \cdot 025$
-	-	-				-	$9 \cdot 130$
-		-	-		-	-	19.840
-	-		-		-	-	1.605
•	-	-	-	-	•	-	$2 \cdot 445$
id		-	-	-	•	-	8.910
1	•	•	-	-	•	•	$2 \cdot 450$
id no	t det	ermin	ed.				
							98.060
	1	l -	d		d • • • •	d	d • • • • •

VIII. BULRUSH (Juncus effusus).

Gathered in a swamp three miles south of Albany: cut June 22; height 40-43 inches, just out of flower. One stalk weighed 46 grs.

							Per centum.
Water -		-	•		•	•	46.586
Dry matter			-	-	-	-	$53 \cdot 414$
Ash .		-	-	-		•	0.978
Ash calculat	ed dry	-	-				1.831
Organic mat	ter calc	ulated o	dry		•	-	98·169 S.

Ash saline. The proportion of inorganic matter in this plant is very small.

IX. SLENDER CLUBRUSH (Scirpus tenuis).

Gathered in a swamp three miles south of Albany: cut June 22, in flower.

										Per centum.	
	Water	-	-	•		-	-	-	-	38.241	
	Dry ma	atter	-	-		-		-	•	61.759	
	$\mathbf{A}\mathbf{s}\mathbf{h}$	•	-	-	•	•	-	-		2 ·663	
	Ash ca	lculat	ed	dry	-		•	-		4.312	
	Organi	c mat	ter	calcu	ılated	dry	-			95.688	S.
$\mathbf{A}\mathbf{s}\mathbf{h}$	saline.					•					

Analysis of the ash.

							Per centum.
Silex	•	-	-	-	-	-	27.000
Phosphates of	lime,	iron an	d ma	gnesia	•	-	16.800
Carbonate of l	ime	-	•	-	-	•	$5 \cdot 450$
Magnesia -	-	-	-	-	-	•	$0 \cdot 100$
Potash	-	-	-	-	•	•	1.610
Soda	•	-	•	-	-	-	$24 \cdot 120$
Chlorine -	•	-	•	-	-	•	$3 \cdot 360$
Soluble silica	-	•	-	•	-	•	1.000
Sulphuric acid	-	-	•	-	•	•	11.880
Carbonic acid	-	-	-	-		٠	0.710
Organic matte	r -	-	•	-	-	-	$5 \cdot 260$
							$97 \cdot 290$

In this analysis, we find large amounts of soda and sulphuric acid, and only a small quantity of potash. It is an unimportant grass, and it is not sought after by any kind of stock, though it is occasionally eaten. If it can be obtained, however, for the litter of the barn yard, or for the compost heap, it would form a valuable addition.

X. BLACKISH-GREEN CLUBRUSH (Scirpus atrovirens).

Gathered three miles south of Albany: cut June 22; height 30-36 inches, just out of flower. One stalk weighed 112.62 grs., and gave

								Per centum.
Water -	-	•	-	-	-			40.881
Dry matter	-	•	-	•	-	•	-	$59 \cdot 119$
Ash -	•	•	•	•	-	•	•	$2 \cdot 353$
Ash calculat	ed o	dry	•	-	-	•	•	3.980
Organic ma	tter	calcul	ated d	lrv			•	$96 \cdot 020$

Ash saline. The percentage of inorganic matter in these two last species of Scirpus is greater than in the Juncus effusus, and less than in the Glyceria fluitans, and the other grasses given.

XI. The Carices form another family of coarse grasses, in which the species are very numerous. I subjoin a single analysis of a *Carex*, a coarsish swamp grass, cut before the spikes had appeared.

							Per centum.
Silex	•	-	-	-	-	-	51.250
Phosphates of lim	ie and	iron	-	-	-	-	2.600
Chlorine -	-	-	-	-	-	-	0.560
Carbonate of lime)	-	-	-	-	-	$3 \cdot 500$
Magnesia -	-	-	-	-	-	-	0.400
Potash	-	-	-	-	-	-	26.060
Soda	-	-	-	-	-	-	4.065
Sulphuric acid	-	-	-	-	-	-	3.755
Carbonic acid	-	-	-	-	-	-	$6 \cdot 170$
							$98 \cdot 190$

XII. PANICUM CRUS-GALLI (Cocksfoot grass).

Grown in the yard of the Old State House: height about 3 feet, in flower. 100 grs. of the plant, nearly dry, gave

Water -	-	_		-	-		-	Per centum. 4.737
Dry matter	-	-	-	-	-	-	-	$95 \cdot 263$
Ash -	-	-	-	-	-	-	-	11.479
Ash calculate	ed d	lrv	-	-	-	-	-	$12 \cdot 049$

100 grs. of ash gave the following results:

							Per centum.	Elements removed in one ton of dry hay.
Silicic acid	1	-	-	-	-	•	$17 \cdot 325$	45.777 lbs.
Coal	-	-	-	-	-	-	1.850	$4 \cdot 995$
Phosphates	s:							
Phosp	hate	of i	ron	-	-	0.325		
Lime	-		-	-	-	0.625		
Magn	esia		-	-	-	2.831		
Silicio	e acid	١ .	-	•	•	0.625		
Phosp	horic	aci	d		-	6.894		
1							11.300	$30 \cdot 510$
Lime	-	-	-	-		•	3.060	$8 \cdot 262$
Magnesia	-	-	-	-	-	-	2.618	7.068
Potash	-	-	-	-	-	-	$36 \cdot 656$	98.971
Soda -	-	-	-	-	-	-	1.885	$5 \cdot 089$
Chloride of	f sod	ium	-	-	-	-	$5 \cdot 723$	15.452
Sulphuric a	acid	•	-	-	-	-	8.524	S. 23·015
Carbonic a	cid a	nd o	rgan	ic ma	atter	not obtai	ned.	

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OF THE DIFFERENT VARIETIES OF CLOVER, THEIR COMPOSITION, AND PROPORTIONS OF THEIR ELEMENTS.

XIII. LARGE RED CLOVER (Trifolium pratense).

Gathered in the yard of the State House: cut May 12; height 15 inches, 5 and 6 leaves above ground. Four stalks weighed 234 625 grs., and gave

								Per centum.
Water -	-	-	•	-	•	-	-	$86 \cdot 136$
Dry matter	-	-	•	-	-	-	-	13.864
Ash -	-	•	•	•	-	-	-	1.381
Ash calculat	ed d	lry	-	•	•	-	-	9.961
Organic mat	ter	calcul	ated d	lry	-	-	-	90·039 S.

The upper half of the root (two years old) of the above clover plant, with the cuticle taken off to free it of adhering soil, gave

								Per centum.
Water -	-	-	-	-	-	~	-	81.060
Dry matter	-	-	-		-	-	-	18.940
Ash -	-	-	-	-	-	-		1.529
Ash calculat	ed d	lry	-	-	-	•	-	8.073
Organic mat	ter	calcul	ated d	lry	-		-	91·937 S.

The lower half of the same root gave

								Per centum.	
Water -	-	-	-	-	-	-	-	81.298	
Dry matter	-	-	-	-	-			18.702	
Ash -	-	-	-	-	-			1.365	
Ash calculat	ed d	ry	-	-	-	-	-	7.298	
Organic mat	te r c	alcul	ated d	ry	-	-		92.702	S.

In this specimen of clover, the percentage of inorganic matter, calculated dry, is greater in the stalk and leaves than in the root, and less in the lower half of the root than in the upper.

XIV. RED CLOVER*.

1. First specimen. Gathered May 20, 1848: whole height 9 inches; buds just beginning to form.

			PROPO:	RTIONS	; ,		
Stalk	•		•	-	-	-	$63 \cdot 75$
Leaf	-	-		-	•	-	$36 \cdot 25$
							100.00

^{*} Determined by Mr. Ball, Hoosic-falls, Rensselaer county.

ANALYSES OF THE CLOVERS.

Water in the	stalk		-	•	-	-	-	88.30
Dry matter	•	-	-	-	-	-	-	10.95
Ash -	•	-	•	-			-	0.75
Ash calculate	d dry	-	-	-	-	-		6.85
Inorganic ma	tter in	a ton	ı, 159	3.44	lbs.			
Water in the	leaf	•	~		-	ь	~	78.00
Dry matter	•	-	-	•	••		-	20.00
Ash -		10	•	-	9	•		2.00
Ash calculate	ed dry	-	-	-	-		-	10.00
Inorganic ma	tter in	a tor	n, 22	4 lbs.				
Water in the	leaf a	and st	alk	-	-			84.15
Dry matter	•	-	10		-	•	**	14.48
Ash -	-		-	-	-		-	1.37
Ash calculate	ed dry	*	-	-	-			9.46
Inorganic ma	tter in	a tor	n, 21	1.681	bs.			

2. Second specimen. Gathered June 10: whole height 24 inches, in full blossom.

FROPORTIONS.

Stalk	•	9			-	•	58	$\cdot 12$
$\mathbf{L}_{\mathbf{eaf}}$	**	•		•	-	-	22	•12
Blossom	-	*	•	9	w		19	•76
							100	.00
Water in the	stalk	-	•		6		-	80.88
Dry matter	-	-	•		-	•		18.06
Ash -						-		1.06
Ash calculate	d dry	-	-	-	-	-		5.87
Inorganic mar	•		on, 13	1 • 04	lbs.			
Water in the	leaf	-	-		-	-		73.56
Dry matter	•		•	-		-		$23 \cdot 56$
Ash -	-	-	-	-		•	-	2.88
Ash calculate	d dry	-	•			-	-	$12 \cdot 22$
Inorganic mat	ter in	a to	on, 27	3.72	lbs.			
Water in the	blosso	m	-			-	-	79.06
Dry matter	•	-	• .	• .			•	19.57
Ash -		-		•			-	1.37
Ash calculate	d dry	-	-	-	•	-	-	7.00
Water in the	whole	pla	nt		-		-	77.83
Dry matter	-		•	-	•	•	•	20.40
Ash -		-	•	-	**	•		1.77
Ash calculale	d dry	-			-	•	•	8-67
Inorganic ma	•		on, 19	3 76	lbs.			

ANALYSES OF THE CLOVERS.

3. Third specimen. Gathered July 15, 1848: stalk 24 inches long, fully ripe.

		P	ROPORT	rions.						
Stalk	•	-	•	-	•	•	60.9	∌ 6		
Leaf	•	•	•	•	u .	ø	14.	94		
Head		-	•	-	•	•	24	10		
							100 ·	00		
Water in the	stalk	.	-	•			-	25.19		
Dry matter	e	ar .	•	-	•	•		70.78		
Ash -	v	-	-	•	e	-	-	4.03		
Ash calculate	d dry	•	ď	•	•		•	5.69		
Inorganic matter in a ton, 127.68 lbs.										
Water in the	leaf		•	w		•	•	33.77		
Dry matter		-	-	•	-	-	-	$57 \cdot 36$		
Ash -	-	-		-	-		•	8.87		
Ash calculate	d dry	-	•	-	•			$15 \cdot 48$		
Inorganic ma			n, 346	3·08 l	bs.					
Water in the	head	-	-	-	-	-		13.00		
Dry matter	•	-	e	•	-	•		81.37		
Ash •	•	w	•	•	•		49	5.63		
Water in the	whole	e plan	t	•	•	-	-	23.99		
Dry matter	•	-	-			•	•	69.83		
Ash -	•	v	-	-	œ	•	•	6.18		
Ash calculate	d dry		•	•	-	-	E -	8.85		
Inorganic ma			n, 198	3.24	lbs.					

XV. WHITE CLOVER*.

Gathered at Hoosic-falls, July 1, 1848: whole height 9 inches, in full blossom, second growth this season.

Water in the	who	ole pla	ınt		ø		•	81.50
Dry matter	-	•	•	-	•	•	-	16.75
Ash -	-	•	•	•	•	•	•	1.75
Ash calculate	ed dr	y -	•	•	•	-	•	10.44
Inorganic ma	atter	in a t	on, 2	34 • 08	lbs.			

^{*} By Mr. Ball.

4 7			. 7	,
Analy	2521	nt	the	ash.

		4.1	two if or	301 0	ic asi	•		
			•	•				Per centum.
Silica -	٠	•	•	٠	•	-		28.075
Phosphate	of lin	ne, m	agne	sia an	d iron	4	٠	19.325
Carbonate	of lin	ne	•	•	. •	•	•	16.730
Magnesia	-	•	-		•	•	ě	$2 \cdot 175$
Potash	٠	•	•	•	•	•	-	10.880
Soda -	-	•	-	-	•	-	-	$15 \cdot 640$
Sulphuric	acid	٠		•	eò		6	$2 \cdot 305$
Chlorine	-	•			•	-	6	0.615
Carbonic a	acid		•	•	٠	•	•	$4 \cdot 234$
								99.979

The white clover differs, as will be observed, from the red, in containing much more silica. This, however, is the only analysis which has been made; and it may appear, on a reëxamination, that it is too high.

The cultivation of red clover is regarded by the farmers of New-York as one of the important means for increasing the immediate fertility of the soil. Its cultivation forms a part of a system of rotation of crops, in which the decaying clover plant in the soil constitutes the manure for the coming crop. As a part of a plan of cropping, it is cheap and useful. Its value is supported by an analysis of its inorganic part, which is well worthy an attentive consideration. The following are all the analyses which I am able to present: the first is of clover in hits young state, before the development of its blosoms.

XVI. YOUNG CLOVER.

PROPORTIONS.

1. Young clover, growing upon a lot one mile southwest of the State House.

Wat	er -	•	•	-	-	•		ø	80.31
Dry	matter	•	-	-	•	-	-	•	19.69
\mathbf{Ash}				•	•		•	-	0.40
Ash	calculate	ed dr	v -		-		•	٠	$2 \cdot 03$

2. White clover, growing upon the same lot.

Wate	r -	-	٠	•	•	•	•	•	$90 \cdot 49$
Dry r	natter	•	•	•	-	•	•	•	9.51
Ash	-	-	•	•	-		•	•	0.85
Ash c	alculate	ed dry	•	à	car	•	o		8.94

3. Analysis of the ash of young clover.

Silica		ь			• ~		Per centum. 0.981	Removed in a ton of hay. 0.446 lbs.
Phosphates -		•					30.245	13.459
•		-	_			_		
Carbonate of lim	e	•	•	1	•	•	7.642	$4 \cdot 400$
Magnesia	•	-	•	-		•	$2 \cdot 285$	1.015
Potash .		•	٠	٠	6	-	33.688	$14 \cdot 991$
Soda · -		•	•	•	-	-	$7 \cdot 164$	$3 \cdot 147$
Chlorine		-	•	•	•	-	3.642	1.620
Sulphuric acid -		۰	-	-	•	-	6.723	2.991
Carbonic acid -		•	•	•	-	•	5.744	$2 \cdot 556$
								
							98.114	44.625

XVII. CLOVER HAY (Cut when ripe).

100 grs. of the sun-dried clover lost in the water bath - 12°73

And gave ash - - - - 5°56

1. Analysis of the stems and leaves, or the whole plant.

						Per centum.	Removed in a ton.
Silica	•	•		6	•	0.850	1.054 lbs.
Phosphates .	•	-	-	-	-	20.600	$25 \cdot 544$
Carbonate of lim	ie -	-		-	-	30.950	$38 \cdot 378$
Magnesia -	-	-	•		•	3.930	4.873
Potash	-	-	-		•	25.930	$32 \cdot 153$
Soda	•	٠	-	•		14.915	18.394
Chlorine	•		•	•	-	1.845	$2 \cdot 288$
Sulphuric acid	•	-	-	-	-	0.495	0.624
•							
						$99 \! \cdot \! 515$	$123 \cdot 308$

2. Analysis of the upper part of the stem, with the superior leaves and heads.

Silica -	•		•		•	•	Per centum. 0.810	Removed in a ton. 1.003 lbs.
Phosphates -	•		-	•	•		21.900	$27 \cdot 156$
Carbonate of lim	.e	6	-	6	•	-	$32 \cdot 333$	39.969
Magnesia -	•		•		•	-	0.200	2.480
Potash -	-		•		•	•	27.940	34.645
Soda	•	•	•	•		•	$6 \cdot 753$	8.374
Chlorine -	•	•	•	•	•	-	3.780	4.533
Sulphuric acid		•	•	•	•	•	3.366	$4 \cdot 174$
							98:682	122:364

Ú.

The two preceding analyses were of the ash as formed from well dried hay, quite ripe, which grew in Bethlehem, the soil of which is often sandy upon a basis of Albany clay. It is less rich than the soil of the State House lot; hence the large amount of phosphates and potash in the ash of the young clover.

XVIII. SWEET-SCENTED CLOVER, var. alba (Mellilotus officinalis).

Introduced and cultivated in gardens, but grows spontaneously in rich ground about Albany.

Con	iposi	tion of	fthe	upper	part	of the	e plan	t.
	•	•			•	•	1	Per centum
Silica -	•	-	-	•	-	•	-	$5 \cdot 600$
Phosphates	-	•	-	-	-	•	•	$33 \cdot 050$
Carbonate of	of lir	ne	•	•	-	•	•	28.000
Magnesia	-	-	•	•	-	-	-	$6 \cdot 040$
Potash	-	-	-	•	-	•	•	$20 \cdot 305$
Soda -	-	•	•	•	•		-	$2 \cdot 625$
Sulphuric a	cid	-	-	•	•		-	$4 \cdot 125$
Chlorine	-	•	-	•	•	-	-	1.355
								$101 \cdot 100$

One hundred grains of the sun-dried plant lost in the water bath 6.22 grains, and gave ash 2.58.

RECAPITULATION AND REMARKS.

- 1. In view of the effects of the hay crop on soils, it is evident that there are three substances which are removed in large quantities, viz: silica in a soluble condition; phosphoric acid in combination with iron, lime and magnesia; and potash in combination with silicic acid. Lime is not found in quantities so large as potash. It is to be noted, however, that grasses differ in composition. In the red-top, a favorite grass, there is ten per centum of carbonate of lime. In the Triticum repens, which grows luxuriously in the yard of the Old State House, where there is no deficiency of lime in the soil, the lime amounts to only 0.96 per centum, and the potash to 24. Soda occasionally forms a large percentage; and from the facts, it would seem probable that the plants take up soda or potash indifferently.
- 2. Of the grasses, timothy and red-top seem to have a composition well adapted to the wants of animals which feed upon hay. The proportion of silica in timothy is less than 45 per centum; and the phosphates amount to 16 or nearly 17 per centum. The carbonate of lime is 0.2 per centum; while the potash amounts to 30 per centum, or \frac{1}{2} of the whole ash.
- 3. It will be observed that the amount of water in the stem and leaf differ, being less in the leaf and head than in the stem; consequently the amount of nutriment is pro-

portionally greater in the leaf and head, than in the stem. There is less than one-third of dry matter in the stalk. Hay, in making or drying in the sun, loses more than one-half its weight in water. Sun-dried hay loses about 12 per centum, when exposed to 212° in a water bath. It does not differ materially in this respect from the straw of grains.

- 4. In a ton of dry hay, about 150 lbs. of silica, phosphates and potash, are removed from the soil. So if two tons are yielded per acre, 300 lbs. of these important substances are taken off. From this fact, it is evident that but few soils can produce an abundant harvest, unless the removed matter is returned in some form.
- 5. For a crop of hay, it is well determined what manures are required. In treating a field for grass, however, we may furnish manures indirectly; that is, apply those matters which will unlock the potash from its combination with silica. This is often effected by the use of hydrate of lime. Immediate results follow from the application of ashes, leached or unleached.
- 6. Sulphur is another element which is always present in hay: it may be in combination with lime or other alkaline earths, or in its simple state in the albuminous matters.
- 7. An interesting fact is brought out by the examination of the constituents of grasses at different periods; thus in the *Triticum repens*, the inorganic matter decreases with the age, while the organic matter increases.

The coarse grasses of swamps, and those usually known as aquatics, contain large quantities of water, while the inorganic part or ash is quite small. They do not seem to be valuable for cattle, and but few are eaten. Some, however, which are early, are sought after; and then again some of the later grasses of a coarse texture are eaten, as the cocksfoot grass, which continues green till frost. The most important use to which the coarser grasses can be applied is for litter; for which purpose, they may be placed in stables or yards, where, from the absorbent nature of their stems, they will necessarily take up the urine and other moist manures with which they come in contact.

- 8. It will be observed that clovers differ essentially in composition from the grasses: they are strictly lime and potash plants. Phosphoric acid is also equal in importance to any of the other elements. The silica amounts scarcely to one per centum.
- 9. When clover is removed from a field, and no returns are made, it is evident that it exhausts a soil quite as much as timothy. Thus 123 lbs. of inorganic matter are removed in a ton of clover hay, which consists mainly of expensive elements, viz. phosphoric acid, lime and potash, amounting in those three substances to 100 lbs.
- 10. It is also evident that we may apply the proper manures for clover. Gypsum is well known for this purpose; and the large quantity of lime which the plant yields on analysis, seems to indicate that gypsum is taken up immediately by the roots.

The use of clover to the wheat crop is indicated in the composition of its ash. Besides acting mechanically in loosening the soil, it also furnishes the elements which the wheat plant requires; and it is probable, from the quantity of water in the stems

and leaves, and the small amount of silica in the clover plant, that it decays with more rapidity and more perfectly than the grasses, and hence its effects are more immediate than those of other green manures, excepting sea weeds. We have, then, in the large stems and roots of clover, a large amount of food for other plants, and which is so combined and constituted that it is soon ready for their uses after it is covered with earth.

In many instances, the crop of clover is too large to be ploughed in to advantage. In this case, farmers are in the habit of feeding it to their cattle. The proper course to be pursued is indicated by the crop itself. If it is small, the whole ought to be given to the succeeding crop: on the contrary, if large, it is useful to feed it down wholly or in part.

CHAPTER IV.

THE CEREALS,

CONSISTING OF OATS, BARLEY, MILLET, RYE, WHEAT AND INDIAN CORN.

The importance of the cereals to man has never been over-estimated. They constitute his food in all climates; and, without them, it would be difficult for the race to subsist. In their composition they contain the materials for bone, muscle and fat, and from these all the tissues are formed and supported. Hence the cultivation of the cereals has ever constituted one of our main employments, and indeed will, if any thing, become more and more the object of our attention. In consequence of their great value as sources from whence we derive our chief nutriment, we find that the soil upon which they grow is soon deprived of some of its essential elements, which, if not replaced, will end at last in so diminishing its fertility that the seeds will cease to be perfected; when the crop necessarily fails, and the farmer is obliged to lose his labor. It is for this reason that it becomes necessary to know what elements are removed from the soil, and how much is carried away in the crop when disposed of in a distant or even in a home market.

The possession of these facts enables the farmer to supply the specific losses he is continually sustaining. He need not be at a loss on this point, inasmuch as it is mainly the inorganic matter which is removed. It is that part which is contained in the ash, or which is left after the combustion of the plant. It may appear in many instances to be inconsiderable; yet, in ordinary soils, the effect of the crop is often perceptible in one year. In the highly favored counties of Western New-York, a long period has elapsed in the cultivation of wheat, without the accompanying signs of barrenness. This fact has been already explained in the first volume. It is well known that a very large portion of the western and central counties, those constituting the Fourth district, are underlaid by decomposing shales, which supply almost continually new and fresh matter for the wheat crop. If to this is also added the fact that gypsum and clover are liberally employed, the secret of constant fertility is explained. It can not be doubted that many large tracts of country in New-York have the best soil for wheat, all things considered, in the world.

Those tracts, however, are not the best for indian corn. This cereal requires a soil constituted somewhat differently from that which is most suitable for wheat: there should be a larger amount of the phosphates. There is a much larger growth of the plant itself;

or, the number of tons of the herbage of indian corn exceeds that of wheat. From this circumstance, alone, it is evident that it must take a greater amount yearly from the soil.

Wheat contains a larger amount of silica in its straw, than either of the other cereals. Barley is allied more intimately to lime plants, than wheat or oats.

It will be observed that more attention has been given to the composition of the ash of the cereals, than to their proximate elements, as gluten, albumen, casein, etc. It is true that the latter are important matters to be considered and determined: they, however, may be placed under the head of varieties, where it becomes necessary to compare two or more in order to determine their relative powers to supply nutriment. The analysis of the ash informs us more particularly of the effects of crops on soils, and lays the foundation for a natural system of culture. It does more than this; for we also learn from analysis the proportions of the phosphates and alkalies, which are of themselves so essential to the constitution of nutritious matters.

I. OATS.

The oats which have been subjected to analysis, were grown in four of the agricultural districts: the Southern, or Fifth; the Wheat, or Fourth; the Hudson-river, or the Third, and the Eastern, or Second, which is based upon the taconic slates.

From the Southern district I received specimens of oats from Mr. Randall of Cortlandville, and from Mr. N. Salisbury of Scott, Cortland county. These oats grew upon the slates of the upper part of the Hamilton or Lower Chemung group. For an analysis of the soil, see p. 341, Vol. I.

1. Mr. RANDALL's oats, straw and chaff.

			PRO	PORTI	ons.			
Grain	-					-		491.270 grs
Straw	•	-		-	-	-	•	438.500
Chaff	-	-	•	-	•	•		$92 \cdot 220$
Ash of t	he grai	n -			•			10.400 grs.
Ash of th	_		-	-	-	-	-	21.120
Ash of th			-	•	•	-	-	$3 \cdot 566$
								Per centum.
Water of	the su	ın-dri	ed gr	ain	-			10.820
Water of	the hu	ısk		-	-	•		$13 \cdot 280$
Water of			undet	ermir	ied.			

2. Analysis of the grain.

								Per centum.
Silica		-	-	-	-	-	-	30.706
Phosphates ·	-	-	-	-	-	-	-	32.826
Lime -	.	-	•	-	-	-	-	0.221
Magnesia -	•	-	-	-	-	-	-	0.143
Potash	•	-	-	-	-	-	-	18.332
Soda	-	-	-	-	-	-	-	3.973
Sulphuric acid	ì	-	-	-	-	-	-	1.674
Chlorine -	•	-	-	-	-	-	-	0.463
Organic matte	er	•	-	-	•	•	-	$5 \cdot 447$
				5				
								93.785

The loss in this instance is too great to pass unnoticed, but I am unable to account for it, unless it is to be attributed to water in the ash, which was not dried previous to analysis.

3. Analysis of the straw.

								Per centum.	
Silica -	-	-	-	-	-	-	-	$24 \cdot 450$	
Phosphate	s of	lime, 1	nagne	esia ar	nd iror	1 -	-	31.650	
Carbonate	of li	ime	•	•	-	-	-	6.050	
Magnesia	-	-	-	-	-	-	-	0.200	
Potash -	-	-	-	-	-	-	-	17.990	
Soda -	-	-	-	-	•	-	-	$3 \cdot 140$	
Chlorine	-	-	-	-	-	-	-	0.490	
Sulphuric	acid	-	-	-	-	-	-	12.571	
Organic ac	ids	-	-	-	-		-	3.672	
_									
								100.213	В.

4. Analysis of the chaff.

							Per centum.
•	-	-	-	-	-	-	66.50
-	•		-	-	-	-	20.50
lime	-	-	-	-		-	0.20
-	-	-		-	-	-	0.10
-	•	•	-	-	•	•	5.62
-	-	-	-	-	-	•	$3 \cdot 73$
d	-	-	-	-	-	-	$3 \cdot 72$
•		-	-	-	-	-	0.40
							$100 \cdot 47$
	- - d	d -	lime	lime	lime	lime	lime

In the chaff of all grains, a larger amount of silica is found than in the other parts. It seems to be required for the purpose of protecting the grain from casual injury, or it may be from insects.

The ash of the chaff amounts to - 3.56 per centum; And of water to - - - 7.00.

The chaff, from its thinness, is usually well dried in the sun, though it seems to absorb moisture from the atmosphere. These specimens had been exposed to the dry air of the heated room for some time previous to their analysis.

5. Analysis of two specimens of oats, straw and its chaff: one of the ripe oat, and the other immature, or cut when in blossom. It is the variety often known under the name of Tantain oats. Its stalk is large and vigorous: the grain is blackish. It was furnished by Mr. Nathan Salisbury, of Scott, Cortland county, and was grown upon the Chemung shales above the Tully limestone.

PROPORTIONS.

Unr	ipe Ta	ntain	oats (in blo	ssom)	sun-d	ried.	
Water -	-	•					-	9.48
Ash -	-	-	-		•	-		$3 \cdot 15$
Calculated d	lry -	•	-	-	•	•	•	3.48
		Ripe	e Tan	tain d	oats.			
Water -		•	-	•	-	•		9.53
Ash -		-	-	-		-	-	2.37
Calculated d	ry -	-	-	-	-	-	-	2.61
	S	Straw	of the	e unri	pe oat			
Water -	•	-	-	-	-	-	-	8.40
Ash -	•		•	-	-	-		5.06
Calculated d	ry -	-	-		•	•	-	5.52
		Strau	of the	he rip	e oat.			
Water -			-	-	~	-	-	8.33
Ash -	-	-		-	-	•		7:30
Calculated d	ry	•	-	-	-	•	-	7.96
	Cha	uff of	the u	nripe	oat.			
Water -	•	-	•	-	-	-		$9 \cdot 10$
Ash	-	-	-	-	-	-		5.94
Calculated d	ry -	•	-	-	•	•	-	6.53
		Chaff	of th	he rip	e oat.			
Water -	•	•	•		•	•	-	$11 \cdot 99$
Ash	•	-	-	-	-	•	-	5.61
Calculated d	ry -	•	-	-	-		•	6.38

6. Analysis of the ripe straw of the black Tantain oats.

						30.000 grs.	Calculated without carbonic
~							acid or organic matter.
Silica •	•	-	-	•	-	2.815	13.399
Phosphates	-	•	-		•	1.870	8.902
Carbonate of li	me	-	•	•	•	1.525	$7 \cdot 259$
Magnesia -	-	6	•	-	-	0.094	0.448
Potash -	•	-	-	-	-	12.613	60.035
Soda - •	-	-	-	-	-	0.761	3.622
Sulphuric acid	-	-	-	-	-	1.209	$5\cdot 754$
Chlorine -	-		•	•	•	0.581	0.581
Carbonic acid	-	-	-	-	-	$6 \cdot 140$	
Organic acids	•	•	•	-	•	$2 \cdot 400$	
						·	
						30.018	$100 \cdot 000$

7. Analysis of the unripe straw of the same.

			•	,	•	•	
						30 · 000 .grs.	Calculated without carbonic acid or organic matter.
Silica	6	-	•		•	$4 \cdot 355$	19.540
Phosphates .	•	•	•	•	•	$3 \cdot 345$	14.666
Carbonate of lime	•	-	-	•	-	1.628	$7 \cdot 140$
Magnesia -	-	•	•	-	-	1.064	4.666
Potash -	۰		•	•	۰	9.256	40.598
Soda	-	•	-	•		1.437	$6 \cdot 302$
Sulphuric acid	-	-	-	•	-	1.364	5.912
Chlorine -	6	-	-	-	•	0.350	$1 \cdot 176$
Carbonic acid	•	-	•	-	-	$5 \cdot 939$	
Organic acids	-	•	•	•	•	1.300	
						30.099	100.000
	Re	moved	in a to	on of :	ripe st	raw. Rem	oved in a ton of unripe straw.
Silica	•	-	•	21:	907	lbs.	22·100 lbs.
Phosphates -	-	-	-	14 \cdot	555		16.587
G 1 C1'				111.	റവ		0.085

			CILLO 1 C.		ton or alpo belan.	removed in a ton or unitipe bu
Silica -	•	-	-	-	21:907 lbs.	22·100 lbs.
Phosphates	-	-	-	-	14.555	16.587
Carbonate of	f lim	ıe	-	-	11.868	$8 \cdot 075$
Magnesia	•	-	•	•	0.732	5.277
Potash	•		•	-	98.157	45.915
Soda -	•	-	-	-	5.921	7.128
Sulphuric a	cid	-	•	•	9.408	6.686
Chlorine	-	•		۰	0.950	1.330
					100 100	110,000
					$163 \cdot 498$,113.098

I propose giving in this place an analysis of the straw of the same oat, divided into two parts, bottom and top; the division being made about midway between the two points. It shows a fact which we have been conversant with a long time, namely, that parts of

the same organ may and do differ in composition. This difference is no doubt due in part to the ascent of fluids, after the roots have performed their office; or, in other words, to the law of the upward and outward movements of the sap, which has been already referred to.

I. ANALYSIS OF THE STRAW OF THE TANTAIN OAT.

			1. Un	ripe e	at str	·aw.		
				-			Top.	Воттом.
Silica -	-	-	-	-	-	-	$14 \cdot 939$	19.823
Phosphates	-	-	-	-	-	-	$23 \cdot 959$	$13 \cdot 039$
Carbonate of lir	ne	-	-	-	-	-	5.921	11.336
Magnesia -	-	-	-	-	-	-	1.206	$3 \cdot 125$
Potash -	-	-	•	-	-	-	$45 \cdot 072$	$43 \cdot 541$
Soda -	-	-	-	-	-	-	$2 \cdot 200$	1.251
Sulphuric acid	-	-	-	-	-	-	$4 \cdot 206$	$5 \cdot 016$
Chlorine -	-	-	-	-	-	-	$2 \cdot 497$	2.866
			2. R	ipe od	it stre	w.		
				•			Top.	Воттом.
Silex -	-	-	-	•	-	-	14.839	$12 \cdot 417$
Phosphates	-	-	-	-	-	-	10.637	$9 \cdot 355$
Carbonate of lin	ne	-	-	-	-	-	8.765	8.939
Magnesia -	-	-	-	-	-	-	0.774	0.604
Potash -	-	-	-	-	-	-	50.623	$53 \cdot 484$
Soda -	-	-	-	-	-	-	$7 \cdot 727$	$5 \cdot 994$
Sulphuric acid	-	-	-	-	-	-	$4 \cdot 044$	6.800
Chlorine -	-	-	-	-	-	-	2.586	$2 \cdot 395$

It will be observed, that in the four preceding analyses, the percentage is calculated without organic matter or carbonic acid, both of which necessarily vary in every kind of ash.

3. Analysis of the phosphates of the unripe straw.

Рноѕрн	ATES	-		-				Top. 4·290	Воттом. 2·220
Phosphate o	f per	oxide	e of in	ron	-	-	_	${0.210}$	0.180
Lime -	-	-	•		_	-	-	0.698	0.698
Magnesia	-	-	-	-	-	_	-	0.840	0.597
Silicic acid	-	-	-	-	-	-	-	trace.	0.060
Phosphoric a	acid	-	-	-		-	-	$2 \cdot 542$	0.685

II. ANALYSIS OF THE CHAFF OF THE TANTAIN OAT.

		1 1	<i></i>	7.		7	<i>r</i>	
		1. 1	o grs.	. ash c	y um	npe c	iay.	Per centum.
Silex	-	-	-	-	-	-	4.750	31.660
Phosphates	-	-	-	-	-	-	$3 \cdot 300$	$22 \cdot 000$
Carbonate of lir	ne	-	-	-	-	-	0.850	5.660
Magnesia -	-	-	-	-	-	-	0.384	2.560
Potash -	-	-	-	-	-	-	2.081	13.860
Soda	-	-		-	-	-	0.341	$2 \cdot 260$
Chlorine -	-	-	-	-	-	-	0.423	2.820
Sulphuric acid	-	-	-	-	_	-	0.841	5.600
Carbonic acid	-	-	-	-	-	-	1.383	$9 \cdot 220$
		2. 20	grs.	ash o	f the	ripe c	chaff.	
Q.1							* 000	Per centum.
Silex	-	-	-	-	-	-	7.060	$35 \cdot 300$
Phosphates -	-	-	-	-	-	-	$3 \cdot 102$	$15 \cdot 510$
Carbonate of lin	ne	•	-	-	-	-	$3 \cdot 254$	$16 \cdot 270$
Magnesia -	-	-	-	•	-	-	0.244	1.220
Potash -	-	-	-	-	-	-	$2 \cdot 150$	10.650
Soda	-	-	_	-	-	-	0.431	$2 \cdot 155$
Chlorine -	-	-	-	-	-	•	1.234	$6 \cdot 170$
Sulphuric acid	•	-	-	-	•	-	1.920	9.600

3. Analysis of the phosphates of the chaff of the ripe oat.

Pноsрна	TES	-	-	-	-	-	-	$3 \cdot 10$
Phosphate of	perox	xide o	f iron	-	-	-	•	0.50
Lime -	-	-	-	-	-	-	-	0.83
Magnesia	-	-	-	-	-	-	-	0.99
Silicic acid	-	-	-	•	-	-	-	trace.
Phosphoric ac	cid	-	•	-	-	•	-	0.78

III. ANALYSIS OF THE RIPE GRAIN OF THE TANTAIN OAT.

								20.000 grs.
								1 000
Silica -	•	-	-	-	-	-	-	$4 \cdot 220$
Phosphates ·	•	-	-	-	-	-	-	7.360
Carbonate of li	me	-	-	-	-	-	-	0.050
Magnesia	•	-	-	-	-	-	-	0.016
Potash - ·	•	-	-	-	-	-	-	$2 \cdot 947$
Soda -		-	-	-	-	-	-	0.534
Sulphuric acid		-	-	-	-	-	-	0.207
Chlorine	-	-	-	-	-	•	-	0.297
Phosphate of p	otash	and	soda	-	-	-	-	1.023
Carbonic acid	-	-	-	-	-	-	-	1.757

IV. ANALYSIS OF OATS FROM GENESEE COUNTY.

Raised by Judge Peters. Variety unknown. The soil is that peculiar clay loam which is adapted to wheat, and belongs to the Onondaga-salt group.

					_			-	•	_	-		
							PROP	ortic	NS.				
		Straw	-	-		6	-		•		•	$703 \cdot 20$	grs.
		Grain	-	-		-			6		-	$765 \cdot 50$	
		Chaff	-		•	•			-	-	•	143.70	
100 grs.	dried i	n the	sun	, ga	ve								
		Straw				**				-		4.875	grs. of ash.
		Chaff	-		-	-	•	-	-	-	-	6.875	
		Grain	-		•	-	-	-	•	-	•	2.750	
			An	unfir	iishe	d an	alysi	s of	the gr	ain a	nd chaff	·	
	~										Grain.		Chaff.
	Silica		-	-	-	-		•	-	-	39.80		45.00
	Phosph	nates	-	-	-	-	•	-	**	-	27.50		11-00
	Carbon	ate of	lim	e	-	•		•		-	0.10		trace.
	Magne	sia	-					-		_	0.09		0.01
	Potash		-	-	-			-	-	-	12.65		$8 \cdot 22$
	Soda	-	-	-				-	-	•	5-32		$4 \cdot 52$

EXPERIMENT WITH THE OAT CROP.

In accordance with the wish of the State Agricultural Society, I have made several analyses of soils and oats growing upon them, for the purpose of throwing additional light on the value of manures, and the exhausting power of different crops. The Society had offered a premium for experiments on three acres of ground, with three successive crops. These experiments were to be conducted in the following manner: 1. The acres to be contiguous to each other: 2. One acre was to be manured with not less than ten cords of common barnyard manure the first year, and plowed under; the second acre to be manured with fermented or composted manure, and the other acre to receive no manure: 3. The three acres to be planted with corn the first year; the second, to be sowed with barley or oats; and the third year, to be cultivated with winter grain. Only one gentleman undertook the execution of the experiment: Mr. I. F. Osborn, Port Byron, Cayuga county.

It is proper to remark, that I have been supplied with the soil since the first crop was taken off. The soil which was supplied for the purpose of examination, to show what it was before the experiments began, was taken from the outskirts of the field. It was the

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intention, in the selection of the soil, to show, first, what it was; and then to show what it became after the removal of the crops. Accordingly, two specimens of soil from each acre were sent me for analysis: the first, to show what it was; and the second, to show what it is; but as the selection was made as I have stated, it is evident that the analyses will be less satisfactory than if they had been made in a different manner.

The only crop which has been forwarded for analysis, is the oat crop, a specimen from each of the three acres; but the oat was incomplete, inasmuch as it was cut off about eighteen inches from the ground, leaving about one half or two-thirds of the straw. It is important to state these imperfections in the experiments. Besides it is stated that the acre which is unmanured is naturally the richest; a fact which was suspected in the analysis of the oat.

I will now proceed to the details of the analyses; premising, however, that my object in engaging in them was to throw more light on the composition of the soil of the Wheat district. The soil, as decribed by Mr. Osborn, is a gravelly loam, with a clay subsoil from eight to sixteen inches deep. For further details of the experiments, the reader is referred to the State Agricultural Society's Transactions for 1847.

ANALYSIS OF THE SOILS REFERRED TO IN THE PRECEDING REMARKS.

I. The east acre was manured with ten cords of barnyard manure, which was hauled on while wet, and spread as fast as it could be ploughed in. One hundred grains of the soil were saturated with moisture; 26.886 grs. of water were absorbed, leaving 73.114 of dry soil.

Another 100 grs. gave

							Sol	il as it was.
Organic mat	ter	-	-	•	•	-		$5 \cdot 390$
Water -	-	-	-	-	-	-	-	1.905
Silica -	-	-	-	-	-	-		$84 \cdot 375$
Alumina, pe	roxid	e of	iron a	nd p	hosph	ates	-	5.870
Soluble silica	a	-	-	-	-	-	-	0.145
*Carbonate of	' lime	-	-	-	-	-	-	0.243
$_{ m Magnesia}$	-	•	•	-	•	a		0.025
Potash -	-	-	-	-	-	-	-	0.164
Soda -	-	-	-	-	-	-	-	0.485
Chloride of	sodiui	n	-	-	-	-	-	0.012
Sulphuric ac	id	-	-	-	-	-	-	0.004
Carbonic aci	d not	app	reciab	le.				

98.618 S.

According to this analysis, the soil, to the depth of one foot, contained

Organic matter	-	-	-	-	-	366856 · 875 lbs.
Lime	-	-	-	-	-	$16539 \cdot 187$
Magnesia -	-	-	-	-	-	$1701 \cdot 562$
Potash	-	-	-	•	-	$11182 \cdot 250$
Soda	-	-	-	-	-	$33010 \cdot 312$
Chloride of sodium	ı	-	-	-	-	816.750
Sulphuric acid	-	-	-	-	-	$272 \cdot 250$
Soluble silica	-	-	-	-	-	$9869 \cdot 062$

The phosphates were appreciable, but too small in amount to be weighed.

It may be repeated in this place, that the soils of the true Wheat district are not so rich in phosphates as the soils of the Hudson river and Eastern districts.

The following exhibits the soil of the east acre as it is, or after a crop of corn and oats (I believe) have been removed, but which had been highly manured as stated in the foregoing remarks.

One hundred grains, saturated with moisture, gave

	Water - Dry soil	-	-	-	-	-	-	-	Soil as it is. 28:686 71:314
100 grs. of dr	y soil gave								
	Organic m	atter	-	-	-	-	-	-	6.935
	Water -	-	-	-	-	-	-	-	$2 \cdot 765$
	Silica -	-	-	-	-	-	-	-	$82 \cdot 725$
	Alumina a	nd pe	roxide	e of in	on	•	-	-	5.360
	Soluble sil		-	-	-	-	-	-	$0 \cdot 025$
	Lime -	-	-	-	-	-	-	-	0.217
	Magnesia	-	-	-	-	•	-	-	0.095
	Potash -	-	-	•	-	-	-	-	0.035
	Soda -	-	-	-	-	-	-	-	0.444
	Chloride o	f sodi	um	-	-	-	-	-	0.025
	Sulphuric	acid	-	-	-	•	-	-	0.013
	Phosphates	appr	eciabl	e.					
	Carbonic a	cid no	ot app	reciat	ole.				
									98·630 S
									9×10311 ×

98.630 S.

According to this analysis, the acre, to the depth of one foot, contains

Lime	-	-	-	-	-	14769 · 562 lbs.
Organic matter	-	-	-	•	-	$472013 \cdot 437$
Magnesia -	-	-	-	-	-	$6261 \cdot 750$
Potash	-	-	-	-	-	$2178 \cdot 000$
Soda	-	-	-	-	-	$30219 \cdot 750$
Chloride of sodiu	m	-	-	-	-	$1497 \cdot 375$
Sulphuric acid	-	-	-	-	-	$884 \cdot 812$
Soluble silica	-	-	-	-	-	$1701 \cdot 562$

For obvious reasons, the calculations are confined to those elements which exist in the smaller quantities. We have only to compare the analyses and the calculations, to form an opinion of the effects of cultivation. It is supposed that the greater absorbent power of the soil as it is, is due to the organic matter in part added in the manure; inasmuch as we have fully shown, in another place, that the absorbent power of a soil is mostly due to the presence of organic matter.

II. The soil of the middle acre was left unmanured; but it is remarked that it is naturally the best soil of the three acres, as it receives the wash of adjacent lands.

1. Soil of the middle acre as it was.

One hundred grains, saturated with moisture, gave

Water -

	Dry soil	-	-	-	-	-	-	-	$74 \cdot 177$
100 grs. of dry	y soil gave								
	Organic mat	tter	-	-	-	-	-	-	4.085
	Water -	-	-	-	-	-	-	-	$2 \cdot 220$
	Silica -	-	-	-	-	-	-	-	$85 \cdot 185$
	Alumina an	d per	oxide	of ir	on	-	-	-	5.610
	Soluble silic	a	-	-	-	-	-	-	0.075
	Lime -	-	•	-	-	-	-	-	0.198
	$\mathbf{Magnesia}$	-	-	-	-	-	-	-	0.044
	Potash -	-	•	-		-	-	-	0.158
	Soda -	-	-	-	-	-	-	-	1.093
	Chloride of	sodi	ım	-	-	~	-	-	0.019
	Sulphuric ac	cid	-	-	-	•	•		trace.
	Phosphates	appre	ciabl	e.					

98.687 S.

25.823

According to the analysis,	the middle acre contains.	in the depth of one foot,

Organic matter	-	-	-	•	-	278035·312 lbs.
Lime	•	-	-	-	•	$13476 \cdot 375$
Magnesia -	-	-	•	-	-	$2994 \cdot 750$
Potash	-	-	-	-	-	$10753 \cdot 875$
Soda	-	-	-	-	-	$74392 \cdot 312$
Chloride of sodin	ım	-		-	-	$1293 \cdot 875$

2. Soil of the middle acre as it is.

100 grs. saturated with moisture, gave

Water -	-	-	-	-	-	-	-	23.641
Dry soil	-	-	-	-	-	-	-	$76 \cdot 359$

In one hundred grains of the dry soil, there was found

Organic matter	•	-	-	-	-		3.660
Water	-	-	-	-	-	-	1.875
Silica	-	-	-	-	-	-	87.975
Alumina and pe	roxid	e of i	ron	-	-	-	3.880
Soluble silica	-	-	-	•	-	-	0.302
Lime	•	-	-	•	-	-	0.155
Magnesia -		-	-	-	-	-	0.044
Potash -	-	-	-	-	-	-	0.044
Soda	-	-	-	-	-	-	0.772
Chloride of sodi	ium	-	-	-	-	-	0.020
Sulphuric acid	-	-	-	-	-	-	trace.
Phosphates appr	eciab	le.					

98·730 S.

There is, therefore, contained in the acre, to the depth of one foot,

Organic matter	-	-	-	-	-	249004 · 750 lbs.
Lime	-	-	-	-	•	$10549 \cdot 687$
Magnesia -	-	-	-	-	-	$2994 \cdot 750$
Potash	-	-	-	-	-	$2994 \cdot 750$
Soda	•	-	-	-	-	$52544 \cdot 250$
Soluble silica	-	•	-	-	-	$224946 \cdot 562$
Chloride of sodiu	m	-	-	-	-	$1361 \cdot 250$

III. The soil of the west acre was prepared with rotted or partially rotted manure.

1. Soil of the west acre, as it was.

One hundred grains, saturated with moisture, gave

Water -	-	-	-	-	-	-	-	$25 \cdot 161$
Dry soil	-	-	-	-	-	-	-	74.839

100 grs. of the dry soil gave

Organic matter	-	-	-	-	•	-	$4 \cdot 115$
Water		-	-	-	-	-	1.470
Silica	-	-	-	-	-	-	86.095
Alumina and per	roxid	e of i	ron	-	-	-	5.615
Soluble silica	-	-	-	-	-	-	trace.
Lime	-	-	-	-	-	-	0.172
Magnesia -	-	-	-	-	-	-	0.242
Potash	-	-	-	-	-	-	0.035
Soda	-	-	-	-	-	-	0.208
Chloride of sodia	ım	-	-	-	-	-	0.010
Sulphuric acid	-	-	•		-	•	0.005
Phosphates appro	eciab	le.					

97·960 S.

The soil of this acre contained, in the depth of one foot,

Organic matter	-	-	-	-	-	300077·187 lbs.
Lime	-	-	-	-		$11706 \cdot 750$
Magnesia -	-	-	-	-	-	$16471 \cdot 125$
Potash -	-	-	-	-	-	$2178 \cdot 000$
Soda	-	-	-	-	-	$14157 \cdot 000$
Chloride of sodiu	m	-	-	-	-	$680 \cdot 625$
Sulphuric acid		•		-	-	$136\cdot 125$

2. Soil of the west acre, as it is.

One hundred grains, saturated with moisture, gave	One	hundred	grains,	saturated	with	moisture,	gave
---	-----	---------	---------	-----------	------	-----------	------

Water -	۵	-	-	-	-	-		27.851
Dry soil	-	-	•	•	•	•	-	$72 \cdot 149$
the dry soil	ma mo							

100 grains of the dry soil gave

Water	•	۵	-	-	$2 \cdot 450$
Organic matter	-	-			6.080
Silica	-	•	•		$83 \cdot 435$
Alumina and peroxide of iron	n	-	-	-	$6 \cdot 125$
Soluble silica	-	-	•	-	trace.
Lime	-	-	•		0.185
Magnesia	•	-	-	-	0.255
Potash	-		-		0.057
Soda	••	-	-	sa	0.258
Chloride of sodium -	-	-	-	-	0.005
Sulphuric acid	-	ø		-	0.009
Phosphates appreciable.					

98.856 S.

The acre, in the depth of one foot, contains

Organic ma	atter	-	-	-	-		413820 000 lbs.
Lime -		-	-	-	-	•	$12585 \cdot 625$
Magnesia		•	-	-	-	•	$17355 \cdot 937$
Potash .			•	۰	-	•	$3879 \cdot 562$
Soda -	-	-	-	-	-	-	$17560 \cdot 125$
Chloride of	f sodiı	ım	•	-	-	-	$136\cdot 125$
Sulphuric:	acid		-	-		•	$612 \cdot 562$

ANALYSIS OF THE CROPS RAISED ON THE THREE EXPERIMENTAL LOTS.

Having given a full statement of the composition of the soil which was furnished from the three experimental acres, I now proceed to give the composition of the crops which were produced upon them respectively. The crop obtained was the common variety of oats, as usually seen in market, namely, a mixture of white and black oats. Nothing, however, has been stated respecting the variety. The samples of the oats were furnished in small bags properly labelled, and in small bundles with about eighteen inches of the upper part of the straw. The proportions of straw, grain and chaff, therefore, could not be determined. The proportions of water and inorganic matter were determined as the following statement will show:

PROPORTIONS.

				1. E.	ST A	CRE.		
				1. 122		ORB.	Quantity.	Per centum.
GRAIN	6	•	•	-	۵	•	$510 \cdot 300$	
$\mathbf{W}_{\mathbf{ater}}$	•	•	-	-	-	•		$9 \cdot 412$
$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	•	•	12.780	$2 \cdot 504$
CHAFF	-	-	•	•	•	•	$67 \cdot 580$	
$\mathbf{A}\mathbf{s}\mathbf{h}$	60	٥		•	•		3.675	$5 \cdot 438$
STRAW	-	-		-	•	-	$225 \cdot 100$	
\mathbf{Ash}	-		-	-	•	-	$13 \cdot 459$	$5 \cdot 975$
			2	. Mid	DLE	ACRE		
~							Quantity.	Per centum.
GRAIN -	•	•	•	-	۰		631.000	
Water	•	•	-	-	-	-		$10 \cdot 400$
Ash -	•	-	•	•	•	-	$15 \cdot 120$	2.396
CHAFF -	-	•	u	-	-	•	80.880	
Ash -	•	•	•	•	•		5.675	$7 \cdot 016$
STRAW	-	•		•		•	$244 \cdot 000$	
Ash -		-	-	-	-	-	$12 \cdot 500$	$5 \cdot 122$
				3. Wı	EST A	CRE.		
GRAIN -	-	-	_	-	_	-	Quantity. 712:300	Per centum.
Water					-			$10 \cdot 120$
Ash	-	-		-		۰	16.690	2.343
CHAFF -		-	-	-	-		95.800	
Ash •				0			5.680	5.925
STRAW					_		359.800	5 0.43
Ash -			6		6		$24 \cdot 940$	$6 \cdot 931$
							~1 010	2 001

I. Composition of the oats of each acre respectively.

1. EAST ACRE.

1. A	lsh of	the	grain.	
------	--------	-----	--------	--

				J	Ų	20.000 grs.	Per centum: calculated without
Silica -	•		-	•	•	$5 \cdot 140$	coal or carbonic acid. $30\dot{\cdot}382$
Coal	-	•	•	-	•	0.180	
Phosphates	-	•	•	-	•	4.740	$28 \cdot 017$
Carbonate of l	ime	•	•	•	•	0.020	0.115
Magnesia -			-	-	•	0.010	$0\!\cdot\!056$
Potash -	-	•	•	-	•	3.800	$22 \cdot 461$
Soda -	-		•	-	•	1.012	$5 \cdot 923$
Sulphuric acid		•	-	-	-	0.501	2.661
Chlorine -		•	•	-	•	0.305	1.802
Phosphate of	ootash	and	soda	-	-	1.500	8.330
Carbonic acid				ter	-	2.613	
						19.821	99 · 444

2. Ash of the straw.

						25.0	00 gr	s.		
Silica -						1.2	75		Calculated without organic matter. 5.587	
Phosphates	-	•	•	•	•					
Carbonate of 1	• •	•	-		-	$\frac{5\cdot 1}{0\cdot 9}$			$22\cdot 450 \\ 4\cdot 247$	
	ime	•	-	•	•					
Magnesia Potash -	-	•	•	•	-	0.1			0.788	
	-	•	•	-	-	12.5			55.048	
Soda -	-	•	•	•	-	0.9			$4 \cdot 132$	
Chlorine	•	•	-	•	-	0.8			3.983	
Sulphuric acid		•	-	•	-	0.8			$3 \cdot 788$	
Organic matter	r	-	•	•	-	1.3	80			
						$\frac{-}{24 \cdot 2}$	16		$\frac{-}{100 \cdot 023}$	
						24 2	10		100 020	
			3.	Ash of	the	chaff.				
Silica			-	_	-	-	_		11.360	
Phospha	ates	-	-	-	-	-	-	-	$3 \cdot 250$	
Carbona		`lime		-	-	-	-	-	0.385	
Magnes			_		_		-		0.196	
Potash						-	_	-	$3 \cdot 292$	
Soda	_	_		_	_				0.269	
Sulphu			_		_	_			0.423	
Chlorin		-	_		_	_	_	_	0.268	
Organi		tor	_	_	_	_	_	_	0.443	
Olgani	Cina	.tc1	_	-	-	-	-	-	0 449	
									19.886	
			•							
			POSIT	ON OF	THE I	PHOSPH	ATES.			
\mathbf{P}_{H}	IOSPH.	ATES	•	-		-	-	•	3.250	
Silica	-	-	-	-	-	-	-	-	0.060	
Phospha	ate of	pero	xide	of iro	n -	-	-	-	0.250	
Magnes		٠.	-	_	-	-	-	-	0.207	
Lime	-	-	-	-	-	-	-	-	0.924	
Phosph	oric a	icid	-	-	-	-	•	-	1.809	
•										

2. MIDDLE ACRE.

1. Ash of the grain.

						Per cent		ulated on the clements as obtained at carbonic acid or coal.
Silica -	-	-	-	•	-	-	-	37.527
Phosphates	-	-	-	-	-	-	-	33.581
Carbonate of	lime	-	-	-	-	-	-	0.171
Magnesia	-	-	-	-		-	-	0.070
Potash -	-	-	-	-	-	•	-	$15 \cdot 172$
Soda -	-	-	-	-	-	-	-	5.438
Sulphuric aci	id		-	-	-	-	-	1.288
Chlorine	-	-	-	-	-	•	-	1.623
Phosphates o	f pota	ash ai	nd sod	la	-	-	•	4.210
-	•							
								99·0S0

2. Ash of the straw.

							25.000 grs.	Calculated without carbonic acid or organic matter.
Silica -		-	-	-	-	-	8.890	37.160
Phosphates		-	-	-	-	-	1.365	$5 \cdot 703$
Carbonate of	f lin	nе	-	-	•	-	3.080	12.874
Magnesia -	•	-	-	-	-	-	0.328	1.368
Potash -		-	•	-	-	-	$9 \cdot 134$	38 · 178
Soda		-	-	-	-	-	0.407	1.698
Chlorine -		-	-	-	•	-	$0 \cdot 129$	$0\cdot 535$
Sulphuric ac	eid	-	-	-	-	-	0.577	$2 \cdot 409$
Organic mat	ter	-	-	-	-	-	0.800	
Carbonic aci	d	-	•	-	•	-	1.601	
							26.311	$\phantom{00000000000000000000000000000000000$

3. Ash of the chaff.

						25·000 grs.	Per centum: calculated without carbonic acid or organic matter.
Silica	-	-	-	-	-	$14 \cdot 240$. 58.976
Phosphates	-		-	-	-	1.620	$6 \cdot 698$
Carbonate of		e	-	-		0.850	$3 \cdot 520$
Magnesia	-	-	-	-		0.248	1.026
Potash	-	-	-	-	-	$6 \cdot 765$	$27 \cdot 015$
Soda -	-	_	-	-	-	0.199	0.822
Sulphuric a	cid	-	-	-	-	0.201	0.829
Chlorine		-	-	-		$0 \cdot 131$	0.541
Organic ma	tter a	nd	carbon:	ic ac	id	$1 \cdot 639$	
							
						25.893	$99 \cdot 467$

COMPOSITION OF THE PHOSPHATES.

Soluble silica	-	-	-	-	-	*	•	0.160
Phosphate of	perox	ide o	firon	-	•	•	-	0.180
Magnesia	-	-	-	-	-	-	-	0.071
Lime -	-	-	-	-	-	-	-	0.259
Phosphoric ac	id		_	_		_	_	0.950

3. West Acre.

1. Ash of the grain.

								m: reckoned without carbonic cid or foreign matter.
Silica -	-	•	-	•	-	-	•	$33 \cdot 129$
Phosphates	of	lime, i	ron aı	nd ma	gnesia	-	-	29.463
Lime -	-	-	•	-	•		•	0.070
Magnesia	-	-	-		-	-	-	0.084
Potash -				-	-	•	•	18.904
Soda -	-			-	-	-	-	5.751
Sulphuric a	cid	-	-	-	-	-		$2 \cdot 135$
Chlorine	-	-	-	-	-	-	-	0.028
Phosphates	of	potash	and s	oda	-	10	-	10.291
_		_						
								99.855

2. Ash of the straw.

Silica			_	_		_	$\frac{20\cdot000}{2\cdot800}$ grs.	Calculated without carbonic acid or organic matter. 16.656
Phosphates:								
of iron	-	-	-	-		0.120		
of lime	-	-	-	-		0.188		
of magn	esia	-	-	-		0.025		
Phosphoric	acid			-		0.667		
_					-		1.000	$5 \cdot 952$
Carbonate of lin	ne	-	-	-	-	-	1.470	8.745
Magnesia -		-	-	-	-	-	0.475	2.827
Potash	•	-	-	-	-	-	8.818	$52 \cdot 483$
Soda	•	-	-	-	-	-	1.088	$6 \cdot 475$
Sulphuric acid		-		-	-	-	0.721	$4 \cdot 291$
Chlorine		-	-	-	-	-	0.321	$1 \cdot 905$
Organic matter			-	-		_	0.960	
Carbonic acid -		-	-	-	-	-	$3 \cdot 106$	
							20.759	$99 \cdot 334$

3. Ash of the chaff.

					J	авн ој	, ine	спау.				
								20	000 g	grs. (Calculated	without coal, carbonic
	Silica		-	-	-	-	-	7.	500			organic matter. 14.876
	Coal -	9	**	a		-	-	1.	120			
	* Phospha	ites	-	-	-	•	-	2.	900]	17.348
	Carbona	ate of	lime	-	-	-	-	1.	090			6.520
	Magnes	ia -	•	•		-	-	0.0	098			0.582
	Potash	-	-	-	-	-	-	4.9	200		2	25 · 130
	Soda -	-	-	-	-	-	•	0.8	S90			5.324
	Carboni	c acid	and o	organi	c mai	tter	-	1.	742			
								10./				
								19.6	040		9	9.800
				COMI	ositio	ON OF	THE	рнозрнат	res.			
	S	Silica	•		a	-	-	**	-	a	0.03	
	I	Phospl	nate o	f iron	-	-	-	-	•	-	0.04	
	L	ime	-		-	-	a	-	-	-	0.05	
	I	Iagne	sia	-	-	-	-	-	-	-	0.31	
	F	Phosph	oric a	cid	-	-	•	-	-		1.91	
	COMPOS	TTION	י אס ז	erre r	HOSI	DII A TU	es o	F THE	CED A 7	N OF	EACH	ACDE
	COMITOD		or .	. 1112 1	11051	. araca:	23 0	West.			DLE.	EAST
	Soluble sil	ica		-		-	-	0.024			030	0.030
	Phosphate		roxide	of ir	on	a		0.37			190	0.190
	Magnesia	-			•			1.234			314	1.234
	Lime -	-	-			-	-	0.727		_	365	0.694
	Phosphoric	acid	-		-	-	-	2.839		-	201	2.592
												55.5
II.	Organic c	ONSTI	TUEN	rs of	THE	OAT	or	GRAIN C	OF E	ACH A	CRE RE	ESPECTIVELY.
					1.	East	r Ac	RE.				
236 • 4	0 grs. oats	yield	ed of							Doroant	arro.	
\mathbf{P}	aleæ or hus	sks -		-	-			69.30		Percenta 29:3		
K	ernels -	4		-	-	-	1	67.10		70.68	36	
								_				

236.40 grs. oats yielded of					
Paleæ or husks	-	•	$69 \cdot 30$	Percentage. $29 \cdot 314$	
Kernels	-	-	$167 \cdot 10$	70.686	
194·16 grs. kernels gave of					
Fine white flour	-		$115 \cdot 12$	$59 \cdot 291$	
Bran (about like that separated at	the	mills	9.02	40.709	
Flour.					
115·12 grs. flour gave of					
Starch	•	-	$89 \cdot 91$	7 7·232	
Epidermis, pilose*, gluten and oil	l	-	$4 \cdot 32$	3.752	
Percentage of ash in the epiderm obtained from this specimen		•	ose (mostly]	pilose) 2·315.	Nothing more was

^{*} The pilose is the short hairs appended to the epidermis.

	Perce	ntage	of w	ater,	dry m	atter	and a	ish ir	n the kernel	s.	
100 grs. of kerne	els vie	elded	of								
Water -	_				a		6	٠	9.412		
Dry kernels			-				-		90.588		
Ash -	-					_		-	$5 \cdot 242$		
Ash calculate	d drv	-	-			-		-	$2 \cdot 475$	•	
Organic matt	_		d dry	-	•		۰	•	97.525		
-			•								
		H	Ternel.	8.							
100 grs. of kernels	gave	of									
Starch -		-	•		-	•			$55 \cdot 000$	Percentage: calculated dry. 59 · 291	
Albumen	-		-		٠	-	•	-	1.800	1.932	
Casein or ave	nine	•			٠	_			15.769	16.984	
Dextrine, or			ucilae	e	*	_	•		7.280	7.861	
Oil •	•	-	:	•	-	-	-	-	3.380	3.644	
Gluten -	-	-			•		-	-	1.680	1.803	
Sugar and ex	tractiv	e mai	tter	-	-	-		-	$2 \cdot 120$	$2 \cdot 278$	
Epidermis an			•			-	_	_	5.750	6.207	
Water -	-	-	•	-	6	•		۵	9.412		
· ·											
									$102\cdot 582$	$100 \cdot 000$	
				2. 1	Middl	E AC	RE.				
289 '06 grs. gave	of										
Paleæ or husl			_	_	_	S	9.06		Percentage. 30 · 810		
Kernels -			-	-	-		0.00		69.190		
						~0	0 00		00 100		
196·40 grs. kern	els vi	elded	of								
	_	craca	01				000		WO 04 W		
Fine white flo		•	•	1	•11		.000		58.045		
Bran (about a	s it is	obtai	ned a	t the	mills)	82	•400		41.955		
100 mms of the fl	01: H . 67:	O.V.O. O.	$_{c}$ Flo	ur.							
100 grs. of the fl	our ga	ave o.	L								•
Starch -	•	-	•	•	•	•	-	•	82.70		
Epidermis, gl	uten,	oil and	d pilo	se, m	ostly p	oilose	-		2.28		
										centage of silicic acide trace of So ₃ .	1,
Casein -		18			-	-		~	2.16	ŭ	
Dextrine or gr	ım			_				_	1.68		
Water -		-	-		•	_			$10\cdot40$		
	vtro c t	ivo m	ottor	oil.	alutan	and	n lhu -	~ ~ ~		montitativaly aht-is	1
(very small					gruten	and	arour	nen	were not q	uantitatively obtaine	a

				70								
82.	40 grs. of	bran æ	vo of	Bran	ı.							
	Starch	oran ge	ive or					35.155		42.664		
			-	•	-	•						
	Casein an		ige -	-	•	•		9.175		11.135		
	Dextrine of		-		.1	-		6.630		8.046		
	Epidermis	s, phose,	gruten	ana	011	-		13.870		16.832		
								64.830		78.577		
	tains co of phos lime, n efferves The albu	mostly e nsiderable phates, a nagnesia cence on men, su	piderm le silex and a s and su adding agar, e	is. 7, a lar small alphur g acid extrac	This ge propric active	ash coronarion cid. sh. matt	on- ion of No ter,					
		and wa				obtain -	ed.	17.570		21.323		
								S2·400		100.000		
				Kern	els.							
100	grs. of ke	ernels ga	ave of									
	Starch .		-	-		-		-		57.220	Percentage: calcula 65.412	ited dry.
	Casein or	avenine	-	-	-		-	-	_	12 · 120	13.897	
	Albumen	_	-	-	_	-	_	-	-	0.820	0.941	
	Gluten -		-	-	_	-	_	•	-	1.400	1.605	
	Dextrine,	or gum a	and mu	cilage	2	-	_	-		6.340	$7 \cdot 269$	
	Sugar and				_	-	_	_	_	1.260	1.445	
	Oil -		-	-	_	-	_			4.740	5.435	
	Epidermis	and nile	ose	_	_	_	_	-	_	3.310	3.796	
	Water -	_	550	_	_			_	_	10.145	3 (00	
	water -	•	-	-	•	-	•	-	-	10 149		
										97.355	100.000	
					3.	Wes	T A	CRE.				
				Kerne	ls.							
100	grs. kern	els gave										
	Starch	_			_	_			_	55.34	Percentage: calcul 60:336	ated dry.
	Casein or	avanina	_	_		_				14.36	15.656	
	Albumen			_	_	_			_	0.94	1.025	
	Gluten		_	_		_			_	1.62	1.766	
	Dextrine,	or gum :	and mu	cilace	ـ د					9.26	10.096	
	Sugar and					_			_	2.28	2.486	
	Oil -	· CAHUON	C IIIdl	_		-		-	-	7.50	8.177	
	Epidermis	and pile	nse.	_	_					0.42	0.458	
	Water	a a	-	_						10.12	0 190	
											-	
										101.84	$100 \cdot 000$	

gave	of						_	
-				-	-	-	$79 \cdot 360$	Percentage: calculated dr 92.637
-	-	-	-	-	-	-	$2 \cdot 120$	$2 \cdot 475$
-	-	-	-		-	-	1.660	1.939
glute	en an	d oil	-	-	-	-	2.500	$2 \cdot 919$
-	-	-	-	-	-	-	$9 \cdot 714$	
							${95 \cdot 354}$	
	-				gluten and oil	gluten and oil	gluten and oil	

The sugar, extractive matter and albumen were not obtained.

				_			
			1	leæ and kernels.			
Paleæ or h	nusks	-	-	-	-	-	$29 \cdot 052$
Kernels	-	-	-	-	-	-	70.948
			Prop	portio	n of j	lour a	and bran in kernels.
Flour	-	-	-	-	-	-	58.958
Bran -	-	-	-	-	-	-	$41 \cdot 042$

ANALYSIS OF OATS FROM LEWIS COUNTY.

This variety, which I have noticed under this name, grew upon the black slates of the Hudson-river group, which is usually a mixture of flat and round gravel and loam. The following is its composition:

	WHITE O	AT.	
Water of the grain -		10.10	
		15.000 grs.	Percentum: calculated without organic matter.
Silica		5.860	$43 \cdot 030$
Phosphate of iron -		0.160	$1 \cdot 174$
Phosphate of lime -		0.250	1.836
Phosphate of magnesia		1.431	10.503
Phosphoric acid		2.030	14.904
Carbonate of lime -		0.020	0.141
Magnesia *		0.010	0.067
Potash		1.998	14.667
Soda		1.346	9.892
Chlorine		0.261	1.800
Sulphuric acid		0.253	1.836
Organic matter		1.203	
Carbonic acid		trace.	
			00.050
		14.822	$99 \cdot 850$

The following is the composition of a black oat from the same locality, cultivated in the same manner, and upon the same soil:

]	Вьаск	OAT.		
Water in the grain	-	-	•	•	9.88	
					15.000 grs.	Per centum: calculated without organic matter.
Silica	-	-	-	-	6.030	43·177
Phosphate of iron	-	-	-	-	0.180	1.287
Phosphate of lime	-	-	-	-	0.330	$2 \cdot 357$
Phosphate of magne	esia	-	-		0.673	4.817
Phosphoric acid -	-	-	-	-	1.997	$14 \cdot 295$
Carbonate of lime	-	-	-	•	0.040	0.284
Magnesia	-	-	-	-	0.010	0.067
Potash	-	-	-	-	1.491	10.670
Soda	-	-	-	-	1.342	$9 \cdot 606$
Phosphate of potash	and	soda	-	-	1.680	$12 \cdot 025$
Chlorine	-	-	-	-	0.125	0.894
Sulphuric acid -	-	-	-	-	0.065	$0\cdot 440$
Carbonic acid -	-	-	-	-	trace.	
Organic matter -	-	-	-	-	0.800	
					$\overline{14\cdot 760}$	$\overline{99\cdot 919}$

ANALYSIS OF A CAREX FROM BEMENT'S FARM.

[Received too late to be inserted in its proper place. See p. 81.]

Cut June 10th, just before the flower spikes made their appearance: from 24-36 inches high.

100 grs. of fresh grass gave

Water -		-	-	-	-	-	71.783
Dry matter	-	-	-	-	-	-	28.217
Ash -		-	-	-	-	-	1.538
Ash calcula	ted dry	-	-	-	-	-	$5 \cdot 451$
Organic ma	tter calcul	ated o	dry	-	-	-	94·549 S.

ANALYSIS OF OATS FROM PITTSTOWN.

Furnished by Mr. Newcomb. It was grown upon the soil resting upon the Taconic slates. The oat was fully ripe, with bright grain and straw.

1. Relative proportion of grain, straw and chaff.

								Actual quantities.
Grain	-	-	-	•	-	•	-	636.000
Straw	-	-	-	-	_	-	-	$566 \cdot 000$
Chaff' -	-	_			_	-		60.000

2. Analysis of the grain.

						20.000 grs.	Per centum: calculated without
Silica -	-	-	-	-	-	8.100	organic matter. $41\cdot449$
Phosphates of	lime	and m	agne	sia	-	$6 \cdot 110$	$31 \cdot 264$
Carbonate of li	me	-	-	-	-	trace.	
Magnesia -	-	•	-	-	-	none.	
Potash -	-	-	-	-	-	$2 \cdot 051$	$10 \cdot 495$
Soda	-	•	-	-	-	1.966	$10 \cdot 059$
Chlorine -	-	-	-	-		0.017	0.100
Sulphuric acid	-	-	-	-	-	0.013	0.065
Phosphate of p	otasl	ands	soda	-	-	1.280	$6\cdot 545$
Carbonic acid	-	-	-	-	-	none.	
Organic matter	-	-	-	-	-	0.412	
-							
						$19 \cdot 949$	$100 \cdot 077$

3. Analysis of the straw, divided into two equal parts, top and bottom.

								Top.	Воттом.
Silica -	-	-	-	-	•	-	-	7.070	4.200
Phosphates -	-	-	•	-	-	-	-	2.615	$2 \cdot 240$
Carbonate of	lime		•	-	-		•	0.990	0.280
Magnesia	-	-	-	-	-	-	-	0.043	$0 \cdot 108$
Potash -	-	-	-	-			-	$6 \cdot 559$	7.833
Soda -	-	-	-	-	-	-	-	1.233	1.463
Sulphuric acid	d	-	•	-	-	-	-	0.310	0.338
Chlorine -		-	-	-	•		-	0.093	0.061
Carbonic acid		-	-	-	•	-	-	1.012	$3 \cdot 145$
								$19 \cdot 925$	19.668

4. Analysis of the chaff.

Silica -	•	-	-	-	-	-	•	$10 \cdot 375$
Coal -	-	-	-	-	-	-	-	0.480
Phosphate	es -	-	-	-	-	-	-	0.935
Carbonate	of lin	ne	-	-	-	-	-	0.800
Magnesia	-	-	-	-	-	-	-	0.570
Potash -	-	•	-	•	-	-	-	0.430
Soda -	-	-	-	-	-	-	-	0.309
Sulphuric	acid	-	-	•	-	-	-	0.447
Chlorine	-	-	-	-	-		-	$0 \cdot 123$
								$14 \cdot 569$

ANALYSIS OF THE STRAW OF OATS GROWN BY MR. PETERS*.

Silica	-	•	-	•	-	-	$12 \cdot 350$
Phosphate of per	oxid	e of i	ron	-	-	-	0.925
Lime	-	•	-	-	-	-	$6 \cdot 247$
Magnesia -	-	•	-	-	-	-	$.1 \cdot 945$
Silicic acid -	-	-	-	-	-	-	$5 \cdot 575$
Phosphoric acid	-	-	-	•	-	-	4.933
Lime	-	-	-	-	-	-	0.761
Magnesia -	-	-	-	-	-	-	1.750
Potash	-	-	-	-	-	-	38.871
Soda	-	-	-	-	-	-	3.792
Chloride of sodiu	ım	-	-	-	-	-	0.214
Sulphuric acid	-	-	J -	-	-	-	0.048
Carbonic acid	•	•	-	-	•	-	10.160
Organic matter	-	-	-	-	-	-	$3 \cdot 300$
J							
							99·871 S.

^{*} This analysis should have been inserted at page 97. The foregoing organic analyses of the oat of the three experimental acres were made by Mr. Salisbury, assistant in the laboratory.

REMARKS UPON THE OAT CROP.

The oat crop is very properly regarded, in all the temperate and more northerly countries, as one of the most important. In some it is highly esteemed in domestic life as a grain for bread, and everywhere as one of the best kinds for cattle. Hence it is extensively cultivated: it therefore becomes proper to make a few inquiries as to its exhausting powers upon the soil, and as to the grounds upon which its reputation rests as an article of food for man and beast.

The weight of this grain varies from thirty to thirty-four pounds per bushel, and a fair crop will not vary much from fifty bushels to the acre. Premium crops have been given, amounting from eighty-five to ninety bushels per acre. From these data, we may determine the amount of mineral matter removed from the soil in a given crop of oats. The per centum of mineral or inorganic matter will be as follows, taking 5.25 as the average per centum of the ash. Hence there will be removed

							In one bushel of grain.	From an acre.
Silica or silici	c acid	-	•	-		•	8.960 oz.	28·150 lbs.
Phosphates	•	-	-	-		-	8.000	25.000
Carbonate of	lime	•	•	-	-	-	0.019	0.059
Magnesia	-	-	-	-		10	0.052	0.071
Potash -	-		-	-	-	-	$5 \cdot 141$	16.068
Soda -			-	-	-	-	1.564	4.888
Chlorine -		-	-	-		-	0.007	0.023
Sulphuric acid	1 -		-	-	-	-	0.580	1.814
Phosphates of	potas	h and	soda	-		-	$2 \cdot 799$	8.747
							27·092 oz.	84·820 lbs.

For the foregoing calculation, I have taken the analysis of the ash of the oat of the middle acre (p. 107). Each analysis will give a result differing somewhat from this; but this single calculation will be sufficient for our purpose, that is to show how much a given crop will remove of these valuable substances from the soil.

The oat crop, it will be seen, may be regarded as one of the exhausting crops. We are not, however, prepared here to enter upon a comparison of this crop with those of the other cereals, as this will necessarily come up for consideration after I have given the analyses of the other grains. But I may now proceed to the important question, What elements does the oat possess, which render it a valuable article of food?

It seems proper, in the first place, to state that the value of food, or of matters to support and sustain life, do not depend upon any one single element. The idea that nitrogen is the important one, is entirely fallacious. It is true that high authorities are wedded to the notion that nitrogren is the body which sustains animal life. Hence when the quantity of nitrogen has been determined, the substance has a station given it in the list of nutriments, according to the amount of nitrogen it contains. It is, however, no more important than starch, or any of the other respiratory products. I assume this position, because I believe

there is a mutual relation subsisting among the elements. Life does not consist simply in the formation of muscle, nor of bone. The acts of life are seen equally in respiration, as in digestion; in supplying heat, as in supplying nutriment; in supplying a bony frame-work, as in giving it motion. Each element has its use, and each organ its function; and not one of them can fail in the performance of its office, without deranging the whole fabric. Sulphuric acid, or oil of vitriol, is wanted in the animal economy; so are brimstone and phosphorus, two highly combustible bodies; and no doubt there are instances when life is too fully charged with them, still they are as necessary as nitrogen, and life would not manifest itself without their aid.

We may therefore regard a body important to animal life, in proportion as it contains the requisite number of all those elements which a living body requires to give it the greatest degree of strength and energy. We may, with this view, look first to the inorganic bodies, potash, soda, lime and magnesia. In the same list, too, we may put the acids which combine with these bases, as the phosphoric and sulphuric, together with chlorine. These by themselves are useless in the absence of oxygen, carbon, hydrogen and nitrogen. The four lest, but especially carbon and oxygen, enter largely into the composition of living beings. The tissues of vegetables, as well as of animals, seem to be mostly composed of carbon, oxygen and hydrogen, which, always taken in combination with the above bases, constitute a frame-work, delicate indeed, but yet essential to the performance of any function the tissue in its organic capacity is designed for. Nitrogen enters only into a certain class of organs; in others it is entirely absent, as in the bones. In the red and fleshy parts, nitrogen is an essential constituent; but even in these parts, iron, oxygen, hydrogen and the phosphates are equally important. A cereal, then, which contains the elements of bone and muscle in combination, or matter for the formation of the tissues generally, together with the maintenance of respiration, is the fittest nutriment in the economy of life.

Of the inorganic constituents of the oat, we have found the phosphates in the ratio nearly of $\frac{1}{6}$ or $\frac{1}{5}$; potash in about the same quantity, in some instances in a greater ratio. Soda and sulphuric acid are also always present. Lime and magnesia, together with silica, make up an important though short list of matters which are essential to the constitution of bone, muscle and brain.

In the list of bodies which are found by approximate organic analysis, are, starch, which exists in the ratio of $\frac{1}{2}$ of the products, is a respiratory and fat-producing substance; avenine, a nitrogenous element, in the ratio of $\frac{1}{8}$; gluten and albumen, two other nitrogenous bodies, in a much smaller ratio; oil and sugar, two other respiratory and fat-producing bodies, are in the ratio of $\frac{1}{12}$; dextrine and mucilage, which probably contribute to the same end, but which also act favorably as demulcents upon the mucous membrane of the alimentary canal. With such a composition it is satisfactorily made out that the oat is a valuable nutriment, taking the word in its widest sense. It has the materials for bone and muscle, as well as for respiratory matters; and it is not deficient in the elements of fat, that substance which gives fulness and beauty to the form. Experience, however,

had in generations long past established the same facts; and man, by his steady cultivation of the crop, has shown the estimation in which it has ever been held by him. It becomes still more important, from the fact that it may be cultivated in higher latitudes than some of the other cereals, and rarely in a northern climate disappoints the farmer.

The amount of organic elements in a bushel of oats weighing 32 lbs. is as follows:

								lbs.	oz.
Starch	•	•	-	-	-	٠	•	17	4
Casein or a	veni	ne	-	-	-	-	-	4	9
Albumen	-	•	•	•	-	-	-	0	4
Gluten	-	•	-	-		•		0	8
Dextrine	-	-	-		-	-	-	2	15
Sugar and	extra	ct	•	•	-	-	-	0	11
Oil -	-		•		•	•	-	2	6
Water -	-		-		-	•	-	3	4
								32	0

DISEASES OF THE OAT.

The oat, in certain fields, especially those that are weedy, is subject to disease; or rather to the attack of a fungous plant, whereby almost every kernel upon the panicle or head is totally destroyed. This fungus is called by authors *Uredo avenæ*. It is a species of smut. The kernels, or the places where they would have grown, is filled with a black smutty substance, which finally becomes dry enough to send out a powder of extreme fineness, which is really a cloud of seeds capable of germinating and growing, and producing, like other seeds, individuals similar to the parent.

The smutty heads are well known undoubtedly to farmers, inasmuch as many may be found in most fields of oats. In addition to the smutty panicle, the whole plant is stinted, being shorter and more erect than the healthy one. The coverings of the grain, the husks, are apparently corroded through, so as to show upon the side the black mass within them. The spores, or seeds which have been already alluded to, are extremely minute rounded particles, whose forms are only made known by the aid of a powerful microscope. They are globular grains, not uniform in size, but with a diameter which does not exceed $\frac{1}{3320}$ of an inch. These minute atoms are scattered over the field, and lodged in the soil; and, without doubt, if we may rely upon analogy, they retain for a long time the power of germinating, and will do so when a favorable opportunity occurs. This fact renders it highly desirable that every smutty plant should be eradicated and burnt, when it first shows any symptoms of the disease in question; for undoubtedly the spores are taken up by the roots, and conveyed to heads of the grain, where they at once begin the work of destruction.

The fungous plant, which appears so much like a gangrene of the grain, is represented on Plate LV, fig. 23, in the first stage of its growth. It, however, more frequently appears

as in Plate LIV, fig. 2, where the plant is upright, rigid and dwarfish, as if the whole system had been poisoned. Figs. 24 and 25 represent the spores, under a high magnifying power; and so also Plate LIV, fig. 3 represents them as seen in plants by myself this season. In many fields, the smutty heads are very numerous.

II. RYE.

The specimens were procured from Arbor Hill, near the city of Albany, the soil being a gravelly loam based upon Albany clay. Heavy beds of drift frequently occur, which aid in giving a rolling surface to the country.

The plant was divided into parts, as grain, leaves, and top and bottom straw.

1. UNRIPE RYE.

Plant in blossom, and cut June 18, 1847.

PROPORTIONS.

								Per centum.
Water	•	•	•	•	۰	•	•	67.5600
Dry heads	-	•	•	•	-	-	-	$32 \cdot 4400$
Ash -	-	•	•	-	-	-	-	1.5300
Ash calcula	ated	dry	-	•	-	•	•	4.7165

2. Straw divided in two parts, top and bottom, and separated between the 3d and 4th joints.

							Top.	Воттом.
Water -	-	•	•	•	-	٠	55.2600	61.7000
Dry straw	•	-	-	•	-	-	44.7400	38.3000
Ash -	•	-	-	-	•	-	1.0010	1.2500
Ash calcula	ited d	ry	•	•	•	•	$2 \cdot 2351$	3.2637

3. Leaves and leaf-sheaths.

Water -		•	-	-		•	69.8400
Dry leaves -	•	•	•	-			$30 \cdot 1600$
Ash		•	•	•	-	•	1.6300
Ash calculated	dry						5.4045 S.

2. RIPE GRAIN AND STRAW.

Growth large: straw divided between the 3d and 4th joints.

PROPORTIONS.

				PROF	ORTIO.	NS.				
Grain Chaff	-	-	-	-	•	-	-	-	Per centum 43:206 8:356	
1. Top of the Bottom Leaves as	-		-	• •	-	- -		15·89 23·15 9·30		69
2. St	raw.									
Water - Dry straw - Ash Ash calculated	- dry	-	-		-		ç	Top. 8·127 91·878 3·572 3·888	!	Воттом. 7·603 92·397 2·343 2·536
3. Le	aves.									
Water Dry lea Ash Ash cal	ves •	-	- - y	•			-	- - -	7·933 92·066 5·383 5·847	
4. CA Water Dry cha	•			-	-			•	9·018 90·982	
Ash Ash cal		- ed dr	y	-	-	-	•	-	9.042 9.885	
5. G1										
Water Dry gra Ash	in -	•	-	•	•	-	•		9.8450 90.1550 1.0445	
Ash cal	culate	ed dr	у	•	•	-	•	•	1 · 1585	S.

3. Analysis of the ash of the heads of Rye. Cut June 18, when in blossom.

Sili	ca	-		-	-	-	-	-	-	61.800	
Coa	l	-	-	-	-	-	-		-	4.000	
Pho	spha	ates	:								
	C	of p	eroxi	de of	iron	-	-	-	1.500		
	Lin	ne	-	-	-	-	-	-	$3 \cdot 102$:	
	Ma	gne	esia	-	-	-	-	-	2.700		
	Sil	icic	acid	-	-	-	-	-	trace.		
	$\mathbf{P}\mathbf{h}$	osp]	horic	acid		-	-	_	9.798		
		•								17:100	
$_{ m Lim}$	e	-	-	-	-	-	-	-	-	0.507	
Mag	nes	ia	-	-	-	-	-	-	-	trace.	
Pota	ash	-	-	-	-	-	-	-	-	0.266	
Sod	a		-	-	-	-	-	-	-	none.	
Chl	orid	e of	sodi	um	-	-	-	-	•	12.600	
Sul	phur	ic a	cid	-	-	-	-	-	-	4.296	
Chl	orid	e of	pota	ssiun	ı -	-	-	-	-	1.586	
Org	anic	ac	ids	-	-	•	•	•	-	6.080	
										$104 \cdot 211$	S

The heads, in this period of growth, are remarkable for the small amount of potash, and the absence of soda, except in combination with chlorine forming common salt.

The composition of the leaves and leaf-sheaths is as follows:

Silica	-	-	-	-	-	-	35.900
Coal	-	-	-	-	-	-	4.500
Phosphates:							
of iron -		-	-	-	-	2.900	
Lime		-	-	-	-	3.396	
Magnesia -		-	-	-	-	0.600	
Silicic acid -		•	-	-	-	trace.	
Phosphoric ac	cid	-	-	-	-	8.704	
-							15.600
Carbonate of lime)	-	-	•	-	-	$5 \cdot 132$
Magnesia -	•	-	-	-	-	-	2.600
Potash	•	-	•	-	-	•	21.672
Soda	-	-	-	-	-	-	$2 \cdot 292$
Sulphuric acid		•	-	-	-	-	5.090
Carbonic acid	-	-	-	_	-	-	0.500
Organic acids	-	-	-	•	-	-	$2 \cdot 400$
							103·535 S.

RYE FROM ARBOR HILL, ALBANY (Growth large).

Analysis of the ripe straw, which was divided into two parts between the third and fourth nodes or joints.

Silica	•	ь				Top. 70.700	Воттом. 28·250
Phosphate of peroxi	de of	iron		ь	ь	0.250	1.350
Lime	-	-	•	ь		7.304	$8 \cdot 489$
Magnesia -	ь		ъ	-	-	trace.	0.050
Silicic acid -		**	•	10	•	trace.	trace.
Phosphoric acid		•	ъ	in .	-	$7 \cdot 296$	5.711
Lime	-		10	ъ		2.849	1-410
Magnesia	-	-	-	-		0.300	2.600
Potash		•	•		-	$3 \cdot 732$	$28 \cdot 402$
Soda		•	-	ь	60	1.324	17.615
Chloride of sodium	-	•		•		1.716	0.489
Sulphuric acid -	•	•		-	-	0.343	1.546
Organic acids .		ь			•	1.305	4.500
						97·118 S.	98·412 S.

There will be removed, in a ton of straw, the following amount of mineral matter in pounds:

								Tor.	Воттом.
Silica		•			•		6	56.560	14.803
Phosphat	es	-		-		•		11.880	8.169
Lime	•	•	10	-	-	•	-	2.279	0.738
Magnesia	. •	•				-		0.240	1.362
Potash	-	-	-	•	•	-	•	$2 \cdot 985$	14.882
Soda	-	•	-	•	-	-		$1 \cdot 059$	9.230
Chloride	of so	dium		•	-	ь	•	1.372	0.256
Sulphuric	acid	-	-	-	-	-	-	0.274	0.810
Organic a	icids	•	-	-	-	•		1.044	$2 \cdot 358$

It is not forgotten, however, that it is a common practice to return a large part of the straw to the field again: it is not therefore lost. But in the vicinity of large towns, the straw is generally sold in bundles for various purposes.

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The composition of the ripe leaves and leaf-sheaths is as follows:

·							Removed in a ton.
Silica	-	UA.	~	đ	•	71.650	85·980 lbs.
Phosphates (14.250):						
Phosphate of perc	oxide	of ir	on	9	•	0.250	0.300
Lime	•	•	•	-	•	8.658	$10 \cdot 389$
Magnesia -	-	-	-	-	-	trace.	
Silicic acid -	-	•	•	-	ø	trace.	
Phosphoric acid	-	•	-	-	•	$5 \cdot 342$	6.410
Carbonate of lime	-	•	-	-	•	0.817	$1 \cdot 000$
Magnesia	-	-	-	ď	•	0.500	0.600
Potash	-	-	-	-	-	2.720	$3 \cdot 264$
Soda	-	-	-	-	-	$3 \cdot 333$	3.999
Chloride of sodium	-	-	-	ø	•	1.184	1.420
Sulphuric acid 🕝		ø	•	e#	-	0.100	0.120
Carbonic acid .	ø	-	•	•	•	trace.	
Organic matter -	•	-	ø	æ	-	$2 \cdot 650$	3.180
						97·204 S.	$\overline{116.662}$

Analysis of the ripe chaff of the rye of Arbor Hill.

Silica -	-	-				-		85.600
Phosphates	(4.60)):						
Phosphate	of p	erox	ide of	iron	•	-	•	0.350
$_{ m Lime}$		-	•	-	-	•	-	3.078
Magnesia	-	-	-	-	-	-	-	trace.
Silicic aci	d	-	•	•	•	-	-	trace.
Phosphori	c aci	d	-	ď	-	-	-	$1 \cdot 182$
Carbonate of	lime	Э	ď	-	-	-	-	1.664
Magnesia	-	-	-	•	-	-	-	0.500
Potash -	-	-	-	•	•	-	-	$3 \cdot 922$
Soda -	-	-	•	•	-	-	•	1.994
Chloride of	sodiu	m	-	-	•	-	•	0.245
Sulphuric ac	id	-	•	•	-	•	-	0.206
Carbonic aci	d	-	•	-	•	-	-	trace.
Organic acid	s	-	8	•	•	-	-	$1 \cdot 150$
								99·891 S.

Analysis of the ripe grain of the same rye.

Silica -	-	-	-	•	•	•	•	$3 \cdot 450$
Phosphat	es (4 8	8.85):						
Phospl	nate of	perox	ide of	iron	•	-	•	6.350
Lime	-	•	-	•	-	-	-	6.458
Magne	sia -	-	-	•	•	-	-	0.816
Silicic	acid	•	-	•		- 20	•	0.250
Phospe	oric ac	id and p	oh o spl	hates o	f sod	a and	potasl	a, 34·676
Carbona			•	•	-	•	-	0.582
Magnesia	a -	•	-	•	-	•	-	0.400
Potash .	-	•	-	•	-	•	•	18.217
Soda -		•	-	•	-	-	•	8.931
Chloride	of so	dium	-	-	-	•	-	0.709
Sulphuri	c acid	-	•	.•	-	•	-	4.724
Carbonic	acid	•	-	-	-	•	-	trace.
Organic	acids	•	-	•	-	•	-	$6 \cdot 125$
Phosphae	es of	soda ar	d pot	ash	•	•	-	6.600
								97·990 S.

A bushel of rye, weighing 60 lbs., will contain the above elements as follows:

						In a bushel,	In the produce of one acre, at 20 bushels.
Silica -	•	-	-	-	-	0·331 oz.	0.414 lbs.
Phosphates	•	-	-	•	-	4.660	5.826
Lime -	-	-	-	•	•	0.027	0.033
Magnesia	•	-	-	•	•	0.038	0.048
Potash -	•	-		-	•	1.748	$2 \cdot 186$
Soda -	-	-	-		-	0.857	1.071
Chloride of	sodium	-		-	•	0.068	0.085
Sulphuric a	cid -	-	•	-	•	0.453	0.966
Organic aci	ds -	-	-	-	-	0.588	0.735
Phosphates	of soda	and	potash	ı -	-	0.633	$0 \cdot 732$
_			_				
						9·403 oz.	11.756 lbs.

ORGANIC ANALYSIS OF RYE.

Raised near H	Ioosic F	alls, l	Renss	elaer c	ounty	. Full b	earing.	
Specific gra	vity	-	•	-	-	- 3	.282.	
		PRO	PORT	ions,				
Per centum of	water					-	22.400	
16	dry m	atter	-	-			87.600	
6.6	ash	-	-	_		-	1.030	
66	ash c	alcula	ted d	lry -	•		1.175	
				alcula	ted d	ry -	98.825	S.
~ .								culated dry.
Starch		•	-	-	-	53.93	(30.737
When dry, translucent, horny Dextrine	; iractu	re vitr	eous.			2.10		0.967
Albumen	-	•	•	•	•	2.51		2.367 2.828
Casein	-	•	-	•	•	0.40		0.450
Matter dissolved out	of anid	- ormie	- and	othor	- bodio			0.490
insoluble in water,	•							
bles gluten -	by 501.	g a	-	, It .	i esem	1 · 17		1.318
Matter dissolved out	of kerr	els h	v dia	esting	then			1 010
for some time in bo		•			-	6.58		7.415
It has a slight empyreumatic t					rye flot			. 110
Matter dissolved out								
insoluble in water a	ınd boil	ling al	lcoho	l, by a	weal	k		
solution of caustic	potash	: albu	men	•	-	$2 \cdot 34$		2.636
Epidermis, after treat	ing it v	vith b	oiling	g alcoh	ol and	d		
a weak solution of	potash	-	-	-	-	11.44	1	12.890
Water	- -	-	•	•	-	12.40	-	
Extractive matter, su	gar, an	d oth	er bo	odies r	iot de	-	•	
termined : has a pu	ingent,	sligh	tly bi	itter ta	ste	8.28		9.329
					_	101 · 15	S. 10	000.000

DISEASES OF RYE.

Rye is a hardy plant, and resists the attacks of disease and insects to a great extent. A fungous plant, or a degeneration of the kernel, well known as *ergot*, is, however, common to almost every field of rye, but some seasons favor its production more than others.

Various opinions have been expressed of the substance, known as ergot. Some writers maintain that it is a distinct plant; others, that it is a degeneration of the rye kernel. Corda, who is probably the best authority, maintains that the tissues of the kernel are crowded out, and replaced by those of the ergot, and hence that it is a plant.

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The structure of this remarkable plant is exhibited on Plate LVI, figs. 21, 22, 23, etc.* Figs. 21 and 22 represent the spurred rye in its natural size. Fig. 24 is the top of a spurred rye-kernel, greatly magnified so as to show the spores a and b; c the outer skin, and d the substance of the plant. Other figures exhibit the structure of the rye-kernel: fig. 28, a thin section of a magnified kernel; a, skin or cuticle; b, inner seed-skin; c, cellular layers of gluten; d, albuminous substance, formed from the cells that bear the starch meal. Fig. 29, a cell of albuminous substance, with grains of amylum magnified. Fig. 30, single grains of amylum magnified.

Another fungus, of rare occurrence however, appears upon rye which is grown in elevated positions, or near the borders of rye culture. It is a deep red gelatinous mass, often half an inch in length. It attacks the outer skin of rye-kernels and seeds of grasses, which appear swollen and deformed. The outer skin is wholly destroyed, but the albuminous matter remains unaffected by it (Figs. 16, 17, 18, Pl. LIV). The fungus itself, as seen under the microscope, is a confused reddish white layer, bearing the spores, fig. 19. The cells of the layer form on their tops, spindle-shaped spores or seeds with four cells. This fungus is not poisonous, like that of ergot.

Figs. 16, 17, 18, single degenerated seeds of rye, affected by fungus of the natural size. Fig. 19, a thin section magnified: a, fleshy bearer; c, spores. Fig. 20, single spores magnified.

III. BARLEY.

Barley, which, in the appearance of its flour resembles wheat, but is quite different in composition, is susceptible of a wider cultivation than wheat. Hence in those districts where wheat, from some cause or other, becomes an unprofitable crop, barley is often substituted for it, and even becomes an advantageous substitute. Its range of culture is very wide, and its liability to a failure is certainly much less than that of wheat.

The barley which I have analyzed, has been derived from only three districts: the Hudson-river, the Southern, and the Western districts. It has not received so much attention as it deserves, for the reason that sufficient time could not be bestowed upon it.

I shall first take up the barley of the Western district. It was furnished by Mr. Peters, of Genesee county, who is one of the most distinguished agriculturists of Western New-York. It was the two-rowed variety, and was well filled. The general characters of the soil have been alluded to, but its particular characteristics have not been given: neither is the mode of culture known.

^{*} Plates and figures from Corda's work, translated from the German by E. Goodrich Smith.

I. BARLEY FURNISHED BY MR. PETERS.

Straw bright; 18 to 24 inches long.

	Specifi	ic gra	vity	•	•	-	-	•	•	1.1	76.	
					PROP	ortio	NS.					
	Constan								A	ctual qua		
	Grain Straw		-	~	•	-	•	•	-	1046		
	Chaff		•	-	•	•	•	•	•	800 · 236 ·		
	Chan	•	•	-	•	•	•	•	-	230	00	
~ .								_		Water.	Ash.	
Grain	-	•	-	•	•	-	10			12 · 16	3.20	
Straw	-	•	-	•	-	•	10				4.04	
Chaff	•	•	•	•	•	•	10	U			$5 \cdot 46$	
									Si	lica.	Phosphates.	
The as			s. of	grain		•	-	•		•37	1.43	
4.6		"		straw	66	-	•	-		.37	1.40	
4.6	•	46		chaff	"	-	-	•	3	$\cdot 38$	0.75	
				1 /	17		42	1.				
				1. 2	inaiys	rs of	the as	n.		Per cen	tum: without car	rbonie
Silica	_	_	_	_		_		91.8	เกก	acid	or organic matte	er.
Phosph	ates of	Lim	e m	aonesi	a and	iron		31.8			36.188	
Carbon						-		9.5			0.569	
Magne		•		-	_		_		140		1.638	
Potash			-	•	_	-	_	22.7			25.912	
Soda	_	_		-	_			2.4			2.822	
Chlorir	ne		_	•	_			1.0			$1 \cdot 172$	
Sulphu		d				۰			314		9.461	
Organi								5.4				
Coal	-	•		4				3.5				
Carbon	ic acid					-	•	3.				
								100 · 2	19		100.294	
				o		•	. 47					
			•	2. Con	iposiii	on oj	ine se	raw.		Per cent	um : without car	bonic
Silica								53 · 1	120	acid	or organic matte	er.
Phosph	ates							14.6			15.761	
Carbon		lime						1.0			1.087	
Magne		•			•			0.1			0.173	
Potash								4.3		,	4.674	
Soda		_		•		-		5.0			5.456	
Sulphu		d		•	•			11.7			12.717	
Chlorin		•	•	•	•		•	2.4	_		2.676	
Carbon		and	orga	nic m	atter		-		113		•••	
			5								*****	
								100 :	373		$100 \cdot 283$	

3. Composition of the chaff.

Silica -	•	•		-	•	•	•	65.360
Phosphate	-	-	•		•	-	•	$13 \cdot 260$
Carbonate of	lime		-	•	•	-	•	4.200
Magnesia	•	•	•	-	•	-	•	0.512
Potash -	•	-	•	•	٠	-	-	4.740
Soda •	•	•	-	•	•	•	•	$3 \cdot 751$
Chlorine	•	•	-	-	•	•	•	1.210
Sulphuric aci	id	-	•	•		-	•	8.967
Organic mat	ter	•	•	٠	•	•	•	4.268

II. BARLEY RAISED IN ALBANY COUNTY.

Harvested July 25.

PL. LVII, Fig. 1, two-rowed barley; fig. 2, four-rowed barley; fig. 3, smut; fig. 4, spores.

						PI	ROPORT	cions,			
145.50	grs. of	the g	grain	gave	•						
	Water								•	63.90	Per centum 43.90
		•	•	•	•	•	•	•	•		
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	•	•	-	•	-	•	•	5 ·25	. 3.60
	Ash ca	lcula	ted di	y	•	•	-	•	-	$9 \cdot 46$	
				•							
100 grs.	of the	strav	v gav	e							
		Dry	matte	r -	•	•	•	•	•	•	49.500
		Ash	-	•	•	-	•	-	-	•	$2 \cdot 375$
		Sp	ecific	grav	ity	•	-	-	•	•]	l·234.

The specific gravity of this barley, although the same variety, the two-rowed, is greater than that grown in Western New-York in the Wheat district. The plant seemed to be of a larger growth, having a length of straw 3 feet.

1. Comparative proportion of grain to straw, etc.

								Actual quantities.	Per centum.
Grain	-	•	•	•	•	•	•	$666 \cdot 00$	100.00
Straw	•	•	• .	•	-	•	•	$732 \cdot 00$	$109 \cdot 90$
Chaff	•	•	•	•	•	•	•	$124 \cdot 00$	18.61

2. Analysis of the ash of the grain.

Silica -						23.807	Amount of elements in 50 lbs. of grain. 6.854 oz.
Phosphates (5	7 · 849):					0 001 02.
of lime,	iron an	id ma	gnesia			41.603	11.981
of soda a	nd pot	ash	•	٠	•	16.246	4.678
Carbonate of	lime	•	-	•	•	none.	
Magnesia -	•	-		-	~	none.	
Potash .	•	-	•	•	•	$12 \cdot 035$	3.466
Soda	•	•	•	•	•	5.460	1.572
Chlorine -	-	•	•	•	•	none.	
Sulphuric acid	d -	-	-	-	-	none.	
							Name of the second of the seco
							28·551 oz.

Calculating that an acre will yield 30 bushels, and that a bushel weighs 50 lbs., there will be removed from it the following amounts of the elements, taking the foregoing analysis as correct:

In	Silica	-	-	•	•	10	•	-	12.855 lbs.	
	Phospha	tes	-	•	-	•	•	-	31.148	
	Potash	•	-	•	•	•	•	-	6.498	
	Soda	•			•	•	-	-	2.948	
									$53 \cdot 449$ lbs.	

It will be observed that this barley differs somewhat in composition from that grown in the Wheat district of this State; but as the analyses which have as yet been made are few, it will not be safe to make generalizations from them. Barley is regarded as a lime plant, when compared with wheat. The lime and magnesia, in this last analysis, seem to be combined entirely with phosphoric acid, as usually is the case in wheat.

3. Analysis of the straw and chaff.

			Perc	entag	e of	ash, a	5.00.	
Silica •	•		•	•			70.001	Mineral matter removed in a ton of straw and chaff. 78.401 lbs.
Phosphates	-	-	-				10.678	11.959
Lime -	•	-	•	-		•	0.081	0.090
Magnesia		•	•	•			trace.	
Potash -	-	•	-	-		٠	10.678	$11 \cdot 959$
Soda -		-		-			2.983	$3 \cdot 340$
Sulphuric ac	id	•	-			•	$5 \cdot 733$	$6 \cdot 420$
Chlorine	-	•		٠	-	-	none.	
								112·169 lbs.

The amount of silica is increased by the mixture of the chaff with the straw: the chaff gives $5\frac{1}{2}$ per centum of ash, the straw about 4.

I remarked, in the foregoing paragraph, that this sample of grain gave a result somewhat different from that of barley grown in Western New-York. It will be observed, also, that the straw is different, having less lime, magnesia and chlorine. This straw differs also from a sample received from Col. H. S. RANDALL, of Cortland county.

III. BARLEY RECEIVED FROM Col. H. S. RANDALL.

Raised in Cortlandville, Cortland county. Rock of the county, shales of the Ithaca group, greenish gray, and subject to decomposition; surface undulating; beds of northern drift not unfrequent.

			_						_				
					P	ROPO	RTIC	INS.					
							Ac	tual quant	ities.	A	sh.	Percentage of as	h
Grain	ь	-	•	•		-		$344 \cdot 3$	1	10	•66	3.096	
Chaff	•	b.	-		ь	-		62 · 1	8	3	•50	5.628	
Straw		•	•	•	•			210.0	0	7	05	$3 \cdot 357$	
				A	nals	ıçiç n	f t	he strai	37.				
	Silie	a -		-	navg	•	•	•	~•	•	37.80	00	
	-		es (2	5:0):									
				f per	oxide	e of	iror	ı •		>	12.5	00	
	\mathbf{L}	ime	-	-		-					12 · 1	50	
	\mathbf{M}	lagne	sia -	-			•			ь	0.36	00	
		bonat		lime					6		11.9	00	
	Mag	nesia	ւ -	-		6	to		6	-	$4 \cdot 9$	50	
	-	ash •		to			•	-		ь	8.5	05	
	Sod	a •		•		-	ь	-	•	-	nor	ne.	
	Chlo	orine				-	-	-	-	-	trac	e.	
	Sul	huri	acid			-	-	-		6	5.3	12	
	-	anic 1				•	-	•	-	-	4.6	30	
											99.0	— 97 В.	

The lime and magnesia, in this sample of barley, exceed considerably that of Judge Peters, which was grown upon the shales of the Onondaga-salt group, in which the amount of lime exceeds that either of the Hudson-river district, or the higher range of seils from the slates and shales of the Hamilton and Chemung groups in the southern tier of counties. I have not been furnished, in either case, with the mode of culture, or the manures which were used; facts which, perhaps, would throw some light upon the cause of the differences in composition of the three samples of barley.

IV. ANALYSIS FOR OBTAINING THE PROXIMATE ELEMENTS OF BARLEY.

The method pursued was the mechanical one, by which, from its imperfections, it is impossible probably to represent the true composition. This objection, however, bears with greater force upon the separation of the starch and products representing the gluten. Raised and furnished by Judge Peters, of Genesee county.

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044 4				. 1		
341 · 1	lυ	grs.	10	the	grain	gave

011 10 515. 01	the grain g	ave							
	Fine flour	•	-	•	•	-	-	-	285 50
	Bran -	-	•	-	-	•	•	-	55.60
100 grs. of thi	s flour gave								
	Starch -	-	-	-	-			•	$54 \cdot 02$
	Gluten -	-	_	-	-		-	-	1.98
	$\mathbf{Albumen}$	-	-	-	-	-	-	-	1.67
	Oil -	-	-	-	-	-	-	•	0.64
	Casein -	-	-	-	-	-	-	•	$2 \cdot 44$
	Dextrine		-	-	-	-	-	-	$5 \cdot 44$
	Sugar and e	extract	-	-	-	-	•	-	8.19
	A substance	insolu	ble i	n wat	er or	alcol	nol	•	18.79

In regard to the foregoing analysis, I have reason to suppose that the oil exists in a larger proportion than that which I have given. It is, however, much like that of maize, of a fine yellow, and a smell not unlike that of cakes taken recently from the oven. After the starch is separated, there remains a brown brittle mass unlike the adhesive gluten of wheat, but from which alchohol dissolved out nearly 2 grains, or ten per cent. The small quantity of gluten renders barley flour unsuitable for bread; that is, light bread, as there is not a sufficient quantity of gluten present to give a permanent consistency to the dough. This brittle substance, or substance without cohesion, might with propriety receive a distinct name, when its composition has been determined. The dextrine forms quite a large proportion, when compared with other grains.

TWO-ROWED BARLEY GROWN NEAR ALBANY.

			I	Iarves	ted Ju	ly 25.			
275 59 grs. g	ave of					•			
	Fine flour	-	-	-	-	-	-		$218 \cdot 30$
	Bran -	•	-	•	-	-	•	-	57 ·29
100 grs. of th	e flour conta	ined							
	Starch -			•		-	4	-	40.420
	Albumen	•	-		-	-	-	-	3.970
	Casein -	-	-	-	-	-	-		2.080
	Matter insol	luble	in wa	ater o	r alco	hol	-	-	21.500
	Gluten -		-	-	-	-	đ		2.480
	Dextrine	-	-	•	-	-	-	-	7.720
	Oil -	-	-	-	-	-	-	-	2.340
	Extract and	suga	ır	-	-	-	-	-	6.920
	Water -		•	•	•	•	•	-	$13 \cdot 220$
									87.430

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DISEASES OF BARLEY.

There is a strong resemblance, in some of the diseases of the barley, to those of rye. Barley, for example, bears the ergot, which can not be distinguished from that which infests rye. Another diseased growth resembles the smut which appears upon the oat, and is called barley brand (Uredo hordei). It has its seat in the ears of barley, particularly the floral parts. The effect of the barley brand upon the kernel, is quite different from that upon the oat or wheat. In these, the fruit is wholly obliterated. In the barley, there is developed new woody tissues, or woody bundles; these are formed between the layers of the brand, or fungous plant (See Pl. LVI, fig. 7, d, d). Fig. 3, a, a, exhibits the structure of a sound kernel highly magnified, in a section cut across it near its middle. Fig. 6 is a cross section of a kernel attacked with the disease: the outer skin remains sound; but the interior is filled with a mass of black substance intersected with white lines or veins. The spores form a dark smutty mass, inclining to an olive green: they are oval bodies. The spores ripen early, and are shed before the grain is harvested. Those diseased heads, therefore, ought to be removed and burnt, that they may not infest the succeeding crops.

The barley brand, like all others of the same class, is more frequent in cool moist seasons, and on wet soil. It is also much more common in fields sown to barley, when the grain has not been thoroughly cleansed. Fig. 8, Pl. LVI, shows the spores, or reproductive bodies, greatly magnified.

In order to present as perfect a view of the value of the barley as possible, I have copied an article from a late number of the Journal of Agriculture and Transactions of the Highland and Agricultural Society of Scotland, containing two analyses of the brewers' grains so much in use in this country for feeding stock, particularly milch-cows.

Of the composition of brewers' draff or grains, and its value as a food for milk-cows.

A difference of opinion having arisen between the buyers and sellers of brewers' draff in Edinburgh, regarding its value, and the price that ought to be paid for it, Mr. Girdwood thought that some light might be thrown upon this question by a chemical analysis. He caused some of it, therefore, to be sent to the laboratory, where it was submitted to both an organic and an inorganic analysis, with the following results:

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1. Organic analysis.

A hundred pounds of the fresh draff were found to contain

Water		•		-			-	•			75.85
Gum -		-	-		•	•	_	-		-	1.06
Other org	anic r	natter	(chi	efly h	usk)	-	•		-	-	21.28
Organic n	atter	, conta	ining	g nitro	gen (protei	ne co	mpou	nds)		0.62
Inorganic	matte	r or as	h	•	-	•	-		•	-	1.19
											100:00

From the above analysis we see that, during the digestion of the malt in the mash-tub, the proteine compounds—those which contain nitrogen, and are necessary to the production of muscle in the body, and curd in the milk—are nearly all dissolved out.

2. Inorganic analysis.

The ash left on burning the draff was found to consist of

An 1: 1: (11 :1 :1 :1 :1 :1	Per cent of ash.	In 1000 parts of wet draff.	In 1000 parts of dry draff.
Alkaline salts (chlorides, with a small quantity			
of sulphates) and alkali	7.60	0.80	$3 \cdot 72$
Phosphoric acid in combination with the alkali -	2.11	0.25	1.04
Earthy phosphates	48.00	5.81	24.06
Silica	41.51	4.94	20.46
		·——	
	99.22	11.90	49.28

An examination of the above numbers suggests the following remarks in regard to the value of draff in comparison with other kinds of food.

- a. The quantity of water present in it is 76 per cent; in this respect it approaches very near to potatoes, and some other varieties of green food. It contains, however, considerably less than the turnip or the cabbage.
- b. The greater part of the solid matter, as we should suppose, consists of husk. This is not wholly insoluble in the stomach of the cow, nor without considerable nutritive power. It is impossible, from theoretical considerations, to assign any definite value to this husk; but the experience of the cowfeeder seems to show that it is not by any means worthless in the feeding of milk-cows. In the grains of the brewers it is generally understood that the proportion of nutritive matter left is much less than in those of the distiller.
- c. One important result is the small proportion of proteine compounds, amounting only to two and a half per cent in the dry grains. No doubt this proportion will vary in different samples. It is probably attached to the husk in the form of coagulated albumen, which, however, is dissolved and appropriated in the stomach of the animal.

The draff weighs about 46 lbs. to the bushel, and costs 3d. to 3½d. for this weight. One hundred and sixty pounds of draff contain one of albumen, and this weight costs 1s. Five

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pounds of oil-cake contain one of albumen or other proteine compounds, and costs 5d. This ingredient of the food, therefore, is cheaper in the form of oil-cake than in that of brewers' draff.

d. Again, the quantity of ash left by the dry draff is about five per cent. One hundred pounds in the wet state contain 0.6 lbs. of phosphates; or 160 lbs., costing 1s., contain 1 lb. of phosphates. But 33 lbs. of oil-cake contain 1 lb. of phosphates at a cost of nearly 3s.; thus the phosphates are cheaper in the form of draff.

Hence a mixture of other food, such as oil-cake or beans, along with it, is recommended where the draff is to be used most economically and with least waste.

e. A part of the feeding value found in the draff by past experience, is probably to be ascribed to the steeping it has undergone, rendering the otherwise innutritive or slowly nutritious matter soluble in the stomach, and thus admitting of less waste.

Turnips are the kind of food most usually given with brewers' grains. The following letter from Mr. Caird, Baldoon, near Wigton, shows the profit of feeding milk-cows upon draff and turnips, compared with beans and turnips, according to his experience.

"I have fed for the last two winters a large dairy stock in the following manner, for between 6 and 7 months, or 200 days:

Each cow half a bushel of draff (23 lbs.) per day, for 200 days,	100			
bushels at 3d.	•	£1	5	0
Each cow (22 lbs. per day), 2 tons of turnips, at 10s		1	0	0
		£2	5	0

"And the dairyman to whom my cows are let prefers this feeding to the following, which is the usual allowance in this district to a dairy stock:

Each cow	4 to	ns of	turnips at	10s.	•		•	•		•	£2	0	0
**	2 bu	shels	of beans,	ground,	at 4s	. 6d.	•	-	•	-	0	9	0
											£2	9	0

In both cases the same fodder is given.

"On the draff and turnips the cows give fully more milk and butter, both of which are well tasted, and they keep themselves in better condition than when fed on turnips alone. With this feeding they are a very healthy stock.

"On the turnips and beans, the butter and milk are always strong tasted, and the cows are not kept in such high condition as on the draff and turnips.

"If cows get an unlimited supply of turnips, they may yield more milk than on the quantity of draff and turnips mentioned above. They do not eat beyond a certain quantity of draff, while it is difficult to satisfy their appetite for turnips.

"The question as to which of the two is the cheapest food, depends altogether on the respective money values of draff and turnips in any given locality."

IV. MILLET.

This grain is rarely cultivated, so far as my observation extends; and I am informed that it is not held in as much esteem now as formerly. There are probably good reasons why its cultivation is not regarded with favor, but I am not able to assign sufficient causes why it should be neglected. Its analysis, it is true, shows that it is an exhausting crop; but an exhausting crop is also relatively valuable. Its growth is large; and it is highly probable that when left in the field for fodder, the stalks and leaves lose their value. If cut when in blossom, its fodder is valuable, but its seed valueless, and it is then no better than the common grasses. The grain of millet is useless as flour for bread, where maize, rye and wheat, and barley, can be obtained. Its flour, if it has any, is rather coarse, and brownish yellow, or the color of the seed. Millet seed, however, is rich in nutritive elements, and I believe exceeds all others. It is productive: an acre, when properly tilled, yields seventy bushels. In addition to this, the stalk, though not as valuable as timothy, still is equal to oat straw for fodder, and probably superior to it. It requires a rich soil, or at least produces in proportion to its richness.

The specimens for analysis were furnished by Mr. Bement, who has often raised it, and who entertains a high opinion of it, either as food for cattle when cut young, or for its grain when ripened. The plants were of spontaneous growth in a field of potatoes.

ANALYSIS OF THE MILLET.

PROPORTIONS.

1. Stalk and leaves.

Ash

				Actual quantilies.	
				2670 · 00 grs. ripe, and s	lightly wilted, gave
\mathbf{Dry}	-	-	-	1309.00	48.65 per centum.
Ash	•	-	-	99.73	3.73
	2.	Mill	et head	ds in blossom.	
				260.00 grs. gave	
Dry	-	-		167.00	64.23 per centum.
Ash	-	-	-	$15 \cdot 74$	6.05
	3.	Mille	et gra	in.	
				840.000 grs. gave	
Dry	-	-	-	not obtained.	

28.615

3.40 per centum.

Millet heads, cut while in blossom, gave the following results:

Silica -	۰	•	•	•				45.083
Phosphate	es of li	me, n	nagne	sia and	l iron		-	20.000
Carbonate	of lim	ıe	•	٠	•	-	-	0.160
Magnesia		•	•	-	٠	•	•	0.130
Potash -	-		-	•	-	•	-	4.250
Soda -	•	•	•	-		•	•	14.650
Chlorine		•	•	-		ю	•	trace.
Sulphuric	acid	•	•	•	•	-	ø	$12 \cdot 390$
Organic r	natter a	ınd ca	irboni	ic acid		•	•	$3 \cdot 145$
								99.788

The large quantity of soda, compared with potash, is quite remarkable. I have no opportunity for repeating the work for the purpose of confirming or disproving this result. The almost total absence of chlorine where so much soda exists, is also unfrequent.

ANALYSIS OF MILLET SEED OR GRAIN.

Silica -	-	-	-	-	-	-	•	$44 \cdot 294$
Phosphates	of lin	ne, r	nagne	sia a	nd iron	•	-	34.555
Carbonate	of lim	е	•	-	-	-	-	none.
Magnesia	-	-	•	-	-	-	-	none.
Potash -	-	•	-	•	-	•	-	$7 \cdot 178$
Soda -	-		•		•	-		8.239
Sulphuric a	icid	•	-	-	•	•	•	trace.
Chlorine	•	-	•	-	•	-	•	trace
Phosphate	of the	alka	lies	-	-	•		0.620
Organic ma	atter	•	-	•	٥	-	•	1.459

From five grains of the ash of millet seed, only a trace of sulphate of barytes was observed. The presence of sulphur is, however, clearly indicated in the organic analysis, though only feebly in the ash analyses; the large quantity of nitrogenous matter proving that it must be present in a larger amount than appears in the inorganic analysis.

ANALYSIS OF THE STRAW OF THE RIPE MILLET.

								Elements in a ton of millet straw.
Silica -	-	-	-	-	•	a	53.750	44.908 lbs.
Phosphates	of lim	e, m	agnes	ia and	l iron	-	18.150	$15 \cdot 164$
Carbonate of	f lime	-	-	-	-	-	0.400	0.334
Magnesia	-	•	•	-	-	•	0.430	0.360
Potash -	-	-	-	-	•	•	$17 \cdot 395$	14·533
Soda -	-	•		•	•		1.525	1.274
Sulphuric a	cid	-	•		•	•	2.500	2.088
Chlorine	•	-	-	•	•	•	0.512	0.427
Organic ma	tter	•	•		-	-		
Carbonic ac	id	۰		-	-		1.056	

The straw, especially that which is thoroughly ripened, is larger, coarser and harsher than any of the grasses used for fodder. It is also quite siliceous, its ash being composed of more than one half of silica.

ANALYSIS OF THE GRAIN FOR THE PROXIMATE ELEMENTS.

100 grs. of the comparatively pure flour gave

Starch -	٠	•	-	٠	-	6,	•	34.840
Albumen	•	-	•	4	6	-	è	8.225
Casein -	•	•	•	•	•	۰	٠	4.765
Dextrine	4	•	•	•		-	4	4.080
Gluten and	oil	•	-	4	-		•	$6 \cdot 240$
Insoluble m	atter	in wa	ter or	alcoh	ol, in	part f	ibre	20.230
Sugar and	extra	ct	-	•	-	•		10.200
Water -	-	-				•		11.061

The oil, though not accurately determined, exists in about 2 per centum.

The analysis for the proximate elements shows that millet is one of the most nutritive grains we have. The albumen and casein exist in three times the amount they are found in the oat; the gluten in about the same, and oil a little less.

Whether the nutritive elements in these large proportions will recommend this grain to a more favorable notice of the agriculturists of New-York, is a question of some importance. For feeding stock, when ground, it can scarcely be doubted that it must be more valuable than maize, especially for working oxen, or for horses. In addition to the nitrogenous substances, it is certainly not deficient in respiratory nor in bone-forming elements, as phosphoric acid, and lime and magnesia. So in whatever light its composition is regarded, it can not be otherwise than favorable. It is, however, necessary when it is designed for cattle, that-it be ground, or made into meal. When it is fed to hogs or canary birds, it may be given whole, or without grinding. Still for the former it must be better when the seeds are crushed, as they are invested with a siliceous cuticle, quite as unfit for digestion as the chaff of other grains.

In order to compare the exhausting power of millet upon the soil, with other grains, it may be stated in this place, that with a yield of 70 bushels to the acre, each weighing 54 lbs., there will be removed,

In Silica	•	-	-	•	٠		62·451 lbs.
Phosphates -	-	•	-	-	•-	•	47.310
Potash and soda		-				٠	19.813

The lime and magnesia is in combination with phosphoric acid; and after the precipitation of the phosphates by ammonia, the filtrate remains nearly undisturbed by the addition of oxalate of ammonia, or phosphate of soda and ammonia. The phosphates, however, unlike those of wheat and indian corn, are tribasic.

DISEASES OF MILLET.

Millet is subject to the attack of fungous plants, like rye and other cereals. The smut of millet, inasmuch as it is rarely cultivated, has never fallen under my own observation. It is, however, described by Corda as affecting the whole panicle or head, the original implantation of the fungus taking possession of the whole while invested in the leaf-sheath. The spores are oval, globular, and olive brown. As the whole panicle is infested, and becomes thereby a prominent object, it is easy to destroy or remove the fungus from the field, by removing the infected individuals and burning them; for if suffered to remain in the field, it will propagate itself for a long time afterwards.

For illustrations of the millet smut, see Pl. LVII, fig. 3-7, Uredo destruens. Fig. 3, immature brand or smut; 4, mature, and shedding its spores; 5, single fibres, natural size; 6, magnified: a, cells of the inner bark; b, spiral vessels of the woody bundles; 7, spores strongly magnified.

V. WHEAT.

It is unnecessary to speak of the importance of this grain: its value is appreciated by community; yet in saying this, we doubt much whether the influence of its culture and use is fully felt by a large portion of society. The doubt is expressed, from the belief that it has had no inconsiderable share of influence on the progress of civilization. It is the food of civilized man. It is the bread of refinement and taste. The loaf of the Genesee flour is the extreme luxury of polished life; while the hoecake of indian corn, baked in ashes, is the symbol of savage fare. The former has become the standard food of the wealthy; while the latter finds its place on his table, rather as a curiosity than one of common resort.

The culture of wheat, too, is a civilized act: it is almost incompatible with savage life. A perfect crop requires the exercise of the highest skill of the husbandman's art: it is the ne plus ultra of agriculture. It is true, that thousands of acres wave their heads to the breeze in the west, under the management and culture of ignorance it may be; but it is in a virgin soil, which nature has made, and whose elements are mixed in right proportions to bring forth the crop. But soon this power fails, and, with its failure, the plant is a profitless product, until knowledge and art restore it to its standard value.

In the present state of society, the place of wheat could not be compensated by either of the other cereals. None of them are really fitted for use alone; while wheat is so constituted, that a mixture of the flour of any other grain is not required either for bread or pastry: it is fit in itself for the purposes for which bread, in its widest meaning, is required.

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The history of the wheat, then, is in part the history of the progress of nations in civilization. It has kept pace with this progress, and has had its influence in promoting it.

The wheat culture of New-York has always held a high place in its husbandry. It is eclipsed, perhaps, by that of many of the western States in their early settlement; but for real adaptation, both in soil and climate, a large portion of New-York is not excelled. It is not to be concealed, however, that the means of communication over wide areas have been so far increased in the last twenty years, that the profits of the crop in this State have apparently diminished. This fact has undoubtedly had its influence upon the extent of its cultivation; still it is among the first in the amount of its productions, and undoubtedly it ever will continue a leading and prominent grain in its husbandry.

Wheat seems to be easily influenced by soil and climate. To this cause, we are indebted for many varieties which possess peculiar characters, or in which there is an excess of one of the proximate elements over the others. Gluten, the pasty part of the grain, is largely developed in some, and starch in other varieties. Some are protected by a thick cuticle, which forms the bran; in others, it is thin and delicate. The former are dark colored usually; the latter, white. Some again are armed with stiff and rigid awns; others are awnless. Silex is more abundant in some than in others. Those in which this element is in full proportion, stand erect, and never lodge. Even in the course of a few years' culture, changes occur in the constitution of wheat, which are easily observed. Thus the Mediterranean wheat, which has the property of hardiness to recommend it, and which has a thick skin and dark color, becomes assimilated in a few years to the more delicate and whiter varieties. This is especially the case in Central New-York; and from this fact, alone, we might draw the inference that Central New-York is the true land for the cultivation of the finest varieties of this grain.

I have supposed that farmers will be interested in making a thorough acquaintance with the varieties of wheat which have been in culture in this State. I have therefore prepared drawings of several of the most important kinds, whether they have succeeded or failed, whether they were worthy or unworthy of cultivation. The facts upon this branch of the subject I have derived from farmers of experience, and particularly from Rawson Harmon, esquire, of Monroe county.

A. WINTER WHEAT.

1. White Flint. In this State it is an esteemed variety, and is supposed to be of Spanish origin. It has been cultivated here about thirty years. Its color is white, and the awns of the chaff are soft, and confined to the upper part of the head. The straw is of a medium length, and is quite solid and wiry at the root, and hence is not liable to lodge. The heads are only of a medium length, but they are well filled out with white thinskinned kernels, which strongly adhere to the spike. The straw too is white and clean, with only a few leaves. This quality, the adherence of the kernel, which was once an

objection when flails were in use, has become a recommendation since threshing machines have been introduced. This variety has a specific gravity* 1.312.

This variety has other recommendations: It succeeds in loam soils, bears the frost well, and resists the attacks of the fly. The kernel, from its hardness, or siliceous cuticle, is less injured by fall rains than many other varieties; and hence, too, it may stand in the shock with very little danger of growing. It is still disposed to vegetate soon after sowing, and hence may be ranked with the hardy and vigorous kinds.

- 2. Improved White Flint Wheat. This variety resembles very closely the preceding. It is considered by Mr. Harmon as new; having been produced by himself, by a selection of the best seed, and liming and sowing it upon a limestone soil. It is larger than the White Flint; and yet the cuticle of the kernel is equally thin, delicate and white. It weighs, according to the statement of Mr. Harmon, when prepared for seed, 64 lbs. to the bushel. The specimen in the Agricultural Society's collection has a specific gravity of 1.310, and was furnished by the improver of the White Flint, and hence may be regarded as authentic. The specific gravity, however, is rather less than I should have expected from the weight per bushel. Two bushels and eighteen pounds of this wheat produced 106.8 lbs. flour and 31 lbs. of bran: loss ½ lb., equalling in the whole 138 lbs.
- 3. White Provence Wheat (Pl. XXVIII, fig. 7). This is a French variety, and is regarded as one of the finest kinds of wheat. It is without beards, and has a large white kernel with a thin skin. It grows rapidly, has larger blades, and sends out a greater number of straws from a root than most varieties. The straw, however, is weak, and does not support itself well. Specific gravity, 1.297. From its low specific gravity, I infer that it weighs less to the bushel than the White and Improved Flints. Specimens in the Society's collections.
- 4. Old Red-chaff Wheat. The Old Red-chaff has been cultivated nearly half a century. It has been one of the most esteemed kinds under cultivation in this State. It is free from a beard or awns; has a long straw and slightly brown head, which arises from the color of the chaff: its flour is, however, white. It weighs from 60 to 63 lbs. to the bushel. Specific gravity, 1.313. It stands up well; but of late, it is remarked by Mr. Harmon, that it has suffered from rust and mildew, and has been injured by frosts or winterkills especially on old lands.
- 5. Kentucky White-bearded, Canada Flint, Hutchinson Wheat. In Western New-York, it has become a favorite variety. It is considered as less valuable than the White Flint, by Mr. Harmon. The bran, he says, is thicker. It spreads but little, and therefore requires more seed. This, however, can not be regarded as an objection to the wheat. Its

^{*}The true weight of wheat is determined by its specific gravity. The weight of a bushel of wheat will vary with the size of the kernel, and from other circumstances; while its relative weight, or that found by comparing it with an equal bulk of water at a given temperature, depends upon its composition. The heavy varieties, or those with a high specific gravity, contain more gluten than the light: the latter contain the most starch.

straw is strong; and hence, on rich loamy lands, it will succeed better than those with a weaker straw. The straw too having more substance, the grain matures or fills out after it has been cut.

- 6. Indiana Wheat (Pl. XXVIII, fig. 12). It was introduced from Indiana. It is awnless, with a large white kernel; cuticle thin; weight per bushel, 64 pounds. It ripens a few days earlier than the White Flint, but it shells out easily when ripe. It has yielded 33 bushels to the acre, and is adapted to strong soils. Specific gravity, 1.334.
- 7. Velvet-beard or Crate Wheat (Pl. XXVII, fig. 12). It has been cultivated about twenty years in Western New-York. It has long awns, a red chaff and kernel. It requires a strong soil; weighs from 60 to 64 lbs. to the bushel; flour yellowish. It is said to be inferior to many of the varieties now cultivated in New-York.
- 8. Soule's Wheat (Pl. XXVIII, fig. 9). Mr. Harmon regards this variety as one made of the Old Red-chaff and White-chaff, bald. It is spoken of as an excellent kind; and in some parts of Western New-York, it is the kind most esteemed. Its kernels are large and white, and yield a superior white flour. Specific gravity, 1.333.
- 9. Virginia White May Wheat (Pl. XXVII, fig. 2). It resembles the White Flint. It appears to have deteriorated by culture in New-York. It ripens six or eight days earlier than the White Flint, and has not been injured by rust.
- 10. Wheatland Red Wheat (Pl. XXVIII, fig. 3). This is another variety which has been brought out by the skill of Mr. Harmon, from the preceding kind. Its chaff is red; head bald, and of a medium length. It is said to weigh 66 lbs. to the bushel. Its specific gravity is 1.321. The objection to this kind is its red berry: its recommendation is that it does not rust.
- 11. Tuscan Bald Wheat. This kind, which was introduced from Tuscany in 1837, has been laid aside in consequence of its liability to be injured or destroyed by frost. Its flour is fine and white, and its heads well filled.
- 12. Mediterranean Wheat. The recommendations which this kind bears, are, that it is early, heavy, and escapes all disasters, the fly not excepted. It is objectionable from its dark color and inferior flour. It is heavy, having a specific gravity of 1.360. It is bearded; shells very easily, if it is suffered to stand until it is ripe. It is not cultivated in Western New-York, where the better kinds succeed well.
- 13. Valparaiso Wheat (Pl. XXVIII, fig. 4). Introduced from Valparaiso. Without awns; seed white, cuticle thin, and resembles the White Flint.
- 14. Skinner Wheat. With awns; chaff white; straw short and stiff; weight 64 lbs. to the bushel. It is not in so much esteem as to displace other kinds.
- 15. Golden-drop Wheat. Awnless, with a red chaff and rather thick cuticle. It is inferior to other well known kinds in Western New-York.
- 16. White Blue-straw Wheat. This kind has been received from Maryland, at the Society's room. It is a beautiful kind, and yields a white and fine flour. Specific gravity, 1.344; with the cuticle removed, 1.379. It is worthy of observation that the specific gravity is increased by the removal of the cuticle.

- 17. Aguira Wheat. This kind was brought, two or three years since, from Spain, by F. Townsend, esquire, of Albany. It is a very beautiful kind, the kernel being large and white. Specific gravity, 1.394. Its weight approximates more closely to the celebrated English kinds, than any of the preceding.
- 18. Verplanck Wheat (Pl. XXIX). In richness of appearance, this wheat excels most others which I have noticed. Its kernel is very large and white; the head long, large, and well filled. The straw is large, and tall in proportion, being at least four and a half feet. The grain, however, is light, as will be seen from its low specific gravity, which only attains 1.261. I am not informed how it is regarded by the wheat-growers of Western New-York.

B. SPRING WHEAT.

- 1. Italian Spring Wheat (Pl. XXVII, fig. 18). This kind, which at first was esteemed, has so far deteriorated as to be neglected.
- 2. Tea Wheat, Siberian Wheat (Pl. XXVII, fig. 3). As a spring wheat, it is regarded as a very good variety; giving a white berry, and fine white flour. It is not subject to rust.
- 3. Black-Sea Wheat. The advantages arising from the culture of this wheat, are, that it escapes the fly, ripens early and rarely mildews. Its disadvantage is that it yields a dark flour of an inferior quality. Its specific gravity is 1.341. In Vermont, Massachusetts and Maine, it is often sown, as it is less liable to a failure than the finer varieties.
- 4. Rock Wheat (Pl. XXVIII, fig. 1). This is a Spanish wheat, and has been cultivated more than forty years. The chaff is white, but the berry is red. Where the finer kinds are uncertain, this has been cultivated with success.
- 5. Black-bearded Wheat (Pl. XXVII, fig. 14). Awns long and stiff; heads heavy; straw large, and berry red and large; hardy.
- 6. Red-bearded Wheat. Awn red, and standing out from the head; kernel white; chaff red. Yields a good flour. A bushel weighs from 60 to 62 pounds. It succeeds best on stiff clay loams. It has yielded 44 bushels to the acre. Its beard is objectionable.
- 7. Scotch Wheat (Pl. XXVIII, fig. 11). Its origin is unknown. Berry large, and resembles the Indiana; straw large.
- 8. Egyptian Wheat, California Wheat. The peculiarities of this kind are, that it has a large branching head, as many as six or seven branches, bearded; berry small, with a thick cuticle or bran. Its flour has a harsh coarse feel and a yellowish color, which resembles that of barley. It has not met with much favor.
- 9. Talavera Wheat (Pl. XXVII, fig. 16). Awnless; chaff white; straw long, white and stiff; heads large, long and well filled. Specific gravity, 1·306. It is not sufficiently hardy to stand severe winters. It is frequently injured by the fly.

Additional varieties of wheat which have been somewhat cultivated in this State.

- 1. Velvet-chaff Bald (Pl. XXVII, fig. 4). Chaff greenish brown and dotted, without beard or awns.
- 2. Wheatland Yellow (Pl. XXVII, fig. 5). Chaff pale yellow, with short awns; heads large and berry large.
 - 3. Virginia Blue-straw (Pl. XXVII, fig. 6). Chaff yellowish, without a beard; head small.
 - 4. Hume's White (Pl. XXVII, fig. 8). Heads rather long and slender; chaff yellow.
- 5. Bearded Baltic (Pl. XXVII, fig. 7). Heads thick and heavy; chaff yellowish brown, bearded; beards moderately long.
- 6. Skinner's Club (Pl. XXVII, fig. 9). Kernels clustered in whorls; chaff greenish yellow, bearded.
- 7. Old Bearded Tuscany. Kernels clustered, and with long beards, greenish yellow; heads rather long.
- 8. Pale Red-chaff (Pl. XXVII, fig. 11). Kernels thickly clustered; chaff pale brown, beardless; heads rather long.
 - 9. Baltic Downy (Pl. XXVII, fig. 13). Chaff brown, quite downy; heads long, beardless.
 - 10. Old Black Bald (Pl. XXVII, fig. 14). Kernels irregularly clustered; chaff brown, bearded.
- 11. Poland White Bald (Pl. XXVII, fig. 15). Berry irregularly clustered; chaff greenish yellow, awned, or with shortish beards.
 - 12. New Velvet-chaff (Pl. XXVII, fig. 17). Kernels very thickly clustered, bearded.
 - 13. Black Velvet-chaff (Pl. XXVIII, fig. 5). Kernels closely set and thick; chaff very dark.
 - 14. Bald Baltic. Kernels thickly set in regular rows; chaff light brown; heads thick, heavy.
- 15. Old White-flint (Pl. XXVIII, fig. 10). Kernels clustered; chaff yellow, beardless; heads rather short.
 - 16. Early Velvet-beard. Kernels clustered in whorls; heads long and yellow.
- 17. Italian Spring Wheat (Pl. XXVIII, fig. 14). Kernels clustered, irregularly arranged upon the spike; chaff greenish yellow, thickly bearded.
 - 18. Bearded Valparaiso. Kernels in rows, regularly arranged; head short and thick, bearded.
- 19. Washington Wheat (Pl. XXIX). Heads very large and long; chaff brown; beards long; berry rather dark, but numerous, amounting to 70 or 80.
 - 20. Verplanck Wheat. Heads quite large and beautiful; berry of the largest size.
- 21. Club Wheat, Pennsylvania Wheat (Pl. XXIX). Heads short; kernels in regular rows, bearded.
- 22. Spring Red-chaff (Pl. XXIX). Kernels clustered; heads long; chaff reddish brown, bearded.
- 23. Spring Wintington Wheat (Pl. XXIX). Kernels thickly set, but irregular and large; chaff yellow, bearded.

ANALYSIS OF WHEAT, INCLUDING ITS STRAW AND CHAFF.

Many difficulties exist in the analysis of the grain of the cereals, and particularly in wheat and indian corn. In consequence of this fact in part, I regret that I am unable to give a full account of the composition of the former. But this is not all. I have been poorly supplied with samples of the grain; and not living in a wheat district, I have been unable to procure it, either in a ripe condition, or in its different stages of growth. I made repeated applications, both to the members of the Agricultural Society, and to other individuals, but only in two or three instances have my applications been successful. I availed myself, however, of several fine samples of wheat in the Society's Collection, which were furnished by Mr. Harmon, who has been referred to already. These, although the straw was in sufficient quantity for analysis, the grain itself was insufficient in amount to answer well that object. I have, however, made as good a use of the means within my reach, as I was able; and I propose now to enter upon the details, as far as I am able at the present time, expecting that farther time and opportunity may be given to extend my examinations of this important subject, the results of which will in that case appear at the close of the volume.

I. WINTER WHEAT FROM GENESEE COUNTY.

	Receive	ed fr	om Mr	. Рет	ERS.	The	variet	y not g	iven.
	Specific g	grav	ity	-	-	-	-	•	1.289
				PRO	PORTIC	ons.			
	GRAIN	-	-	-	-	-		-	$100 \cdot 000$
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	1.450
	STRAW	-	-	-	-	-		-	$100 \cdot 000$
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	2.660
	CHAFF		-	-	-	-	-	-	$100 \cdot 000$
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-		-	7.970
From these pro	portions, I	obt	ained	froni	the	ash o	f the	grain	,
	Silica -	-	-	-	-	-	-		0.075
	Phosphates	-	-	-	-	-	-	-	0.810
From the straw	7,								
	Silica -	-	•	-	-	-	-	-	1.285
	Phosphates	•	-	-	-	-	-	-	0.070
From the chaff	,								
	Silica -	-	-			•		•	$6 \cdot 435$
	Phosphates	-	-	-	-	-	-	-	0.080

The phosphates were obtained by precipitation by caustic ammonia, and hence the full amount of phosphoric acid does not appear in the grain.

1. Analysis of the ash of Mr. Peters's winter whea	at.
--	-----

Effervesces slightly on the a	addition	of	acid.
-------------------------------	----------	----	-------

Sand -	•	-	•	•	•	•		$3 \cdot 525$
Silicic acid		-	-	-	•	-	-	1.700
Phosphoric	acid	with j	part o	f the	magn	esia	-	60.725
Lime -	•	-	-	•	-	-	-	0.050
Magnesia	•	-	-	•	•	122		$2 \cdot 880$
Potash -	•	•	-	-	-	•		7.180
Soda -	-	-	-	•	-	-	-	16.920
Sodium	-	-	-	-	•	•	-	0.195
Chlorine	•	-	-	•	-		-	0.295
Sulphuric :	acid	-	۰	٠	-	-	-	0.895
Organic ac	ids		-	-		•	•	$2 \cdot 400$
Carbonic a	cid no	t det	ermin	ed.				

96 · 775 S

2. Organic analysis.

100 grs. gave as follows:

Starch -	•		•			-		$61 \cdot 400$
Albumen	-	•	•	-		-	•	1.215
Gluten	-	•	-	-		-		$4 \cdot 460$
Casein	-		•	۰	•	-	٠	trace.
Matter diss	olved	lout	of e	pidern	nis a	nd ot	her	
bodies in	solub	le in	wate	er an	d alc	ohol,	by	
acetic aci	d		-	-	-	-	-	1.980
Matter diss	olved	lout	of c	pidern	nis aı	nd ot	her	
bodies in	solub	le in	water	, alco	hol ar	nd ac	etic	
acid, by	a we	ak so	lutior	of c	austic	pota	sh:	
comports	itself	like	albur	nen	-	-	-	1.480
Epidermis a	fter o	digest	ing ir	alcol	nol, ac	cetic a	acid	
and potas	h	-	-	-	•	•	-	3.410
Dextrine	•	•	•	•	-	-	-	2.400
Water -	9	-	-	-	•	-	•	9.380
Oil -	-	-	-	-	-	-	-	1.050
Extractive 1	matte	r and	suga	r, and	d loss	-	•	$13 \cdot 225$
								100:000 S.
								TOUTHUU S.

This analysis is not complete: the extractive matter and sugar were not obtained.

PROPORTIONS.

Water	•	•	•	•		۰	-		9.380
Dry mat	ter	•	-		-	4	-	-	90.620
Ash		-	-	-		-	-	-	1.650
Ash calc	culate	ed on	dry	matter		-		•	1·821 S.

II. BLACK-SEA WHEAT FROM LEWIS COUNTY.

Soil slaty, being based upon the Utica slate. 1. Analysis of the ash.

				zrnacy	363 01	1116	wore.		
									Removed from an acre.
Silica ·	•	•	-	-	•	-	•	4.300	0.970 lbs.
Phosphates o	f lin	ae, m	agnes	ia and	iron	•		$45 \cdot 376$	$10 \cdot 240$
Phosphate of	the	alkal	lies		-			$28 \cdot 195$	$6 \cdot 363$
Potash -	•		-	•		-	-	10.830	$2 \cdot 444$

 $98 \cdot 221$

Phosphate of the alkali Potash -Soda -8.110 1.8300.0100.002Lime -0.0200.004Magnesia Organic matter Carbonic acid 0.3011.340

2. Analysis of the earthy phosphates.

							Per centum.
Soluble silica •	•	-	•	-		0.003	0.074
Lime	•	•	-	-	-	1.940	2.380
Phosphate of perc	oxide of	iron	-	•	-	1.880	$4 \cdot 470$
Magnesia		-	•	-	•	2.920	$12 \cdot 440$
Phosphoric acid	•		*	•	-	12.825	30.760

III. BLACK-SEA WHEAT FROM THE SAME COUNTY.

Specific gravity 1.341. Kernel small, and but little lighter colored than the best kinds of rye. Soil based upon limestone.

Analysis of the ash.

Sand -	ь	•	•	•	-	•	•	3.700
Silicic acid	-	•		-	•		-	1.550
Phosphoric	acid	with 1	part of	f the	magn	esia	-	$62 \cdot 075$
Lime -	-	•		-	•	•	-	0.050
Magnesia			•	•	•	•	-	$3 \cdot 435$
Potash -	•	-		•	•	-		$8 \cdot 045$
Soda -	-	-	•	-	-	•	•	14.790
Sodium	-	-	•	•	-	•	6	0.320
Chlorine	•	-	•	-	•	è		0.490
Sulphuric a	cid	•			•	-	-	0.340
Organic aci	ds	-	•		•		-	$2 \cdot 000$
Carbonic ac	id no	t dete	rınine	d.				

96·795 S.

 $22 \cdot 154$

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Effervescence very slight on adding acid to ash.

IV. ANALYSIS OF SUMMER WHEAT.

Received from Mr. PETERS of Genesee county.

										Removed from the acre.	
Silica ·	•	-	•	•	-	-	•	•	2.633	0.687 lbs.	
Sand ·	•	-	-	-	•	-	47	-	1.607	0.419	
Phospha	ites o	f	lime, n	agne	esia aı	nd iron	-		48.000	12.528	
Phospha	ates o	of	the alk	alies	-	-	•	•	19.440	5.073	
Lime an	nd m	ag	nesia	•	•	•	•	•	0.020	0.005	
Potash ·	ď	-	•	•	-	•	•	-	14.720	3.841	
Soda ·	•	-	•	-	•	•	•	٠	$3 \cdot 356$	0.875	
Chlorine	9	•	-		•	-	•	-	none.		
Sulphur	ic ac	id	-	•	-	•	•		0.544	0.141	
Organic	mat	tei	-	•	-	•	•	•	8.480	$2 \cdot 213$	
									$98^{\circ}864$	25 · 782	

Percentage of water of Black-sea Wheat grown on different soils.

On limestone -	-	-	•	•	-		10.52
On slate	•	-	-	-	•	•	10.72
On alluvial gravel	-	-		-	•	-	10.27
On sandy soil -	-	•	•	•	-	-	11 · 10

The variety known as *Harmon Wheat*, grown upon clay loam based upon the rocks of the Salt group, gave water 11.82, after long drying in the water bath. The last had assumed a brown color, and appeared partially charred, although it had never been exposed to a temperature above 212° Fahr.

From the preceding observations, and others of the same kind, I am led to believe that this grain has always in combination about the same quantity of water, and that soil and varieties do not cause it to vary much either way from 12 per centum of water. This amount of water, however, although it is comparatively small, has probably a decided influence upon its preservation in transportation to foreign countries. The hygrometric power of grains and flour has not been determined. The percentage of water may not of itself form an obstacle to its keeping; and if it is not in a situation to imbibe more, it may perhaps remain for years in a sound state.

V. BLACK-SEA WHEAT FROM LEWIS COUNTY.

Grown upon the Trenton limestone.

Analysis of the ash.

Silica and sand .		•	-	-	-	-	14.520
Earthy phosphates		-	-	-	-	-	$43 \cdot 333$
Alkaline phosphates	S	-	•	-	•	-	23.646
Potash		-	•	-	-	-	12.629
Soda		-	•	-	•	-	5.068
Magnesia and lime		-	-	-	•	-	0.030
Chlorine		-	•		•	-	trace.
Sulphuric acid -		•	-	-	-	-	trace.
Carbonic acid -		-		-	-	-	none.

VI. A WINTER WHEAT FROM THE SAME COUNTY.

Grown upon sandy soil. Variety not given. Furnished by Mr. BEACH.

1. Analysis of the ash.

Silica - ·		-	-	-	-	-	$9 \cdot 120$
Sand and coal		-	-	-	-	-	10.000
Earthy phosph	nates	-	-	-	-	•	$48 \cdot 273$
Alkaline phos	phates	3 -	-	-	-	-	15.501
Potash -			-	-	-	-	$23 \cdot 407$
Soda -			-	•	-	-	$4 \cdot 044$
Lime - ·		_	-	-	-	-	0.020
Magnesia -		. <u>-</u>	-	-	-	-	0.002
Sulphuric acid	i -	-	-	-	-	-	trace.
							${100 \cdot 367}$

2. Analysis of the earthy phosphates.

Soluble silica	a -	-	-	-	-	-	-	0.08
Lime -	-	-	-	-	-	-	-	1.98
Phosphate of	` pero	oxide	of iro	n -	-	-	-	4.95
Magnesia	-	•	-	-	-	-	-	6.64
Phosphoric a	cid		•	-	-	-		$28 \cdot 31$

VII. WINTER WHEAT FROM THE SAME COUNTY.

Furnished by Mr. BEACH. Grown upon a gravelly soil.

A_{i}	nalysi	s of t	he ash	2.		
Silica and coal -	-	•	•	-	-	$12 \cdot 134$
Earthy phosphates	-	-	-	-	-	$37 \cdot 072$
Alkaline phosphates	-	-	-	-	-	21.313
Potash	-		-	-	-	$22 \cdot 496$
Soda	-	-	-	-	-	7.348
Chlorine	-	-	-	-	-	trace.
Sulphuric acid -	-	•	-	-	-	trace.
Magnesia and lime	-	•	-	-	-	0.031

Note. I was desirous of repeating all those analyses in which so much foreign matter, as coal and sand, existed. Experience subsequently enabled me to avoid this objectionable state of the ash; still, the results are correct for all the elements except silica. In regard to this, I have been satisfied that it varies from 1.50 to 5 per centum; and it is probable, in those varieties grown upon soils of Lewis county, that they reach the maximum percentage. The grain has a thick cuticle, and is rather dark; and it is in these kinds that the silica is in the largest proportions.

VIII. STRAW AND CHAFF OF WHEAT FROM MR. PETERS.

		1.	Anal	ysis oj	f the	strai	v.	
Silica -	-	-	-	-	-		49.100	Removed in a ton of straw. $29 \cdot 255$
Earthy phosphat	es	-		-	-	-	19.600	11.678
Lime -	-	-	•	•	-	-	3.460	2.061
Magnesia -	-	-	-	-	-	-	0.324	$0 \cdot 193$
Potash -	-	-	-	-	•	-	$22 \cdot 245$	$13 \cdot 253$
Soda -		-	-	•	-	-	$5 \cdot 195$	$3 \cdot 095$
Sulphuric acid	-	-	-	-	-	-	0.876	0.521
Chlorine -	-	-	•	-	-	-	0.121	$0 \cdot 072$
							100.921	60.128
		2	. Ana	lysis (of the	cha	ff.	
Silica -	-				-	•	80.60	Removed in a ton of chaff,
Earthy phosphat	~							14 3 ·893
	es	•	-	-	•	-	8.80	143·893 15·710
Carbonate of lin		-	•		-	-	_	
Carbonate of lin		-	-	-	•		8.80	15.710
			-				8.80	15.710
Carbonate of lin Magnesia -							8·80 4·70	15·710 8·390
Carbonate of lin Magnesia - Potash - Soda -			•	•			8·80 4·70 1·80	15·710 8·390 3·213
Carbonate of lin Magnesia - Potash -			-		-		8·80 4·70 1·80 3·20	15·710 8·390 3·213 5·712

Proportions of grain, straw and chaff of several varieties of wheat.

1.	$Old\ H$	Red- ch	aff W	heat					
			J.					Actual quantities.	Per centum.
Grain	•	-	-	-	•	•	•	724 grs.	$100 \cdot 000$
\mathbf{Chaff}	•	-	-	-	•	-	-	221	30.524
Straw	•	•	-	-	-	-	-	1154	$^{^{\prime\prime}}159\cdot392$
2.	Talar	vera I	Wheat.						
Grain	-	-	-	-	•	-	-	1240	$100 \cdot 000$
\mathbf{C} haff	-	-	-	-	-	-	-	292	$23 \cdot 548$
Straw	-	-	-	•	-	-	-	1444	$116 \cdot 209$
3.	India	na W	heat.						
Grain	-	-	-	-	-			$556 \cdot 50$	100.000
Chaff	-	-	-	-	-	-		$129 \cdot 50$	$23 \cdot 270$
Straw	-	•	-	-	-	•		611.00	109.811
4.	Impro	ved F	lint V	Vheat					
Grain	•	-	•	-	-	-	-	1130	100.000
Chaff	-	-	-	•	•	-	-	272	$24 \cdot 070$
Straw	-	-	-	•	•	-	•	1323	$117 \cdot 079$
5.	Harm	on W	heat.						
Grain	•	-	-		-	-		1207.50	100.000
Chaff		-	-	-	-	•		$300 \cdot 00$	24.844
Straw	-	-	-	-	•	-		$1166 \cdot 50$	$96 \cdot 604$

To determine the foregoing proportions of grain, etc., I took from a small bundle those heads and straw which remained perfect, a certain number, and shelled the grain, and weighed each part by itself. This method of determining the proportions of grain, chaff and straw, has been found as correct, if not more so, as weighing large quantities in the usual way. Due care must, of course, be taken to avoid losses in separating the grain.

IX. IMPROVED WHITE-FLINT WHEAT.

		1	Analys	sis of	the st	traw.			
Silica	•	-	-	-	-	•	•	•	$42 \cdot 60$
Carbona	te of	lime	-	-	-	-	-	-	8.90
Phospha	tes of	lime	, mag	nesia	and	iron	-	-	9.30
Potash	•	-		-	-	-	-	-	22.76
Soda	-	-	-	•	-	-	-	-	5.28
Magnesi	a	•	•	-	-	•	-	-	1.58
Sulphuri	c acid	l	-	-	-	•	-	-	5.85
Chlorine		-	-	-	-	•	-		1.86
									98 · 13

X. OLD RED-CHAFF WHEAT.

Analysis of the straw.

	279									Removed in a ton of straw.	
Silica	-	-	-	-	•	•	-	•	70.00	78·40 lbs.	
\mathbf{Coal}	-	-	-	-	-	-	-	-	0.25	0.58	
\mathbf{P} hospl	hates o	f lime	e, mag	gnesi	a and	iron	-	-	8.89	$9 \cdot 95$	
Carbon	nate of	lime		-	-	-	-	-	1.80	$2 \cdot 01$	
Magne	sia	-	-	-	-	•	-	-	$0 \cdot 15$	0.16	
Potash	_	-	-	-	-	-	-	-	$12 \cdot 12$	13·5 7	
Soda	-	-	-	-	-	-	-	-	$4 \cdot 19$	4.69	
Sulphy	ıric aci	id ·		-	•	-	-	-	$2 \cdot 25$	$2 \cdot 52$	
Chlorin	ne	-	-	-	-	•	-	-	1.75	1.94	
								-	101:50	113.52	
									101 90	110 02	

The straw of the Old Red-chaff is stiff and rigid; and from its characters alone, it would be inferred that it contained a greater percentage of silex.

XI. WHEATLAND RED WHEAT.

Analysis of the straw.

									Removed in a ton of straw.
Silica -	-	-	-	•	-	•	-	$75 \cdot 75$	84.84 lbs.
Phosphates	-	-	-	-	-	-	-	8.21	$9 \cdot 19$
Carbonate o	f lin	ne	-	-	-	-	-	$1 \cdot 05$	1.17
Magnesia	•	-	-	-	-	-	•	$0 \cdot 25$	0.28
Potash -	•	-	-	-	-	-	-	$7 \cdot 20$	8.06
Soda -	-	-	-	-	-	-	-	$2 \cdot 10$	$2 \cdot 35$
Chlorine	•	•	-	-	-	-	-	0.24	0.26
Sulphuric ac	$^{ m cid}$	-	•	•	-	•	-	$2 \cdot 21$	$2 \cdot 47$
								97.01	$108 \cdot 62$

XII. SOULE'S WHEAT.

Specimen taken from the State Agricultural Rooms. Fine plump berry.

Starch	•							62 · 29	Calculated on dry matter. 68.360
Sugar and	extra	ctive	matt	er, wi	th a l	ittle a	acid		
formed d	luring	the	analy	sis	•	-	-	$6 \cdot 40$	$7 \cdot 023$
Dextrine of	r gum		•	-	•	۰	•	1.21	1.328
Epidermis	•	-	-	•	•	-	•	$7 \cdot 20$	$7 \cdot 903$
Matter dis	solve	l ou	of e	pider	mis a	nd of	her		
bodies in	solub	le in	water	and b	oiling	alco	hol,		
by a wea	ak sol	utior	of ca	ustic	potas	h -	a	6.82	7.485
Oil -	•	•	•	-	•	-	-	$1 \cdot 02$	1.119
Gluten	-	-	•	•	•	۰	-	4.51	4.949
Albumen	-	-	-	•			۰	1.67	1.833
Casein	•	-	•	•	•	•	•	trace.	trace.
Water	•	•	•	-	٠	٠	•	9.79	
								100·91 S.	100.000

The gluten in the above analysis is small, though I think correct. The matter insoluble in water was digested in successive portions of boiling alcohol for six hours, till nothing more was taken up. The matter insoluble in water and boiling alcohol was digested in a weak solution of caustic potash, which took up over 7 per centum of the dry grain; which, if albumen, increases that body to a large percentage. The gluten and starch agree nearly with the winter wheat from Genesee, but the albumen and epidermis are much greater.

		PROI	PORTIO	NS.				
Percentage of	water	•	٠	•	•	•	9.790	
"	dry ma	tter	•	•	-		$90 \cdot 210$	
"	ash	-	•	•	-	۰	1.720	
66	ash cal	culate	ed on	dry n	atter		1.906 S	_

XIII. PROVENCE WHEAT.

	Analys	is of	the	straw.
--	--------	-------	-----	--------

Silica -	-	•	•	•	•		۵	$68 \cdot 60$
Phosphates	-	•		•	•		-	4.70
Carbonate of	lime	•	•	ø	•	•	•	$2 \cdot 35$
Magnesia	•	•	•	•	•		•	$1 \cdot 35$
Potash .	•	•	-	-	-	•	•	5.55
Soda -	•	-		•	•	-	6	5.63
Sulphuric acid	ĺ	-	•		•	۰	•	2.83
Chlorine	•	•	-	-	•	•	•	$1 \cdot 34$
Organic matte	r	•	-	•	•	-	-	$4 \cdot 20$
Carbonic acid		•	٠	-	٥	•	**	1.40
								$97 \cdot 95$

XIV. HOPETON WHEAT*.

Length of straw, 44 inches.

1.	Relation	of	grain,	straw	and	chaff.

	1.	Kela	tron (y gr	aın,	strau	an	a cna	U•	1		
								Actu	al quantities.	Perc	entage.	
Grain	•	•	•	•	•	•		•	1207	42	5.30	
Straw	•	•	•	•	•	•	•	•	1423	49	9.86	57.70
Chaff	-	-	-	•	•	•	•	•	224	7	·84 \	01 10
	\mathbf{s}_{p}	ecific	grav	vity o	f the	grai	n	•	• •	1.39	91	
	2.	Per	centa;	ge of	wate	r and	l ash	ı .				
								Wat		Ash.		alculated di
Grain	•	-	•	-	•	•	-	12.	_	$1 \cdot 76$		2.01
Straw	•	•	•	•		-	•	् 13•	7	$4 \cdot 16$		4.82
Chaff	•	-	-	-	•	•	•	11.	5 10	0.36	1	1.70
		n	7	7	•	,		c				
	3.	Proc	iuce,	ana	mine	rai m	atte	rof	an acre.			
Grain										luce. • 16		eral matter $43\!\cdot\!5$
Straw	•	•	•	•	•		•	•		•94		20.4
Chaff	•	•	•	•	•		•	•		•11	1	57·6
Chan	•	ŭ	•	•	Ì	-	•	•	• •	11		0, 0
	4.	Ana	lusis	of th	e ash	of t	he g	rain.				
			•			,	Ü			Rei	noved fr	om an acre
											lbs.	oz.
Silica	•		•		•	•	•	•	$3 \cdot 20$		1	6.6
${f P}$ hosph				•	•	•	•	•	44.44		19	6.0
Sulphu			•	-	•	•	•	•	trace.			
Carbon	ic ac	id .	•	•	•	•	•	•	none.			
Lime	•		•	•	•	•	•	٠	8.21		3	$9 \cdot 2$
Magnes	sia •			•	•	-		•	$9 \cdot 27$		4	$3 \cdot 3$
Peroxid	le of	iron		-	-	•	•	-	0.08		0	0.9
D . 1	•		•	-		-	-	-	$32 \cdot 14$		17	13.8
Potash			_	-	-			•	2.14		1	8.8
	-	•	•									
Potash Soda Chlorid	- le of	sodi	um	-	•	•	•	-	none.			
	- le of	sodi	um	-	•	•	•	•	none.		43	9.7

^{*} Journal of the Royal Agricultural Society.

	5.	Analy	isis of	the s	traw	and chaff.	
			J			ı.	Removed from an acre. lbs. oz.
Silica	-	-	-	-	-	67.10	119 6.8
Phosphoric acid	-		-	-	-	6.05	12 8.7
Sulphuric acid	-	-	-	-	-	5.59	$91 \ 5 \cdot 2$
Lime	•	-	-	-	-	$4 \cdot 44$	7 14.4
Magnesia -	-	-	•	•	-	$3 \cdot 27$	5 13.0
Peroxide of iron	-	-	-	-	-	1.54	2 11.8
Potash	-		-	-	-	$10 \cdot 03$	17 13.6

0.85

99.97

1 8.6

177 11.5

The foregoing extract, exhibiting the proportions of water, grain, composition, etc. of an English variety of wheat, has been copied for the purpose of comparison with wheat of New-York growth. A comparison can be made by any person who feels an interest in the matter. I do not, therefore, propose to enter upon a detail of difference or similarity; observing, however, that in the statement respecting the phosphates and phosphoric acid, I have given the phosphates of the earths and phosphates of the alkalies, by which it will be perceived that the earths, the lime and magnesia, as well as iron, are in combination with phosphoric acid. This fact does not appear in the extract which is given.

The real composition of wheat appears only when an analysis is made of its parts, as bran (which is the cuticle), and its flour. Time, however, has not permitted me to make those analyses. I can therefore make only the following very brief statement:

Shorts, which is mostly a coarse bran, gives

Soda

Ash - - - - 5:115 per centum; which contains

Silica - - - - 0.140 Phosphates of magnesia, lime and iron, 2.380

Fine middlings lost in a water bath - 12.78 of water;

Bran - - - - 12.37 water, which proportions are rather greater than that given by wheat.

The specimen of winter wheat furnished by Mr. Peters - 9.72 water.

Summer wheat - - - 9.62

Proportion of ash and water in the straw of four varieties of wheat.

									Mineral matter in a ton of straw.
	Indiana, water	-	-	-	-	-	-	$8 \cdot 50$	
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	4.40	99·90 lbs.
	OLD RED-CHAFF,	wate	er	-	•	-	-	7.50	
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	•	-	-	$5 \cdot 22$	117.60
	IMPROVED WHIT	E-FLI	NT, V	vater	-	-	-	9.50	
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	•	-	-	-	4.50	160.80
	TALAVERA, wate	r	-	•	-		-	8.00	
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	$5 \cdot 46$	$122 \cdot 30$
[Agri	CULTURAL REPORT	<i>I</i> — <i>I</i>	ol. 1	1.]	20)			

DISEASES OF WHEAT.

Wheat or Smut Brand (Uredo sitophila, Dittmar). Among all the species of brand which infest our grain crops, this is by far the most worthy of notice. It lodges only in the ears of wheat, and is found in no other kind of grain or grass. It migrates with wheat in all climates of the earth, without being subject to local influences, as is almost ever the case with the other cultivated plants. The farmer dreads it most, and justly; for being lodged in the ears when they are brought to be threshed, it is there dispersed by the flail or threshing machine, and thus directly infects the sound grain; while the barley and oat brands are for the most part out on the field, and hence the largest portion of the seed of these latter kinds of brand necessarily falls on meadow, forest, or other kinds of soil, which are not applied to the cultivation of grain, and so, for the want of plants adapted to the infection, are not further spread.

Those halms of wheat which afterwards bear ears affected with brand, may be early distinguished, before their bloom, by their luxuriant growth and their dark green color, as well as by their large, broad, stiff leaves. They apparently bloom much earlier; but very often (yet not always) their anthers contain no grains of pollen (powder of fructification); and the first act of fructification, the shedding of their pollen on the cup of the pistil, is very imperfect; and should the ears affected by the brand and already diseased be dusted with sound pollen, the little balls of pollen usually form no aggregation of pollen on the pistil, or such as are formed do not press into the pistil and down to the ovary. The fructification of the blossoms of wheat affected by brand is therefore imperfect; and in case the grains of pollen form no cluster in the cup, then there is indeed no fructification; but the careful observer finds on almost all the ears ripe for receiving the grains of brand, on the side of every seed corn affected with brand, one or two anthers (Plate LV, fig. 3, 4, b, b), which, on the buds which are well fructified and bear seed, the anthers with their stamens have long since fallen off. The anthers which remain standing on the grains affected with brand are usually destitute of pollen; and sometimes we find that the same, in consequence of efforts at imperfect fructification, are stuck, as it were (fig. 4, b, right side), to the pistils which remain standing (fig. 4, c). The two pistils which remain standing, of the seed affected with brand, are usually covered and joined together by delicate white threads interwoven like mould in a sort of network (Plate LV, fig. 4, c). This mould formation belongs essentially to the wheat brand, and forms as it were a part of the root texture of the fungus constituting the brand. It always exists, only frequently more or less developed, and therefore more or less easily found. It wholly covers the head of the seed, and lies between the chaff-hairs (fig. 16, h), while the sound seed exhibits not a trace of such an interwoven fibrous formation among the chaff-hairs on its head. The same is also the case in respect to the pistils of the sound seed (fig. 16, i). With the development of this outer fibrous mould begins likewise the transformation of the seed, as well in respect to its external form as to its internal structure.

If the transformation which the seed of wheat undergoes by the formation of brand be examined, we find that the particles have undergone either an entire or partial transformation in respect to internal structure; and without here entering on the technological signification of the several parts of the seed, I shall describe the same, in a way generally intelligible, and simply, as should always be done, and so pass over the head as well as the pistil, since they suffer no visible change by the formation of the brand. The same is true of the glumes and petals, the anthers and the spike of the ears themselves.

The fruit or seed of the wheat viewed on the outside, consists of an elongated irregularly egg-shaped body (fig. 16, f), having on the front surface a streak or furrow lengthwise, (fig. 16, k), which bears on the point the pistil (fig. 16, i), and the head (fig. 16, h).

At the bottom, we see on the back the little shield (fig. 15) containing the germ, and the front side the little opening, Feusterchen (fig. 16, g). If the seed is cut across through the middle (fig. 18), we find that it has an outer skin (fig. 18, l), which by bending inward forms the furrow lengthwise (k). Inside of this skin are found white hard transparent bodies containing starch-meal, which natural historians call the albuminous bodies, "the albumen" of the seed (fig. 18, m). If now we cut off as thin as possible a slice perfectly transparent, in the direction already mentioned, and examine the same microscopically, we find that,

- 1. The skin of the seed (fig. 18, l) consists of three layers, to wit:
- a. The outer layer (fig. 19, n);
- b. The middle layer (fig. 19, o); and
- c. The inner layer (fig. 19, p), on which layer immediately lies a large soft cellular stratum, which contains the grains of gluten (fig. 19, q). Nearest the inside lie the amylum cells, containing the starch-meal (fig. 19, r, s).
- a. The outer layer of the seed-skin (fig. 19, n), consists of two layers of thick-walled porous cells, which stand with their longest diameter parallel to the axis of the seed, and the walls of which contain slight hollows or little canals, which, in a section cut lengthwise and very strongly magnified (fig. 20, w), give to the cellular walls a form as if they were formed of oblong figures.
- b. The second layer (fig. 19, o) of the seed-skin, consists of similar cells to those of the first layer, only the walls of the cells are not so thick; and the pores, which these walls contain are much more distinctly (fig. 19, o) to be seen, than is the case in the cellular walls of the first layer of the skin; but the cells of this layer stand with axis of length horizontally to the axis of the first cellular layer and of the seed, and therefore runs as it were parallel to the outer surface of the seed. In a section lengthwise they resemble even to the direction of the cells of the first layer of cells, and are nearly as large as they are (fig. 20. x, x).
- c. The third layer is extremely soft and somewhat confused. Its cells are so small, that we can discern their hollows only indistinctly and in the form of mere streaks (fig. 19, p).
- 2. Directly under this cellular stratum or of the seed-skin in general, we find situated the already mentioned cells of gluten (fig. 19, q). They are large bag-formed cells, with

extremely thin scarcely visible cellular walls, which are filled exclusively with the gluten, a small-grained, greasy, smutty-gray substance, approaching to yellow. Under these cells of gluten lie, first,

3. The albuminous bodies of the seed, which consist of large six-sided prismatic cells (fig. 19, r), the walls of which are soft, clear as glass, and perfectly transparent; and the hollow space in the ripe seed is filled with little grains of starch-meal (fig. 19, s; fig. 22). These latter are round or irregularly egg-shaped, transparent and white, and consist of concentric layers or peels (fig. 22), the outer of which often bursts or springs open. Between the grains of amylum or starch-meal are found still smaller grains which consist almost wholly of starch, and must be regarded as little grains of amylum.

At the base of the seed below the little shield lies the embryo plant or germ; but as the same is scarcely ever found in the bud of wheat which is affected with brand, the consideration of this here does not belong to the province of this essay, since no immediate transforming influence can be referred to it.

If now after the minute examination of the sound seed, we compare with it the structure of that which is affected by brand, we find that the diseased seed (fig. 4) is wholly changed as well in respect to its form as to its structure. It has become shorter and thicker, and not as in the sound seed tapering towards the top (fig. 16), but increased in thickness (fig. 4). On its base, or on the head, the anthers remain hanging or standing, while in the sound seed they have long since fallen off. The head with the pistil (fig. 4, c) is broader, and the outer skin (a) of the seed corn affected by brand is rougher and fine punctured.

Let a seed corn thus affected by brand be cut through horizontally, and it be examined under a magnifying glass (fig. 5), we find outwardly a simple outer skin, and internally a dark black substance often approaching to violet, which is extremely fine grained and greasy, gives out a foul penetrating ammoniacal smell, and on being dried falls to powder. In the middle of the grain affected by brand we generally see a clearer colored square gray spot, which on close examination is found to consist of the remains of the former cellular texture. If now we examine more closely the particular organs of such a kernel affected with brand, we find that the outer skin of the seed thus affected consists of a single stiff layer of cells (fig. 21), the cells of which in respect to their form and size resemble much the outer cellular layer of the seed-skin of the sound seed (fig. 20, w); but their walls are no longer porous, but paper-like, stiff and folded lengthwise: they are not so finely colored as in the sound seed, but are of a smutty earth color. The second and third cellular layer (fig. 19, v, v) of the sound seed, has wholly disappeared in the diseased one: the same is true of the cells of gluten, of which not a single trace remains.

On examining still more minutely the black smutty mass, which fills the space designed for the albuminous bodies, we find that here and there it contains some particles of cellular tissue, like the celluar tissue of the albuminous bodies, but the cells themselves are much widened (fig. 6, d) and folded; but the hollow spaces are filled with grains of brand (fig. 6, e). Should the brand not be fully ripe or developed, we find the cellular tissue still

entirely preserved and connected together, but without any traces of amylum. This latter is scarcely ever developed in diseased seed, but in place of it are formed clear globular cells of the same size (fig. 7), which we instantly distinguish as the young grains of brand. These by form are oily-grained contents (fig. 7), which increases with the advancing growth of the same (fig. 8); and their cellular skin, previously clear as glass, and white, becomes brownish colored. In the later growth we find the entire cells of brand (fig. 9) filled with little oil-drops; and the cellular wall is of pale violet color, but it is still smooth. These cells, natural historians call the spores or seeds of the fungi which constitute the brand; and in the advancing growth the cellular skin, which is the seed skin of the spore, gradually becomes dark colored and covered with fine warts, while at the same time the little oil-drops visibly increase in the space of the spore-skin, and finally flow into a compact yet scarcely discernible body (fig. 10).

But if we thoroughly examine the ripe spores of brand, and we happen to obtain good sections of the same - a problem extremely difficult on account of the minuteness of the body to be cut, and only to be secured by chance — then we see that the spore-skin (figs. 11, 12, t, t) of the brand-spore forms a dark colored single membrane uneven on the outer surface, which encloses in its hollow space a second transparent cell (figs. 11, 12, 13, u, u, u, u), which forms the second or inner spore-skin; but in the space of the second spore-skin we find a waxy, curved body (figs. 11, 13, v, v), which is called the kernel of the spore, and which, in spores not yet fully ripe, appears to be surrounded with little drops of oil. The spores, compared to other of the different kinds of brand, are large, and their linear diameter is from 0.000700 to 0.000730 $\left(\frac{1}{1420}\right)$ of a Paris inch. The spores distinguish this species of brand from all others which habitate wheat, and their specific gravity is greater than that of water: they sink therefore in water, and hence the seed which is affected by brand may be cleaned with running water, as it is thus also clear that well washed seed suffers less from the brand. But the seed must be thoroughly washed before sowing, in order that the spores of the brand, which may still be in the furrow of the seed and among the chaff-hairs of the head, may be removed.

Here is not the place to quote all the various opinions of the husbandmen and natural historians respecting the existence and propagation of the brand in the various kinds of grain generally. The conviction and view of every individual is so peculiar a matter, which rests on such different grounds of representation and positive induction, that opposition to even the crudest ideas (and so-called experience), according to my multifarious observation, is only injurious.

Yet I may be allowed to maintain here as preliminary, that the view which regards the brand merely as a stage of disease, or a disease analogous to the organic diseases of the animals, must indeed be false. I can only compare the parasitic formations which belong to the class of fungi or mushrooms, to the phthiriasis or the louse disease, and in this case no spontaneous generation is supposed. We have one of the most decisive proofs in the case of a majority of exotic plants which are evidently produced from seed, and no parasites (especially eutophytes) have been imported from their native country; while in our glass-

house, all the plants known to me as having been brought alive from the tropic have introduced certain eutophytes peculiar each to its species of plants, and not belonging to this country. The great idea of De Candolle, "the spreading of the species of brand depends on the sowing of the spores," since the beautiful observations which Gleichen published more than sixty years ago, can no more be doubted. This great German natural historian found indeed that the wheat crop strewn and sown with brand dust gave over 50 per cent of ears affected by brand, while the dry and thoroughly washed seed exhibited scarcely any ears affected by brand.

Besides many eutophytes may be transported; and in the kinds of brand of grain we are by no means justified in denying the transmission by spores, and especially as no husbandman can maintain "that he has cultivated wholly clean seed containing not a single brand-spore," for in practice the extraordinary minuteness of the brand-spores lays an insurmountable obstacle in the way of all observations. The parasites which have their abode in the dead parts of plants may easily be propagated by the sowing of their spores; and a careful observer may in this latter case readily follow the germ of the spores sown, and the gradual development of the parasite through all the stages of its formation, as I have already many times shown in other places. But a multitude of eutophytes, besides the sowing by spores, also require peculiar conditions of soil and a moist atmosphere for their development; since otherwise the mother plant is not capable of furnishing the nutrition indispensable for its development, or to perform the secretion of the same from its own fluids.

These organic processes necessary to such formations are yet partially mysteries to natural historians, which may not be laid open by logical phrases, or such as belong to natural philosophy. Only direct observations can here determine; and all views, opinions, belief, and so-called experience are positively injurious, while they are almost ever wanting in any strong induction, and under critical examination sink into their original nothingness. It is therefore the wiser openly to admit that we have not yet observed the direct propagation of the kinds of brand by spores; as we must allow, on a critical investigation of all circumstances, "that the conditions of soil, the influence of cultivation, weather, situation and manure which is required for the spreading of the various species of brand, are not fully known." Such conclusions are more salutary for the advancement of human knowledge, than all the so-called learned or purely empirical talk.

But since Ehrenberg has practically demonstrated the propagation of the infusoria by eggs and division, and I have also the sowing of fungi and mushrooms by spores, we may too hope for a similar proof of the propagation of eutophytes by spores, and until then set aside all speculations on their spoutaneous generation as injurious and unnecessary; and the more so, as nearly every kind of plant has parasites exclusively having their abode in it, and likewise the soil equally necessary to its development.

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Explanation of the Illustrations.

Plate LV, fig. 1, an ear affected with brand of the natural size. Figs. 2 and 3, seed kernels affected with brand, with and without anthers, of the natural size. Fig. 4, a seed kernel affected by the brand, greatly magnified: a, the seed kernel with the fold lengthwise; bb, anthers, which stick to the head of the same, and on the pistils, c, interwoven by fungous fibres. Fig. 5, a horizontal section of same greatly magnified. Fig. 6, a cell of this section, with brand-spores, e, greatly magnified. Fig. 7, an entire young brand-spore. Figs. 8, 9, older brand-spores, before the formation of the kernel and sporeskin. Fig. 10, four ripe brand-spores strongly magnified. Fig. 11, section of a spore very greatly magnified: t, spore-skin; u, inner spore-skin; v, kernel of the spore surrounded with little drops of oil. Fig. 12, a single spore of this kind where the kernel of the spore v, with the inner spore-skin u yet lies in the hollow space of the outer sporeskin t. Fig. 13, a kernel of the spore v, represented alone with the inner spore-skin u, greatly magnified. Fig. 14, a sound, ripe wheat kernel, front view. Fig. 15, the same, back view, with the little shield, natural size. Fig. 16, the same, greatly magnified: f, the seed-kernel, with the seed-skin; k, the fold lengthwise; g, the little opening; h, the head; i, the pistils. Fig. 17, horizontal section of the same through the centre, and fig. 18, slightly magnified: l, the seed-skin; k, the fold; m, the albuminous bodies. Fig. 19, a thinner section of the wheat kernel strongly magnified: n, the cellular layers of the first seed-skin; o, the second; p, the third or innermost seed-skin; g, the cells of gluten; r, the cellular tissue of the albumen with grains of starch-meal, s. Fig. 20, external view of a very delicate vertical section of the seed-skin of a sound kernel of wheat strongly magnified: w, outer cellular layer; xx, inner cellular layer of the same greatly magnified. Fig. 21, the outer skin of a wheat kernel affected with brand of the same ear, strongly magnified. Fig. 22, little grains of amylum very strongly magnified.

VI. MAIZE, OR INDIAN CORN.

American farmers regard indian corn as one of the most important crops; indeed, for general usefulness, it must be regarded as inferior to none. For family use, it occupies a prominent place. Although in the state of society it is not used so much for bread as wheat, still there are so many modes of cooking it, by which dishes are improved, that it is nearly indispensable. For domestic animals, it is far more important than wheat. Its oil, and other elements place it among the most fattening of the cereals: thus its parts, as the leaves and stalks, furnish a fodder superior to straw. Indeed there is really nothing in the crop which may not be turned to account: even the cob should not be thrown away. Were the sugar cane to become extinct, the stalks of indian corn would become a substitute, and sugar would be one of the essential products of the corn crop.

For family use, and for domestic animals, corn is almost indispensable; and every farm which does not show its golden ear, seems to be deficient in one important respect.

Indian corn is not difficult to grow, though it is well known that its perfection depends greatly upon the season; and yet if July and August are favorable, it is sufficient to secure a good crop. Early planting, too, is not less important, especially for the later and larger varieties.

Very little has been done by chemists, which is calculated to throw light upon the composition of corn. All the analyses which have hitherto been published are incorrect, as well as imperfect.

I have been induced, from the foregoing considerations, to enter upon a minute and careful investigation of this grain. Observations have been made upon the plant, commencing with its earliest, and ending with its ripest stage. It requires a succession of seasons, however, in order to acquire a series of facts which can be regarded as full and complete. Two sets of observations have been carried on during the past season. My own, however, are less perfect than those by Mr. Salisbury. Unavoidable interruptions have occurred, which have occasionally interfered with the punctuality which is desirable upon inquiries of this kind. The completeness of Mr. Salisbury's observations supply my own defective opportunities in many respects.

The varieties to which my attention has been directed, are the small 8-rowed yellow corn, and the 8-rowed white flint: both are early and very superior kinds. The proportions have been determined, in one instance, of the large variety of white, mixed with the Tuscarora corn, which are later kinds, and which in this climate are more liable to be destroyed by early frosts.

TIME.	PARTS.					QUANTITY.	PER CENTUM.			
June 27.	1. Plant weighing -	•	-	-	-	374.2 grains.				
	Dry	•	-	•	-	26 ·				
	Water	-	-	-	-	$348 \cdot 2$				
	Ash	-	-	-	-	1.87	· 5 18			
	Ash calculated dry	-	-	-	-		$7 \cdot 19$			
	Dry matter - •	-	-	•	-		$6 \cdot 95$			
	Proportions of the different parts of one plant. Stalk (4 inches long; 7 joints appear) - 100.70 14.184									
July 5.	Height from 33 – 36 inch	es : st	out g	rowth	•					
	Stalk (4 inches long	14.184								
	Tassel (8 inches long, enveloped in the									
	leaf-sheaths) -	-	-	-	-	$61 \cdot 05$	8.599			
	Leaf-sheaths -	-	-	-	•	$181 \cdot 10$	$25 \cdot 509$			
	Leaf blades -	-	-	•	-	367 · 10	51.708			
	Weight of the whol	e plan	t	-	-	709 · 95	100.000			

	Pe	rcentage of as	sh , ϵ	dry m	atter	, etc.	in ed	ach of	the above part	8.
TIME.		PARTS.							QUANTITY.	PER CENTUM.
JULY 5.	1.	Stalks -	•	•,	•	•	-	•	100.7 grs.	
		Water	•	-	•	-	-	-	$92 \cdot 52$	91.877
		Dry matter		•	-	-	-	-	8.18	$8 \cdot 123$
		Ash (decide			soda	. and	potas	sh),	$\cdot 94$	$\cdot 933$
		Ash calcula			-	-	-	•		11-486
		Organic ma	ıtter	calcu	lated	dry	-	-		88.514
	2.	Tassels -	-	-	-	-	-	-	$61 \cdot 05$	
		Water	-	-	•	-	-	•	• 54.79	$89 \cdot 746$
		Dry matter		-	•	-	•	-	$6 \cdot 26$	$10 \cdot 254$
		Ash -	-	-	-	-	-	-	•66	1.081
		Ash calcula	ted	dry	-	-	-	-		10.542
		Organic ma	itter	calcu	lated	dry	-			$89 \cdot 458$
	3.	Sheaths	-	•	-	-	-	-	181 · 1	
		Water	-	•	-	-	-		$166 \cdot 03$	$92 \cdot 231$
		Dry matter		•	-	-	-	•	$15 \cdot 07$	7.769
		Ash (tastes	dec	idedly	salin	ie)	-	-	$1 \cdot 94$	$1 \cdot 077$
		Ash calcula	ted	dry	-	-	•	-		$13 \cdot 863$
		Organic ma			lated	dry	-	-		$86 \cdot 137$
	4.	Leaves	-	-		•	-	•	367.1	
		Water	-	-	-	-	-		$304 \cdot 01$	$82 \cdot 814$
		Dry matter		-	-	•	-	-	63.09	$17 \cdot 186$
		Ash -	-	-	-	-	-	-	$6 \cdot 75$	1.839
		Ash calcula	ted	dry	-	-	-	•		10.701
		Organic ma	tter	calcu	lated	dry	•	-		$89 \cdot 299$
	Ash very saline, with a slight taste of caustic alkali.									
	5. Roots, washed clean in cold water, and freed from adhering water by pressing									
		in paper.	90	grs.	gave	of				
		\mathbf{W} ater	-	-	-	-	•	•	$70 \cdot 35$	$78 \cdot 722$
		Dry matter		-	•	-	-	-	$19 \cdot 15$	$21 \cdot 278$
		Ash -	•	-	-	-	-	-	1.65	1.8
		Ash calcula			-	-	-	-		$8 \cdot 459$
		Organic ma	tter	calcu	lated	dry	-	-		$91 \cdot 541$
July 12.	1.	Leaves -	-	-	-	-	-	•	698 · 1	
		Dry -	-	-		-	-	-	130 ·	
		\mathbf{Water}	-	-	-	_	-	_	$56 \cdot 1$	
		water								
			-	-	-	-		-	14.84	$2 \cdot 125$
		Ash -		- drv	-	-	-	-	14.84	$2 \cdot 125$ $11 \cdot 384$
	2.	Ash - Ash calcula		- dry -		•	- -	-		$2 \cdot 125$ $11 \cdot 384$
	2.	Ash - Ash calcula Sheaths		dry -	-	-	-		561.65	
	2.	Ash - Ash calcula Sheaths Dry -		dry -					561·65 61·	
	2.	Ash - Ash calcula Sheaths Dry - Water		dry - -			•		561·65 61· 500·65	11.384
	2.	Ash - Ash calcula Sheaths Dry - Water Ash -	ted - - -	-					561·65 61·	11·384 1·422
	2.	Ash - Ash calcula Sheaths Dry - Water	ted ted	-				-	561·65 61· 500·65	11:384

					,		
TIME.	PARTS.					QUANTITY	PER CENTUM.
July 18.	1. Leaves	•	•	-	•	886 · 1 grs.	
	Dry	•	•	•	•	16.2	
	Water	•	•	•	-	869.9	1.50
	Ash		-	-	•	13.48	1.52
	Dry matter	•	•	•	•		28.28
	Ash calculated dry	-	•	•	-		$8 \cdot 32$
	2. Sheaths	-	d	•	-	$457 \cdot 3$	
	Dry	-		-	-	$46 \cdot 5$	
	Water	•	-	•	-	410.8	
	Ash	•	-	•	•	$5 \cdot 27$	$1 \cdot 13$
	Ash calculated dry	-	el	-	•		11.333
	Dry matter -	•	-		•		$10 \cdot 167$
	3. Stalks	-		-	-	1084.2	
	Dry		-	-	-	$96 \cdot 5$	
	Water	-		-	-	987.7	
	Ash	•		•		8.04	.74
	Ash calculated dry			-	•		7.474
	Dry matter -	•		•	-		13.025
	4. Tassels					149 · 4	
		•	•	-	•	37.1	
	Dry Water	•	•	•	•	112.3	
	Ash	•	-	•	•	2.77	1.854
		•	•	•	•	2.11	7.466
	Ash calculated dry	•	•	•	•		24.83
	Dry matter	-	•	•	•		24 00
	5. Silks	•	•	-	-	44.48	
	Dry	-	•	-	-	$_{5}\cdot$	
	Water	-	۰	-	-	$39 \cdot 48$	
	Ash	-	•	•	-	$\cdot 405$	$\cdot 933$
	Ash calculated dry	-	•	•	٠		$8 \cdot 3$
	Dry matter -	•	-	-	•		11.24
July 29.	1. Leaves		_			2294	
90M1 20.	Dry			_	-	458.5	
	Water	-	-	_	_	1835 · 5	
	Ash			-	-	41.58	1.812
	Ash calculated dry			_	_	11 00	9.068
	Dry matter	_	_	_	_		20.42
	•	_	-	•	_		20 42
	2. Sheaths	-	•	-	•	3345	9
	Dry	2	-	-	-	399.	-
	Water	-	•	-	-	2846	
	Ash	-	•	-	-	$25 \cdot 49$	•765
	Ash calculated dry	•	-	•	-		6.388
	Dry matter	-	-	-	۰		11.92

ANALYSES OF MAIZE.

TIME.	PARTS.		QUANTITY. PER CENTUM.
July 29.	3. Stalks		3041 grs.
	Dry		370
	Water		2671 ·
	Ash		16.22 .533
	Ash calculated dry -		$4 \cdot 383$
	Dry matter		12 · 16
	4. Tassels		87.2
	Dry		40.
	Water		47.2
	Ash		1.85 2.121
	Ash calculated dry -		4.625
	Dry matter		45.884
	5. Silks		$426 \cdot 5$
	Dry		36.
	Water		$390 \cdot 5$
	Ash		1.92 $\cdot 45$
	Ash calculated dry		5.333
-	Dry matter		8.444
	Dry matter		0 111
August 4.	1. Leaves		2810
	Dry		631 ·
	Water		2179
	Ash		58.97 2.09
	Ash calculated dry -		$9 \cdot 367$
	Dry matter		22.099
	2. Sheaths		3124.5
	Dry		514.5
	Water		2610
	Ash		32. 1.02
	Ash calculated dry -		6.21
	Dry matter		16.466
	3. Stalks		5219
	Dry		654.5
	Water		$625 \cdot 02$
	Ash		29.48 .564
	Ash calculated dry -		4.54
	Dry matter		12.52
	4. Immature grain and cob		433.
	Dry	• • •	36.6
	Water		$396 \cdot 4$
	Ash		3.49 .806
	Ash calculated dry -	• •	$9 \cdot 53$
	Dry matter	• • •	8.45

TIME,	PARTS.		QUANTITY. PER CENTUM.
August 4.	5. Husks		2439 · grs.
	Dry		330.5
	Water		2108.5
	Ash		$12 \cdot 22$ $\cdot 501$
	Ash calculated dry		3.697
	Dry matter		13.59
	6. Tassels		184
	Dry		68.5
	Water		115.5
	Ash		$4\cdot 74$ $\cdot 252$
	Ash calculated dry		$6 \cdot 919$
	Dry matter		37.22
	7. Silks		700
	Dry		70.5
	Water	•	$629 \cdot 5$
	Ash		4 · · · 571
	Ash calculated dry		$5 \cdot 674$
	Dry matter		10.007
August 11.	1. Leaves		1642
	Dry		415.5
	Ash		36.59 2.022
	Ash calculated dry		8.806
	Dry matter		$25 \cdot 29$
	2. Sheaths		2709
	Dry		7 29·
	Water		1980 ·
	Ash	• •	35.86 1.324
	Ash calculated dry		$3 \cdot 92$
	Dry matter		$26 \cdot 99$
	3. Stalks		4577
	Dry		$744 \cdot 2$
	Water		3832.8
	Ash		51.25 1.119
	Ash calculated dry		$6 \cdot 86$
	Dry matter		$16 \cdot 05$
	4. Husks		2375
	Dry		242.5
	Water		2132.5
	Ash		7·35 ·31
	Ash calculated dry		3.031
	Dry matter		41.21
,	2 1 y 22240001		· · · · ·

TIME.	PARTS.					QUANTITY.	PER CENTUM.
August 11.	5. Cobs	-	-	-	•	3000 grs.	
	Dry	•	•	-	-	609 ·	
	Water	•	•	•	•	2391	
	Ash	-	-	-	-	14.61	· 4 86
	Ash calculated dry	-	•	-	-		$2 \cdot 234$
	Dry matter -	•	-	•	•		20.3
	6. Kernels	-	-	•	-	1830 ·	
	Dry	-	•	•	•	$384 \cdot 5$	
	Water	-	-	-	-	$1445 \cdot 5$	
	Ash	•	-	•	-	13.865	·758
	Ash calculated dry	•	•	-	•		$3 \cdot 605$
	Dry matter -	•	•	•	-		21.01
	7. Silks			-		199 ·	
	Dry					28.	
	Water	-				161.	
	Ash	-	•	-	-	1.91	•95
	Ash calculated dry	-	-	•	-		
	8. Tassels	_	_	_	_	123.7	
	Dry					47.	
	Water		_	_	_	76.76	
	Ash				-	4.1	3.31
	Ash calculated dry					T .	8.74
	Dry matter						37.99
August 23.	·					707.	3, 55
August 23.	1. Leaves	•	•	•	•	707· 197·	
	$egin{array}{lll} ext{Dry} & ext{-} & ext{-} \ ext{Water} & ext{-} & ext{-} \end{array}$	•	•	•	•	510.	
	Ash	-	•	-	•	17.84	2.523
	Ash calculated dry	_		-	_	17 04	2 923 8·954
	Dry matter -						27.87
		_				500.5	2. 0.
	2. Sheaths	•	•	•	•	590.5	
	Dry	•	•	•	•	112.	
	Water	-	-	•	-	478.5	1.70
	Ash	-	-	•	•	10.02	1.79
	Ash calculated dry	•	•	-	•		8·946 18·96
	Dry matter -	-	•	•	-		19.90
	3. Stalks	•	-	•	•	2237	
	Dry	•	•	•	•	225.5	
	Water	•	-	•	-	1911.5	
	Ash	-	•	•	-	13.72	•61
	Ash calculated dry	-	•	-	-		4.215
	Dry matter	•	•	•	•		$14 \cdot 55$

TIME.	PARTS.					QUANTITY.	PER CENTUM.
August 23.	4. Husks	•	•	•	•	1218 grs.	
	Dry	•	•	•	•	277	
	Water	•	•	-	•	941 ·	
	Ash	-	•	•	-	$10 \cdot 72$	•88
	Ash calculated dry	•	•	-	-		3.871
	Dry matter - •	•	•	-	-		$22 \cdot 74$
	5. Cobs	-	•	-	•	1481	
	Dry	-	•	-	-	404	
	Water	-	-	_	_	1077	
	Ash	-	•	-	_	$8 \cdot 225$	•556
	Ash calculated dry	•	•	-	-		$2 \cdot 033$
	Dry matter -	-	•	•	•		27.28
	6. Kernels					1840	
		•	•	-	•		
	Dry Water	•	-	•	•	730.3	
		•	•	•	-	1109.7	. MEO
	Ash	•	•	•	•	13.96	·758
	Ash calculated dry	•	•	•	•		1.919
	Dry matter -	•	•	•	-		39.67
	7. Tassels	•	•	•	-	5 8·	
	Dry		•	•	•	14.	
	Water	•	•	•	-	44.	
	Ash	-	•	•	-	3.1	•534
	Ash calculated dry	•	•	•	•		.704
	Dry matter -	-	•	•	-		24.
	8. Silks	-				167.56	
	Dry		_	_		20.	
	Water	•	•	-		147·56	
	Ash	_	_	•		.76	.045
	Ash calculated dry	_	-		-	10	3.8
		-	•	•	•		11.9
	Dry matter -	•	•	•	•		11.9
September 1.	1. Kernels	-	•	•	•	400	
	Dry	•	-	•	-	$249 \cdot 9$	
	Water	-	•	•	•	$150 \cdot 1$	
	Ash • • •	•	•	•	•	4·21 (some	coal).
	2. Cobs	•		•		931 ·	
	Dry	•	•	•		361.5	
	Water			•		569.5	61.16
	Ash		•			5.39	•539
	Ash calculated dry	•	•			3 30	1.491
	Dry matter -	-		•			38.82
	~ ~		-				

TIME.		PARTS.							QUANTITY.	PER CENTUM.
SEPTEMBER 1.	3.	Leaves -	•	•	-	•	•	-	1058 grs.	
		Dry -	•	•	•	•	•	-	449	
		Water		•	٠	•	-	-	609.	
		Ash -	•	•		•	•	-	52.7	4.98
		Ash calculat	ted	dry	•	•		۰		13.96
		Dry matter		•	•	-	•	•		$42 \cdot 53$
	4.	Husks -		•	•	•	•	-	1391 ·	
		Dry -	-	•	•	-	•	•	$405 \! \cdot \! 5$	
		Water .		•	•	•	•	-	$985 \cdot 5$	70.78
		Ash -		•		•	•		19.2	1.38
		Ash calculat	ed	dry	•	-	•	-		4.734
		Dry matter			•	•	•	-		29 · 1
	5.	Sheaths (lost)		•	•	•	•	•		
		Dry -	•	•	•	-	•	•	207·	
		Water	•	•		-	•	-		
		Ash		•	•	•	-	-	$19 \cdot 25$	
		Ash calculat	ed	dry	•	•	•	•		9.29

Note. The plant taken for August 23 happened to be one proportionally smaller than the preceding, yet at this time the leaves began to dry and wither; but this lightness of the herbage will increase with the development of the grain.

II. WHITE-FLINT CORN.

August 22.	1.	Tassels -	•	-	-	•	-	-	130 · 7 grs.	
		Dry -	-	-	•	•	_	•	57·	
		Water	•			•		•	$83 \cdot 7$	
		Ash -	-	•	-	-	•	-	4.37	$3 \cdot 34$
		Ash calcu	ılated	dry		•	•	•		7.666
		Dry matt	er	•	•	•	•	-		43.67
	2.	Stalks -	•	-	-	•	•	•	5766.9	
		Dry -	-	-	-	•	•	-	880.1	
		Water	•	-	•	•	•	•		
		Ash •	-	-	-	•	-	-	$35 \cdot 75$	•691
		Ash calcu	ılated	dry	-	-	-	-		4.006
		Dry matt	er	•	•	•	•	•		$15 \cdot 26$
	3.	Leaves -		•	•	•			2797	
		Dry -	-	•		-	•	•	$763 \cdot 8$	
		Water	-	•	-	•		-	$2033 \cdot 2$	
		Ash -	•	•	•		•	-	81.35	2.908
		Ash calcu	ılated	dry	-	-	-	-		10.637
		Dry matt	er	•	-	-		•		27.31

TIME.		PARTS.							QUANTITY.	PER CENT	UM.
August 22.	4.	Silks -	-	•	-	•	•		57 grs.		
		Dry -	-	•	•	•	•	-	36.		
		Water	-	-	-	•	•	-	21.		
		Ash -	•	•	•	•	-	•	1.72	3.1	
		Ash calcula	ated	${ m dry}$	•	•	-	-	~	4.777	,
		Dry matter	•	-	•	-	-	-		63.1	
	5.	Sheaths	•	-	•	•	•	•	2536		
		Dry -	•		-	-	•	-	494		
		Water	•		-	•	-	•	$2042 \cdot$		
		Ash -	•	-	-	•	-	•	40.46	1.596	
		Ash calcul	ated	$_{ m dry}$	-	-	-	-		8.192	;
		Dry matter		•	•	•	-	-		$19 \cdot 47$	
	6.	Husks		-	-	-	-	-	3534		
		Dry -		•	-	-	-		802		
		Water	-	-	-	-	-	-	2732		
•		Ash -	-	-	-	-	-	-	$29 \cdot 53$.835	j
		Ash calcul	ated	dry	-	•	-	-		3.682	;
		Dry matter		•	-	-	-	-		$22 \cdot 68$	
	7.	Cobs -	-	-	-	-	-		2935.5		
		Dry -	-	-	-	-	-	-	864		
		Water			-	•	-	-	$2070 \cdot 5$	70.53	
		Ash -		-		•	-		18.	•613	}
	-4	Ash calcula	ated (dry	•	-	-	•		2.083	}
		Dry matter		•	-	•	-	-		$29 \cdot 46$	
	8.	Kernels		-		-	_	-	4170.5 (fully	glazed).	
		Dry -	•	-		•	•	•	1633.	,	
		$\mathbf{W}_{\mathbf{a}\mathbf{t}\mathbf{e}\mathbf{r}}^{\mathbf{J}}$	•	-	•	-	-	-	2537.5	60.84	
		Ash -	-		-	-	-	-	32.7	•759)
		Ash calcul	ated	dry	-	•	-	-		2.002)
		Dry matter		-	•	-	-			$39 \cdot 15$	
					_						
III.	CORN	MIXED W	TTH	$\mathbf{L}\mathbf{A}$	RGE	WH	ITE-I	FLIN	T AND TUS	CARORA.	
September :	26. 1.	Leaves -	•	-	-	•	-	-	2198.5 grs.		
		Dry -	-	-	•	-	-	•	476		
		\mathbf{W} ater	-	-	•	•	-	•			
		Ash -		-	-	•	-	-	58.	2.061	Ĺ
		Ash calcul		dry	-	-	-	-		12.05	
		Dry matter	•	•	-	-	•	-		21.64	
	2.	Stalks -	-	-	-	-	-	-	$7552 \cdot 5$		
		Dry -	-	-	•	-	-	•	$1290 \cdot$		
		Water	-	-	-	-	-	-			
		Ash -	-	-	-	•	-	-	44.61	•59	
		Ash calcul		dry	•	-	•	-		3.533	\$
		Dry matter	•	-	•	•	•	•		17.08	

TIME.		PARTS.									QUANTITY.	PER CENTUM.
September 26.	3.			_	-	-					2176 · grs.	
SEI THINDER 20.	٥.	Dry -							-		$512 \cdot 3$	·
		Water							-		1663 · 7	
		Ash -							-		$34 \cdot 1$	1.566
		Ash calcul	ated	drv								6.686
		Dry matter		ury -	_	, _		_	_		4	$23 \cdot 54$
		-		_	_						1000 5	20 0 2
	4.	Husks -	•	a	-			-	a		1972.5	
		Dry -	-	-	-	-		·-	•		440.3	
		\mathbf{Water}	-	-	•	-		-	-		$1532 \cdot 2$	
		Ash -	-	-	•	-		•	CO CO		12.08	·612
		Ash calcula	ated	dry	-	-		-	•			$2 \cdot 743$
		Dry matter		-	-			-	-			$22 \cdot 27$
	5.	Cobs -				-					1773	
	٠.	Dry -									630 ·	
		Water						_	_		1143	$64 \cdot 46$
		Ash -									11.25	.634
		Ash calcula	ated	drv							11 25	1.785
		Dry matter		ury	_			_	_			35.53
		<u>-</u>		-	_	_					0044	05 35
	6.	Kernels of or	ne ea	ır	•	-		•	-		3344	•
		Dry -	•	-	•	•		•	-		1744.7	
		\mathbf{W} ater	•	-	•	•		•	-		1599 · 3	47.82
		Ash -	-	•	•	-		•	•		$23 \cdot 625$	1.706
		Ash calcula	ated	dry	-	-		•	-			1.354
		Dry matter	•	-	-	•		-	•			$52 \cdot 14$
		337 . 1	, .								1 <i>ECH</i> O. <i>E</i>	
		Weight of fo	nage	•		-			-	•	15672.5 grs.	
		Dry	•	-		•	•		-	•	3548.6	
		Ash	•	•		•	•		-	•	$160\cdot04$	
		Weight of k	ernel	s -		-	•		-	-	$3344 \cdot$	
		Dry kernels	-	-		-	-		-	•	$1744 \cdot 7$	
		m 1				,					10010 5	
		Total weight				plan	t		-	•	19016.5	
		Total weight				-	-		-	•	5293 · 3	
		Total amount	t of v	water		•	•		•	-	$183723 \cdot 2$	

REMARKS ON THE FOREGOING OBSERVATIONS.

A variety of facts relating to the progress of the crop may be deduced from the observations, which may be regarded as correct in the main.

The increase in weight, in ten days, of a single plant, an individual stalk, is as follows, taking the interval between the 18th and 29th of July, during which period corn probably grows as rapidly as at any other time:

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Increase in weight	-	-	6571.52 grs.
Increase in solid organic matter -	-	-	$1112 \cdot 2$
Total weight of the plant July 29	•	-	9193
Containing water	-	-	7889.5

The increase, therefore, of the plant every day, amounts to 657 grs., of which 546 are water and 111 dry matter.

August 10. The individual plant weighed	-	-	-	-	16455 grs.
Containing water •	-	-	-	-	$13255 \cdot 3$
Dry matter	-		-	-	$3199 \cdot 7$
Total increase in weight from Ju	aly 29	to A	ugust	11,	7262 ·
Increase per day for 11 days	•	•	•	•	660.
Inorganic matter assimilated Aug	ust 1	1		195	54
Inorganic matter assimilated July		-	-	87	06
Increase from July 29 to August 1	11	•	-	108	
Increase of inorganic matter abou	t 10	grs. p	er day	у.	

It will be observed that the increase in weight for two periods, from July 18 to July 29, and from July 29 to August 11, is virtually equal, only three additional grains remaining in favor of the last period (See Meteorological Observations and Tables at the end of the volume). The perfect solution of the inorganic matter is maintained by the large amount of water always circulating in the plant: thus, in August 11, there was 13255 grs. of water, and only 165½ grs. of mineral matter.

IV. EARLY WHITE-FLINT CORN.

Relation of the different parts of the plant to each other; observations, and the quantity of water, dry matter and ash in the several parts at different stages of its growth.—S.

June 3.	OBSERVATIONS, PARTS AND PROPORTIONS. Corn planted.	QUANTITIES.	PERCENTAGE.
June 11.	Plants begin to show themselves above the soil.		
June 15.	Plants 5 inches high. Average weight of each		
	plant	10·8925 grs.	
	PROPORTIONS.		
	Weight of four plants cut close to the roots -	43.57	
	Water	$39 \cdot 05$	$89 \cdot 626$
	Dry matter	$4 \cdot 52$	10.374
	Ash	•59	1.354
	Percentage of ash calculated on the dry matter,		$13 \cdot 053$

TIME. JUNE 21.	observations, farts and proportions. Average height of plants $10\frac{1}{2}$ inches. Growth during the last 6 days, $5\frac{1}{2}$ inches, which is a little less than 1 inch per day.	QUANTITIES.	PERCENTAGE.
	Average weight of each plant	22·1 grs.	
	Average increase of weight of each plant during	J	
	the last 6 days	11.2075	
	PROPORTIONS.		
	Weight of four plants cut close to roots	88.4	
	Water	78.86	89.208
	Dry matter	9.54	10.792
	Ash	1.83	$2 \cdot 07 \\ 19 \cdot 182$
	Percentage of ash calculated on the dry matter,		19 102
June 28.	Average height, 18 inches. Increase in height during the last week, $7\frac{1}{2}$ inches. Before this, the plants have been made up entirely of leaves. The sheaths now begin to form on the longest leaves. Average weight of each plant	84·15 grs.	
	Average increase in weight during the last week, being about 9 grs. per day.	$62 \cdot 05$	
	being about o gis. per day.		
	Took three plants: divided each plant into the part above the soil, and the part in it.		
	Relation of the part above the soil, to the part in it:		
	Part above the soil	$206 \cdot 7$	81.877
	Part in the soil	$45 \cdot 75$	$18 \cdot 123$
	Whole weight of the three plants	$\overline{252 \cdot 45}$	
	Divided the part above the soil into leaves and she	eaths.	
	Relation of leaves and sheaths:		
	Leaves	163.3	79.003
	Sheaths	43.4	20.997
		$206 \cdot 7$	100.
	PROPORTIONS.		
		163.3	
		141.44	86.613
	Dry matter	21.86	13.387
	Ash	$2 \cdot 34$	1.433
	Percentage of ash calculated on dry matter -		10.704
	Ash tastes decidedly of caustic potash.		

TIME.	OBSERVATIONS, PARTS AND PROPORTIONS.	QUANTITIES.	PERCENTAGE.
June 28.	2. Sheaths of three plants weighed	43·4 grs. 40·12	09.449
	Water	3.28	92.442
	Dry matter	•48	7·558
		40	1.106
	Percentage of ash calculated on dry matter - Ash saline, with a slight taste of caustic potash.		14.628
	Part of plant in the soil contains a large percentage of sugar. This part, from six plants, weighed	91.5	
	Water	$84 \cdot 39$	$92 \cdot 229$
	Dry matter	7.11	7.771
	Ash	1.20	1.311
	Percentage of ash calculated on the dry matter, Ash very saline, greenish, more difficult to burn than the leaves.		16.878
July 5.	Average height, 26 inches, Growth during the last week, 8 inches: a little over 1 inch per day.		
	Weight of one plant	237.5 grs.	
	Increase of weight during the last week -	$152 \cdot 35$	
	Average increase of weight of each plant per day,	21.764	
	The stalk begins to form.		
	Relation of the different parts to each other:		
	Stalk (3 inch long, with 3 joints)	6.75	2.844
	Sheaths	$62 \cdot 5$	$26 \cdot 312$
	Leaves	$168 \cdot 25$	70.844
		237.5	100 ·
	PROPORTIONS. 1. Stalks	6.75	
	Water	6.11	90.518
	Dry matter	•64	9.482
	Ash	.09	1.333
	Percentage of ash calculated on dry matter -	03	14.101
	Ash very saline.		14 101
	2. Sheaths	62.5	
	Water	57.92	92.672
	Dry matter	4.58	$7 \cdot 328$
	Ash	·65	$1 \cdot 04$
	Percentage of ash calculated on dry matter - Ash saline.		$14 \cdot 056$

TIME,	OBSERVATIONS, PARTS AND PROPORTIONS.	QUANTITIES. PERCENTAGE.
July 5.	3. Leaves	168·25 grs.
	Water	146.5 81.01
	Dry matter	21.75 18.99
	Ash	2·1 1·248
	Percentage of ash calculated on dry matter - Ash saline, with a slight taste of caustic potash.	6 • 566
	Roots washed thoroughly with cold water, and the adhering moisture separated by pressing in paper.	
	4. Roots of one plant	77.
	Water	$64 \cdot 69$ $84 \cdot 013$
	Dry matter	12.31 15.987
	Ash	1.22 1.584
	Percentage of ash calculated on dry matter -	$9 \cdot 908$
July 12.	Average height of plants, 35 inches. Average increase of height of each plant during the last week, 9 inches: over 1 inch per day. Corn has been hoed twice; the first time on the 15th of June, the last on the 5th of July. Stalks about 1½ inches long, with 4 joints. The old kernels attached to the roots completely absorbed, leaving nothing but the epidermis. Longest roots extend from 5-8 inches downwards, and from 8-10 inches laterally. Each plant has generally 12 leaves; 7 of these have sheaths. Growth moderately thrifty. Weight of one plant Increase in weight of each plant during the last	681·9 grs.
	week	$432 \cdot 7$
	Average increase in weight of each plant per day, being $2\frac{1}{2}$ grs. per hour.	61.814
	Relation of the different parts of one plant to each other:	
	Stalk	$40 \cdot 4$ $5 \cdot 926$
	Sheaths	216 · 1 31 · 678
44.	Leaves	413.7 60.679
	Roots	11.7 1.717
	Weight of one plant	681.9 100.

TIME.	OBSERVATIONS, PARTS AND PROPORTIONS.	QUANTITIES.	PERCENTAGE.
July 12.	PROPORTIONS 1. Stalk	10:1 mg	
90L1 12.	Water	$40 \cdot 4$ grs. $38 \cdot 06$	$94 \cdot 208$
	Dry matter	$2 \cdot 34$	5.792
	Ash	•42	1.039
	Percentage of ash calculated on the dry matter,		$17 \cdot 949$
	Ash tastes decidedly of caustic potash.		
	2. Sheaths	$216 \cdot 1$	
	Water	$204 \cdot 55$	$94 \cdot 655$
	Dry matter	11.55	$5 \cdot 345$
	Ash	1.23	•569
	Percentage of ash calculated on the dry matter, Ash saline, with a slight taste of caustic potash.		10.649
	3. Leaves	413.7	
	Water	366.81	88.665
	Dry matter	46.89	11.335
	Ash	5.12	4.288
	Percentage of ash calculated on dry matter -	0 12	11.37
	Ash decidedly saline.		11 07
	4. Roots (treated the same as those on the 5th) -	11.7	
	Water	$9 \cdot 48$	$81 \cdot 026$
	Dry matter	$2 \cdot 22$	18.974
	Ash	.26	$2 \cdot 222$
	Percentage of ash calculated on dry matter - Ash tastes of caustic potash.		11.711
July 19.	Average height of corn, 43 inches. Increase in height during the last week, 8 inches: a trifle over 1 inch per day.		
	About the average weight of each plant -	875·48 grs.	
	Increase in weight during the last week -	$177 \cdot 197$	
	Increase in weight per day	$25 \cdot 317$	
	which is a little over 1 grain per hour.		
	The longest roots extend into the soil 12 inches.		
	8 leaves have sheaths. Stalk $2\frac{1}{2}$ inches long, with 7 joints.		
	Relation of the different parts of one plant to		
	each other:		
	Stalk	$60 \cdot 4$	6.899
	Sheaths	242.5	27.699
	Leaves	544.5	62 · 194
	Roots	28.08	3.208
	Weight of one plant	\$75·48	100 ·

TIME.	OBSERVATIONS, PARTS AND PROPORTIONS.	QUANTITIES.	PERCENTAGE.
T 10	PROPORTIONS. 1. Stalk	60:4 mg	
July 19.	Water	60·4 grs. 56·16	92.98
	Dry matter	$4 \cdot 24$	7.02
	Ash	·68	1.126
	Percentage of ash calculated on the dry matter,	00	16.038
	Ash tastes decidedly of caustic potash.	10 000	
	2. Sheaths	$242 \cdot 5$	
	Water	$225 \cdot 75$	$93 \cdot 092$
	Dry matter	16.75	$6 \cdot 908$
	Ash	$2 \cdot 62$	1.081
	Percentage of ash calculated on the dry matter,		$15 \cdot 642$
	Ash tastes of salt and potash.		
	3. Leaves	$544\cdot 5$	
	Water	$479 \cdot 25$	88.016
	Dry matter	$65 \cdot 25$	11.984
	Ash	$6 \cdot 64$	1.219
	Percentage of ash calculated on the dry matter,		10.176
	Ash saline, with a slight taste of caustic potash.		
	4. Roots (treated as before described)	140.4	
	Water	121.87	86.802
	Dry matter	18.53	13 · 198
	Ash	1.76	1.253
	Percentage of ash calculated on the dry matter,		$9 \cdot 498$
July 26.	Average height of corn, 49 inches. Increase in height during the last week, 6 inches: a little less than 1 inch per day. Length of stalk from $4-5\frac{1}{2}$ inches, with 8 joints. Tassels from $1-3$ inches in length.		
	About the average weight of each plant	2039 grs.	
	Increase in weight during the last week -	1191.6	
	Average increase in weight per day	$170\cdot 22$	
	Average increase in weight per hour	$7 \cdot 09$	
	Relation of the different parts of one plant to each other:		
	Stalk	338.8	16 · 125
	Sheaths	690.5	34.056
	Leaves	1002.5	49 • 466
	Tassels	7.2	•353
		2039	100

TIME.	OBSERVATIONS, PARTS AND PROPORTIONS.	QUANTITIES.	PERCENTAGE.
July 26.	PROPORTIONS. 1. Stalk	338.8 grs.	
JULI 20.	Water	317·12	93.601
	Dry matter	$\frac{317}{21.68}$	6.399
	Ash	3.83	1.131
		9.00	17.666
	Percentage of ash calculated on the dry matter, Ash tastes decidedly of caustic potash.		17.000
	rish tastes decidedly of caustic potash.		
	2. Sheaths	$690 \cdot 5$	
	Water	646 03	91.822
	Dry matter	$44 \cdot 47$	8.178
	Ash	$7 \cdot 33$	1.615
	Percentage of ash calculated on the dry matter,		$16 \cdot 483$
	,		
	3. Leaves	$1002 \cdot 5$	
	Water	857 · 15	$85 \cdot 501$
	Dry matter	$145 \!\cdot\! 35$	$14 \cdot 499$
	Ash	$15 \cdot 3$	1.526
	Percentage of ash calculated on the dry matter,		$15 \cdot 263$
	4. Tassels	$7 \cdot 2$	
	Water	$6 \cdot 45$	89.583
	Dry matter	·75	$10 \cdot 417$
	Ash	•11	1.528
	Percentage of ash calculated on the dry matter,		14-667
	5. Roots	91.4	
	Water	78·57	85.962
	Dry matter	12.83	14.038
	Ash	.80	·875
		00	6· 2 35
	Percentage of ash calculated on the dry matter,		0 200
•			
August 2.	Average height of corn, 58 inches. Increase in		
	height during the last week, 9 inches: being		
	about $1\frac{1}{3}$ inches per day. Tassel 12 inches		
	long: does not show itself yet above the		
	sheaths. Stalk 12 inches in length, with 11		
	joints. Eleven or all of the leaves have		
	sheaths.	2000	
	About the average weight of each plant	3308·4 grs.	
	Increase in weight during the last week	1269 • 4	
	Average increase in weight per day	181.34	
	Average increase in weight per hour	7 ·556	

TIME.	р	observation of							4 4.	QUANTITIES,		PERCENTAGE.
August 2.		elation of t each other		imer	ен р	arts c	n the	pian	ı ıo			
		Stalk	•					•		000		28.167
			•	-	-	-	-	•	-	982 · grs.		
		Sheaths		-	•	•	-	-	•	847.		24.294
		Leaves		-	-	-	-	-	-	1358		38.951
		Tassels	•	-	-	-	•	-	-	121.4		3.482
		Roots	-	-	-	-	-	-	-	178.		5 · 106
	W	eight of 1	plan	t	-	-	-	-	-	$3486 \cdot 4$		100 ·
			-	ROPORT	nons.					-		
	1. S	talks -	-	-	-	-	-	-	-	$982 \cdot$		
		Water	-	-	-	-	-	-	-	$929 \cdot 15$		$93 \cdot 999$
		Dry mat	ter	-	-	-	-	-	-	$52 \cdot 85$		6.001
		$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	$7 \cdot 95$	÷	$\cdot 809$
	$P\epsilon$	ercentage o	f ash	cal	culate	d on t	he dr	y mat	ter,			$13 \cdot 481$
	2. S.	heaths	-	-	-	-	-		-	847		
		Water	-	_	-	-	-	-	_			
		Dry mat	ter	-	-	_	_	_	_			
		Ash	-	_	-	-	-	_	_	$9 \cdot 22$		1.088
	\mathbf{P}_{ϵ}	ercentage o	f ash	calc	ulate	d on t	he dr	v mat	ter.			
		nitted weig							,			
		eaves -	3	, 420	-1 -1110			3 •		1358		
	5. L	Water	-	-	-	-	-	-	-	1333 1178·5		06.700
			• • • • •	-	-	•	•	-	-	179.5		$86.782 \\ 13.218$
		Dry mat	ter	-	•	•	-	-	•	179.3 17.35		1.277
	D.	Ash	- c - 1	- 1	- 1-4-		1	- 4	-	17.99		
		ercentage o	ıasn	caro	ulate	u on t	ne ar	y mat	ter,			9.666
	4. T	assels	-	-	-	-	-	-	-	121 • 4		
		Water	-	-	-	-	-	-	-	$109 \cdot 72$		$90 \cdot 379$
		Dry mat	ter	-	-	-	-	-	-	11.68		9.621
		$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	1.03		.848
	\mathbf{Pe}	rcentage of	f ash	calc	ulated	l on t	he dry	y mati	ter,			8.818
	5. Re	oots -	-	-	-	-	-	-	-	178.		
		Water	-	-	-	-	-	-	-	$155 \cdot 22$		$87 \cdot 202$
		Dry mat	ter	-	-	-	-	-	-	$22 \cdot 78$		12.798
		${ m Ash}$	-	-	-	-	-	-	-	1.73		$\cdot 972$
	\mathbf{Pe}	rcentage of	ash	calc	ulated	l on tl	he dry	mat	ter,			$7 \cdot 595$
		, ,			0.5	. ,	T .					
August 9.		erage heig										
		height dur										
		inch per da										
		selves from										
		Average le										
		in length		_								
		Each stalk					ne sa	me ni	ım-			
г.		ber of leav										
AGRICU	LTURAL	REPORT-	— v o	L. II	-1	23						

TIME.		OBSERVATIO	ons, 1	PARTS	AND	PROPO	RTIONS			QUANTITIES.	PERCENTAG
August 9.		out the av		3827·5 grs.							
		rease in w						k -	•	$286 \cdot 1$	
		erage incr						-	-	$11 \cdot 92$	
	Ave	erage incr	ease	in we	eight	per h	our	-	•	•49	
	Rei										
		Stalk	_			-	-	-	-	1241	$32 \cdot 413$
		Sheaths		-	_		-	-		$904 \cdot 5$	23.63
		Leaves		-		_	-	_	-	1216	31.769
		Tassels			-	_		-	-	233.	6.094
		Roots	-	-	-	-	-	-	-	233.	$6 \cdot 094$
	0	Weight	of 1	plant	: -	-	-	-	-	3827.5	100.
			P	ROPORTI	enoi.						
	1. St.	alk -	-	-	-	-	-	-	-	1241 ·	
		Water	•		-	-	-	-	-	11 56 ·3	$93 \cdot 09$
		Dry mat	tter	-	-		-	-	-	84.7	$6 \cdot 91$
		Ash	-	-	-	-	•	-	-	$9 \cdot 58$.707
	Pe	rcentage o	of ash	ı calcı	ulated	l on t	he dry	y mat	ter,		11.301
	2. Sh	eaths	-	-	-	-	_		-	904	
		Water	-	-	-	-	-	-	-	$814 \cdot 06$	$90 \cdot 05$
		Dry mat	ter	-	-	-	-	-		$89 \cdot 94$	$9 \cdot 95$
		$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	$10 \cdot 16$	1.01
	Pe	rcentage o	of asl	ı calc	ulateo	l on t	the dry	y mat	ter,		11.02
	3. Le	aves -	-	_		_	_		-	1216	
			-	-	-	-	-	-	-	$1000 \cdot 38$	82.268
		Dry mat	ter	-	-	-	-	-	-	215.62	17.732
		Ash		-		_		-	-	20.66	1.699
	$\mathbf{P}\mathbf{e}$	rcentage o	f asl	ı calcı	ulated	l on t	he dry	mat	ter,		9.581
	4. To	zssels					_			233 ·	
		Water		_	_	_	-	_	-	191.5	82.104
		Dry mat	tter		-	_		_		41.5	17.896
										2.16	.902
			-	-	-	-					
	Pe	Ash rcentage o	• of asl	- h calc	- ulate		the dr	y mat	ter,		5.201
		Ash rcentage o	of asl	- h calc			the dr	y mat	ter,		
		Ash recentage of cots -	of asl	- h calc -			the dr	y mat	ter,	233•	5.201
		Ash rcentage c oots - Water	-	- h calc - -			the dr	y mat - -	ter,	233 · 194 ·	5·201 83·205
		Ash recentage of cots -	-	- h calc - - -			the dr	y mat - -	ter,	233•	5.201

TIME. AUGUST 16.	Average height of corn, 72 inches. Increase in height during the last week, 7 inches, or 1 inch per day. Nearly the whole tassel is above the upper sheath, in the plant taken for proportions; length, 19 inches. Length of stalk, 46 inches. Increase in length during the last week, 24 inches, or $3\frac{1}{2}$ inches per day. Two ears begin to form. Ears 3 inches long. Silks from $1-3$ inches long. Each ear has 9 sheaths, 5 of which have leaf-blades. Length of husks, from $10-17$ inches. Each kernel is entirely enveloped in husks of its own.	QUANTITIES.	PERCENTAGE.
	Ear-stalks $1\frac{1}{2}$ inches long.		
	About the average weight of each plant	6780°85 grs.	
	Increase in weight during the last week	$2953 \cdot 35$	
	Average increase in weight per day	$436 \cdot 19$	
	Average increase in weight per hour	$18 \cdot 16$	
	Relation of the different parts of the plant to each other:		
	Stalk	2738	$38 \cdot 343$
	Sheaths	1450	20.306
	Leaves	1640	22.967
	Tassels	330.5	4.624
	Sheaths of husks of 2 ears	350.9	4.914
	Leaves of husks of 2 ears	$140 \cdot 25$	1.964
	Two ear-stalks	64.	.897
	Two ears	67.5	•943
	Roots	360	5.043
	Exact weight of the plant taken for proportions,	7140.85	100 •
	proportions.		
	1. Stalk	2738	
	Water	$2496 \cdot 5$	$91 \cdot 179$
	Dry matter	241.5	8.821
	Ash	25.87	$\cdot 945$
	Percentage of ash calculated on the dry matter,		10.712
	Q 51-41-	1450.	
	2. Sheaths Water	1450	00.000
		$1289 \cdot 53$ $160 \cdot 47$	88.933
	Dry matter Ash		$11 \cdot 067 \\ 1 \cdot 281$
		18.58	
	Percentage of ash calculated on the dry matter,		11.578

								_				
TIME.		OBSERVATI	ons, P	ARTS	AND	PROPO:	RTIONS			QUANTITIES.	PERCENTAGE.	
August 16.	$3. L_{\epsilon}$	eaves -	-	-	•	•	-	-	-	1640 · grs.		
		\mathbf{W} ater	-	-	-	-	•	-	-	$1334 \cdot 2$	81.353	
		Dry ma	tter	-	-	-	•	٠.	-	$305 \cdot 8$	18.647	
		$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	•	-	-	$34 \cdot 14$	$2 \cdot 081$	
	${ m Pe}$	rcentage o	of ash	calc	ulated	l on t	he dr	y mat	ter,		11 · 164	
	4. To	accale	_	_		_	_	_		330 · 2	5	
	4. 1	Water	_	_	_	_	_	_	_	213 · 1	$64 \cdot 503$	
		Dry ma	ttor	-	_	_	_	_	_	107 · 1	35.497	
		Ash	-	_	_	_	_	_	_	4.95	1.041	
	D _o		e aab	- 	- la.e.a.	- 1 an 1	- the du	• ••• ••• • •	- 	4 90	4.602	
	re	rcentage o	n asn	carc	urateo	1 On 1	ine ar	y mai	ter,		4.002	
	5. S	heaths of h	husks	-	-	-	-	-	-	$350 \cdot 9$		
		Water	-	-	-	_	-	-	-	$311 \cdot 72$	88.802	
		Dry ma	tter	-	-	-	-	-	-	$39 \cdot 18$	$11 \cdot 198$	
		Ash	-	-	-	-	-	-	-	$2 \cdot 27$.604	
	\mathbf{Pe}	rcentage o	of ash	calc	ulate	d on	the dr	y mat	ter,		$6 \cdot 04$	
	a T									140.07		
	6. L	eaves of hi		-	-	-	-	-	-	140.25	60 601	
		Water		-	-	-	-	-	-	$116 \cdot 71$	83.201	
		Dry ma	itter	-	-	-	-	-	-	$23 \cdot 54$	16.799	
		Ash	-	-	-	-	-	•	-	$1 \cdot 36$	•906	
	Percentage of ash calculated on the dry matter, 5.70											
	7. E	ar-stalks	-	_	_	_	-	_	_	64.		
		Water	_			-	_	_	_	57.53	89.89	
		Dry ma				_	_	_		6.47	10 · 11	
		Ash	-	_	_	_	_	_	_	•37	•578	
	Pa	ercentage	of ach	۔ مام	ulato	d on	dry m	attor	_	01	5.718	
	1	arcentage (or asn	carc	urate	u on	ary III	atter	_		9 110	
	8. E	ars -	-	-	-	-	-	-	-	$67 \cdot 5$		
		Water	-	-	-	-	-	-	-	$59 \cdot 67$	88.4	
		Dry ma	tter	-	-	**	-	-	-	7.83	11.6	
		\mathbf{Ash}	-	-	-	-	-	-	-	. •56	.829	
	$\mathbf{P}\epsilon$	ercentage	of asl	h calo	culate	d on	the dr	y ma	tter,		7 · 152	
		sh tastes d							,			
										222		
	9. K	Roots -	•	-	•	-	•	-	,-	360	00.0	
		Water		-	-	-	-	-	-	296.63	82.397	
		Dry ma	atter,	-	-	-	-	-	-	63.37	17.603	
	_	$\mathbf{A}\mathbf{s}\mathbf{h}$	•	. .	•			-	-	$4 \cdot 9$	1.361	
	$\mathbf{P}\epsilon$	ercentage	of as	h cal	culate	ed on	the di	ry ma	tter,		$7 \cdot 732$	

TIME.	OBSERVATIONS, PARTS AND PROPORTIONS.	QUANTITIES.	PERCENTAGE.
August 23.	Average height of corn, 76 inches. Increase in		
	height during the last week, 4 inches. In full		
	flower. Silks show themselves. Stalks 59		
	inches long. Increase in length during the		
	last week, 13 inches: or almost 2 inches per		
	day. Tassel 17 inches long. The plant taken		
	for proportions has 1 ear silked out: two		
	others are just beginning to form.		
	About the average weight of each plant	8170 · 7 grs. 1389 · 85	
	Increase in weight during the last week		
	Average increase in weight per day	$198 \cdot 55$	
	Average increase in weight per hour	8.27	
	Relation of the different parts of the plant to		
	each other:		
	Tassel	$173 \cdot 2$	$2 \cdot 12$
	Top stalk	1007	$12 \cdot 335$
	Butt stalk*	$1489 \cdot 5$	$18 \cdot 239$
b	Sheaths	$1273 \cdot 5$	$15 \cdot 596$
	Leaves	$1540 \cdot 2$	18.812
	Ear-stalks	$338 \cdot 5$	$4 \cdot 143$
	Ears (each kernel is enveloped yet in its		
	husks)	$312 \cdot 2$	3.821
	Silks	$130 \cdot 4$	1.596
	Sheaths of husks	$933 \cdot 2$	11.421
	Leaves of husks	$379 \cdot$	4.638
	Roots	594•	7 • 279
	Weight of 1 plant	8170.7	100
	proportions.		
	1. Tassels	173.2	
	Water	$119 \!\cdot\! 95$	$69 \cdot 255$
	Dry matter	$53 \cdot 25$	$30 \cdot 745$
	Ash	$3 \cdot 36$	1.939
	Percentage of ash calculated on the dry matter,		$6 \cdot 309$
	2. Top stalk	1007	
	Water	$900 \cdot 25$	89.399
	Dry matter	$106 \cdot 75$	10.601
	Ash	7.31	.726
	Percentage of ash calculated on the dry matter,		6.847
	3. Butt stalk	1489.5	
	Water	1320 •	88.62
	Dry matter	$169 \cdot 5$	11.38
	Ash	15.43	1.036

^{*}The stalk is divided at the insertion of the upper ear.

TIME.		OBSERVATIONS	, PART	s AND	PROPO	RTIONS	·.		QUANTITIES.	PERCENTAGE.		
August 23.	4. Sh	eaths -	-	-	-	•	-	-	1273 · 5 grs.			
		Water -	•	•	•	•		-1	$120\cdot$	$87 \cdot 947$		
		Dry matter	-	-	•	-	-	-	$153 \cdot 5$	$12\cdot053$		
		Ash -	-	•	-	-	-	•	$19 \cdot 75$	1.551		
	Per	centage of a	sh calc	ter,		12.866						
	5. Le	aves - •	•	-	•	-	-	-	$1540 \cdot 2$			
		Water -	-	-	•	-	-	-	$1270 \cdot 4$	$82 \cdot 482$		
	~	Dry matter	· -	-	-	•	•	•	$269 \cdot 80$	17.518		
		Ash -	-	-		-	-	-	$24 \cdot 37$	1.582		
	Per	rcentage of	ash cal	culate	ed on	dry m	atter	-		$9 \cdot 032$		
	6. E	ar-stalks -	_	-			-		$338 \cdot 5$			
		Water -		-	-	-	-	-	$315 \cdot 64$	$93 \cdot 246$		
		Dry matter			-	-	-	-	22.86	$6 \cdot 754$		
		Ash -			-		-	•	1.63	•481		
	Per	centage of a	sh cal	culate	d on t	he dr	v mat	tter.	- 40	7.13		
				• • • • • • • • • • • • • • • • • • • •			,	,	0.10	. 10		
	7. Ea		-	•	-	-	•	•	312.2			
		Water -	-	-	•	•	•	•	$284 \cdot 4$	91.095		
		Dry matter	•	•	-	•	•	-	27.8	8.905		
		Ash -					•	-	1.81	· 57 9		
	Per	rcentage of a	ish cal	culate	d on	the dr	y mat	ter,		$6 \cdot 511$		
	8. Si	lks	-	-			-	-	$130 \cdot 4$			
		Water -	-	-	-	-	-	-	$120 \cdot 32$	$92 \cdot 269$		
		Dry matter	· -	•	-	•	-	-	10.08	$7 \cdot 731$		
		Ash -	•	•	•	•	-	-	•4	·306		
	Percentage of ash calculated on the dry matter, 3.968 Ash saline.											
	9. SA	neaths of husi	ks -				-		$933 \cdot 2$			
		Water -					-		831.33	89.083		
		Dry matte		-	-				101.87	10.917		
		Ash -			_				6.1	.654		
	Pe	rcentage of a	sh cal	culate	d on t	he dr	v mat	ter.	. .	5.988		
		_			u 011 0	no an	,	,,,,	0.50	3 000		
	10. $L\epsilon$	aves of husk:	s -	-	-	•	-	-	379			
		Water -	•	•	-	•	•	-	315.15	$83 \cdot 153$		
		Dry matte	r -	•	•	-	-	•	63.85	16.847		
		Ash -	•	-	-	•	-	-	$6 \cdot 22$	1.641		
	Pe	rcentage of a	ish cal	culate	d on t	he dr	y mai	tter,		9.741		
	11. Re	oots	-			-	-		$594 \cdot$			
		Water -	-	•	-	-	-	-	$486 \cdot 13$	81.84		
		Dry matter		-	-	-	-	-	107.87	18.16		
		Ash -		-	-	-			$9 \cdot 35$	1.574		
	Pa	rcentage of a	sh cal	culate	d on i	the dr	v ma	tter.		8.667		
	. C.	gc or c	car	Juliu	JII	c ui	Jina	,		J 00.		

TIME.	OBSERVATIONS, PARTS AND PROPORTIONS.	QUANTITIES.	PERCENTAGE.
August 23.	12. Anthers filled with pollen grains	35 · grs.	
	. Water	8.70	$24 \cdot 857$
	Dry matter	$26 \cdot 30$	75 ·143
	Ash	1.67	4.771
	Percentage of ash calculated on the dry matter,		6.349
	Ash tastes decidedly of caustic potash.		
August 30.	Average height of plants, about 78 inches.		
	The increase in height during the last week		
	is but very little. Pollen nearly all fallen.		
	About $\frac{1}{3}$ of the bulk of each kernel protrudes		
	from its husks. The plant taken for propor-		
	tions is 83 inches high, and has 3 ears; two		
	of which are very small, the silks having not		
	yet appeared: the third is $S_{\frac{1}{2}}$ inches long.		
	About the average weight of each plant	$10580\cdot 2$	
	Increase in weight during the last week	$2409 \cdot 5$	
	Average increase in weight per day	$344 \cdot 21$	
	Average increase in weight per hour	$14 \cdot 34$	
	Stalk 78 inches long. Increase in length during		
	the last week, 19 inches. Average increase in		
	length per day, about $2\frac{2}{3}$ inches.		
	Relation of the different parts of the plant to each other:		`
	Tassel (length 15 inches)	150.6	1.363
	Top stalk (length 46 inches)	1187.5	10.748
	Butt stalk (length 22 inches)	2169	19.632
	Sheaths	1561.3	$14 \cdot 132$
	Leaves	1664.4	15.065
	Sheaths of husks	$2349 \cdot 2$	21.263
	Stalk of ear	409.1	3.703
	Ear	889	8.047
	Silks	200 ·	1.81
	Roots	468.1	4.237
	Weight of one plant	11048.2	100 ·
	PROPORTIONS. 1. Tassels	$150 \cdot 6$	
	Water	98.98	$65\!\cdot\!724$
	Dry matter	51.62	34.276
	Ash	4.3	2.855
	Percentage of ash calculated on the dry matter,	± 0	S·33
	i crochage of asir calculated on the dry matter,		0.00

August 30.	െ										
	2.	Top stalks	-	-	-	-	-	-	-	1187.5 grs.	
		Water		-	-	-	-	-	-	1009:38	85.
		Dry ma	tter	-	•	-	-	-	-	178 · 12	15.
	_	$\mathbf{A}\mathbf{s}\mathbf{h}$	•	٠,	•			-	-	$5 \cdot 42$	•456
	ŀ	Percentage o	of ash	ter,		$3 \cdot 043$					
	3.	Butt Stalks	-	-	-	•	-	-	-	$2169 \cdot$	
		\mathbf{W} ater	-	-	-	-	-	-	-	$1858 \cdot 75$	85.696
		Dry ma	tter	-	-	-	-	-	-	$310 \cdot 25$	$14 \cdot 304$
		$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	$14 \cdot 56$	·671
	I	Percentage c	of ash	calc	ulated	on t	he dry	mat	ter,		4.689
	4.	Sheaths		-	-	-	-	-	-	$1561 \cdot 3$	
		Water	-	-	-	-	-	-	-	$1349 \cdot$	$86 \cdot 402$
		Dry mar	tter	-	-	-	-	-	-	$212 \cdot 3$	$13 \cdot 598$
		$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	16.8	$1 \cdot 076$
	I	Percentage o	of ash	calc	ulated	l on t	he dry	mat	ter,		7.913
		Leaves	_	_	-	_		_		1664 · 4	
	٥.	Water		_	_	_	_	_	_	1331.4	$79 \cdot 993$
		Dry mat	ter	_	_		_	_	_	333.	$20 \cdot 007$
		Ash	-	_	_	_	_	_	_	30.62	1.839
	ī	Percentage o	of ash	calc	nlater	loni	he dr	v mai	ter	00 0 2	$9 \cdot 195$
		_		· our	diane			<i>y</i> 11100	,,,,	0240.0	200
	0.	Sheaths of h		•	-	•	-	-	-	2349 · 2	OP . 501
		Water		-	•	•	•	-	•	2057.7	87.591
		Dry mat	ter	•	-	•	•	•	•	291.5	$12 \cdot 409$
	7	Ash	- 1-	- 1-	- 1-4	- ! 4	- .bl		•	$8 \cdot 42$	•358
		Percentage of		carc	uiatec	i on t	ne ar	y mai	ier,		2.888
	7.	Stalks of ea		-	-	-	-	-	-	$409 \cdot 1$	
		Water		-	-	-	-	-	-	$376 \cdot 24$	91.967
		Dry ma	tter	-	-	-	-	-	-	32.86	8.033
		Ash	-	-	-	-	-	-	-	.7	•171
		Percentage o	of ash	calc	ulated	on t	he dry	y mat	ter,		$2 \cdot 13$
	8.	Ears -	-	-	-	-	-	-	-	889.	
		Water	-	-	-	-	-	-	-	812.13	$91 \cdot 353$
		Dry mat	ter	-	-	-	-	-	-	76·S7	8.647
		\mathbf{Ash}	•	-	-	-	. •	•	-	$3 \cdot 02$.339
	1	Percentage o	of ash	calc	ulated	l on t	he dry	y mat	ter,		3.912
	9.	Silks -	-	-	-	-	-	-	-	$200 \cdot$	
		Water	-	-	-	-	-	-	-	184.6	$92 \cdot 3$
		Dry mat	ter	-	-	-	-	-	-	$15 \cdot 4$	$7 \cdot 7$
		Ash	-	-	-	-	-	-	-	.77	•385
	1	Percentage o	of ash	calc	ulated	l on t	he dr	y mat	ter,		5.
	1	Ash saline, v	vith a	slig	ht tas	te of	causti	c pot	ash.		

TIME.	OBSERVAT	nons,	PARTS	AND	PROPO	RTIONS	i.		QUANTITIES.	PERCENTAGE.
August 30.	10. Roots -	-	-	-	-	-	-		468·1 grs.	
	Water	-	-	-	-	mi	-	-	$383 \cdot 23$	81.878
	Dry ma	atter	-	-	-	-	-	-	84.87	$18 \cdot 122$
	Ash	-	-	•	-	-	-	-	4.36	$\cdot 931$
	Percentage	of asl	h calc	ulate	d on t	he dr	y mai	ter,		$5 \cdot 137$
	Relation of	kerne	els to	cob:						
	Kernel	s -	-	_	-		-		$202 \cdot 65$	27.566
	Cob	-	-	-	-	-	•	-	$532 \cdot 5$	$72 \cdot 434$
									735·15	100
			PF	ROPORTIO	ons.					
	1. Kernels -	-	-	-	-	-				
	Water	-	-	-	-	-	-	-		90.8
	Dry ma	after	-	-	-	-	-	-		$9 \cdot 2$
	Ash		-	-	-	-	-	-		•41
	Percentage	of asl	n calc	ulated	d on t	he dry	y mat	ter,		$4\cdot 456$
	Ash tastes o	f pota	ash ai	nd chl	loride	of so	dium.			
	2. Cobs -			_	_	_		_	332 · 5	
	Water	-	-	-	_	-	_	_	299.3	80.015
	Dry ma	tter	_	_	-	-	-	-	33.2	$19 \cdot 985$
	Ash	_	_		-		-	-	· 7 9	.237
	Percentage	of asl	n calc	ulated	on t	he dry	z mat	ter	• •	2.379
	Ash tastes o					-				~ 310
	11011 140100 0	. 0.110	1140	01 500		and P		,		

September 6. Average height of plants, about 78 inches.

Corn in early stages of milk. Plant taken for proportions is 75 inches in length: it has 2 ears, one about $5\frac{1}{2}$ inches long. Kernels about $\frac{1}{3}$ protruded from their husks, and about $\frac{1}{3}$ grown. The other is 10 inches long: kernels $\frac{2}{3}$ protruded from their husks, and about $\frac{2}{3}$ grown.

Specific gravity of kernels - - - 0 · 9917

About the average weight of each plant - 12717 · 18

Increase in weight during the last week - 2136 · 98

Average increase in weight per day - - 305 · 28

Average increase in weight per hour - - 12 · 72

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TIME.		OBSERVATIONS,	PARTS	AND I	ROPOR	TIONS.			QUANTITIES.	PERCENTAGE.
SEPTEMBER 6.		Relation of the	differe	nt pa	rts of	the	plant	to		
		each other:		_						
		Tassel (len	gth 14	inch	es)	-	-	-	147.98 grs.	$1 \cdot 163$
		Top stalk (l	_			-	-	•	1128·S	8.886
		Butt stalk (1	length	26 in	ches)	-	-	-	$2084 \cdot 7$	$16 \cdot 394$
		Sheaths -	•	-	-	-	-	-	$1239 \cdot 8$	$9 \cdot 759$
		Leaves -	-	-	-	-	-	-	$1970 \cdot 2$	$15 \cdot 592$
		Ear-stalks	•	-	-	-	-	-	$1217 \cdot 2$	9.581
		Silks -	-	-	-	-	•	~	$260 \cdot 3$	$2 \cdot 047$
		Sheaths of	husks	-	-	-	-	-	2019.1	$15\cdot729$
		Leaves of h	usks	-	-	-	-	-	$466\cdot 4$	3.667
		Kernels of	the lar	gest e	ear	-	-	-	926.8	7.297
		Cob of the l	largest	ear	-	-	-	-	$1255\cdot 9$	9.885
		Weight of o	one pla	ant	-	-	-	-	12717 · 18	100
			PROPOR	RTIONS.						
	1.	Tassels			-	-	•	-	$147 \cdot 98$	
		Water -		-	-	-	-	-	$94 \cdot 38$	$63 \cdot 779$
		Dry matter	-	-	-	-	-	-	$53 \cdot 6$	$36 \cdot 221$
		Ash -	-	-	-	-	-	-	$3 \cdot 39$	$2 \cdot 297$
		Percentage of as	sh calc	ulated	l on tl	he dry	mat	ter,		$6 \cdot 325$
	2.	Top stalks -		-	-	-	_	-	1128.8	
,		Water -	-	-	-	-	-	-	$913 \cdot 6$	80.935
		Dry matter		-	-	-		-	215.2	19.065
		Ash -		-	_	-	-	-	$7 \cdot 22$	0.639
		Percentage of as	h calcu	ulated	on th	ne dry	mati	er,		3·35 5
	3.	Butt stalks -	-	-	-	_			$2084 \cdot 7$	
	•	Water -	-	-	-	-		-	1710.3	82.041
		Dry matter	-	_	-	-	-		$374 \cdot 4$	17.959
		Ash -			-	-	-	-	23.66	1.135
		Percentage of as	sh calc	ulated	l on t	he dry	y matt	er,		$6 \cdot 319$
	4.	Sheaths -	-	-	-	-	-	-	1239 · 1	
		Water -		-	-	-		-	1039.6	83.899
		Dry matter	_	-	-	-	-	-	$199 \cdot 5$	16 · 101
		Ash -		-	-	-		-	17.7	1.428
		Percentage of as	sh calc	ulated	l on t	he dry	y mati	ter,		8.872
	5.	Leaves		-					1970 · 2	
	·	Water -	-	-	_		-	-	1578.2	80.103
		Dry matter		•	-	-	-	-	392	19.897
		Ash in part	was a	ccider	ıtallv	lost.				
		P			· J					

TIME.	OBSERVA	TIONS,	PART	s AND	PROPO	RTION	s.		QUANTITIES.	PERCENTAGE.
September 6.			-	-	-	•	-	-	1217 · 2 grs.	
	Wate		-	-	-	-	-	-	$1055 \cdot 8$	86.74
	Dry n	natter	-	-	-	•	-	-	$161 \cdot 4$	$13 \cdot 26$
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	•	-	-	•	-	-	$5 \cdot 95$	$\cdot 489$
	Percentage	e of as	h cal	culate	d on t	he dr	y mat	ter,		3.686
	7. Silks -	-	-	-	-	-	-	-	$260 \cdot 3$	
	Wate	r -	-	-	-	-	-	-	$232 \cdot 3$	$89 \cdot 243$
	Dry n	natter	-	-	-	-	-	-	28.	10.757
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	1.66	$\cdot 638$
	Percentage	e of as	h cal	culate	d on t	he dr	y mat	ter,		5.929
	S. Sheaths of	husk.	s -		-				2019 · 1	
	_	r -			-		-	-	$1624 \cdot 25$	80.444
	Dry m			-	-	-	-		394.85	19.556
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-		-	-	-	-	10.56	•523
	Percentage	e of as	h cal	culate	d on t	he di	v mat	ter.		2.674
	9. Leaves of		_		_		_		466 · 4	
	Wate:		_	-	-	-	-	•	363.	77.83
	Dry n		_	_	_	-	-	•	103 • 4	22 · 17
	Ash	latter	_	_	_	_	-	-	7.73	1.657
	Percentage	of acl	h calc	nlate	l on t	ha dr	z met	tor	7 70	7 · 476
	_							ισι,		7 470
	Relation of		eis ar	ia cob	to ea	ch ot	her:		004.0	40.401
	Kerne	Is •	•	•	-	•	-	-	926.8	42.461
	Cob	•	•	-	-	-	-	-	1255 · 9	57 · 5 39
	Weigh	ht of c	one e	ar -	-	-	•	-	2182.7	100 ·
	1. Kernels of	the la	rgest	ear	-	-	-	-	676.8	
	Water	r -	-	-	-	-	-	-	587.62	8 6·823
	Dry m	atter	•	-	-	-	-	-	89 · 18	$13 \cdot 177$
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	3.31	.488
	Percentage	of asl	h calc	ulated	l on t	he dr	y mat	ter,		3.693
	2. Cob of the	larges	st ear	-				-	1105.9	
	Water	_	•	-					866 · 4	7 8 · 343
	Dry m				-		-	_	239.5	21.657
	$\mathbf{A}\mathbf{s}\mathbf{h}$	•		•	-		-		3.91	·354
	Percentage	of asl	h calc	ulated	l on t	he dr	v mat	ter.		1.632
	Ash tastes							,		
1	0. Roots -		-				-		247.2	
_	Water		-	-		-		_	171.9	$69 \cdot 134$
	Dry m		-	-		-		-	75 ·3	30.866
	,								-	
	${f Ash}$	-	-	-	-	_	-	-	$4\cdot 52$	1.828

TIME.	OBSERVA	TIONS,	FARTS	AND	PROPOR	TION	š.		QUANTITIES.	PERCENTAGE.
September 13.	Corn in the	e adva	nced					els	4,0111111111111111111111111111111111111	I EROBITAGE.
	nearly fu									
	Specific gr	avity o	of keri	nels,	1.055	5966	•			
	Relation of	the o	liffere	nt pa	arts of	the	plant	to		
	each oth	er:								
	\mathbf{T} assel	s	-	-	-	-	-	-	195 grs.	$2 \cdot 239$
	Top st	alk	-	-	-	-	-	-	$709 \cdot 1$	$8 \cdot 145$
	Butt s	talk	-	-	-	-	-	-	1842 · 1	$21 \cdot 169$
	Sheatl	ıs -	-	-	-	-	-	_	$849 \cdot 6$	$9 \cdot 759$
	Leave	S *	-	-	-	-	-	-	$1317 \cdot 2$	$15\cdot 139$
	Stalk	of ear	-	-	-	-	-	-	$373 \cdot 1$	$4 \cdot 285$
	Silks	-	-	-	-	_	-	-	98.8	1.019
	Sheat	ns of l	nusk	-	-	_	-	-	$1021 \cdot 25$	11.730
	Leave	s of h	usk	-	-	-	-	-	$53 \cdot 5$.614
	\mathbf{Kernel}	s of o	ne ear	٠ -	-	-	-	_	$1150 \cdot$	$13 \cdot 309$
	Cob	-	_	-	-	**	-	-	$1096 \cdot 3$	$12 \cdot 592$
	Wajal	st of o	no nlo	n é					8705.95	100:
	Weigh This plant		-		- ho avo	- ra.co	-	-	0100 90	100
	•					_				
	Relation of		ernels	and	cob to	eac	h othe	r :	1470	71 107
	Kerne	ls -	-	-	•	-	-	-	1150*	51.195
	\mathbf{Cob}	-	-	•	-	-	-	-	1096 · 3	48.805
	Weigl	nt of o	ne ear		-	_	_	-	$2246 \cdot 3$	100 ·
	J		ROPORTI				•			
1	. Kernels	_		_		_	_	_	750 ·	
	Water		-	-	-	-	-	-	590·6 5	78.753
	Dry m	atter	-	-	-	-	-	-	$169 \!\cdot\! 35$	$21 \cdot 247$
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-		-	-	-	-	-	$4 \cdot 72$.629
	Percentage	of ash	calcu	lated	d on th	e dr	y matt	er,		$2 \cdot 787$
	Ash tastes									
9	. Cobs -	_	_	-		_		_	$896 \cdot 3$	
~	. Water		-	_	_	_			683 · 2	$76 \cdot 224$
	Dry m		_	_	_			-	213 · 1	23.776
	Ash	-	-				-	-	3.6	•402
	Percentage	of asl	ı calcı	late	d on th	e dr	v matt	er.		1.689
	Ash tastes						-	- /		
3	. Tassels	-	-	_	-		-	-	195	
	Water		-	-	_	-	_	_	$126 \cdot 5$	$64 \cdot 359$
	Dry m		-	-		-	-	-	69.5	35.641
	Ash	-		-	-	_	-		5.83	2.983
	Percentage	of asl	h calci	ılateo	d on th	ie dr	v matt	er.	3 33	8.374
	- creemage	O. 1401					,	,		0 011

TIME.	OBSERVATIONS,	PART	S AND	PROPO:	RTIONS			QUANTITIES.	PERCENTAGE.
September	13. 4. Top stalks -	-	-	•	~	-	-	709·1 grs.	
	Water -	-	*	-	-	-	-	568.48	$80 \cdot 169$
	Dry matter	-	-	-	-	-	-	$140 \cdot 62$	$19 \cdot 831$
	Ash -	-	-	-	-	-	-	$6\cdot 1$.86
	Percentage of asl	ı cal	culated	l on t	he dr	y mat	ter,		4.338
	5. Butt stalks -	-	-	-	-	-	-	$1842 \cdot 1$	
	Water -	-	-	-	-	-	-	$1493 \cdot 35$	$81 \cdot 067$
	Dry matter	-	-	-	-	-	-	348.75	18.933
	Ash -	-	-	-	-	-		$16 \cdot 1$.879
	Percentage of asl	ı cal	culate	d on t	the dr	y mat	ter,		4.617
	6. Sheaths -	-	-	-	-	-	-	$849 \cdot 6$	
	Water -	-	-	-	-		-	$679 \cdot 8$	80.014
	Dry matter	_	-	-	-	-	-	169.8	19.986
	Ash -	-	_	-	-	_	-	15.72	1.85
	Percentage of asl	ı cal	culate	d on t	the dr	v ma	tter,		$9 \cdot 252$
	7. Leaves		_	_	_		_	1317.2	- 202
	Water -	_	_	_	_			$968 \cdot 42$	73.521
	Dry matter	_		_	_	_		348.78	$26 \cdot 479$
	Ash -	-			_			$32 \cdot 1$	2.437
	Percentage of asl	ı cal	culated	l on t	he dr	v mat	ter.	0.2 I	9.203
	8. Stalk of ear -	_	_	_		_	_	$373 \cdot 1$	0 200
	Water -	_	_	_			_	$311 \cdot 26$	83 · 426
	Dry matter	_	_	•	-	_	-	61.84	16.574
	Ash -	-	_	Ī	_	_	_	1.86	•498
	Percentage of asl	r cal	- culate	d on t	ha dr	v mat	ter	1 00	3.008
	_	ı car	caracc	a on t	inc ai	y mac	,,	00.00	9 000
	9. Silks	-	•	-	-	•	-	98·80	OB 771
	Water -	-	-	-	-	-	-	86.5	87.551
	Dry matter	-	-	-	-	•	-	12.3	12 · 449
	Ash -	- 1 1	-	3	• 41		-	.66	$egin{array}{c} \cdot 668 \ 5 \cdot 366 \end{array}$
	Percentage of as	n cai	cuiaze	a on	tne ar	y ma	tter,		9.900
	10. Sheaths of husks	-	-	-	-	-	-	$1021 \cdot 25$	TIO TOO
	Water -	-	-	-	-	•	-	812.2	79.529
	Dry matter	-	-	•	-	-	-	$209 \cdot 05$	20.471
	Ash -				. •	•	•	$6 \cdot$	•587
	Percentage of as	h cal	culate	d on 1	the dr	y mai	ter,		$2 \cdot 865$
	11. Leaves of husks	-	-	-	-	-	-	5 3·5	
	Water -	-	-	-	-	-	-	38.65	72.243
	Dry matter	-	-	-	-	-	-	14.85	27.757
	Ash •	•				-	-	1 · 16	2.168
	Percentage of as	h cal	culate	d on t	the dr	y mat	ter,		7.811

TIME.	OBSERVATIONS, PART	s AND	PROPO	RTIONS	·.		QUANTITIES.	PERCENTAGE.					
OCTOBER 18.	Corn ripe. The amo												
	leaves and sheaths,												
	since the 13th of 3												
	have gradually inc	have gradually increased in specific gravity											
	since their first app												
	Relation of the differ	Relation of the different parts of the plant to											
	each other:												
	Tassels	-	-	-	•	-	133 · grs.	1.068					
	Top stalk -	•	•	•	-	-	1026 ·	$8 \cdot 239$					
	Butt stalk •	-	-	•	-	•	$2786 \cdot$	$22 \cdot 375$					
	Sheaths	-	•	-	-	-	744 ·	$5 \cdot 975$					
	Leaves	-	•	•	-	-	$1584 \cdot$	12.721					
	Sheaths of husk	s -	-	-	-	-	$763 \cdot$	$6 \cdot 126$					
	Stalks of ears	-	-	-	-	•	$299 \cdot$	$2 \cdot 401$					
	Silks	-	•	-	-	-	81.	•651					
	Roots	-	-	-	•	-	556	$4 \cdot 465$					
	Kernels	•	-	-	-	-	$3468 \cdot$	27.852					
	Cob	-	-	-	•	-	1012.	8.127					
							12452	100.					
	Number of kernels or	ı the	above	ear,	180.								
	Average weight of ea	ich ke	ernel	-	-	-	$7 \cdot 225$						
	Specific gravity of ke												
	the plant, 1.233378	3.											
	Specific gravity of ke	rnels	after	being	depri	ved							
	of water, 1.2753.												
	Depriving the kernel				his c	ase,							
	has increased their	speci	fic gra	avity.									
	Relation of the kerne	ls to t	he co	b:									
	Kernels	-	-	-	-	-	$3468 \cdot$	77.411					
	Cob	-	•	~	-	-	1012.	$22 \cdot 589$					
							4480	100					
	PROPOR'	TIONS.					4400	100					
	1. Kernels	-	-	_	-	-	1000						
	Water	-	-	-	-	-	374.6	$37 \cdot 46$					
	Dry matter -	•	-	-	-	-	$625 \cdot 4$	$62 \cdot 54$					
	Ash	•	-	-		-	8.06	.806					
	Percentage of ash cal-	culate	d on	the dr	y mat	ter,		1.288					
	2. Cob	_			-		<i>5</i> 06·						
	Water		_	_	_	-	295	$54 \cdot 348$					
	Dry matter -			-		_	211.	45.652					
	Ash			_		-	2.94	.581					
		culate	d on t	he dr	v mat	ier.	~ 01						
	-				-	,		1 000					
	Percentage of ash call Ash tastes decidedly				-	ter,	~ UX	1.393					

стовек 18.	It will be	noticea rom the								
	large p					Journ	.a 101	me		
	3. Tassels	-	-	-	-	-	-	-	133 · grs.	
	Wate	er -	-	-	-	•	-		$54 \cdot 90$	41.278
	\mathbf{Drv}	natter	-	-	-	-	-		78.1	58.725
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	$9 \cdot 20$	6.91
	Percentag	e of asl	ı calc	culate	d on t	he dr	y mat	ter,		11.77
	4. Top stalk	: -	-	-	-	<u>-</u>	-	-	1026.	
	Wate	er -	-	-	-	-	-	-	$866 \cdot 3$	$84 \cdot 43$
	Dry :	matter	-	•	-	-	-	-	$159 \cdot 7$	15.56
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	$11 \cdot 25$	1.09
	Percentag	e of asl	ı calc	culate	d on t	the dr	y mat	ter,		$7 \cdot 04$
	5. Butt stal	k -	-	-	-	-	-	-	2786	
	Wate	er -		-	-	-	-	-	$2425\!\cdot\!95$	87.07
	Dry 1	natter	-	•	•	•	-	•	$360\cdot 05$	$12 \cdot 92$
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	•	-	-	-	-	$26 \cdot 65$	•95
	Percentag	e of ash	calc	ulate	d on t	he dr	y mat	ter,		$7 \cdot 12$
	6. Sheaths	-	-	-	-	-	-	-	744 ·	
	Wate	er -	-	-	-	-	-	-	$525 \cdot 7$	70.65
	\mathbf{Dry}	matter	-	-	-	-	-	-	$218 \cdot 3$	$29 \cdot 34$
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	$22 \cdot 5$	$3 \cdot 02$
	Percentag	e of asl	ı calo	culate	d on t	he dr	y mati	ter,		10.30
	7. Leaves -	•	-	-	-	-	-	•	1584	
	Wate	er -	-	-	-	-	-	-	948.7	59.89
	Dry i	natter	-	-	-	-	-	-	$635 \cdot 3$	$40 \cdot 10$
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	•	-	-	-	-	-	79.8	5.03
	Percentag	e of ash	ı calc	culate	d on t	he dr	y mat	ter,		12.56
	8. Sheaths of	-	· •	-	-	-	-	•	7 63·	
	Wate		-	-	-	-	-	•	529 · 2	69.35
	•	natter	•	-	-	-	-	•	233.8	30.64
	Ash	•	•.	•	. •	· ·	-	-	$13 \cdot 46$	1.76
	Percentag	e of ash	ı calc	ulate	l on t	he dr	y mati	ter,		5.75
	9. Ear-stalk		-	-	-	-	-	-	399	
	Wate		•	-	-	-	-	-	$354 \cdot 6$	88.87
	_	matter	•	-	-	-	-	-	44.4	11 · 12
	Ash	•	•		. •	-	-	-	1.28	•32
	Percentag	e of asl	n calc	culate	d on	dry m	atter	-		2.88

OCTOBER 18. 10. Silks	TIME.	OBSERVATIONS, PARTS AND PROPORTIONS.		QUANTITIES.	PERCENTAGE.
Dry matter 14 · 6 18 · 025 Ash · · · · · · · · · · · · ·	OCTOBER 18.	10. Silks		81 · grs.	
Ash		Water		_	81.975
Percentage of ash calculated on the dry matter, Ash saline, with a slight taste of caustic potash. 11. Nodes or joints of stalk 257.6 Water 216.35 83.987 Dry matter 41.25 16.013 Ash 2.8 1.087 Percentage of ash calculated on dry matter - 6.787 Ash saline, with a slight taste of caustic potash. 12. Roots 556. Water 427.6 76.906 Dry matter 128.4 23.094 Ash 553 .953		Dry matter	-	14.6	$18 \cdot 025$
Ash saline, with a slight taste of caustic potash. 11. Nodes or joints of stalk 257.6 Water 216.35 83.987 Dry matter 41.25 16.013 Ash 2.8 1.087 Percentage of ash calculated on dry matter - 6.787 Ash saline, with a slight taste of caustic potash. 12. Roots 556. Water 427.6 76.906 Dry matter 128.4 23.094 Ash 5.3 .953		Ash	-	•99	$1 \cdot 222$
11. Nodes or joints of stalk 257.6 Water 216.35 83.987 Dry matter 41.25 16.013 Ash 2.8 1.087 Percentage of ash calculated on dry matter - 6.787 Ash saline, with a slight taste of caustic potash. 12. Roots 556. Water 427.6 76.906 Dry matter 128.4 23.094 Ash 5.3 .953	•	Percentage of ash calculated on the dry m	atter,		6.781
Water 216·35 83·987 Dry matter 216·35 16·013 Ash 2·8 1·087 Percentage of ash calculated on dry matter - 6·787 Ash saline, with a slight taste of caustic potash. 12. Roots 556· Water 427·6 76·906 Dry matter 128·4 23·094 Ash 5·3 ·953		Ash saline, with a slight taste of caustic 1	potash.		
Dry matter 41·25 16·013 Ash 2·8 1·087 Percentage of ash calculated on dry matter - 6·787 Ash saline, with a slight taste of caustic potash. 12. Roots 556 Water 427·6 76·906 Dry matter 128·4 23·094 Ash 5·3 ·953		11. Nodes or joints of stalk	. <u>-</u>	257.6	
Ash 2.8 1.087 Percentage of ash calculated on dry matter - 6.787 Ash saline, with a slight taste of caustic potash. 12. Roots 556 Water 427.6 76.906 Dry matter 128.4 23.094 Ash 5.3 .953		Water		$216 \cdot 35$	$83 \cdot 987$
Percentage of ash calculated on dry matter - 6.787 Ash saline, with a slight taste of caustic potash. 12. Roots 556. Water 427.6 76.906 Dry matter 128.4 23.094 Ash 5.3 .953		Dry matter	-	41.25	16.013
Ash saline, with a slight taste of caustic potash. 12. Roots 556 Water 427.6 76.906 Dry matter 128.4 23.094 Ash 5.3 .953		Ash	•	2.8	1.087
12. Roots 556· Water 427·6 76·906 Dry matter 128·4 23·094 Ash 5·3 ·953		Percentage of ash calculated on dry matte	er -		6.787
Water 427·6 76·906 Dry matter 128·4 23·094 Ash 5·3 953		Ash saline, with a slight taste of caustic p	otash.		
Dry matter 128·4 23·094 Ash 5·3 953		12. Roots		$556 \cdot$	
Ash 5·3 ·953		Water		$427 \cdot 6$	$76 \cdot 906$
		Dry matter	• /	$128 \cdot 4$	$23 \cdot 094$
Percentage of ash calculated on the dry matter, 4.128		Ash	-	$5 \cdot 3$	$\cdot 953$
		Percentage of ash calculated on the dry m	atter,		$4 \cdot 128$

COMPOSITION OF MAIZE OR INDIAN CORN.

The variety of corn, the analysis of which follows, is the small 8-rowed yellow corn, intermediate in size between the Canada corn and the 8-rowed white-flint. It was planted early in May, and was nearly ripe the 11th of August. It was strongly glazed at this time, and filled all the spaces upon the cob; but the chit shrank some on drying. I am particular to mention the variety examined, inasmuch as varieties differ in composition, both in the proximate elements and in the composition of the ash. Starch, in some varieties, is a very prominent element, and the phosphates in some occur in greater quantities than in others, while some varieties contain more oil than others. So during the periods of growth, the elements continually vary, and yet indian corn has a composition peculiarly its own.

I am not sure that I have obtained more than a proximate determination of the composition of this cereal; still, I have obtained many facts of some importance, which I shall proceed to lay before the reader. One of the great difficulties met with in the outset of the examination, or analysis of corn, is the want of a good ash. It is one of the most difficult of substances to burn. It is quite fusible when combustion has gone so far as to form an ash of a part of the material; and when this fusion occurs, it is better to begin anew, than to proceed farther with the matter. The ash, however, may be obtained in a perfectly white color, and free from coal, by proceeding in a certain manner. It should be burned at a low temperature, in a wide-mouthed crucible or capsule, and sufficient time be given it to consume. After the first part of the combustion, or the oil has been burnt off, it may be kept at a low red heat; but if it becomes quite red from a slight inattention,

it is very liable to end in fusion. The best mode of proceeding, when the ash begins to appear, is to remove it from the fire which is in common use, and which usually requires to be kept in a lively state while the work is going on; and then at the approach of evening, when the heat of the stove is considerably diminished, to replace it, and suffer it to remain during the night. In the morning it will be found to have made considerable progress. Proceeding in this way for a week or more, a good ash for analysis will be obtained. Another requisite is a perfectly dry state of the corn: at least, very dry corn has burnt better than that which was moist at the beginning. The same mode of procedure will be successful in wheat and its flour, or indeed in any of the cereals. Oats and barley, however, are not very difficult to burn.

1. Analysis of the ash of the leaf of the small 8-rowed corn: cut August 4.

Brisk effervescence of the ash on the addition of acids.

									Per centum.
Silica	-	•	-	-	-	•	-	-	$27 \cdot 375$
Earthy	phosp	hates	-	-	-	-	-	-	23.780
Carbon	ate of	lime	-	-		-	-	-	0.500
Magnes	sia	-	-	-	-	-	-		0.365
${\bf Potash}$	-	-	-	•	•	-	-		$22 \cdot 825$
Soda	•	-	-	-	-	-	-	-	5.845
Chlorin	e	-		-	-		-	-	1.750
Sulphu	ric aci	d	-		•		-	-	2.580
Carbon	ic acio	d	-	-	-	-	-	-	10.615
Organic	e matt	er	•	-	•	-	-	-	4.000
									99.665

2. Analysis of the ash of the corn-leaf, cut August 11; the results of which may be stated as follows:

Quantity	_	_	_	_	_	_		20.000 grs.	
Quality	-	-	-	-	•	-	•	20 000 grs.	-
~									Per centum.
Silica	-	-	-	-	-	-	-	$5 \cdot 475$	$27 \cdot 375$
Phosphates	s of l	ime,	iron	and m	agne	sia	-	4.756	$23 \cdot 780$
Carbonate	of li	me	-	-	-	-	-	0.460	1.840
Magnesia	-	-	-	-	-	-	-	0.078	0.390
Potash	-	-	-	-	-	-	-	4.565	20.825
Soda	•	-	-	-	-	-	-	1.169	5.845
Chlorine	-	-	-	-	-	ь	-	0.350	1.750
Sulphuric a	acid	-	-	-	-	-	-	0.516	2.580
Carbonic a	.cid	-	-	-	-	-	-	$2 \cdot 123$	10.615
Organic m	atter	-	-	-	-	-	-	0.500	2.500
Soluble sil	ica	-	-	-	-	•	-	0.080	0.400
								20.0%2	071 000
								$20 \cdot 072$	$97 \cdot 900$

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3. Analysis of the ash of the corn-sheath of the 8-rowed yellow corn, cut August 11.

Calculated without carbonic acid or organic matter.

						Per centum.
Silica	_	-	-	_	æ	20.000
Phosphates of lime, m	agn	esia an	d iron	a	-	18.750
Carbonate of lime -	•	-	-	-	-	0.900
Magnesia	•	-	-	-	-	0.040
Potash	-	-	-	-		$24 \cdot 405$
Chloride of potassium	-	-	**	-		14.051
Chloride of sodium	-	-		-	-	0.469
Sulphuric acid -	•	-	•	-	-	$3 \cdot 120$
						81.735

4. Analysis of the corn-stalk, cut August 11.

Calculated without carbonic acid.

								Per centum.
Silica -	-	-	۰	-	-		-	$6 \cdot 457$
Phosphates	of lim	ie, m	agnes	ia and	l iron	-	-	$25 \cdot 625$
Carbonate o	f lime	e	-	-	-	-	-	0.600
Magnesia	-	-	-	-	-	-	-	0.050
Potash -	-	•	-	-	-		-	40.540
Chloride of	potass	sium	-	-	-	-	-	$8 \cdot 355$
Chloride of	sodiu	m	-	-			-	$3 \cdot 430$
Sulphuric a	cid	-	-	-	-		-	$6 \cdot 190$
Organic ma	tter	•	-	-	-	-	-	$8 \cdot 134$
								00.001
								98.881

5. Analysis of the husks, cut August 11.

Calculated without organic matter or carbonic acid.

								Per centum.
Silica -	•	-	-	-	-	-	-	$32 \cdot 137$
Phosphates .	of li	me, n	nagne	sia ar	id iron	•	-	$29 \cdot 431$
Carbonate of	f lin	ne		-	-	-	-	0.271
Magnesia	-	-	-	-	-	-	-	0.239
Potash -	-	-	-	-	-	-	-	21.854
Soda -	-	-	-	-	-	-	-	$2 \cdot 047$
Chlorine	-	-		-	-	-	-	1.117
Sulphuric ac	id	-	-	-	-	-	-	11.117
•								
								$98 \cdot 213$

6. Analysis of the cob of the same date.

Calculated without organic matter or carbonic acid.

									Per centum.
Silex	-	-	-	-	-	-	-	-	10.768
Phosph	ates	of the	earth	s and	iron	-	-	-	$36 \cdot 570$
Lime	-	-	-	-	-	-	-	-	0.244
Magnes	ia	-	-	-	-	-	-	-	0.539
Potash	-	-	-	-	-	-	-	-	37.855
Soda	-	-	-	-	-	-	-	-	1.837
Chlorin	e	-	-	-	-	-	-	-	$2 \cdot 955$
Sulphui	ric ac	cid	-	-	-	-	-	-	$9 \cdot 200$
									$99\!\cdot\!968$

7. Analysis of the kernels of the same date, and borne upon the same stalk and cob.

								Per centum.
Silex -	-	-	-	-	-	-	-	1.730
Phosphates o	f the	earth	s and	alkal	lies	-	-	52.759
Lime -	-	-	-	-	-	-	-	trace.
Magnesia	-	-	-	-	-	-	-	trace.
Potash -	-	-	-	-	-	-	-	27:357
Soda -	-	-	-	-	-	-	•	5.796
Chlorine	-	-	-	-	-	-	-	$4 \cdot 101$
Sulphuric aci	id	-	-	-	-	•	-	$3 \cdot 485$
								95.228
								JU 220

The small quantity of ash employed in this analysis gave only feeble traces of lime and magnesia.

8. Organic analysis of corn of the same date and stage of growth.

								Per centum.
Starch -	-	-	-	-	-	-	-	$30 \cdot 186$
Sugar -	-	-	-	-	-	-	-	4.968
Albumen	-	-	-	-	-	-	-	1.969
Casein -	-	-	-	-	-	-	-	3.897
Oil -	-	-	-	-	-	-	-	$4 \cdot 415$
Fibrin -	-	-	-	-	-	-	-	$19 \cdot 150$
Dextrine	-	-	-	-	-	-	-	1.950
Gluten -	-	. -	-	-	-	-	-	1.795
Water -	-	-	-	-	-	-	-	27:300
								$95 \cdot 630$

9. Analysis of the phosphates of the cob.

Рноѕрнать	es -	-	-	-	-	-	2.840
Soluble silica -	_	_		_	_		0.020
Lime			-	-		-	0.250
Magnesia ·		-	-	-	-	-	0.579
Phosphate of p		of ire	on	-	-	-	$0 \cdot 120$
Phosphoric acid	l -	-	-	-	-	-	1.871
							2.840

On consulting the proportion of inorganic or earthy matter in the cob, it will be seen that it is small; still when the question respecting the nutritive value of the cob is discussed, it is proper to take this into account. It is particularly rich in phosphates and other alkalies: so we have an additional reason for feeding the cob in conjunction with the grain.

10. Analysis of the cuticle of the small 8-rowed yellow corn*.

								Per centum.
•	-	-	-	•	-	-	-	3.000
tes c	of the	earth	s and	iron	-	-	-	$54 \cdot 200$
tes o	of the	alkal	ies	-	-	-	-	$12 \cdot 100$
	-	-	-	-	-	-	-	0.050
a	-	-	-	-	-	-	-	0.010
	-	-	-	-	-	-	-	$22 \cdot 570$
	-	-	-	-	-	-	-	6.510
c ac	id	-	-	-	-	-	-	1.400
	-	-	-	-	-	-	-	none.
aci	d	-	-	-	-	-	-	none.
								$99 \cdot 790$
	tes o	tes of the - a -	es of the alkal	tes of the alkalies a c acid	a	es of the alkalies	es of the alkalies	es of the alkalies

11. Organic analysis of the dry cob of the small 8-rowed yellow corn: cut August 11.

_		-				-		
Insoluble ma	tter or	fibrin	ı	-	-			Per centum. 86.74
Albumen	-		-	-	-	-	-	1.94
Casein -	-	-		-	-	-	-	0.64
Dextrine or	gum	-	-	-	-	-	-	1.21
Sugar and e	xtract	-	-	-	-	-	-	$7 \cdot 25$
Oil and resin	ı -	-	-	-	-	-	-	1.00
								98.78

The amount of soluble matter in the cob exceeds what was expected when I commenced the analysis; and it may still exceed considerably what is here stated, inasmuch as it is

^{*} Obtained by sifting out the cuticle from coarsely ground corn : hence not entirely free from the farinaceous matter.

affected by the time occupied in washing. It shows, however, that the cob is not by any means destitute of value; and that the practice of many farmers, of grinding the cob with the corn for feeding stock, is the most economical way of disposing of it. There is still another reason for the practice, namely, to increase the bulk of the food; for it is unsafe to feed concentrated nourishment to herbivorous animals. They require a bulky food, for the purpose of effecting a moderate distension of the alimentary canal; and hence if the cob was less nutritive than is here represented, it would still be the proper plan to be pursued in feeding cattle or horses, though not so necessary to the latter as to the former.

It will be observed particularly that the nitrogenous compounds amount to 2.58, and the gum, sugar, extract and oil, to 9.46; and if the 86 per centum of insoluble matter had been subjected to the action of fluids equal in solvent power to the gastric juice, this would still have been diminished some four or five per centum.

12. Analysis of the leaves of Pennsylvania Dent corn.

									Per centum.
Silica -	-	-	-	-	-	-	-	9.900	49.500
Phosphates	s -	-	-	-	-	-	-	3.300	16.500
Lime -	-	-		-	-	-	-	1.297	$6 \cdot 485$
Magnesia	-	-	-	-	-	-	-	0.360	1.800
Potash	-	•	•	-	-	-	-	$3 \cdot 360$	16.800
Soda -	-	-	-	-	-	-	-	0.350	1.750
Sulphuric	acid	-	-	-	-	-	-	0.822	4.510
Chlorine	-		-	-	-	-	-	0.463	$2 \cdot 315$
								$19 \cdot 852$	$99 \!\cdot\! 660$

13. Analysis of the stalk of a variety mixed with the Tuscarora, White-flint and Sweet corn: cut August 11.

Slight effervescence of the ash with acids*.

									Percentage without organic matter and carbonic acid.			
Silica -	-	-	-	-	-	-	-	$2 \cdot 325$	$12 \cdot 398$			
Phosphates of lime, magnesia and iron - 3.260 17.386												
Carbonate	of li	me	-	-	-	-	-	1.250	6.663			
Magnesia	-	-	-	-	-	-	-	0.025	0.131			
\mathbf{Potash}	-	-	-	-	-	-	-	8.615	$45 \cdot 949$			
Soda -	•	-	-	-	-	-	-	0.916	4.882			
Sulphuric	acid	-	-	-	-	-	-	0.561	$2 \cdot 989$			
Chlorine	-	-	-	-	-	-	-	1.800	$9 \cdot 596$			
Organic m	atte	r and	carbo	nic ac	id	-	-	1.020				
								$19 \cdot 772$	$99 \cdot 994$			

^{*}This variety, before ripening, was supposed to be the Pennsylvania Dent. It has the red cob of the Tuscarora, and about ‡ of the kernels were of this kind.

14. Analysis of the ash of the sheaths of the same variety of corn.

									Percentage without organic matter or carbonic acid.
Silica -	-	-	-	-	-	-	-	$6 \cdot 205$	$33 \cdot 326$
Phosphates	of	lime,	magr	nesia a	and in	on	-	3.712	$19 \cdot 932$
Lime -	-	~	-	-	-	-	-	0.312	1.671
Magnesia	-	-	-	-	-	-	-	0.361	$1 \cdot 938$
Potash	-	•	-	-	-	-	-	6.945	$37 \cdot 298$
Soda -	-	-	-	-	-	-	-	0.474	$2 \cdot 541$
Sulphuric a	ıcid		-	-	-	-	-	0.399	$2 \cdot 137$
Chlorine	-	-	-	-	-	-	-	0.213	$1 \cdot 142$
Organic ma	tte:	r and c	arbor	ic aci	d -	-	-	0.478	
								$19 \cdot 099$	$99 \cdot 985$

15. Analysis of the ash of the husks of the mixed corn, Tuscarora Sweet and White-flint.

							Per centum.
Silica	-	-	-	-	-	-	36.331
Earthy phosphate	s	-	-	-	-	-	18.060
Lime	-	-	-	-	-	-	1.053
Magnesia -	-	-	-	-	-	-	0.205
Potash	-	-	-	-	-	-	21.061
Chloride of potas	sium	-	-	-	-	-	19.587
Soda	-	-	-	-	-	-	0.953
Sulphuric acid	-	-	-	-	-	-	1.105
Chlorine -	-	-	-	-	-	-	1.316
							99.671

16. Analysis of the ash of the kernels of the foregoing mixed corn.

							Per centum.	Elements in one bushel.
Silica -	-	-	-	-	-	-	2.417	0·333 oz.
Earthy phosphat	es	-	-	-	-	-	45.839	6.600
Potash -	-	•	-	-	-	-	$32 \cdot 726$	4.616
Soda	-	-	-	•	-	-	$4 \cdot 454$	0.614
Lime	-	-	-	-	-	-	0.160	$0 \cdot 022$
Magnesia -	-	-	-	-	-	-	0.080	0.011
Phosphoric acid	-	-	-	•	-	-	$14 \cdot 330$	1.977
Chlorine -	-		-	-	-	-	none.	
Carbonic acid	-	-	-	-	-	-	n o ne.	
Sulphuric acid	-	-	-	-	-	-	none.	
								
							$100 \cdot 006$	14·173 oz.

The ash contained between four and five per centum of organic matter, which was not reckoned.

17. Analysis of the ash of the cob of the same corn.

								Per centum.
Silica -	-		-		•		10	17.626
Earthy phos	sphate	es	-	-	•	-	-	$16 \cdot 140$
Phosphates	of th	e alk	alies	-	-	-	-	4.693
Magnesia	-	-	-	-	-	•		0.240
Lime -	-	-	-	•	-	-	-	0.250
Potash -	-	-	-	-	-	-	-	29.693
Soda -	-	•	•	-	-	-	-	8.768
Chlorine	-	-		-	-	-	-	10.212
Sulphuric a	cid	-	-	•			•	11.332
-								$\frac{-}{98 \cdot 954}$

The analyses of the Rocky Mountain corn, of the Pennsylvania Dent sweet corn, and of the middle size 8-rowed yellow corn, were made from crops which grew in 1847. The analyses were conducted with as much care as in any of the subsequent cases; but the ash or materials were not in a condition as pure, and free from coal as the latter. As a whole, too, there are more discrepancies in the results, especially in the grain; and greater losses were sustained, in consequence of leaving out of the determination the phosphates of the alkalies. Still I am satisfied that the general results are worthy of confidence, and hence ought not to be omitted in consequence of a loss in the analysis, which seems to be too great for a work which is conducted with care, and after the most approved modes. It is, however, mostly in the grain that these remarks apply, and it would have been easy to have supplied the losses by calculation; but the rule which has been adopted, is, to note the actual results, and to take nothing for granted, unless it has been for organic matter or carbonic acid; and these have in most instances been determined, for the purpose of confirming the accuracy of results, or for satisfying our minds as to the correctness of the whole work.

THE ROCKY MOUNTAIN CORN.

This variety has its kernels enveloped in a husk. The cob is loose and spongy. After a few years' cultivation, its husk disappears, and it differs in no important point from some one of the common varieties now under cultivation. There are two kinds of the Rocky Mountain corn. The one removed the fartherest from the cultivated varieties, has its kernels arranged on a panicle like oats.

I. ANALYSIS OF THE ROCKY MOUNTAIN CORN.

Kernels arranged on a cob or axis of growth. Grown in Cortland county, by Nathan Salisbury, Esq. Growth large. The crop was perfectly ripened.

	-	-	•	-				
		1.	Stall	es.				
Silica	-	-	-	-			1.375	
PHOSPHATES:					•		-	
Phosphate of peroxide of iron -						$2 \cdot 375$		
Lime -	-	-	•	-	-	0.719		
Magnesia	tu tu	•	•	-	-	0.060		
Silicic acid	-	-	-	•	-	1.250		
Phosphoric a	acid	-	•	-	-	3.875		
							7.275	
Lime	-	-	-	-	-	-	1.142	
Magnesia -	-	-	•	-	-	-	0.175	
Potash	-	•	-	•	•	-	52.977	
Soda	-	-	-	-	-	-	4.713	
Chlorine -	-	-	-	-	- ,	-	0.869	
Sulphuric acid	•	-	-	-	-	•	0.361	
Carbonic acid	-	-	-	-	-	-	26.000	
Organic matter	•	-	~	•	-	•	$2 \cdot 235$	
							${98 \cdot 132}$	S
							JO 102	ν.
	_	r		, ,	. 7			
	2.	Leav	es and	t shea	ths.			
Silica	-	-	-	-	-	•	14.000	
PHOSPHATES:								
Phosphate o	f per	oxide	of ir	on	•	3.125		
Lime -	-	-	-	-	-	4.216		
${f Magnesia}$	-	-	-	•	-	0.075		
Silicic acid	-	•	-	•	-	8.350		
Phosphoric :	acid	-	-	-	•	5.684		
							21.450	
Lime	•	•	•	-	-	-	$5 \cdot 145$	
Magnesia -	-	-	*	a	-	-	2.080	
Potash	-	-	-	-	•	-	39.124	
Soda	•	-	-	-	-	-	3.152	
Chlorine -	-	-	-	-	-	-	0.555	
Sulphuric acid	-	-	-	-	-	-	0.652	
Carbonic acid	-	•	-	-	-	-	10.000	
Organic matter	•	-	-	-	-	-	3.450	
							99.608	S.

3.	Cobs	and	husks	of	the	kerneis.

Silica PHOSPHATES:	•		-	•	•	-	11.950	
Phosphate o	f ner	oxide	of in	on		6.650		
Lime -	-	•	•		-	2.058		
Magnesia	-				•	0.410		
Silicic acid	-			•	_	0.350		
Phosphoric :		-		•		$9 \cdot 482$		
							18.950	
Lime	-	-				-	0.140	
Magnesia -	-	•	-	•	-	•	0.270	
Potash	•	-	-	-	•	•	$52 \cdot 060$	
Soda	-	-	•	-	-	-	$9 \cdot 202$	
Chlorine -	-		-		-	•	3.034	
Sulphuric acid	-	-			-	-	0.722	
Chloride of sodiu	ım	-	-	-	•	-		
Carbonic acid	•	-	-	•		=	trace.	
Organic matter	-	•		-	6	•	$2 \cdot 100$	
Coal	-	•	-	•	-	-		
							00. 400	~
							98.428	S.
		4.	Kern	els.				
Silica	-	4.	Kern -	els. -			3.050	
		4.	Kern -	els. -	-	v	3.050	
PHOSPHATES:	- f pei	-	•	-	-	1.450	3.050	
	- f per	-	•	-		1·450 5·358	3.050	
Phosphates: Phosphate o Lime -	- f per	-	•	-	-		3.050	
PHOSPHATES: Phosphate o Lime	- f per	-	•	-	-	5·35 8	3.050	
PHOSPHATES: Phosphate o Lime - Magnesia - Silicic acid		-	•	- ron -	-	5·358 14·650	3.050	
PHOSPHATES: Phosphate o Lime - Magnesia -		-	•	- ron -	-	5·358 14·650 trace.	3·050 42·150	
PHOSPHATES: Phosphate o Lime - Magnesia - Silicic acid		-	•	- ron -		5·358 14·650 trace.		
PHOSPHATES: Phosphate o Lime Magnesia - Silicic acid Phosphoric		-	•	- ron -	-	5·358 14·650 trace.	42.150	
PHOSPHATES: Phosphate o Lime - Magnesia - Silicic acid Phosphoric:		-	•	- ron -	-	5·358 14·650 trace.	42·150 0·085	
PHOSPHATES: Phosphate o Lime - Magnesia - Silicic acid Phosphoric: Lime - Magnesia -		-	•	- ron -	-	5·358 14·650 trace.	42·150 0·085 0·050	
PHOSPHATES: Phosphate o Lime - Magnesia - Silicic acid Phosphoric: Lime - Magnesia - Potash -		-	•	- ron -	-	5·358 14·650 trace. 20·692	$42 \cdot 150$ $0 \cdot 085$ $0 \cdot 050$ $24 \cdot 350$	
PHOSPHATES: Phosphate of Lime Magnesia - Silicic acid Phosphoric: Lime - Magnesia - Potash - Soda -	acid - - -	-	•	- ron -	-	5·358 14·650 trace. 20·692	$42 \cdot 150$ $0 \cdot 085$ $0 \cdot 050$ $24 \cdot 350$	
Phosphates: Phosphate of Lime Magnesia - Silicic acid Phosphoric: Lime - Magnesia - Potash - Soda - Chlorine - Sulphuric acid Chloride of sodia	acid - - - -	roxide	•	- ron -	-	5·358 14·650 trace. 20·692	42·150 0·085 0·050 24·350 3·394	
PHOSPHATES: Phosphate of Lime - Magnesia - Silicic acid Phosphoric: Lime - Magnesia - Potash - Soda - Chlorine - Sulphuric acid	acid - - - -	roxide	•	- ron -	-	5·358 14·650 trace. 20·692	42·150 0·085 0·050 24·350 3·394 1·210	
Phosphates: Phosphate of Lime Magnesia - Silicic acid Phosphoric: Lime - Magnesia - Potash - Soda - Chlorine - Sulphuric acid Chloride of sodia	acid - - - -	roxide	•	- ron -		5·358 14·650 trace. 20·692	$42 \cdot 150$ $0 \cdot 085$ $0 \cdot 050$ $24 \cdot 350$ $3 \cdot 394$ $1 \cdot 210$ $1 \cdot 020$	
PHOSPHATES: Phosphate of Lime Magnesia - Silicic acid Phosphoric: Lime - Magnesia - Potash - Soda - Chlorine - Sulphuric acid Chloride of sodio Carbonic acid	acid - - - -	roxide	•	- ron -		5·358 14·650 trace. 20·692	42·150 0·085 0·050 24·350 3·394 1·210 1·020 trace.	
PHOSPHATES: Phosphate of Lime Magnesia - Silicic acid Phosphoric: Lime - Magnesia - Potash - Soda - Chlorine - Sulphuric acid Chloride of sodio Carbonic acid Organic matter	acid - - - -	roxide	•	- ron -		5·358 14·650 trace. 20·692	42·150 0·085 0·050 24·350 3·394 1·210 1·020 trace. 6·500	

II. ANALYSIS OF THE PENNSYLVANIA DENT CORN.

The plant had just past its period of flowering. Grown upon Mr. Bement's farm, Albany county. Soil a sandy loam.

1. Leaves and sheaths.

Silicic acid -	•	-	-	-				53.450	
Phosphates:									
Phosphate of	per	oxid	e of	iron	-		0.350		
Lime -	•		•	-			1.466		
Magnesia -	۰		-	•	•		0.150		
Silicic acid	-		-	-	-		0.250		
Phosphoric a	cid -		-	•	•		$12 \cdot 184$		
•								$14 \cdot 400$	
Lime	-	-	-			-	-	$3 \cdot 468$	
Magnesia -		-	-	-		-	-	0.320	
Potash	-	•	-	-		-	-	$12 \cdot 335$	
Soda	-	-	-	-		-	-	5.860	
Chloride of sodium	m	•		-		-	-	$5 \cdot 450$	
Sulphuric acid	•	-	-	-			-	0.670	
Carbonic acid	•	**	-	-		-	-	trace.	
Organic matter	-	-	-			-	-	1.050	
<u> </u>									
								97·003 S.	

2. Stalks, of the same date as the leaves.

From Bement's 3-hill farm. Soil a sandy loam. Growth very large.

Silica	-	•	-		-	•	•	-		-		-	2.385	
PHOSPHA'	res:													
\mathbf{Pho}	sphat	e of	pe	roxi	ide o	of i	ron		-		0	$\cdot 625$		
$_{ m Lim}$	е	-		-	-		•		-		3	$\cdot 497$		
Mag	nesia	ı -		•	-		-		-		0	$\cdot 075$		
Silic	ic ac	id		-	-		-		-		5	.075		
\mathbf{P} hos	sphor	ic a	id	-	-				-		19	$\cdot 785$		
	•												29.075	
Lime -		-	-		-	-		-		-		•	2.025	
Magnesia	ι .	•	•		-	-		-		-		-	trace.	
Potash -	-		-			-		-		-		-	$44 \cdot 943$	
Soda -	-		-			-				•		-	0.005	
Chloride	of so	diur	n					•		-		-	10 ·190	
Sulphurie	c acid	1	-		•	-		-		-		-	0.283	
Carbonic	acid		-			-		-		•	•	-	trace.	
Organic :	matte	er	-		-	-		-		-		-	9.750	
													98.638	S.

3. Husks of the same corn.

Silica PHOSPHATES:	-	6				•	-	21.000	
Phosphate of	of perc	oxide	of ire	on	-		3.200		
Lime			-	-	-		6.092		
Magnesia -	• .		•	-	•		0.200		
Silicic acid	•		•	-	-		0.700		
Phosphoric :	acid ·		-	-	-		26.608	0 m . 000	
.								37.608	
Lime	-	-	-	-		-	-	0.280	
Magnesia -	•	-	-	-		-	-	0.600	
Potash	-	•	-	-		-	-	27.860	
Soda	-	-	-	-		•	-	$9 \cdot 400$	
Chloride of sodiu	ım	-	-	-		-	-	0.043	
Sulphuric acid	-	-	-	-		-	-	1.244	
Carbonic acid	-	-	-	-		-	-	trace.	
Organic matter	-	-	-	-		-	•	$4 \cdot 200$	
							<u>.</u>	100 · 727	S.
4	4. Co	bs of	the s	same	cor	n.			
Silica	4. Co	bs of -	the s	same -	cor	n. -		5.200	
Silica PHOSPHATES:	-	-	•	-	cor	n. -	0.900	5.200	
Silica PHOSPHATES: Phosphate of	-	-	•	-	cor	n. -	0.300	5.200	
Silica PHOSPHATES: Phosphate of Lime	-	-	•	-	<i>cor</i>	n. -	4.680	5.200	
Silica PHOSPHATES: Phosphate of Lime Magnesia	-	-	•	-	- -	n. -	4·680 0·700	5.200	
Silica PHOSPHATES: Phosphate o Lime Magnesia - Silicic acid	f perc	oxide	•	-	- - -	n. -	4.680 0.700 6.500	5.200	
Silica PHOSPHATES: Phosphate of Lime Magnesia	f perc	oxide	•	-		n. -	4·680 0·700		
Silica PHOSPHATES: Phosphate of Lime Magnesia Silicic acid Phosphoric	f perc	oxide	•	-	- - -	n. -	4.680 0.700 6.500	10.400	
Silica Phosphates: Phosphate of Lime Magnesia - Silicic acid Phosphoric	f perc	oxide	•	-	- - -	n. -	4.680 0.700 6.500		
Silica PHOSPHATES: Phosphate of Lime Magnesia Silicic acid Phosphoric	f perc	oxide	•	-	- - -	n. -	4.680 0.700 6.500	10.400	
Silica Phosphates: Phosphate of Lime Magnesia - Silicic acid Phosphoric	f perc	oxide	•	-	- - -	n. -	4.680 0.700 6.500	10·400 0·160	
Silica Phosphate o Lime Magnesia Silicic acid Phosphoric Lime Magnesia -	f perc	oxide	•	-	- - -	n.	4.680 0.700 6.500	10·400 0·160 trace.	
Silica PHOSPHATES: Phosphate of Lime Magnesia Magnesia Magnesia Potash	of perconacid	oxide	•	-		n.	4.680 0.700 6.500	10·400 0·160 trace. 38·332	
Silica Phosphate o Lime - Magnesia - Silicic acid Phosphoric Lime Magnesia - Potash Soda	of perconacid	oxide	•	-		n.	4.680 0.700 6.500	10·400 0·160 trace. 38·332 21·844	
Silica Phosphate of Lime - Magnesia - Silicic acid Phosphoric Lime Magnesia - Potash Soda Chloride of sodia	of perconacid	oxide	•	-	- - - -	n.	4.680 0.700 6.500	10·400 0·160 trace. 38·332 21·844 2·350	
Silica Phosphate of Lime - Magnesia - Silicic acid Phosphoric Lime Magnesia - Potash Soda Chloride of sodio Sulphuric acid	of perconacid	oxide	•	-		n.	4.680 0.700 6.500	10·400 0·160 trace. 38·332 21·844 2·350 0·688	

5. Young soft	ker	rnels c	of Per	ınsy	/lvani	a Dent	corn.
Silica		40	-				1.950
PHOSPHATES:							
Phosphate of	ner	oxide	of iror	ı		1.450	
Lime -	Ι.,				-	0.084	
Magnesia -			_		_	12.298	
Silicic acid					-	0.650	
Phosphoric ac	eid •		-		-	16.706	
Soda					-	5.382	
Potash -					_	0.695	
							$36 \cdot 665$
Lime	-	•			_	-	0.141
Magnesia -		-	-	-	-	-	$9 \cdot 640$
Potash	-	-	-	-	-	-	$14 \cdot 232$
Soda	-	-	-	•	-	-	10.113
Chloride of sodiur	n		-	-	-	-	0.203
Sulphuric acid	-	-	-	_	-	-	0.395
Carbonic acid		-	-			-	trace.
Organic matter an	id lo	oss		-	-	-	20.537
8							
							100·000 S.
6. Stalks and leaves of	f th	Dom	n cailm	an i 0	Dom		out Tum o 11
						r com	
							cui June 11.
14 inches							
							9·800
14 inches Silica Phosphates:	higł -	, but ;	grew i	nas -		place.	
14 inches Silica PHOSPHATES: Phosphate of	higł -	, but ;	grew i	nas -		place 2:500	
I4 inches Silica Phosphates: Phosphate of Lime -	higł -	, but ;	grew i	nas -		place. - 2.500 4.456	
14 inches Silica PHOSPHATES: Phosphate of	higł -	, but ;	grew i	nas -		place 2:500	
I4 inches Silica Phosphates: Phosphate of Lime -	higł -	, but ;	grew i	nas -		place. - 2.500 4.456	
I4 inches Silica Phosphates: Phosphate of Lime - Magnesia -	per	, but ;	grew i	nas -		2.500 4.456 0.400	9.800
I4 inches Silica PHOSPHATES: Phosphate of Lime - Magnesia - Silica - Phosphoric ac	per	, but ;	grew i	nas -		place. 2.500 4.456 0.400 trace.	9·800 19·000
I4 inches Silica PHOSPHATES: Phosphate of Lime - Magnesia - Silica - Phosphoric ac	per	, but ;	grew i	nas -		place. 2.500 4.456 0.400 trace.	9.800
I4 inches Silica Phosphates: Phosphate of Lime - Magnesia - Silica - Phosphoric ac Lime Magnesia -	per	, but ;	grew i	nas -		place. 2.500 4.456 0.400 trace.	9·800 19·000 2·256 trace.
I4 inches Silica PHOSPHATES: Phosphate of Lime - Magnesia - Silica - Phosphoric ac	per	, but ;	grew i	nas -		place. 2.500 4.456 0.400 trace.	9·800 19·000 2·256 trace. 25·429
I4 inches Silica PHOSPHATES: Phosphate of Lime Magnesia - Silica - Phosphoric ac Lime Magnesia - Potash Soda	per	, but ;	grew i	nas -		place. 2.500 4.456 0.400 trace.	9.800 19.000 2.256 trace. 25.429 4.607
I4 inches Silica PHOSPHATES: Phosphate of Lime - Magnesia - Silica - Phosphoric ac Lime Magnesia - Soda Chloride of sodium	per	, but ;	grew i	nas -		place. 2.500 4.456 0.400 trace.	9·800 19·000 2·256 trace. 25·429 4·607 8·953
I4 inches Silica PHOSPHATES: Phosphate of Lime - Magnesia - Silica - Phosphoric ac Lime Magnesia - Chloride of sodium Sulphuric acid	per	, but ;	grew i	nas -		place. 2.500 4.456 0.400 trace.	9.800 19.000 2.256 trace. 25.429 4.607
I4 inches Silica PHOSPHATES: Phosphate of Lime - Magnesia - Silica - Phosphoric ac Lime Magnesia - Soda Chloride of sodium	per	, but ;	grew i	nas -		place. 2.500 4.456 0.400 trace.	9·800 19·000 2·256 trace. 25·429 4·607 8·953
I4 inches Silica PHOSPHATES: Phosphate of Lime - Magnesia - Silica - Phosphoric ac Lime Magnesia - Chloride of sodium Sulphuric acid	per	, but ;	grew i	nas -		place. 2.500 4.456 0.400 trace.	9·800 19·000 2·256 trace. 25·429 4·607 8·953 1·976 13·200
I4 inches Silica PHOSPHATES: Phosphate of Lime - Magnesia - Silica - Phosphoric ac Lime Magnesia - Chloride of sodium Sulphuric acid	per	oxide	grew i	n a s		place. 2.500 4.456 0.400 trace.	9·800 19·000 2·256 trace. 25·429 4·607 8·953 1·976
I4 inches Silica Phosphates: Phosphate of Lime - Magnesia - Silica - Phosphoric ac Lime Magnesia - Chloride of sodium Sulphuric acid Organic matter	per	oxide	grew i	n a s		2·500 4·456 0·400 trace. 11·644	9·800 19·000 2·256 trace. 25·429 4·607 8·953 1·976 13·200 98·221 S. Per centum.
I4 inches Silica Phosphates: Phosphate of Lime - Magnesia - Silica - Phosphoric ac Lime Magnesia - Chloride of sodium Sulphuric acid Organic matter	per	oxide	grew i	n a s		2·500 4·456 0·400 trace. 11·644	9·800 19·000 2·256 trace. 25·429 4·607 8·953 1·976 13·200 98·221 S. Per centum. 92·660
I4 inches Silica PHOSPHATES: Phosphate of Lime - Magnesia - Silica - Phosphoric ac Lime Magnesia - Potash Soda Chloride of sodium Sulphuric acid Organic matter Water Dry stalks, etc.	per	oxide	grew i	n a s		2·500 4·456 0·400 trace. 11·644	9·800 19·000 2·256 trace. 25·429 4·607 8·953 1·976 13·200 98·221 S. Per centum. 92·660 7·380
I4 inches Silica Phosphates: Phosphate of Lime - Magnesia - Silica - Phosphoric ac Lime Magnesia - Chloride of sodium Sulphuric acid Organic matter	per eid	oxide	grew i	n a s		2·500 4·456 0·400 trace. 11·644	9·800 19·000 2·256 trace. 25·429 4·607 8·953 1·976 13·200 98·221 S. Per centum. 92·660

III. ANALYSIS OF SWEET CORN.

1. Stalk, cut below the ears.

Plant in blossom. Soil a clay loam.												
	Plant	in blo	ossom.	Soi	lac	lay	loan	1.				
Silica -	•	-	•	•	-		-	-	$3 \cdot 100$			
PHOSPHATES	:											
${f P}$ hospha	ate of	` pero	xide (of iro	n	-		0.300				
Lime	-	-	-	•	•	-		2.651				
Magnes	ia -	-	•	-		-		0.500	•			
· Silicic a	cid	-	-	-		•		4.300	•			
Phospho	oric a	cid -	-	-		-	2	$2 \cdot 049$				
							_		29.500			
Lime -	•	-	-	•	-		-	-	0.169			
$\mathbf{Magnesia}$	-	•	-	-	-		-	•	0.150			
Potash -	-	-	•	-	-		-	-	44.912			
Soda -	-	-	-	-	-		-	•	5.302			
Chlorine	-	-	-	-	-		-	-	1.089			
Sulphuric ac	id	-	-		-		-	•	11.885			
Carbonic aci	d	•	-	-	-		-		trace.			
Organic mat	ter	-	-	-	-		-	-	5.200			
J												
									101·307 S.			
Per centum	of m	tor	PROP	ORTION	s		_	_	81.840			
Dry stalks	or wa	itei	•	•	-		•	•	18.160			
Ash -	-	-	-	-	-		-	_	0.660			
Ash calculat	• •d d=		•	•	•		-	•	3.634			
Asii calculat	eu ui	У	•	•	-		•	•	0 004			
	2.	Stal	ks cu	t abo	ve t	he e	ar.					
Silica -									1.000			
		•	-	-	•		-	•	1.000			
PHOSPHATES:			:	£ :				4.000				
Phospha	ite or	pero.	xiue c)1 1f0	11	•		4.600				
Lime		-	-	-	•	•		0.423				
Magnesi		-	•	•	•	•		0.015				
Silicic a		.,	-	•		-		3.800				
Phospho	ric a	cid -	-	•	•	-	2	7.462				
Lime -							_		$36.300 \\ 0.344$			
	-	-	•	•	•		-	-				
Magnesia	-	•	-	•	•		•	•	0.015			
Potash -	•	-	•	-	-		-	•	19.682			
Soda -	-	-	-	-	-		-	-	31.589			
Chlorine	-	-	-	•	•		-	•	4.736			
Sulphuric ac		-	-	•	•		•	-	1.516			
Carbonic aci		-	-	•	•		-	•	trace.			
Organic mat	ter	•	-	-	-		-	•	4.500			
									99·721 S.			
									JJ 121 D.			

					PRO	portio	Ns.				
	TT7 .									Per centum	
	Water	-	-	-	-	-	•	-	-	91.660	
	Dry	-	-	-	-	-	-	-	•	8.340	
	Ash	-	-	-	-	-	-	-	-	0.680	
	Calcula	ated o	dry	-	-	-	•	•	-	8.153	
	3. She	eaths	of the	he sar	ne co	rn, ϵ	ut A	ugus	t 6 : in	flower.	
	Silica			_	-	•	-	-		16 · 100	
	PHOSPH	ATES	:								
	Pl	osph	ate o	f per	oxide	of ire	on	-	1.000		
		me	-	٠.			-	-	1.579		
	\mathbf{M}_{i}	agnes	sia -				•	-	4.800		
		licic :					-		5.100		
				acid -			_	_	12.621		
		~ I								25.100	
	$_{ m Lime}$	-	_	-	_	_		-	_	2.707	
	Magnes	sia	-	-	_	-	-		-	0.900	
	Soda	-	_	-	_	-	_	_	_	0.406	
	Potash	_	_	_		_		_	_	31.881	
	Chlorid		sodir	ım	_					7.883	
	Sulphu			-				_	_	lost.	
	Carbon			_	_	_	_	_	_	trace.	
	Organi			-		_	_	_	_	4.000	
	- 18	- 1110									
										92.633	S.
					PRO	PORTIO	NS.				
	Water									Per centum 79 · 130	
			-	•	•	•	•	•	•		
	Dry	•	-	-	•	•	-	•	•	20.870	
	Ash	-	-	-	-	•	•	•	-	1.360	
	Ash cal	culat	ed di	ry	-	•	•	-	-	6.516	
4.	Leanes	of su	veet r	orn.	cut n	vhile	in blooms	osson	ı, Augu	st 6. 18	347.
•	Silica	_	-	-	-	-	-	•	•	19.700	
	Рноsрн			-	_	_	_	-	_	10 100	
				f ===	oxide	- f :-	o v o		3.100		
		_	iaie C	n per	oxide	01 11	OH	•	8.010		
		me	.:	•		•	•	-			
			sia -	•	•	•		•	2.920		
		lica		٠, ٠	•	•	-	-	5.500		
	Pl	nosph	oric	acid ·		-	-	-	14.370	33.900	
										00 800	
	Carried	forw	ard		-	•	•	-	-	53.600	

Brought forwar	d		-	-	-	-	$53 \cdot 600$	
Lime	-	-		-	-	-	2.538	
Magnesia -	-	-	-	-	n	-	$4 \cdot 480$	
Potash	-	-	-		-		17.712	
Soda	-	-	-	-	-	-	12.645	
Chloride of sodium	m	-	-	-	•	-	2.451	
Sulphuric acid	-	•	-	-	•			
Carbonic acid	-		-	-	-	-	trace.	
Organic matter	-	-	-		-	-		
5								
							$93 \cdot 426$	S.
		PRO	PORTIC	ons.				
							Per centum	
Water	-	-	-	-			65.820	
Dry	-	-	•	-	-	-	$34 \cdot 180$	
Ash	-	-	•	-	-	•	2.880	
Ash calculated dry	У	-	-	-	-	•	8.431	
5. Tassels o	f su	eet c	orn.	cut.	Аион.	st 6.	1847.	
	,		,			,,		
Silica	•	-	-	•	•	-	17.800	
PHOSPHATES:		. 1	٠,			0.0	00	
Phosphate of	per	oxide	oi ir		•	2.6		
Lime -	•	•	-	-	•	1.8		
Magnesia -	•		-	-	•	$5 \cdot 4$		
Silica -	٠, ٠		-	•	•	trac		
Phosphoric a	cid •	•	-	•	•	4.9	56 — 14·800	
Lime								
	-	•	•	-	-	-	3.158	
Magnesia -	-	-	•	-	-	-	0.400	
Potash -	•	•	•	•	•	•	19.484	
Soda		•	•	-	•	-	12.118	
Chloride of sodium	m	-	•	-	-	n	7.528	
Sulphuric acid	-	-	•	-	•	•		
Carbonic acid	-	-	-	-	•	-	trace.	
Organic matter	•	•	-	•	-	-	11.304	
							86.592	S.
							00 00%	~.
		PRO	PORTIC	NS.			Per centum	
Water		-	-			-	67.750	•
Dry -	-	-	-			-	$32 \cdot 250$	
Ash	-		-	-		-	1.900	
					_			
Calculated dry			_	_	-	_	5.891	

2.

IV. ANALYSIS OF MIDDLE-SIZED 8-ROWED YELLOW CORN.

Cut August 6. Just out of flower.

			J	1.	Shear	ths.				
Silic	a -	-	-	-			-	-	31.300	
	SPHATE	s:								
	Phospl		of per	oxide	of iro	n	-	1.900		
	Lime		•		-	-	_	3.384		
	Magne	esia	-	-	_	-	-	1.840		
	Silica			•	-	-	-	0.500		
	Phosp	horic	acid	•	-		-	6.876		
	-								14.500	
Lim	e -	-	-	-	-		-	-	2.989	
Mag	nesia	-	-	-	-	•	-	-	5.920	
Pota	ısh -	-	•	-	-	-	-	-	$30 \cdot 363$	
Sod	a -	-	-	-	-	-	-	-	$2 \cdot 499$	
\mathbf{Chlc}	oride of	sodi	ium	-	-	-	-	-	4.738	
Sulp	huric a	cid	-		-	-	-	-		
Carl	oonic a	cid	-	-	-	-	-	-	trace.	
Orga	anic ma	atter			-	-	-	•		
									00.100	~
									92 · 139	S.
				PROP	ORTIONS	5.			Per centum.	
Wat	er -			-			-	-	82.730	
Dry			-			-	-	-	17.270	
Ash		-		-				_	2.200	
	calcula	ated o	lrv			-		•	12.738	
			•							
Stalk	of the	midd	$lle ext{-}siz$	ed 8-	rowed	l yello	w con	rn, cut	below the	e ear
Silic	a -	•	-	۰	-	-	-	•	8.950	
Рно	SPHATE									
	Phosp	hate	of per	oxide	of ire	on	-	0.300		
	Lime		-	•	•	•	-	$1 \cdot 144$		
	Magn	esia		-	-	-	-	0.100		
	Silica		-	-	-	•	-	0.350		
	Phosp	horic	acid,	potas	h and	soda	-	17.406		
									19.300	
\mathbf{L}_{im}	e -	•	•	•	-	-	-	-	$3 \cdot 187$	
Mag	gnesia	•	•	•	•	-	-	-	trace.	
Pota	ısh -	-	•	-	-	-	•	-	41.623	
Sod	a -	-		-	-	-	-	-	7.726	
	oride of	f sod	ium	-	-	-	-	•	4.943	
	phuric		-		-	-	-	•	$12 \cdot 475$	
	bonic a		-				-	-	trace.	
	anic m		-	***	-			-	2.797	
٠ <u>.</u> ۶	111									
									100.901	S.

		PROPO	RTIONS.				
***							Per centum.
Water	-	-		•	-		81.220
Dry stalks -	•	-	-	ь	-	-	18.780
Ash	-	-	•	4	-	-	0.420
Ash calculated dry	•	**	-	-	•	-	$2 \cdot 236$
3. Hush	ks of	the 8	-rowe	ed yel	low	corn.	
Silica	-	-	•	-		•	$23 \cdot 000$
PHOSPHATES:							
Phosphate of	perox	ide o	f iron			trace.	
Lime -	٠.,		••			$3 \cdot 271$	
Magnesia -		-	-	_		0.150	
Silica -	-	-		-		3.910	
Phosphoric ac	eid -	-	_			18.569	
p							$25 \cdot 900$
Lime	-	-	-		-	-	1.353
Magnesia -		-		-	10	-	0.800
Potash	**		-		-		$14 \cdot 267$
Soda	-	-	-	-		-	19.267
Chloride of sodiur	n	-	-		6	-	$3 \cdot 145$
Sulphuric acid	-	_				-	$5 \cdot 001$
Carbonic acid	-	-	_		-	Ð	4.765
Organic matter				-		-	$3 \cdot 456$
0							
							100·745 S.
		PROPO	RTIONS				
XX7							Per centum.
Water	-	-	-	-	-	•	74.490
Dry	-	-		•	-	-	25.550
Ash	•	-	•	•	-	•	0.810
Calculated dry	-	-	-	-	-	-	$3 \cdot 175$
Proportions of	elemer	nts in	the to	isse l s	of t	he same	
Water							Per centum.
Water	•	•	•	•	•	-	56.500
Dry	-	-	•	•	•		43.500
Ash	-	•	•	•	•	-	2.780
Ash calculated dry	7	•	•	•	-	•	$6 \cdot 393$

4. Stalks above the ear.

Silica	•	-	-	-	-	-	19.600	
Phosphates:								
Phosphate	of per	oxide	of ire	on	-	$6 \cdot 100$)	
Lime -	-	-	-	-	-	10.943	}	
Magnesia	-	-	•	-	•	0.800)	
Silica	-	-	-	-	-	1.000		
Phosphoric	acid	-	-	-	-	14.657		
							33.500	
Lime	-	-	-	-	-	-	3.215	
Magnesia -	-	-	-	•	•	-	0.100	
Potash	-	-	-	-	•	-	20.874	
Soda		-	-	-	-	-	17.047	
Chloride of sod	ium	•	-	-	•	-	8.603	
Sulphuric acid	-	-	-	-	-	-	$2 \cdot 329$	
Carbonic acid	-	-	-	-	-	-	trace.	
Organic matter	~	-	-	-	-	68		
							105.268	s.
		PRO	oportio	NS.				
117							Per centum.	
Water	•	•	-	•	-	-	78.840	
Dry	•	-	-	-	-	•	21.160	
Ash	-	-	-	-	-	-	0.580	
Calculated dry	-	-	•	•	-	•	2.741	
Į	5. Le	aves o	of the	same	corn	·•		
Silica	-	-	-	-			37.550	
PHOSPHATES:								
Phosphate	of per	oxide	of ire	n	-	8.800		
Lime -	-	-		-	_	6.289		
Magnesia	-	-	-	-	-	trace.		
Silica -	-	-	-	-	-	2.350)	
Phosphoric	acid	•			-	7.210	•	
							24.649	
Lime	-	-	-	-	-	-	$6 \cdot 133$	
Magnesia -	-	-	-	-	-	-	2.820	
Potash	-	-	-	-	-	-	18.660	
Soda	-	-	-	-	-	-	8.860	
Chloride of sod	ium	-	-	-	-	•	0.838	
Sulphuric acid	-	-	-	4	-	-	1.030	
Carbonic acid	-	-		-	-		trace.	
Organic matter		-	ex		-	-	2.280	
Ü								~
							102·S23	S.

PROPORTIONS.

							Per centum.
Water	-	-	-	-	-	-	80.750
Dry	-	-	-	-	-	-	$19 \cdot 250$
Ash	-	-	-	-	-	-	2.800
Calculated dry	•	-	-	-	-	-	$14 \cdot 545$

INORGANIC ANALYSES OF THE SEVERAL PARTS OF THE PLANTS OF THE EARLY WHITE FLINT CORN, AT DIFFERENT STAGES OF THEIR GROWTH.

The dates on which the plants were taken for analysis, correspond with those on which they were taken for proportions. I have hence, to save repetition, omitted to give descriptions of the stage of advancement of the plants with the following analyses, as these are given in full in connection with the proportions, and can be easily referred to.

It would have been desirable to have made this series of analyses more complete; but the plant was divided into so many parts, that it was quite impossible for one to do it, and carry on organic analyses of the same parts at the same time. Considerable, however, has been done. The analyses were conducted during several successive weeks, when the plants were growing most vigorously. They were also analyzed after they had ripened their grain. The roots were freed of foreign matter, by washing them carefully in cold water. The potash was obtained with bichloride of platinum. It is proper to state this here, inasmuch as by this process less potash has been obtained, and of course more soda than by absolute alcohol. Both methods have been followed at different times; but in this series, the bichloride of platinum has been exclusively resorted to for separating the two alkalies.

JULY 12.

i.	Leaves	(Description	of plant o	n page 173).

2300000 (25000.	-P 0101	L	10110	o P.	*5° *	• • , •		
Carbonic acid	-	-	-	-	-	-	-	$7 \cdot 50$
Silicic acid	-	-	-	-	-	-	-	$13 \cdot 30$
Sulphuric acid	l	-	-	-	-	-	-	2.06.
Phosphates	-	-	-	-	-	-	-	$25\cdot 00$
Lime -	-	-	-	-	-	-	-	0.68
Magnesia	-	-	-	-	-	-	-	0.27
Potash -	-	-	-	-	-	-	-	9.88
Soda -	-	-	-	-	-	-	-	$32 \cdot 17$
Chlorine	-	-	-	-	-	-	-	$5 \cdot 08$
Organic matte	r	-	-	-	-	-	-	4.60
								93·54 S.

2.	Sheaths.								5
	Carbonic acid	1 -	-	-	-	-	-	-	6.85
	Silicic acid	-	-	-	-	-	-	-	11.00
	Sulphuric aci	d	-	-	-	-	-	-	$6 \cdot 54$
	Phosphates	-	-	-	-	-		-	$10 \cdot 40$
	Lime -	-	-	-	-	-	-	-	$7 \cdot 42$
	Magnesia	-	-	-	-	-	-	-	1.16
	Potash -	-	-	•	-	-	-	-	$7 \cdot 64$
	Soda -	-	-	-	•	-	-	•	$34 \cdot 48$
	Chlorine -	-		•	-	-	-	-	8.14
	Organic mat	ter			-	-	-	-	5.50
									99·13 S.
3.	Roots (The	follor	wing	analy	ysis is	inco	mple	te).	
	Silicic acid	•	•	_	-	-	•	•	$31 \cdot 00$
	Phosphates	-	-	-	-	-	-	-	7.00
	Lime -	-	-	-		-	-	-	$3 \cdot 40$
	Magnesia	-		-		-	-	-	$2 \cdot 40$
	Potash -	-		-	-		-	-	$9 \cdot 25$
	Soda -	-	-	-	-	-	-	-	20.55
	Carbonic and	l sulp	huric	acids	3 -	-	-	-	
	Chlorine and	_				lloss	-	-	$26 \cdot 40$
									100·00 S.

JULY 19.

The corn plants are now growing rapidly. The roots have extended themselves from 7 to 12 inches into the soil. The stalk has just begun to form. A further description of the corn plants, at this date, will be found on page 174.

1. Leaves.

Carbonic acid	l	-	-	-	-	-	-	$5 \cdot 40$
Silicic acid	-	-	-	-	-	-	-	$13 \cdot 50$
Sulphuric aci	d	-	-	-	-	-	-	2.16
Phosphates	-	-	•	-	-	-	-	21.60
Lime -	-	-	-	-	-	-	-	0.68
Magnesia		-	-	-		-	-	0.27
Potash -	-	-	-	-	•	•	-	$9 \cdot 98$
Soda -	-	-	-	-	-	-	-	$34 \cdot 39$
Chlorine	-	-	-	-	-	-	-	4.55
Organic mat	ter	-	-	-	-	-	-	5.50
								98·03 S.

2. Sheaths.								
Carbonic acid		-	-	-	-	-	-	4.00
Silicic acid	-	-	-	-	-	-	-	$15 \cdot 60$
Sulphuric aci	d	-	-	-	-	-	-	9.84
Phosphates	-	-	-	-	-	155	-	7.60
Lime -	-	-	-	-	-	-	-	5.06
Magnesia	-	-	-	-	-	-	-	1.64
Potash -	-	-	-	-	-	-	-	$9 \cdot 96$
Soda -	-	-	-	-	-	-	-	$32 \cdot 12$
Chlorine	-	-	-	-	-	-	-	8.04
Organic matt	er	-	-	-	•	-	-	$5 \cdot 40$
								99·26 S
3. Roots (This a	analy	sis is	inco	mple	te).			
Silicic acid	-	-	-	-	-	-	-	$36 \cdot 60$
Phosphates	-	-	-	-	-	-	-	4.66
Lime -	-	-	-	-	-	-	-	1.56
Magnesia	-	-	-	-	-	-	-	$0 \cdot 34$
Potash -	-	-	-	-	-	-	-	17.48
Soda -	-	-	-	-	-	-	-	15.82
Carbonic and	l sul	phuric	acid	s, ch	lorine	, orga	anic	
matter and	-	-	-	•	-	•	-	$23\cdot 54$
								100.00

JULY 26.

Corn plants are increasing rapidly in size. For a further description, refer to page 175.

1. Leaves.

Carbonic acid -	-	-	-	-	-	-	$5 \cdot 40$
Silicic acid -	-	-	-	-	-	-	16.70
Sulphuric acid	-	-	-	-	-	_	2.51
Phosphates -	-	-	-	-	-	-	$20 \cdot 50$
Lime	-	-	-	-	-	-	1.30
Magnesia -	-	-	6.	-	-	-	0.75
Potash	-	-	-	-	-	-	13.12
Soda	-	-	-	-	-	-	$27 \cdot 41$
Chlorine -	-	-	-	-	-	-	3.06
Organic matter	-	-	-	-	-	-	6.60
							97·35 S

2. Sheaths (Owing to an error in obtaining the chlorine and sulphuric acid, this analysis is incomplete).

-			-	,					
Carbonic acid		-	-	-	-	-	-	$5 \cdot 20$	
Silicic acid	-	-	-	•	-	-	-	8.30	
Phosphates	-	-	-	•	-	-	-	10.50	
Lime -	-	-	-	-	-	-	-	5.76	
Magnesia	-	-	-	-	_	-	-	2.52	
Potash -	-	-	-	-	-	-	-	$13 \cdot 39$	
Soda -	-	-	-	-	-	-	-	29.68	
Chlorine, sulp	huri	acid,	organ	nic ma	itter a	and lo	ss,	26.65	
			_						
								$100 \cdot 00$	S.

I can not account for the small percentage of silicic acid in the above analysis. It is much less than was obtained either the preceding or the succeeding weeks.

AUGUST 2.

					AU	1001	4.				
1.	Leaves	(Desc	rip	tion of	f pla	nt on	page	176)			
	Carbon	ic aci	$^{-}$	-	-	-	•	-	-	$2 \cdot 850$	
	Silicic	acid	-	-	-	-	-	-	-	19.850	
	Sulphu	ric aci	d	-	-	-	-	-	-	1.995	
	Phosph	ates	-	-	-	-	-	-	-	16.250	
	Lime	-	-	-	-		-	-	-	$4 \cdot 035$	
	Magne	sia	-	4 -	-	-	-	-	-	2.980	
	Potash	-	-	-	-	-	-	-	-	11.675	
	Soda	-	-	-	-	-	-	-	-	$29 \cdot 590$	
	Chlorin	ne .	-	-	-	-	-	-	-	$6 \cdot 020$	
	Organi	c acid	s	-	-	-	-	-	-	$2 \cdot 400$	
										97.645	S.
2.	Sheaths	•		-							
	Carbon	ic acio	ł		-	-	-	-	-	9.60	
	Silicic	acid	-	-	_	-	-	-	-	$16 \cdot 30$	
	Sulphu	ric aci	id	-			~	-	-	$3 \cdot 64$	
	Phosph		-	-	-		-	-	-	8.90	
	$\overline{\text{Lime}}$	-	-	-	-		-	-	-	3.27	
	Magnes	sia	-	-	-	-	-	-	-	$2 \cdot 45$	
	Potash	-	-	-	-	-	-	-	_	10.46	
	Soda	-	-	-	-	-	-	-	-	33.60	
	Chlorin	ie	-	-	-	-	-	-	-	8.68	
	Organi	c acids	s -	-	-	-	-	-	-	2.80	
										99.65	s.

З.	Stalks.	

Carbonic acid	-	-	-		-	-	•	$13 \cdot 40$
Silicic acid	-		-	-	•	•	-	2.50
Sulphuric aci	d	-	•	•	-	-	-	$3 \cdot 18$
Phosphates	-	-	-	40	۰	-	-	$11 \cdot 50$
Lime -	-	-	-	-		-	-	3.82
Magnesia	•	-	•	-	-	-	-	$3 \cdot 60$
Potash -	-	-	-	-	•	-		$12 \cdot 35$
Soda -	-	-	۰	•	-	10	-	$35 \cdot 09$
Chlorine	-	•	-	-	-	-	-	$7 \cdot 21$
Organic acids	· -	-	-	-		•	-	4.85
								97.50 S.

The two following analyses are not completed, there being too small a quantity of ash.

4. Tassels.

Carboni	c acid	-	•	•	-	•	•	-	$2 \cdot 40$
Silicic a	cid	-	•	-	-	-	-	-	2.00
Phospha	ates	-	-	-			-	-	21.67
$_{ m Lime}$	-	-	-	•	-	•	-	-	1.50
Magnes	ia	-	-	-	-	•	-	-	0.67
Potash	-	-	•°	-	-	•	-	-	10.70
Soda	-	-	-	-	-	-	-	-	37.77
Sulphur	ic acio	1	-	-	-	-	•	-	
Chlorine	e, orga	anic n	natter	and l	oss	-		-	$23 \cdot 29$
									100.00 S.
									100 00 D.

5. Roots.

1.00000								
Silicic acid	-	•	-	-	•	-	-	16.916
Phosphates	-	-		-	-	-	•	14.160
Lime -	-	-	-	•	-	-	-	2.634
Magnesia	-	-	-	-	-	•	-	1.433
Potash -	-	-	-	-	-	-	-	8.950
Soda -	-	-	-	-	. •	-	-	31.700
Carbonic ac	id, s	ulphu	ric ac	id, ch	lorine	, orga	anic	
matter an	d los	ss -	-	-	-	-	-	$24 \cdot 207$
								100.000

100**·00**0 S.

AUGUST 9.

1. Leaves (Descripti	ion of	f pla	nt on	page	177)	١.	
Carbonic acid			•			-	1.200
Silicic acid -	-	-	-	-	-	-	$24 \cdot 750$
Sulphuric acid	•	-	-		-	-	$2 \cdot 475$
Phosphates -		-	•	-		-	16.900
Lime	-	-	-	-	-	•	$4 \cdot 365$
Magnesia -	-	-		-	•	۰	$3 \cdot 400$
Potash	-	-		-	-	-	$8 \cdot 240$
Soda		-	-	-	-	-	26.690
Chlorine -	-	-	-	-	-	-	6.700
Organic acids	-				-	•	$2 \cdot 025$
							97·750 S.
2. Sheaths.							
Carbonic acid -	-	•		-	-	-	4.85
Silicic acid -	-		-	-	-	-	14.80
Sulphuric acid	-	-	-	-	e	-	3.75
Phosphates -	•	-	•	-	-	-	$13 \cdot 15$
Lime	-	-	-	•	-	-	$3 \cdot 27$
Magnesia -	•	-	-	-	-		$5 \cdot 10$
Potash	-	-	-	-	•	•	8.21
Soda	-	•		-	-	-	$32 \cdot 04$
Chlorine -	-	-	-	-	-	-	7.21
Organic acids -	-	-	-	-	-	-	5.70
							98·08 S.
3. Stalks.							
Carbonic acid		•		•	-	-	13.50
Silicic acid -	-	-	-	-	-	-	2.90
Sulphuric acid	-	-	-	-	-	-	$2 \cdot 15$
Phosphates -	-	-	-	-	-	-	11.80
Lime	-		-	-	-		4.67
Magnesia -	-	٠.	-	-	-	•	2.72
Potash	-	•	-	-	-	-	10.88
Soda	-	-	-	-	-	-	29.66
Chlorine -		-	-	-	-		· 14·90
Organic acids	-	•	-	-		-	4.65
							97·63 S.

4.	Tassels (The fo	llowin	g an	alysis	is in	comp	lete)		
	Carbonic acid						•	2.892	
	Silicic acid -		-	-	-	-		1.335	
	Phosphates -					_	-	20.248	
	Lime	-	-	-	-	-	-	$0 \cdot 445$	
	Magnesia -	-	-		-	-	-	$0\cdot445$	
	Potash		-	-	-	-	-	11.081	
	Soda	-	-	-	-	-	-	$39 \cdot 494$	
	Chlorine, sulphu	ric aci	d, org	anic r	natter	and	loss,	$24 \cdot 060$	
								100.000	S.
								100 000	
			AUG	UST	16.				
1.	Leaves (Descrip	tion o	f pla	nt on	. page	179)) .		
	Carbonic acid		-	10	-	•	60	$3 \cdot 025$	
	Silicic acid -	-	-	-	-	-	•	$21 \cdot 100$	
	Sulphuric acid	-	-	-	•	-	۰	6.220	
	Phosphates -	-	~	-	-	-	-	14.650	
	Lime		-		-	-	-	$5 \cdot 347$	
	Magnesia -	-	-	-	-	•	•	1.510	
	Potash		-	-	-	-	-	$10 \cdot 170$	
	Soda	-			-	-	-	31.390	
	Chlorine -	•	-	-	-	-	-	1.975	
	Organic acids	-	-			-	-	$2\cdot800$	
								98 · 187	S.
2.	Sheaths.								
	Carbonic acid	•	-		•		-	2.60	
	Silicic acid -	-	-	-	•	-	-	$18 \cdot 20$	
	Sulphuric acid		-	-	•	-	-	$3 \cdot 01$	
	Phosphates -	-	-	-	-	-	-	11.20	
	Lime	-	-	•	•	-	•	0.62	
	Magnesia -	-	-	-	-	-	-	1.02	
	Potash	-	-	-	-	-	•	10.73	
	Soda	-	-	-	-	-	-	$32 \cdot 62$	
	Chlorine -	-	-	-	-	-	-	$16 \cdot 42$	
	Organic acids -	-	-	•	-	•	-	3.25	
								99.67	S.

3. Stalks.								
Carbonic acid -	-	-	-		-	•	12.05	
Silicic acid	-	d	-	-	-	-	4.70	
Sulphuric acid	-	-	-	•	-	-	$5 \cdot 91$	
Phosphates -	42	-		-	**	_	11.20	
Lime	-		-	esi	-		2.82	
Magnesia -	-	4	-	-		-	1.73	
Potash	-		-		4	-	10.31	
Soda			-	-	-		$35 \cdot 54$	
Chlorine -			•				9.03	
Organic acids -	_		-				4.40	
Ü								
							97·69 S	•
		AUG	GUST S	23.				
1 Lagran / Dogavir	ation			_	101)			
1. Leaves (Descrip	puon	_		page	101)	٠	0.00	
Carbonic acid	-	-	-	•	a	-	0.65	
Silicic acid -		•	-	•	•	•	34.90	
Sulphuric acid				•	-	-	4.92	
Phosphates -		es	-	•		-	17.00	
Lime			•	•	•	•	2.08	
Magnesia -	. 4	9		•	-		1.59	
Potash	-	•	-	-		-	10.85	
Soda	-	•	•		-	•	21.23	
Chlorine -		-		-	•	-	3.06	
Organic acids -			-		-	-	3.38	
							00.66 0	
							99·66 S	•
2. Sheaths.								
Carbonic acid -							4.82	
Silicic acid -					_	_	23.70	
Sulphuric acid							4.26	
Phosphates -			_	_	_	_	12.70	
Lime -			_	_	-	_	1:13	
Magnesia -			-				2.16	
Potash				i.	-	-	5.21	
Soda				•	-	-		
	-		•	44		•	$33 \cdot 44 \\ 4 \cdot 69$	
Chlorine -				-		•		
Organic acids	• .		-	d	-	-	5.00	
							97·11 S	
							J. 11 W	

3.	Butt stalk.								
	Carbonic acid	-	-	-	-	-	-		4.99
	Silicic acid	-	-	-	•	•	•	-	6.20
	Sulphuric acid	l	_		-	-	-	-	$5 \cdot 53$
	Phosphates	-	-		-	-	-	•	17:30
	Lime -	-	-	-		-	-	-	1.69
	Magnesia	-	-	-	-	-	-	-	$2 \cdot 23$
	Potash -	-		-	_	•	-	-	10.85
	Soda -	-	-	-	-	-	-	-	$33 \cdot 21$
	Chlorine -	-	-	-	-	-	-	-	$10 \cdot 16$
	Organic acids	-	•	-	•	o	-	-	$5 \cdot 50$
	J								
									97·66 S.
4.	Top stalk.								
	Carbonic acid			•	-	•	-	-	$3 \cdot 10$
	Silicic acid	-	-		-	-	-	-	10.80
	Sulphuric acid	Ì		-	-	•	-	-	6.88
	Phosphates	-	-	-		-	-	-	$25 \cdot 60$
	Lime -	-	-	-	-	-	-	-	1.18
	Magnesia	-	•	-	-	-	-	-	0.84
	Potash -	-	-	•	-	-		•	$12 \cdot 36$
	Soda -	•	-	-	-	-	-	-	$27 \cdot 48$
	Chlorine	-		-	-	•	-	-	$4 \cdot 74$
	Organic matte	r	-	-	•	•	-	-	$5 \cdot 40$
									98·38 S.

5. Tassel (The four following analyses are not complete: the carbonic and sulphuric acids, chlorine and organic matter, were not determined; the rest were accurately obtained).

Silicic acid	-	-	-	•	-	•	-	$23 \cdot 00$
Phosphates	-	-	-	-	-	-	-	$20 \cdot 40$
Lime -	-	•	-	-	-	-	-	$1 \cdot 14$
Magnesia	-	-	-	-		-	-	1.60
Potash -	-	-	•	-	•		-	$6 \cdot 10$
Soda -	-	-	-	-	•	•	-	$25 \cdot 12$
Carbonic ac	id, sulj	phuric	acid,	chlo	rine,	organ	nic	
matter an	d loss	•	a				•	$22 \cdot 64$
								100 00 0
								100.00 S.

6.	Leaves of hus	sks.								
	Silicic acid	-	-	•	-				$25 \cdot 40$	
	Phosphates	-	-	-	_	-	-	-	15.80	
	Lime -	-	-	-	-	-	-	-	0.56	
	Magnesia	-	_	•	-	-	_	-	0.50	
	Potash -	-	-			-	-	-	6.80	
	Soda -	-		_	-	_	-	-	28.12	
	Carbonic acid	l, sul	phuri	c aci	d, ch	lorine	, orga	ınic		
	matter and			-	·-	-	, <u> </u>	-	22.78	
									100.00 8	S.
7.	Sheaths of hi	usks.								
	Silicic acid	-	-	-	-				15.80	
	Phosphates	-	-	-	-	-	-	-	17.60	
	Lime -	-		-	-	-	-	•	0.46	
	Magnesia	-	-	-	-	-	-	-	0.40	
	Potash -	~	-	-	-	-	-	-	6.56	
	Soda -	-	-	-	-	-	-	-	$32 \cdot 84$	
	Carbonic acid	d, sul	lphuri	c acid	d, ch	lo r ine	, orga	nic		
	matter and	loss	•	-	•	-	•	-	$26 \cdot 34$	
									100.00	S.
8.	Roots.									
	Silicic acid	-	-	-	-	-	-	-	$33 \cdot 25$	
	Phosphates	-	-	-	-	-	-	-	17.38	
	Lime -	-	-	-	-	-	-	-	0.56	
	Magnesia	-	-	-	-		-	-	0.75	
	Potash -		-	-	-	•	-	-	$6 \cdot 94$	
	Soda -	-	-	-	-	-	-	-	$22 \cdot 16$	
	Carbonic aci	d, su	lphuri	ic aci	d, ch	lorine	, orga	ınic		
	matter and	loss	•	-	-	-	-	•	18.96	
									100.00	S.

AUGUST 30.

At this date, for want of time, only the leaves and sheaths have been analyzed. Description of plant on page 183.

1.	Leaves	of	the	stalk	and	husks.	
----	--------	----	-----	-------	-----	--------	--

Carbonic acid	-	-	-	-	-	-	$3 \cdot 50$
Silicic acid -	-	-	-	-	-	-	36.27
Sulphuric acid	-	-	-	-	-	-	5.84
Phosphates -	-	-	-	-	-	-	13.50
Lime	•	-	-	-	-		$3 \cdot 38$
Magnesia -	-	-		-	-	-	$2 \cdot 38$
Potash	-	-	-	-	-	-	$9 \cdot 15$
Soda	-	-	-	-	-	-	22.13
Chlorine -	-	-	-	-	-	-	1.63
Organic matter	-	-	-	-	-	-	2.05
							90·83 S.

3. Sheaths of the stalks and husks.

Dictails	oj in	cocu	uico ui	iw nu	3/63.					
Silicic a	cid	-	-	-	-	•	-	•	$38 \cdot 10$	
Sulphur	ic aci	d	-	-	-	-	-	•	$6 \cdot 36$	
Phospha	ates	-	-	-	-	-	-	-	$12 \cdot 80$	
Lime	-	-	-	-	-	-	-	•	$1 \cdot 20$	
Magnesi	ia	-	•	-	-	-	-	•	$2 \cdot 02$	
\mathbf{Potash}	-	-	-	-	•	-	•	•	$7 \cdot 76$	
Soda	-	•	-	-	-	-	-	-	19.68	
Chlorine	е	-	-	-	-	-	-	•	$4 \cdot 34$	
Organic	matt	\mathbf{er}	-	-	-	-	-	• .	$3 \cdot 40$	
Carboni	c acid	ł	-	-	-	-	-	•	4 · 14	
									99.86 S.	

SEPTEMBER 13.

At this period, only the stalks and sheaths were examined. For proportions and description of corn at this date, refer to page 188.

1. Stalks.

Carbonic acid		-	-	-	-	-	•	0.35
Silicic acid	-	-	-	-	-	-	-	10.90
Sulphuric acid	l	-	-	-	-	-	-	$14 \cdot 02$
Phosphates	-	-	-	•	-	-		13.45
Lime -		•	-	-	-	-	-	2.70
Magnesia	-	-	-	-	-	-	-	1.58
Potash -	-	-	-	-	-	•	-	$19 \cdot 22$
Soda -		-	-	-	-	-	-	$\cdot 26 \cdot 22$
Chlorine	-	•	-	-	-	-	-	3.80
Organic acids	-	-	-	-	-	-	-	3.50
								98·89 S.

വ	Sheath	_
· 2	Shearn	e-

Diecatios.							
Carbonic acid	-	-	-	-	-		$4 \cdot 195$
Silicic acid -	-	-	-	-	-	-	$45 \cdot 250$
Sulphuric acid	-	-	-	-	-	-	2.406
Phosphates -	•	-	-	-	-	-	$10 \cdot 150$
Lime	•	-	-	-	-	-	1.803
Magnesia -	-	-	-	-	-	-	1.300
Potash	-	-	-	-	•	-	$11 \cdot 115$
Soda	-	-	-	-	-	-	17.799
Chlorine -	-	-	-	-	•	-	$2 \cdot 355$
Organic acids		-	•	-	-	-	$2 \cdot 450$
							98·823 S.

OCTOBER 18.

Corn ripe. For proportions and further description of plants at this date, refer to page 190.

1. Stalks.

Carbonic acid		-	-	-	-	•	-	1.850
Silicic acid	-	-	-	-	-	-	-	12.850
Sulphuric acid	d	-	-	-	-	-	-	10.793
Phosphates	-	-	-	-	-	-	-	$15 \cdot 150$
Lime -	-	-	-	-	-	-	-	2.820
Magnesia	-	-	-	-	-	-	-	0.936
Potash -	-	-	-	-	-	-	-	16.210
Soda -	-	-	-	-	-	-	-	24.699
Chlorine	-	-	-	•	-	-	-	10.953
Organic acids	;	-	-	-	-	-	-	$3 \cdot 200$
•								
								99·461 S.

2. Sheaths.

Carbonic acid		-	-	-	-	-	-	trace.
Silicic acid	-	•	-	-	-	-	-	51.250
Sulphuric acid	d	-	-	-	-	-	-	$12 \cdot 270$
Phosphates	-	-	-	-	-	-	-	9.750
Lime -	-	-	•	•	-	-	-	$2 \cdot 139$
Magnesia	_	-	•	-	-	-	-	0.792
Potash -	-	-	-	-	-	-	-	7.488
Soda -	-		-	-	-	-	-	$12 \cdot 449$
Chlorine	-	-	-	-	-	-	-	2.960
Organic acids	:	-	-	-	-	-	-	trace.

99·098 S.

3.	Sheaths of h	usks.								
	Carbonic aci	id	-		-	-	10	٠	trace.	
	Silicic acid	-	-	•	-		-	-	47.650	
	Sulphuric ac	cid	-	-	-	•	-	-	6.674	
	Phosphates	-	-		٠	-		-	$26 \cdot 250$	
	Lime -	•	*	•	-	٠	-	_	0.450	
	Magnesia	-	-	•	-	-	-		0.072	
	Potash -		•	•		ь	-		3.512	
	Soda -	-	-	-	-	-	-	-	9.832	
	Chlorine, or	ganic	acids	and	loss	-		9	5.560	
		Ŭ								
4	Leaves.								100.000	S.
4.									4 0 70	
	Carbonic act	ıd			•	•	•		4.050	
	Silicic acid	-	-	-	-	-	-	-	58.650	
	Sulphuric ac	ad	•	-	•	-	•	-	4.881	
	Phosphates	-	-		10	•	-	-	5.850	
	Lime -	•	-	10	•	4	-	-	4.510	
	Magnesia	-	•	•	•	•	ю.		0.864	
	Potash -	*	-	-	-	-	-	-	7· 333	
	Soda -	4	-	-	*	-	•	•	8.520	
	Chlorine	-	-	•	•	-	-	•	2.664	
	Organic acid	ls	-	-				•	2.200	
									${99 \cdot 342}$	C!
_	27 7 (71)									ι,
5.	`	e two	follo	wing	anal	yses a	are no	ot co	mplete).	
	Silicic acid	-	-	-	•	•	•	-	61.056	
	Phosphates	-	-	-	•	-	-	-	9.833	
	Lime -	-		-	10	•	-	-	2.349	
	Magnesia	-	•	•	-	-	-	-	0.640	
	Potash -	-	-	-	-	•	-	-	6.861	
	Soda -	-	-	-	-	-	•	-	8.899	
	Carbonic and	d sulp	huric	acids	s, chlo	rine,	organ	ic		
	matter and	l loss	-	•			•	-	10.362	
									100.000	Q
6.	Roots.								100.000	Ŋ,
•	Silicic acid	_	_			_	_	_	23.608	
	Phosphates		_	_	_	_	•	-	11.856	
	Lime -	-	•	•	•	•	•	•	4.666	
	Magnesia		•	-	-	••	•	-	1.039	
	Potash -	•	•	•		•	•	•		
	Soda -	-	•	•	-	-	**	•	11.334	
		- 11	- h		- l. l -	<u>-</u>		•	25.450	
	Carbonic and acids and		nulic	acias	, cnio	rine,	organ	1C	ൈക്കാറ	
	acius and	103\$	•	A	-		-	-	22.038	
									100.000	S.

PROXIMATE ORGANIC ANALYSIS OF THE SEVERAL PARTS OF THE EARLY WHITE-FLINT CORN.

AUGUST 2.

(For a description of the stage of growth, refer to page 176.)

1.	Stalk.	\mathbf{A}	transverse	section	was	taken	from	between	the	nodes near	its	base.
	100 g	rs.	gave of									

								•	Calculated without the wa	ter.
Starch	•	•	•	100	•			trace.		
Sugar an	d ext	ractiv	e m ati	ter sol	uble i	in alco	ohol			
and wa										
ble in		•	_		•		-	1.670	41.666	
Fibre an			_	-	_	-	-	0.095	$2 \cdot 400$	
Fibre				-	-	-	-	1.400	$35 \cdot 353$	
Matter d	issolv	ed ou	t of fi	bre b	y a w	eak s	olu-			
tion of	f pota	sh:r	eseml	oles a	lbume	en	-	0.535	13.510	
Albumen	ı -	-	-	ta	-	-	-	0.125	$3 \cdot 157$	
Casein	-	-	-	-	-	-	-	trace.		
Dextrine	or gu	ım	-	-	-	-	-	$0 \cdot 155$	3.914	
								$\frac{-}{3.980}$	100.000	
***									100.000	
\mathbf{W} ater	•	•	•	•		ь	-	95.030		
								99·010 S.		
								99.010 9.		

Sheaths.	100	grs. g	gave	of					
									Calculated without the water.
Starch	•	-	•	•	-	ta	•	none.	
Sugar an	nd ex	tractiv	e ma	tter s	oluble	e in a	lco-		
hol, 1	·65;	insolu	ıble,	0.24	-			1.890	$25 \cdot 233$
Fibre wi	th ch	loroph	yll	-	-	-	•	0.245	$3 \cdot 257$
Fibre	-	-		-	-	-		3.290	$43 \cdot 672$
Matter d	lissolv	ed ou	t of fi	bre b	y a w	eak s	olu-		
tion of	caus	tic pot	ash:	resen	ibles a	albun	ien,	$1 \cdot 120$	14.861
Albumei	n -	-	-	-	-	-	ь	0.210	$2 \cdot 793$
Casein	•	-	-	-	•	•	-	0.180	$2 \cdot 395$
Dextrine	or g	um	-	•	•	•	•	0.585	7.789
								$\frac{-}{7 \cdot 520}$	100.000
								• • • • • •	$100 \cdot 000$
Water	•	24	-	α	•	•	•	91.485	
								00, 005 6	
								99·005 S.	

3. Leaves. 100 grs. gave of			
Starch		none.	Calculated without the water.
Sugar and extractive matter solul mostly sugar, 1:11; insolubl	•		
1.89	• • •	3.000	19.961
Fibre with a little chlorophyll		1.390	$9 \cdot 248$
Fibre		8.620	$57 \cdot 352$
Matter separated from the fibre solution of caustic potash:			
bumen		0.290	$1 \cdot 929$
Albumen	• • •	trace.	
Casein		0.730	4.857
Dextrine or gum		0.850	$5 \cdot 655$
Oil and wax	• • •	$0 \cdot 150$	0.998
Water • • •		$\frac{-}{15 \cdot 030}$ 86 · 782	100 · 000
water			
		100·812 S.	
4. Tassels.			Calculated without the water.
Sugar and extractive matter solub	ole in alcohol,		Calculated without his water.
0.70; insoluble, 2.10 -		2.800	$34 \cdot 272$
Fibre with a little chlorophyll as	nd starch -	0.670	8.201
Fibre		1.830	$22 \cdot 398$
Matter separated from fibre by verally in alcohol and a wea	k solution of		
caustic potash : oil, wax and	albumen -	1.520	18.605
Albumen		0.420	$5 \cdot 141$
Casein		trace.	
Dextrine or gum		0.930	11.383
Water		$ \begin{array}{r} \hline 8 \cdot 170 \\ 90 \cdot 379 \end{array} $	100.000
		98·549 S.	

29

AUGUST 23.

(For description of the plant at this date, refer to page 181.)

1. Butt stalk. A transverse section was taken from between the 3d and 4th nodes.

									Calculated without the
Starch	-	•	-	•	-	-	- n	nere trace.	
Sugar and	l extr	active	matt	er, m	ostly	sugar	•	3.090	$32 \cdot 543$
Fibre with	h a lit	ttle ch	lorop	hyll	•		4	0.295	3 · 106
Fibre	-	-	-	•	-	-	-	4.200	$44 \cdot 234$
Matter ser	oarate	d fro	m fib	re by	a we	ak so	lu-		
tion of c	causti	c pota	sh:1	esem	bles a	ılbum	en,	1.160	12.218
Albumen	-	-	-	•	-	-	-	0.015	0.158
Casein	-	-	-	٠	۰	•	-	0.240	2.528
Dextrine	-	-	•	-	-	-	-	0.495	$5 \cdot 213$
								$\overline{9\cdot495}$	$\frac{100.000}{100.000}$
Water	0	••	•	•	•	٠	•	88.620	
								98·115 S	

2. Top stalk. A transverse section was taken from between the 5th and 6th nodes.

								C	alculated without the
Sugar an	d ex	tractiv	e ma	tter, 1	nostly	y suga	ar -	$2 \cdot 175$	$31 \cdot 224$
Fibre wit	hal	little c	hloro	phyll	-	-	-	0.445	6.389
\mathbf{Fibre}	-		-	•	-	-		$2 \cdot 925$	41.997
Matter di	ssolv	ed out	of f	ibre b	y a w	eak s	olu-		
tion of					•			0.825	11.846
Albumen	-	•	-	-	-	-	-	0.075	1.077
Casein	-	-	-	-	•	-	-	0.060	0.862
Dextrine	-	-		•	-	-	-	0.460	$6 \cdot 605$
								6.695	100.000
Water		•		•	٠	•	-	$92 \cdot 399$	
								99·094 S.	

e³

3	Tassels.										
0.		l o m t m	- ativo	matt	را د د د د	.hio:		h.al		Calculated without the water	٠.
	Sugar and 3.08;						n arco	noı,	7.540	26 • 490	
	Fibre with						tarch		7.640	26 · 852	
	Fibre, oil				- -	ana s			7.250	25·468	
	Matter dis					v a w	eak s	olu-	1 200	20 400	
	tion of				-		-	•	4.260	14.958	
	Albumen		.o po		-	-	_	-	0.110	0.365	
	Casein				_		-		0.150	0.526	
	Dextrine	-		-			-		1.510	5.331	
	TTY								28.460	100.000	
	Water	•	•	-	•	•	•	•	69.255		
									97.715	S.	
4.	Sheaths.									Calculated without the water	,
	Sugar and	l exti	active	mat	ter, n	nostly	suga	ar -	3 • 190	28·687	•
	Fibre with					- "	•	-	0.540	$4 \cdot 856$	
	\mathbf{Fibre}	-	-	-	•	•	-	-	4.780	$42 \cdot 985$	
	Oil and w	ax	-	-	-	•	•	-	0.150	1.348	
	Matter dis							olu-			
	tion of	potas	h : re	semb	les al	bume	n -	-	1.770	$15 \!\cdot\! 927$	
	Albumen	•	-	-	-	•	-	• -	0.020	$0\cdot 179$	
	Casein	•	•	-	-	-	-	-	0.010	0.090	
	Dextrine	•	•	•	-	•	-	•	0.660	$5 \cdot 928$	
									11.120	100.000	
	Water	-	-	-	-	-	-	-	87.947		
									99.067	· ·	
									33 001	D•	
5.	Leaves.									Calculated without the water	
	Sugar and	l exti	ractive	ma	tter	-		-	$3 \cdot 300$	18.856	•
	Fibre wit					-		•	$3 \cdot 350$	$19 \cdot 142$	
	\mathbf{Fibre}	-	-	-	•	-	-	-	6.950	$39 \cdot 720$	
	Matter dis										
	tion of	caust	ic pota	ash:	resem	bles a	albun	ien,	2.380	13.600	
	Albumen	-	-	-	•	-	•	-	0.020	0.113	
	Casein	-	-	-	-	•	•	•	0.100	0.571	
	Dextrine		-	-	-	-	-	•	1.270	7.256	
	Oil and w	ax	•	•	•	•	-	-	0.139	0.742	
									17.500	100.000	
	Water	•	-	-	-		-	•	$82 \cdot 482$		
									99.982	S.	

6.	Sheaths of	hus	ks.								
••	Sugar and			natte	r solu	ble ir	alcol	hol,		Calculated without the	water.
	mostly								4.650	$50 \cdot 434$	
	Fibre with	h a li	ittle chl	orop	yhll	-	-	•	0.280	$3 \cdot 037$	
	Fibre	-	-	<u> </u>		•			1.880	$20 \cdot 391$	
	Matter dis	solve	ed out (of fil	re by	a we	eak so	olu-			
	tion of	causi	tic pota	sh:	album	en	-	-	1.870	$20 \cdot 282$	
	Albumen	•	-	-	•	-	-	-	0.200	2.160	
	Casein	-	•	-	•	•	•	•	0.010	0.108	
	Dextrine	-	•	-	•	•	•	•	0.210	2.278	
	Oil and w	ax	-	•	-	•	-	-	0.120	1.301	
									$\frac{-}{9 \cdot 220}$	100.000	
	337 4			Ü					89.083	100,000	
	Water	•	•	-	•	•	•	•	99.099		
									$98 \cdot 303$	S.	
7.	Silks.										
	Sugar and									Calculated without the	e water.
	mostly : Fibre wit								4.374	$54 \cdot 964$	
	rophyll	-	•		-	-	-	-	0.860	10.806	
	Fibre	-	-	-		-	-	•	1.266	1 5 ·909	
	Matter di	ssolv	ed out	of fil	ore by	a we	eak so	olu-			
	tion of								0.846	$10 \cdot 631$	
	Albumen	-	-	-	•	•	•	•	0.006	0.076	
	Casein	•	-	•	•	-	-		0.226	$2 \cdot 839$	
	Dextrine	-	•	•	•	7	-	-	0.380	4.775	
									7.958	100.000	
	Water								92.269	100 000	
	W atci										
									100 · 227	S.	
8.			cernel						iusks.		
	Sugar and									Calculated without th	e water.
	mostly	suga	r, 4·18	3; i	nsolul	ole in	alco.	hol,	4 040		
	0.707	٠,	. ,	•.	•	•	-	-	4.840	59.827	
	Fibre wit	hal	ittle sta	rch	•	•	•	•	0.606	7.491	
	Fibre	٠,	•	-	• ,	•	•	•	1.006	$12 \cdot 435$	
	Matter di				ore by	a w	eak so	olu-	1 100	14 000	
	tion of	caus	tic pota	sh	•	•		•	1.186	14.660	
	Albumen	•	•	•	•	-	-	-	0.086	1.063	
	Casein	•	•	•	•	•	•	-	trace.	4.504	
	Dextrine	•	•	•	•	-	-	-	0.366	4.524	
									8.090	100.000	
	Water	-	-	-	-	-	-	-	91.095		
										~	
									99•185	S.	

AUGUST 30.

About one-third of the bulk of each kernel protrudes from its husks. For a description of the plant at this date, refer to page 183. A proximate organic analysis was only made of the kernels and cob.

Su man and		:	44		.blo to	1 1	h _ 1		Calculated without the wat
Sugar and									
mostly s	sugai	, 3.	3U; 11	isoiut	ne in	aicoi	101,	9.67	45.905
	•	•	- C1	•	•	•	•	3.67	45.365
Starch wit			nbre	•	•	•	•	1.04	12.855
Fibre or e					•	•	, -	1.42	$17 \cdot 553$
Matter dis								1 00	10
tion of	caust	ic pot	ash:	resem	ibles a	album	ien,	1.03	12.732
Albumen	•	-	•	-	•	-	•	0.21	2.595
Casein	•	•	•	-	•	•	•	0.08	0.989
Dextrine	•	-	-	-	•	•	•	0.64	7.911
								8.09	100.000
Water	-		-	-	•	•	•	90.80	200 000
									
								98·89 S.	
								98·89 S.	
Coh								98·89 S.	
Cob.								98·89 S.	Calculated without the wa
Cob.	l extr	active	e matt	er sol	uble ii	n alco	hol,	98·89 S.	
0111								98·89 S.	
Sugar and								98·89 S. 3·725	
Sugar and mostly	suga •	r, 3·	47; i						Calculated without the wa
Sugar and mostly 0.255	suga •	r, 3·	47; i					3·7 25	Calculated without the way $36\cdot 556$
Sugar and mostly 0.255 Fibre with	suga • nali	r, 3·	47; is tarch	nsolul - -	ole in - -	alco	hol, - -	3·725 1·710	Calculated without the way $36 \cdot 556$ $16 \cdot 782$
Sugar and mostly 0.255 Fibre with Fibre Matter dis	suga • • a li • • •ssolv	r, 3·	47; istarch	nsolul	ble in	alco	hol,	3·725 1·710	Calculated without the way $36 \cdot 556$ $16 \cdot 782$
Sugar and mostly 0.255 Fibre with	suga n a li ssolve	r, 3·	47; istarch	nsolul	ble in	alco	hol,	3·725 1·710 2·890	Calculated without the war 36 · 556 16 · 782 28 · 357
Sugar and mostly 0.255 Fibre with Fibre Matter distion of a	suga n a li ssolve	r, 3·	47; istarch	nsolul	ble in	alco	hol,	3·725 1·710 2·890 1·210	36.556 16.782 28.357 11.875 1.571
Sugar and mostly 0.255 Fibre with Fibre Matter distion of a Albumen	suga n a li ssolve	r, 3·	47; istarch	nsolul	ble in	alco	hol,	3·725 1·710 2·890 1·210 0·160	Calculated without the war 36 · 556 16 · 782 28 · 357 11 · 875
Sugar and mostly 0.255 Fibre with Fibre Matter distion of a Albumen Casein	suga n a li ssolve	r, 3·	47; istarch	nsolul	ble in	alco	hol,	3·725 1·710 2·890 1·210 0·160 0·050 0·445	36.556 16.782 28.357 11.875 1.571 0.491 4.368
Sugar and mostly 0.255 Fibre with Fibre Matter distion of a Albumen Casein	suga n a li ssolve	r, 3·	47; istarch	nsolul	ble in	alco	hol,	3·725 1·710 2·890 1·210 0·160 0·050	36.556 16.782 28.357 11.875 1.571 0.491

SEPTEMBER 6.

Kernels about two-thirds protruded from their husks, and about two-thirds grown. For a description of the plant at this date, refer to page 185. A proximate organic analysis was only made of the kernels and cob.

4.	Kern	LUD.

Sugar and	d ext	ractive	mai	ter, n	ostly	sugai	r -	7.60	Calculated without the water. 39.015
Starch	-	-	•		-	•		8.06	$41 \cdot 376$
Fibre or	epide	rmis,	and	oil	-	-	-	$1 \cdot 24$	6.366
Matter di	ssolv	ed out	of e	epider	mis b	y a w	eak		
solution	of	caustic	pot	ash:	reser	nbles	al-		
bumen	-	•	-	-	-	-	-	1.48	7.597
Albumen	•	•	-	-	-	-	-	0.69	$3 \cdot 542$
Casein	-	-	-	-	-	-	-	$0 \cdot 05$	$0\cdot 256$
Dextrine	-	-	-	-	-	-	-	0.36	1.848
								19.48	100.000
Water	-	•	-	•	•	-	-	$80 \cdot 43$	
								99·91 S.	

Ether dissolved out of 100 grs. of fresh kernels, 2.64 grs. of oil, and a sweetish matter, probably sugar, which was taken up by the water in the corn.

2.	Cob.
~·	C00.

									Calculated without the water.
Sugar and	l extr	active	e mat	ter, n	nostly	suga	r -	$5 \cdot 46$	34.340
Starch	•	-	-	-	•	•	-	0.08	0.503
Dextrine	or gu	m	-	-	-	-	-	0.76	4.779
Fibre and	l resin	a	-	-	•	-	-	5.94	$37 \cdot 359$
Matter dis	solve	d out	of fi	bre by	a w	eak so	olu-		
tion of c	causti	c pota	ash:	resem	bles :	album	en,	1.50	$9 \cdot 434$
Albumen	-	-		-	•	-	-	1.52	9.560
Casein	-	-		-	•	-	-	0.62	$4 \cdot 025$
									
								15.90	$100 \cdot 000$
Water	-	-		-	-	-	-	82.41	
								98·31 S.	

SEPTEMBER 13.

Corn in advanced stage of milk. Kernels nearly full size. For proportions and specific gravity of kernels, refer to page 188.

PROXIMATE ORGANIC ANALYSIS OF THE KERNELS AND COB.

J., ., .,	l	- + -						£.100	Calculated without the water
_	i extra	.ciive	matte	r, m	iostry	sugar	•	5.100	25.053
	-	•	•	•	•	-	•		47.861
	-	-			•	•	•	2.435	11.966
			_		-				
solution	of ca	ıustic	potas	h :	resem	bles a	.]~		
bumen	•	-	-	•		-	•	1.530	$7 \cdot 523$
Albumen	•	•	•	•	-	-	•	0.845	4.149
Casein	-		-	•	•	•	•	0.040	0.196
Dextrine o	or gum	1	-	•	•	•	•	0.660	$3 \cdot 251$
								20:365	100.000
Vater	_						•		200 000
								99·118 S.	
ob.									Calculated without the water
Sugar and	l extra	ctive	matte	r. m	ostly	sugar		PM . O. 4	
		-	********	-,		24541	•	$7 \cdot 34$	$23 \cdot 724$
Starch	-	-	•	•	-	-	•	0.06	$23 \cdot 724$ $0 \cdot 193$
Starch Fibre and	-	-	•	-	-	•	•		•
Fibre and	- resin	-	:	-	•	•	•	0.06	0.193
Fibre and Matter dis	resin ssolved	- • l out	of fibr	e by	a we	ak sol	- - u-	0.06	0·193 49·742
Fibre and Matter dis tion of c	resin ssolved	- • l out	of fibr	e by	a we	ak sol	- - u-	0·06 15·39 6·45	0·193 49·742 20·847
Fibre and Matter dis tion of o	resin ssolved	- • l out	of fibr	e by	a we	ak sol	- - u-	0·06 15·39 6·45 1·21	0·193 49·742 20·847 3·911
Fibre and Matter dis tion of d Albumen Casein	resin ssolved caustic	l out	of fibr	e by	a we	ak sol	- - u-	0·06 15·39 6·45 1·21 0·05	0·193 49·742 20·847 3·911 0·161
Fibre and Matter dis tion of o	resin ssolved caustic	l out	of fibr	e by	a we	ak sol	- - u-	0·06 15·39 6·45 1·21	0·193 49·742 20·847 3·911
Fibre and Matter dis tion of d Albumen Casein	resin ssolved caustic	l out	of fibr	e by	a we	ak sol	- - u-	0·06 15·39 6·45 1·21 0·05	0·193 49·742 20·847 3·911 0·161
Fibre and Matter dis tion of d Albumen Casein	resin ssolved caustic	l out	of fibr	e by	a we	ak sol	- - u-	0·06 15·39 6·45 1·21 0·05 0·43	0·193 49·742 20·847 3·911 0·161 1·422
Fibre and Matter dis tion of c Albumen Casein Dextrine c	resin ssolved caustic	l out	of fibr	e by	a we	ak sol	- - u-	0·06 15·39 6·45 1·21 0·05 0·43 30·94	0·193 49·742 20·847 3·911 0·161 1·422
	Starch Cibre or e fatter dis solution bumen Albumen Casein Dextrine of	Starch - Cibre or epiderr Statter dissolved solution of ca bumen - Clbumen - Casein - Dextrine or gum Vater -	Starch	Starch	Starch	Starch	Starch	Tatter dissolved out of epidermis by a weak solution of caustic potash: resembles albumen	Starch 9.755 Sibre or epidermis, and oil 2.435 Matter dissolved out of epidermis by a weak solution of caustic potash: resembles albumen 1.530 Albumen 0.845 Casein 0.040 Dextrine or gum 0.660 20.365 Vater 78.753 99.118 S.

1. Kernels.

OCTOBER 18.

Corn ripe. For proportions, and specific gravity of kernels, refer to page 190.

PROXIMATE ORGANIC ANALYSIS OF THE KERNELS AND COB.

									Calculated without the water
Oil -	•	•	•	•	•	•	•	3.60	3.896
Zein or gl	uten	•	•	•	•	•	•	1.68	1.818
Starch	•	-	•	•	•	•	•	59.30	$64 \cdot 169$
Sugar and	l extr	active	matt	er	-	•	•	$13 \cdot 32$	14 · 415
Fibre or e	pider	mis	•	•	•	•	•	0.89	0.964
Matter dis	ssolve	d out	of fib	re by	a we	ak so	lu-		
tion of o	austic	c pota	sh:re	esemb	les a	lbume	en,	$5 \cdot 99$	$6 \cdot 482$
Albumen	•	•	•		•	•	•	$4 \cdot 29$	4.642
Casein	•	•	•	•	•	•	•	0.08	0.086
Dextrine of	r gun	n	-	•	•	•	•	3.26	3.528
								$92 \cdot 41$	100.000
Water	•	•	•	-	•	•	-	8.45	
								100·86 S.	
Cob.									Calculated without the water
Sugar and	evir	active	matt	e r		_		4.700	6.791
Starch	-	•	•	•		_		trace.	0.01
Fibre			-		•			44.188	$63 \cdot 845$
Matter dis	solve	d out	of fib	re hv	a we	ak sol	n-	11 100	00 020
	ausnc	: nota:	$\operatorname{sh}:re$	esemb	oles al	lbume	n.	15.712	22.702
Albumen		pota:	sh : re	esemt	les a	lbume •	en,	15·712 0·525	22·702 0·759
Albumen		pota:	sh : re -	esemb •	les a	lbume •	en, •	0.525	0.759
Albumen Casein	•		sh : re	esemb	les a	bume - - -	en,	0 · 525 0 · 100	0·759 0·144
Albumen	•		sh : re	esemb	oles a	lbume	en, -	0 · 525 0 · 100 0 · 800	0·759 0·144 1·155
Albumen Casein Dextrine o Resins	• or gun	n	• •	•	•	•	•	0 · 525 0 · 100	0·759 0·144
Albumen Casein Dextrine o	or gun	n d out	- - - of b	• • • • odies	•	•	•	0 · 525 0 · 100 0 · 800	0·759 0·144 1·155
Albumen Casein Dextrine o Resins Matter di	or gun	n d out	- - - of b	• • • • odies	•	•	•	0·525 0·100 0·800 0·625	0·759 0·144 1·155 0·903
Albumen Casein Dextrine of Resins Matter di water,	or gun	n d out	- - - of b	• • • • odies	•	•	•	0.525 0.100 0.800 0.625 2.562 	0·759 0·144 1·155 0·903
Albumen Casein Dextrine o Resins Matter di	or gun	n d out	- - - of b	• • • • odies	•	•	•	0·525 0·100 0·800 0·625	0·759 0·144 1·155 0·903

OCTOBER 5.

The corneous portion of the kernels is quite firm, or glazed, as it is called. The amylaceous part around the chit is yet soft and milky. At this stage, the kernels burn into a white ash much more readily than when ripe.

Composition of the ash of the kernels.

Carbonic ac	id	•	-	•	•	•	•	trace.	
Silicic acid	-	-	-	-	-	•	-	0.900	
Phosphoric	acid,	with a	little	perox	cide of	f iron	-	50.010	
Lime -	-	-	-	-	-	-	-	0.075	
Magnesia	-	-	-	-	•	-	-	4.240	
Potash -	-	-	-	-	-	-	•	$28 \cdot 405$	
Soda -	-	-	-	•	-	•		11.840	
Sodium	-	-	-	-	-	-	•	0.345	
Chlorine	-	-	-	-	•	-	-	0.520	
Sulphuric a	cid	-	-	-	-	-	-	1.240	
Organic aci	ds	-	-	-	-	-	-	$2 \cdot 100$	
_									
								$99 \cdot 675$	S.

OCTOBER 18.

Corn ripe.

Composition of the ash of the kernels.

Carbonic aci	id	-	-	-	-	-	-	trace.
Silicic acid	-	-	-	-	-	-	-	0.850
Phosphoric	acid,	with	a litt	le pe	roxide	e of in	on,	60.310
Lime -	-	-	-	-	-	-	-	0.075
Magnesia	-	-	-	-	-	-	-	6.500
Potash -	-	-	-	-	-	-	-	$23 \cdot 175$
Soda -	-	-	-	-	-	-	-	3.605
Sodium	-	-	-	-	-	-	-	0.160
Chlorine	-	-	-	-	-	-	-	0.295
Sulphuric a	cid	-	-	-	-	-	-	0.515
Organic acid	ds -	-	-	-	-	-	-	5.700
								99·175 S.

Composition of the soil on which the plants of the Early White-flint corn were grown (Yard back of the Old State House).

Water -	-	-	-	-	-	-	-	$6 \cdot 205$
Organic ma	tter	-	-	-	-	-	-	4.750
Silex -	•	-	-	-	-	-	-	82.620
Alumina an	d per	oxide	of in	on	-	-	-	$4 \cdot 165$
Lime -	-	-	-	-	-	-	-	$1 \cdot 025$
Magnesia	-	-	-	-	-	-	-	0.200
Potash -	•	-	-	•	-	-	-	$0 \cdot 137$
Soda -	-	-	-		-	-	-	0.554
Sulphuric ac	cid	-	-	-	-	-	-	0.160
Chlorine	-	- •	-	-	-		-	0.065
Phosphates	-	-	-	-	-	-	-	0.025
								
								99·906 S.

The following observations upon the growth of indian corn were made by Mr. Ball, of Hoosic Corners, Rensselaer county, this last season (1848):

"Six varieties planted on a field descending to the north, on which a crop of eight-rowed yellow corn was raised the last year.

Composition of the soil before manuring.

Water	-	•	-	-	-	-	-	-	0.00
Organic	mat	ter	-	-	-	-	-	-	7.85
Silex	-	-	-	-	-	-	-	-	86.50
Alumina	and	perox	ide o	f iro	n -	-	-	-	$5 \cdot 50$
Carbona	te of	lime	-	-	-	-	-	-	trace.
Magnesi	a	-	-	-	-	-	-	-	trace.
Loss	-	-	-	-	-	-	-	-	$0 \cdot 15$
									100.00

Compost was applied, containing

20 parts of barnyard manure;

10 "hog manure;

10 " wood ashes, leached;

1 " lime;

1 " plaster.

Composition of the soil after manuring.

Water	-	-	-	-	•	-	-	-	00.00
Organic	matt	er	-	•	-	-	•	-	10.38
Silex	-	-	-	-	-	-	-	-	$69 \cdot 34$
Alumina	and	perox	ide of	iron	-	-	-	-	10.50
Carbonat	e of	lime	-	-	-	-	-	-	6.53
Magnesia	a	-	•	-	-	-	-	-	0.96
Sulphate	of l	ime	-	-	-	-	-	-	1.74
Soluble s	silica	-	-	-	-	-	-	-	0.31
Chloride	of so	odium		-	-	-	-	-	0.06
\mathbf{P} otash	- .	-	-	-	-	-	-	-	0.09
Loss	-	-	-	-	-	-	-	-	0.08
									100.00

Wood ashes used in the hill.

6.00

Charcoal	-	-	-	-	-	-	-	0.57
Silex -	-	-	-	-	-	-	-	9.85
Carbonate	of lime		-	-	-	-	-	66.75
Magnesia	-	-	-	-	•	-	-	$3 \cdot 40$
Phosphate	es:							
of	peroxide	of i	ron	-	-	-	3.62	
Lime	-	-	-		-	-	$3 \cdot 25$	
Magn	esia	-	-	-	-	-	2.19	
								9.06
Potash -	-	-	•	-	-	-	•	3.00
Chloride o	f sodiur	n	-	-	-	•	-	0.44
Soluble si	lica -	-	-	-	-	•	-	0.08
								$99 \cdot 15$

One whole stalk of corn (eight-rowed brown), four weeks after planting.

Whole height 18 inches.

			PROI	ORTION	s.				
Stalk	•	-	•	-	-	-	-	-	60:00
Leaf she	eaths	•	-	-	-	•	•	-	$60 \cdot 50$
Leaf -	-	-	-	-	-	-	•	-	$68 \cdot 50$
									189.00

	ter in stal		-	-	-	•		56.00		
Ash	in do	-	-	-	-	-		0.10		
Org	anic matt	er -	-	-	-	-		3.90		
									$60 \cdot 00$	
Wat	er in leaf	shoot	.h	_	_	_		55.50		
Ash			.11 -	_	-	-				
		do	-	-	-	•		0.25		
Orga	anic matte	er -	-	-	•	•		4.75	00 F0	
									$60 \cdot 50$	
Wat	er in leaf	_	-	-	-	-		57.25		
$\mathbf{A}\mathbf{s}\mathbf{h}$	in do	-	_	_		-		1.37		
	anic matte				_			9.88		
Oig	anic matt	-	_	_	_	_			68.50	
In the	whole sta	ılk:								
Wat	er -	_	_	_		_	-	<u>.</u> .	168.75	
Ash		_	_•	_	_	_	_	_	1.72	
		-	-	_	_	_	_	•		
Org	anic matt	er	•	-	•	•	-	•	18.53	100.00
										189.00
0	. 1 4 77 .	. c	/-	7.4		2	\	.c	1C1 .	7
One wn	ioie staik	of co	rn (e	ignt-r	·owea	orow	n),	, $nve w$	eeks ajte	er planting.
				Whole	height	32 inc	ches			
				P	ROPORT	IONS.				
	Stalk	_	_	_					_	307.50
	Leaf s			_						
		sucam							_	122.00
		_		_	•	•		• -	-	133.00
	Leave	s -	-	•	-	-			-	$133 \cdot 00$ $181 \cdot 00$
	Leave	s -	-	-	•	•			-	181.00
	Leave	s -	-	•	•	-			-	
Wat			-	-		•	9	- - 287·40	-	181.00
	ter in stal				-		<i>5</i>	287·40	-	181.00
Ash	ter in stal in do	k -	-		-	-	4	2.87	-	181.00
Ash	ter in stal	k -	-	-	-	-	ş	2·87 17·23	207.50	181.00
Ash Orga	ter in stal in do anic matte	k - - er -		-		•	<u> </u>	2·87 17·23	307.50	181.00
Ash Orga	ter in stal in do	k - - er -					_	2·87 17·23	- - 307·50	181.00
Ash Orga	ter in stal in do anic matte ter in leaf	k - - er -					_	2·87 17·23	- - 307·50	181.00
Ash Orga Wai Ash	ter in stal in do anic matte ter in leaf in	k - er - Sshea do					_	2·87 17·23 	- - 307·50	181.00
Ash Orga Wai Ash	ter in stal in do anic matte ter in leaf	k - er - Sshea do	- - - th				_	2·87 17·23 119·69 2·00 11·31	307·50 133·00	181.00
Ash Orga Wai Ash Org	ter in stal in do anic matte ter in leaf in anic matt	k - er - shea do er -	- - - th				1	2·87 17·23 119·69 2·00 11·31		181.00
Ash Org Wat Ash Org	ter in stal in do anic matte ter in leaf in anic matte	k - er - shea do er -	- - - th				1	2·87 17·23 119·69 2·00 11·31		181.00
Ash Org: Wat Ash Org	ter in stal in do anic matte ter in leaf anic matte ter in leaf in do	k - er - shea do er -	- - - th				1	2·87 17·23 119·69 2·00 11·31 146·00 4·03		181.00
Ash Org: Wat Ash Org	ter in stal in do anic matte ter in leaf in anic matte	k - er - shea do er -	- - - th				1	2·87 17·23 119·69 2·00 11·31	1 3 3·00	181.00
Ash Org: Wat Ash Org	ter in stal in do anic matte ter in leaf anic matte ter in leaf in do	k - er - shea do er -	- - - th				1	2·87 17·23 119·69 2·00 11·31 146·00 4·03		181.00
Ash Org: Wat Ash Org	ter in stal in do anic matte ter in leaf anic matte ter in leaf in do	k - er - shea do er -	- - - th				1	2·87 17·23 119·69 2·00 11·31 146·00 4·03	1 3 3·00	181.00
Ash Org Wat Ash Org	ter in stal in do anic matte ter in leaf in anic matte ter in leaf in do anic matt	k - er - shea do er - -					1	2·87 17·23 119·69 2·00 11·31 146·00 4·03	133.00	181.00
Ash Org Wat Ash Org Wat Ash Org	ter in stal in do anic matte ter in leaf in anic matte ter in leaf in do anic matt	k - er - shea do er - er whole					1	2·87 17·23 119·69 2·00 11·31 146·00 4·03	133·00 181·00 	181.00
Ash Org Wat Ash Org Wat Ash Org	ter in stal in do anic matte ter in leaf in anic matte ter in leaf in do anic matt	k - er - shea do er - er whole					1	2·87 17·23 119·69 2·00 11·31 146·00 4·03	133·00 181·00 	181.00
Ash Org Wat Ash Org Wat Ash Org	ter in stal in do anic matte ter in leaf in anic matte ter in leaf in do anic matt	k - er - shea do er - er whole					1	2·87 17·23 119·69 2·00 11·31 146·00 4·03	133·00 181·00 	181·00 621·50
Ash Org Wat Ash Org Wat Ash Org	ter in stal in do anic matte ter in leaf in anic matte ter in leaf in do anic matt	k - er - shea do er - er whole					1	2·87 17·23 119·69 2·00 11·31 146·00 4·03	133·00 181·00 	181.00

One whole stalk of corn (eight-rowed brown), six weeks after planting. Whole height 39 inches.

		PR	OPORTIO	Ns.		•	
Stalk -	-	-	-	-			611.00
Leaf sheaths	-	-	-	-			$279 \!\cdot\! 25$
Leaf -	-	•	-	-		-	$427 \cdot 00$
							1317.25
Water in stalk -	-	-	-	-	568.25		
Ash in do -		-	-	-	4.00		
Organic matter	•	-	_	-	38.75		
o .						611.00	
Water in leaf sheath	s	4	-	-	252.00		
Ash in do		•	-	-	2.75		
Organic matter -	-	-	-	-	24.50		
						$279 \cdot 25$	
Water in leaf -	-	-	-	-	$358 \cdot 94$		
Ash in do -	-	-	•	-	$5 \cdot 25$		
Organic matter -	-	-	-	-	$62 \cdot 81$		
0						$427 \cdot 00$	
Water in the whole	-		-	-	-	1179 · 19	
Ash in do	-	-	-	-	•	12.00	
Organic matter -	-	-	_	-		126.06	
5							1317 • 25

One whole stalk of corn (eight-rowed brown), seven weeks after planting. Whole height 53 inches.

		P	ROPORT	ions.				
Stalk -	-	-	-	-	-	-	-	$1230 \cdot 00$
Leaf sheaths	-	-	-	-	-	-	-	$289 \cdot 50$
Leaf -	-	-	-	-	-	-	-	$763 \cdot 62$
					•			$2283 \cdot 12$

Water in stalk	•	-	-	-	1167.00	
Ash in do -	-	-	-	-	8.50	
Organic matter	-	-	-	-	$54 \cdot 50$	
						1230.00
Water in leaf she	aths	-	-	-	255.75	
Ash in do		-	-	-	4.50	
Organic matter	-	-	-	-	$29 \cdot 25$	•
						$289 \cdot 50$
Carried forward				-		${1519 \cdot 50}$

Brought forward	-	-			-			-	1519.50	
Water in leaf -	-	-		-		-		$630 \cdot 37$		*
Ash in do -	-	-		-		-		11.50		
Organic matter	-	-		-		-		121.75		
•									$763 \cdot 62$	
Water in the whole			-		-		-		2053 · 12	
Ash in do		-	-		-		-	-	24.50	
Organic matter -		-	-		•		-	-	$205 \cdot 50$	
_										$2283 \cdot 12$

One whole stalk of corn (eight-rowed brown), eight weeks after planting.
Whole height 68 inches.

					PROPOR	RTIONS.				
	Stalk	-	•	-	-	-	-	-	-	3000.00
	Leaf sh	eaths	; -	-	-	-	-	-	-	$963 \cdot 50$
	Leaf	-	•	-	-	-	-	-	-	$903 \cdot 75$
	\mathbf{T} assel	-	-	-	-	-	-	_	-	$317 \cdot 25$
Water	in stalk		•	-	-	-	2813	3.33		
Ash in	do		-	-	-	-	17	7.96		
Organi	ic matter	:	-		-	-	168	3.71		
J									3000.00	
Water	in leaf	sheat	\mathbf{h}	-	-	•	86	1.50		
Ash in	d	o		-	-	-	9	9.50		
Organi	ic matter	•	_	_	-	-	99	2.50		
-0									$963 \cdot 50$	
Water	in leaf	-	-	-	-	-	73	5·13		
Ash in	do	-	-	-	-	-	19	9.87		
Organ	ic matter	·	-	-	-	-	149	3.75		•
Ü							—.		$903 \cdot 75$	
Water	in Tasse	el	-	-	-	-	230	0.94		
Ash in	do		- '	-	-	-	4	1.62		
Organi	ic matter	•	_	-	-	-	8:	1.69		
0									$317 \cdot 25$	
Water	in the w	vhole		-	-	-	-	-	$4640 \cdot 90$	
Ash in	do	•		-	•	-	-	-	$51 \cdot 95$	
Organi	ic matter			-	-	-	-	-	491.65	
0										5184.50

One whole stalk of corn (eight-rowed brown), nine weeks after planting.

Whole height 78 inches.

			P	ROPORT	ions.				
Stalk	-	-	-	-	-	-	•	-	$4304 \cdot 07$
Leaf sh	eaths	-	-	_	-	-	-	-	$1706 \cdot 25$
\mathbf{Leaf}	-	-	-	-	-	•	-	-	$1789 \cdot 50$
\mathbf{Tassel}		-		-	-	-	-	-	$483 \cdot 50$
Young e	ear	-	-	-	-	-	•	-	668.00
									$8851 \cdot 32$

TABLE DESIGNED TO SHOW THE PROGRESSIVE GROWTH OF SEVERAL VARIETIES OF MAIZE RAISED IN THE TOWN OF HOOSIC, RENSSELAER COUNTY, IN 1848.

Number and name of variety.	Manure in hill. ‡	Number of row.	Day of month corn planted.	Day of month corn begins to appear above ground.	Day of month when all the corn appears above ground.	Day of month each begins to tassel out.	Day of month each begins to flower.	Day of month corn cut up when ripe.	Average height 2 weeks after planting.	Average height 4 weeks after planting.	Average height 6 weeks after planning.	Average height 8 weeks after planting.	Average height 10 weeks after planting.	Whole weight of 32 stalks.	Weight of stalk.	Weight of leaf.	Weight of leaf sheath.	Weight of husks.	Weight of shelled corn.	Weight of cobs.
Sio	H. m. Ashes.	1 2	May 15	May 19 21	May 23 25	July 15	July 21 22					72 in 58 "	İ	18.7	4·1	1.5	1.1	1.0	lbs. 8·0	oz.
8-rowed brown No. 2.	H. m. Ashes.	3		21 22	22 23	4 7	14 16	16	-				78 in		2.4	1.1	0 · 13	1.2	7.14	2.2
low	H. m. Ashes.	5 6		22 24	25 28	7 10	15 16	16	_		}		90 in 82 ''	19.13	4.4	1.7	1.2	1.4	9.6	2.6
Sweet corn. No. 4.	H. m.	7 8		22 23	25	14	19 21	19	_			60 in 57 ''	78 in	18.8	3.6	1.1	0.8	1.0	6.1	1.8
Chinese. No. 5.	H. m.	9 10		22 23	24 26	17 22	28 30	Sept 12		1			108 in 104 ''	26 · 11	9.6	2.3	1 · 10	1.13	7.6	4.5
Jersey White. No. 6.	H. m. Ashes.	11 12		19 21	23 28	17 23	28 31	Sept 14	6 in	1			102 in 96 ''	26 · 12	9 · 11	2.3	1.7	1.15	9-1	3.1

TABLE SHOWING THE CONNEXION OF THE GROWTH OF MAIZE WITH THE TEMPERATURE OF THE SOIL AND THE AIR.

No. of row.	1st and 2d	Mean	3d and 4th	Mean	5th and 6th	Mean	7th and 8th	Mean	9th and 10th	Mean
	week.	temp.	week.	temp.	week.	temp.	week.	temp:	week.	temp.
1 3 5 7 9 11	5 in. 5½ " 5½ " 4½ " 6 "	Air. 76·30 Soil. 67·76	$\begin{array}{c} 10 & \text{in.} \\ 9\frac{1}{2} & \text{``} \\ 6\frac{1}{2} & \text{``} \\ 6\frac{1}{2} & \text{``} \\ 10 & \text{``} \\ 11 & \text{``} \end{array}$	Air. 72·53 Soil. 65·73	27 in. 28 " 30 " 23 " 30 " 29 "	Air. 85·80 Soil. 80·53	30 in. 27 " 22 " 26 " 24 " 29 "	Air. 78·26 Soil. 76·13	12 in. 8 " 26 " 18 " 38 " 27 "	Air. 80·00 Soil. 74·93

Note. The mean temperature is that of the hour of 3 P. M.

"One half the corn was manured in the hill with log manure, and one half with unleached ashes. The first was properly attended to; but in that manured with ashes, owing to the carelessness of the person who had it in charge, the whole sixteen hills were put together, and the varieties so mixed that I was obliged to let it go; and of course have confined what I have done to the portion manured with hog manure.

"During a part of the months of August and September, the observations on the weather were not quite so complete as I intended to have them: winds, clouds, and rain were not always noted."

It was my purpose to have given a more complete analyis of Mr. Ball's varieties of corn. Circumstances, however, do not permit me to do this at this time. Observations on temperature, when connected with other meteorological facts, are highly important and interesting, especially when made for the purpose of illustrating their influence upon the growth and progress of vegetation. The section of country, too, where these observations are made, is remarkable for the growth of this crop. Hoosic, in Rensselaer county, is probably one of the best townships in this respect in the State; the crops being always large, and remarkably sound and perfect.

It is proper to say, also, that the analysis of the soil after manuring gave those results from the hills of corn to which the manure was directly applied. If such an amount of manure had been distributed generally through the soil of the field, its presence could hardly have been detected by analysis.

PROXIMATE ORGANIC ANALYSIS OF FIVE VARIETIES OF RIPE MAIZE.

PROPORTIONS.

1.	Golden Si	oux c	orn,	100 • 0	00 grs	. gav	е			
	Water	-		-	-	-	-	-	-	15.02
	Dry	-	-	-	-	-	-	-	-	84.98
2.	Large 8-7	rowed	yella	nv cor	n, 10	0 grs.	. gave			
	Water	-	-	•	•	-	-	-	-	14.00
	Dry	-	-	-	-	•	-	-	•	86.00
3.	Small 8-1	owed	yello	w cor	n, 10	0 grs.	gave			
	Water	-	-	-	-	-	-	-	-	$14 \cdot 03$
	Dry	•	-	-	-	-	-	-	•	85.97
4.	White-flin	nt cor	n, 10	00 grs	. gave	е				
	Water	-	-	-	-	-	-	-	-	14.00
	Dry	-	-	-	-	-	-	-	-	86.00
5.	Ohio dent	corn	100	grs.	gave					
	Water	-	-	-	-	-	-	•	-	14.50
	\mathbf{Dry}	-	-	-	-	-	-	-	-	85.50

1. Proximate analysis of the Golden Sioux corn.

A bright yellow 12-rowed variety, passing into 14 rows; frequently 14 rows at the base of the ear. It may perhaps be begarded as an improved variety of Buel's Dutton corn; as it ripens earlier, and I believe has a smaller kernel.

Starch	-	-	•	-	-	-	-	-	36.06
Gluten	-	•	-	-	-	-	-	-	5.00
Oil		-	-	-	-	-	-	-	$3 \cdot 44$
Albume	n	-	-	•	-	-	-	-	$4 \cdot 42$
Casein	-	•	-	-	-	-	-	-	1.92
Dextrin	е	•	-	•	-	-	-	-	1.30
Fibre	•	-	-	-	-	-	-	-	18.50
Sugar a	nd	extract	ive n	atter	-	-	-	-	7.25
Water	-	-	-	-	-	•	-	-	15.02
									$\overline{100.05}$

2. Organic analysis of the Ohio Dent corn. One of the largest varieties of maize.

Starch	-	•	-	-		-	-	-	41.85
Gluten	-	-	-	-	-	-	-	-	$4 \cdot 62$
Oil	-	-	-	-	-	-	-	-	3.88
Albumer	n	-		-	-	-	-	-	2.64
Casein	•	-	-	-	-	<u>-</u>	-	-	$1 \cdot 32$
Dextrine)	-	-	-	-	-	-	-	$5 \cdot 40$
Fibre	-	-	-	-	-	-	-	•	21.36
Sugar a	nd	extract	-	-	-	-	•	•	$10 \cdot 00$
Water	-	-	-	-	-	•	-	-	$10 \cdot 00$
									101.05
									$101 \cdot 07$

3. Organic analysis of the small 8-rowed corn.

Starch -	•	-	-	-	-	-	-	30.290
Gluten -	-	-	-	-	-	-	•	5.600
Oil -	-	-	-	-	-		-	3.900
Albumen	-	-	•	-	-	-	-	6.000
Casein -	-	-	•	-	•	•	-	$2 \cdot 200$
Dextrine	-	-	-	-	-	-	-	4.615
Fibre -	-	-	-	-	•	-	-	26.800
Sugar and	extract	-	-	-	-	-	-	5.200
Water -	-	•	-	-	-	-	-	$13 \cdot 400$
								00.000
								98.000

4. Analysis of White-flint corn.

Grown upon a clay loam, and manured in the hill with a mixture of coal, ashes and horsedung, and ashed with unleached ashes twice.

Starch	-	-	•	•	-	-	-	-	40.34
Gluten	-	-	-	-	-	-	-	-	7.69
Oil	-	-	-	-	-	-	-	-	4.68
Albume	n	-	-	-	-	-	-	-	$3 \cdot 40$
Casein	-	-	-	-	-	-	-	-	0.50
Dextrin	e		-	-	-	-	-	-	2.90
Sugar a	nd e	xtract	ive m	atter	-	-	-	-	8.30
Water	-	-	-	-	_	-	-	-	14.00
Fibre	-	-	-	-	-	•	-	-	18.01
									$99 \cdot 72$

The water has been determined from the ground grain, and some oil has been found upon the paper enveloping its contents: probably the water may be stated too high by one per cent. 5. Organic analysis of the larger variety of 8-rowed yellow corn.

Starch -	-	-	-	-	-	-	-	$49 \cdot 22$
Gluten -	-	-	•	-	-	-	-	$5 \cdot 40$
Albumen	-	-	-	-	-	-	-	$3 \cdot 32$
Oil -	-	-	-	-	-	-	-	3.71
Casein -	-	-	-	-	-	-	-	0.75
Fibre -	-	-	-	-	-	-	-	11.96
Dextrine	-	-	-	-	-	-	-	1.89
Sugar and	extract	-	-	-	-	-	-	$9 \cdot 55$
Water -	-	-	-	-	-	-	-	14.00
								99.80

The amount of starch in this variety was unexpected, and a small part may be set down as adherent albumen.

ADDITIONAL ANALYSES OF WHITE-FLINT CORN.

1. Analysis of the kernels of White-flint corn, cut August 22.

Grown upon the same soil as the small 8-rowed yellow, a sandy loam, and manured in part with coal ashes.

						-		
Silica -	-	-	-	-	-	-	-	9.500
Alkaline and	d ear	thy p	hosph	ates	-	-	-	$35 \cdot 500$
Lime -	-	-	-	-	-	-	-	0.160
Magnesia	-	-	-	-	-	-	-	$2 \cdot 410$
Potash -	-	-	-	-	-	-	-	$23 \cdot 920$
Soda -	-	-	-	-	-	-	-	$22 \cdot 590$
Chlorine	-	-	-	-	-	-	-	0.405
Sulphuric a	cid	-	-	-	-	-	•	$4 \cdot 385$
Organic ma	tter	-	-	-	-	-	-	0.367
								$99 \cdot 237$

2. Analysis of the leaves of the White-flint corn, cut August 22.

_	-					_		-	_
Silica	-	-	-	-	-	-	-	-	$53 \cdot 550$
Earthy	· ph	osphate	es	-	-	-	-	-	$19 \cdot 250$
$_{ m Lime}$	-	•	-	-	-	-	-	-	6.092
Magne	esia	-	-	-	-	-	-	-	1.250
Potash	-	-	-	-	-	•	-	-	12.762
Soda	-	-	-	-	-	-	-	-	8.512
Chlori	ne	-	-	-	-	-	-	-	$9 \cdot 762$
Sulphi	aric	acid	-	-	-	-	-	-	$4 \cdot 185$
									101.371
									O O! T

3. Analysis of the cob of White-flint.

Silica	-	-	-	-	-	-	$13 \cdot 600$
Earthy phosphat	es	-	-	-	-	-	$23 \cdot 924$
Lime	-	-	-	-	•	-	0.300
Magnesia -	-	-	-	-	-	-	0.900
Potash	-	-	-	-	-	-	35.802
Soda	-	-	-	-	-	-	$5 \cdot 914$
Chlorine -	-	-	-	-	-	_	0.132
Sulphuric acid	-	-	-	-		-	0.345
Organic matter	-	-	-	-	-	-	$2 \cdot 314$
Carbonic acid	-	-	-	-	-	-	$6 \cdot 134$
							89.365

In each of the foregoing results, the quantity of silica is greater than in the 8-rowed yellow corn growing beside it, and treated in the same way. The ash, in its physical properties, appeared more siliceous than it usually is, and hence I have no doubt the analyses are correct: it goes to show that the same plant may take up and assimilate a greater amount of inorganic matter under some circumstances, than in others. This corn, besides being supplied with manner of the horse, mixed with coal ashes in the hill, was ashed with unleached ashes. The consequence was that the crop gave a remarkably sound hard grain; it would seem that this treatment had some share in producing the excess of silica obtained in the foregoing analyses.

INTERMIXTURE OF VARIETIES OF MAIZE.

Farmers are familiar with the fact that two or more varieties of maize grow upon the same ear. The fact is an interesting one, though it has not elicited much attention. Dr. Jackson, however, in his New-Hampshire Report, notices it, and states what is undoubtedly true, that each kernel of the varieties retains its own powers of appropriating its untriment. Now we may carry back our inquiries a step farther: Whence is derived this power—this special power? This question, it is true, must be answered by reference to its own individual and original condition; but the source of this power is undoubtedly due to the male blossom, or its effective part, the pollen grain. A pollen grain of the Tuscarora variety, falling upon the silk, is transmitted to its insertion, and enters the embryo cell. It is in fact a living cell detached from the Tuscarora kind: it has its beginning within the individual from which it is detached, and implants itself into the nidus formed for it in the Sweet, or some other variety of its own specific kind. Although grown upon an ear of the Sweet corn, it produces, when planted, the Tuscarora variety. This may not be entirely true to the letter, as the grain may have been mixed the first year; but ultimately the

Tuscarora becomes a pure kind, and free from intermixture. The result is much like grafting or budding. The stock may influence the character of the fruit, but still it retains very much of its own individuality.

The question, however, which I wished to determine definitely in bringing up this subject, is, whether these kinds, growing thus together upon one cob or in one habitation, are really as different in the internal constitution as they are in their external appearance. Jackson maintains they are, and with good reason; but his opinion was not founded upon a real analysis. I have therefore attempted to settle the question in this way. I accordingly selected an ear bearing the Tuscarora, a large 8-rowed yellow, and the Sweet corn varieties, which are at the extremes in appearance, here meet upon one ground, and hence were subjects suitable for my inquiry. Another variety, however, seemed to have originated in the relations in which these stood to each other in the field — a variety between the Tuscarora and the 8-rowed yellow.

1. Composition of the Sweet corn, as determined by a proximate organic analysis.

Starch	-	-	-	-	-	-	-	-	11.60
Gluten	-	-	-	-	-	-	-	-	$4 \cdot 62$
Oil	-	-	-		-	-	-	-	3.60
Sugar	-	-	-	-	-	-	-	-	6.62
Albume	n		-	-	•	-	-	-	$14 \cdot 30$
Casein	•	-	-	-	-	-	-	-	5.84
Dextrine	e -	-	-	-	-	-	-	-	24.82
\mathbf{Fibre}	-	-	-	-	-	-	-	-	11.24
Extract	-	-	-	-	-	-	-	-	8.00
Water	-	-		-	-	-	-	-	10.32
									$100 \cdot 96$

2. Analysis of the Tuscarora corn.

Starch	-	-	-	-	-	-	-	-	48.90
Gluten	-	-	-		-	-	-)		
\mathbf{Oil}	-	-	-	-	-	-	- }	undet	termined.
Albumer	n	-	-	-	-	-	- ´	-	8.72
Casein	-	-	-	-	-	-	-		$2 \cdot 32$
Dextrin	e	-	-	-	-	-	-	-	2.00
\mathbf{Fibre}	-	•	-	-	-	-	-		14.00
Sugar a	nd e	extract	-	-	-	-	-	-	$10 \cdot 00$
Water	-	-	-	-	-	-	-	-	13.68
			•						99.62

3. Analysis of the Yellow corn, grown upon the same ear as the two preceding.

Starch	-	-	-		-	-	-	-	50.83
Gluten	-	-	-	-	-	-	-	-	2.58
Oil	-	-	-	-	-	-	-	-	2.18
Albume	n	-	-	-	-	-	-	-	1.00
Casein	-	-	-	-	-	-	-	-	$3 \cdot 42$
Dextrine	Э	-	-	-	-	-	-	-	3.12
Sugar a	nd	extract	-	-	-	-	-	-	$9 \cdot 13$
Fibre	-	-	-	_	-	-	-	-	$14 \cdot 00$
Water	-	-	-	-	-	-	-	undete	ermined.
									$86 \cdot 26$

4. Analysis of a beautiful variety of maize, which, if it has no name, I would call Lady-finger corn. The ear is quite small, slender, delicate and white, but grows upon a remarkably long stalk at least seven feet high. It bears, at three or four joints, an ear, and sometimes two. It ranks among the small varieties, the Pop corn, and this is considered by far the best. It contains

Starch	-	-	-	-	-	-	_	-	46.90
Gluten	and s	sugar	-	-	-	-	-	-	$9 \cdot 24$
Oil	-	-	-	-	-	-	-	-	$6 \cdot 96$
Albume	n	-	-	-	-	-	-	-	$5 \cdot 02$
Casein	-	-	-	-	-	-	-	-	2.50
Dextrine	е	-	-	-	-	-	-	-	$2 \cdot 25$
\mathbf{F} ibre	-	-	-	-	-	-	-	-	8.50
Extract	-	-	-	-	-	-	-	-	$7 \cdot 02$
\mathbf{W} ater	-	-	-	-	-	-	-	-	12 · 12
									$\frac{-}{99.51}$

The Lady-finger corn is one of the finest varieties of pop corn, which has yet been cultivated for that special use.

5. Analysis of the Calico corn.

Starch	-	-	-	-	-	-	-	-	5 3·40
Gluten	-	-	-	-	-	-	-	-	$3 \cdot 22$
Oil -	-	-	-	-	-	-	-	-	2.80
Sugar	-		-	-	-	-	-	-	2.80
Albume	n	-	-	-	-	-	-	-	8.96
Casein	-	-	-	-	-	-	-	-	1.00
Dextrin	e	-	-	-	-	-	-	-	2.41
Extract	-	-	-	-	-	-	-	-	9.60
\mathbf{Fibre}	-	-	-	-	-	-	-	-	3.20
Water	-	-	-	-	-	-	-	-	12.55
									99.88

This may be regarded as one of the most interesting among the varieties. Its kernel seems to be made up internally of meal or starch, and protected only by a thin cuticle. The chit or embryo is oily, and apparently the oil is confined to this part. It is the extreme variety on the side of starch, while the Sweet corn is the extreme on the side of dextrine and sugar.

6. Analysis of the ash of the corn and cob of the large Ohio Dent corn.

o. <i>In</i>	aiysis oj	ine as	sn oj	ine		ana ortion	-	ine	targe	e Onio De	ni corn.
	Weight	of the	ear		,	-	•	• -		5993.00	grs.
	Weight					•		•		2652.36	8
	Weight					-	-	-	-	1114.40	
	8										
2000 •	00 grs. of	kerne	els ga	ave a	ash	-	•	-	17	7·12	er centum. 0 · 855
	40 grs. of		_			-	-	-		9.25	0.911
	•										
				1	Ash o	f the	corn.				
	Silica	-	-	-	-	-	-	-	-	1.80	
	Earthy p	hosph	ates	-	-	•	-	-	-	$60 \cdot 49$	
	Alkaline	phosp	hate	s	-	-	•	•	-	$13 \cdot 12$	
	$_{ m Lime}$	-	•	-	-	-	-	•	-	0.10	
	Magnesi	a	•	-	*	-	-	-	-	0.03	
	Potash	•	•	-		•	-	-	-	20.73	
	Soda	•	-	-	-	-	-	-	-	5.31	
	Chlorine		-	-	-	•	-	-	•	0.10	
	Sulphur	ic acid	l	-	-	-	-	-	-	0.71	
	Carbonio	acid		-	-	-	-	-	-	none.	
										${102 \cdot 39}$	
										10.0	
			2	2. A	sh of	the c	cob.				
	Silica	-	-	-	-	-	-	-	-	$25 \cdot 62$	
	Earthy p	hosph	ates	-	-	-	-	-	•	8.79	
	Carbona					-	-	•	-	0.40	
	Magnesi	a	-	-	-	•	-	•	-	0.20	
		-	-	-	•	•	-	-	-	$29 \cdot 02$	
	Soda	-	-	-	-		-	-		$9 \cdot 40$	
	Chlorine	•	_	-	-		-		-	0.41	
	Sulphuri			-	-	-	-	-	-	12.00	
	Carbonio			orga	nic m	atter	· -	-	•	$13 \cdot 40$	
										99·15	

It appears from the foregoing analyses, that we have the extreme limits in the amount of starch and dextrine in the Sweet, Tuscarora, and Calico varieties of maize: in the former, it is reduced to the lowest limit; in the Calico, it attains its maximum of starch. In the Sweet corn we find the maximum of dextrine, and in the other two only an ordinary amount, probably not an extreme.

In the presence of this large quantity of dextrine in Sweet corn we find the true reason of its shrinking so much, or of its assuming so much the appearance of a green or unripe grain. Dextrine, which, though it resembles very closely a gum, still differs essentially from true gum, of which Gum arabic is a type; but it swells when moistened, and shrinks again like gum when dried. A familiar example of shrinking and expanding is found in the gum which exudes from cherry or peach trees: in a wet day, it is four or five times as bulky as in a dry. So Sweet corn, when recently gathered, is full, plump, and with its skin distended: when it is dried, it shrinks, from the contraction of its gummy matter, the dextrine within the kernel. Probably too all the dent grains, or varieties of maize which exhibit a dent or depression upon the exterior, contain more dextrine than those which remain full and distended. The Ohio Dent, it will be observed, contains 5.40 per centum of dextrine, an amount exceeding that of the common yellow varieties of maize. It adds to the interest of the subject, to find the extremes of starch and dextrine growing or borne upon the same ear; thus proving the correctness of the opinion of Dr. Jackson, already referred to, with respect to the appropriation of nutriment by each individual kernel; for we find, upon analysis, that the difference of composition is as great as that of any two varieties cultivated at a distance and upon different soils. Indeed the facts are more striking in the case of these varieties, inasmuch as they obtain their nutriment from the same identical storehouse, or the same system of vessels. It all goes to prove, too, that each individual kernel modifies its food after its own manner; and it is this power which gives it its individuality.

MIDDLE-SIZED EIGHT-ROWED YELLOW CORN.

The following analyses exhibit the composition of the ash of the kernels and cobs of three specimens, grown on different soils, in Lewis county, in 1847. They were furnished by Mr. Beech.

1. Specimen grown on a clay loam.

Cobs of medium firmness, and well covered with plump kernels.

	Cops of median in	1111622	, and	wen	COVETE	u wi	ա թապ	p Kerners.	
1.	Ash of kernels.								
	Carbonic acid	-	-	-	-	-	-	trace.	
	Silicic acid -	-	-	-	-	-	-	1.450	
	Coal	-	-	-	-	_	-	1.650	
	Phosphate of iron		-	-	-	-	-	4.350	
	Phosphoric acid		-	-	-	-	-	57:365	
	Lime	-	-	-	-	-	•	0.150	
	Magnesia -	-	-	-	-	-	-	10.110	
	Potash	-	-	-	-	-	-	8.286	
	Soda	-	-	-	-	-	-	10.908	
	Chloride of sodium	n	-	-	-	-	-	0.249	
	Sulphuric acid	-	-	-	-	-	-	0.206	
	Organic acids	-	-	-	-	-	-	$3 \cdot 100$	
	· ·								~
								97.824	S.
	Water		PROP	ORTION	ıs.			7.750	
	******	-	•	-	•	-	-		
	Dry matter -	•	-	-	-	-	-	$92 \cdot 250$ $1 \cdot 340$	
	Ash	• 1	1 ()	•	3	-	-		C
	Percentage of ash	caic	uiated	on	ary m	atter	-	1.452	٠٠.
2.	Ash of cob.								
	Carbonic acid	-	-	-	-	-	-	11.500	
	Silicic acid -	-	-	-	-	-	-	7.562	
	Sulphuric acid	-	-	-	•	-	not det	ermined.	
	PHOSPHATES (24·1	25):							
	Phosphate of	-		-		-	-	0.313	
	Lime -	-	•	-	~	-	-	1.797	
	Magnesia	-	-	-	-	-	-	$9 \cdot 435$	
	Silicic acid	-	-	-	-	-	-	0.313	
	Phosphoric ac	eid	-	-	-	_		12.267	
	Lime	-	_	-	-		-	0.740	
	Magnesia -	-	_	-	-	-	•	0.050	
	Potash	-	-	-	-	-	•	$33 \cdot 605$	
	Soda	-	-	-	-	-	NI.	9.250	
	Chloride of sodiur	n		-			•	$5 \cdot 512$	
	Organic acids	-	-	-	-	-	**	$5 \cdot 450$	
	5							-	
								$97 \cdot 794$	S.

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PROPORTIONS.

Water -	29		-	-	-	-		6.890	
Dry matter	-		-	-	-	-	•	$93 \cdot 110$	
Ash -	-		-	•		-		1.560	
Ash calcula	ated o	n the	dry n	natter		-		1.675	3.

2. Specimen grown on a sandy plain, which had a top dressing of manure.

Cobs light and spongy, but well covered with kernels.

RELATION OF KERNELS AND COB.

							Per centum
Kernels		•	•	-	•	1790 ·	6 83.935
Cob	•	-	-	•	-	$329 \cdot$	5 16.065
Weight of one ear	-	-		-	-	2120	1
1. Ash of kernels.							
Carbonic acid	ės.					-	trace.
Silicic acid -		-				-	2.500
Sulphuric acid		236	_		_	-	0.055
Phosphoric acid		_	_			-	$54 \cdot 160$
Phosphate of per		e of i	ron				1.550
Lime	-	• 01 1				_	0.191
Magnesia -				_			11.040
Potash		_	_				13.031
Soda		_			_		16.885
Chloride of sodi	ım	_	_		_		0.215
Organic acids			_		_	_	trace.
Organic acias							
							99·627 S.
		PRO	PORTIC	NS.			
Water	-	-	-	-	•	•	7.69
Dry matter -	-	-	-	-	-	-	$92 \cdot 31$
Ash	-	-	-	-	-	-	1.58
Ash calculated or	the	dry 1	natte	r -	-	-	1·71 S.
2. Ash of cob.							
Carbonic acid	-	-	-	-	-		13.000
Silicic acid -	-	-	-	-	-	-	13.300
Sulphuric acid	-	•	-		•	-	0.413
Phosphates (15:	55):						
Phosphate o		n -	-	-	_	-	trace.
Lime -	-	-	-	-	-	•	1.720
Magnesia	-	-	-	-	-	-	2.584
Phosphoric a	acid	-	-	-	-	-	11.246
Carried forwar	d	-	_			-	${42 \cdot 263}$

Brought f	orwa	rd		-			•	$42 \cdot 263$
Lime -	-	-		-	-	•		0.497
Magnesia	-	•	-	•	-			0.368
Potash -	-	-	-	-	-	-	-	$35 \cdot 993$
Soda -	-	-	69	-	-	40	-	11.894
Chloride of	sodi	um	•		•			$1 \cdot 325$
Organic acid	ls	-	-	-		-		5.850
								9S·190 S.
			PRO	PORTION	vs.			
Water -	•		-	-	-	-	-	$7 \cdot 140$
Dry matter	•	-		-		-	-	92.860
Ash -	-	-	-	-	-	-	•	1.360
Ash calculat	ted o	n drv	matt	er			-	1·464 S.

3. Specimen grown on an intervale without manure. Cobs very firm, and well covered with plump kernels.

RELATION OF KERNELS TO COB.

				141.		1021	,, 1	23421		10 00			er centum.
Kerne	els	-		-		-	-	-		-	1727		80.511
Cob	-	-	-	-		•	-	-		•	418.	08	$19 \cdot 489$
Weig	ht of	one	ear			-	-	-			2145	23 10	00.000
1.	Ash o	f ker	nels.										
		bonic				-			-	-	-	trace.	
		cic ac			9			9		-	6	2.650	
		huri									_	0.132	
		spho				_		-			-	56.050	
		spha				a			-	_		0.750	
	Lim	-				-			_	-		0.450	
		nesi	3.	_	-	_						8.750	
	-	ash •		-	-	4				_		5.191	
	Sod		,	_	_				_			19.184	
		rine				_						0.098	
		anic	acida	3						_	-	3.450	
	Coa		uora		-			_		-	9	1.750	
												${98.455}$	S.
						PI	ROPOL	TION	8.				
	Wa	ter -		-					-		-	7.630	
		mat	ter	-		-		-	-	•	-	$92 \cdot 370$	
	Ash			-		-		-	-	-	-	1.615	
		calc	ulate	d on	the	dry	ma	tter	-	-		1.748	S.

2.	Ash of cob.								
	Carbonic acid	-	-		•	-	-	$9 \cdot 405$	
	Silicic acid -	•	-	-	-	-	-	$9 \cdot 985$	
	Sulphuric acid	-		-	-	-	_	1.306	
	PHOSPHATES (21:	163):							
	Phosphoric a	cid	•	-	-	-	-	$13 \cdot 135$	
	Phosphate of	iron	-	-	-	-	-	0.415	
	Lime -	-	-	-	-	-	-	1.643	
	Magnesia	-	-	-	-	-	-	5.555	
	Silicic acid	-	-	-	-	-	-	0.415	
	Lime	-	-	-	-	-	-	2.190	
	Magnesia -	-	-	-		-	-	$1 \cdot 190$	
	Potash	-	-	-	-	-	-	$34 \cdot 470$	
	Soda	-	-	-	-	-	-	$11 \cdot 425$	
	Chloride of sodiu	m.	-	-	-	-	-	1.960	
	Organic acids -	-	-	-	-	-	-	$6 \cdot 450$	
	S								
								$99 \cdot 544$	S.
			PROP	ORTIONS					
	Water	-	-	-	-	-	-	6.880	
	Dry matter -	-	-	-	-	-	-	$93 \cdot 120$	
	Ash	-	-	-	-	-	-	1.325	
	Ash calculated on	the d	lry m	atter	-	-	-	$1 \cdot 423$	S.

PROXIMATE ORGANIC ANALYSES OF THREE VARIETIES.

1. White Kentucky dent corn.

									Calculated without the water.	
Starch -	-	-	-	-	-	•	-	$50 \cdot 92$	58.710	
Oil -	-	-	-	-	-	-	-	0.64	$0 \cdot 737$	
Zein or glu	en	-	-	-	-	-	-	0.72	0.830	
Dextrine or	gum	-	-	-	-	-	-	$3 \cdot 08$	$3 \! \cdot \! 551$	
Fibre after	boilir	ng se	veral	ly in	alcoh	ol an	d a			
weak sol	ution (of ca	ustic	potas	h -	-	-	9.70	11.184	
Matter disso	olved	out o	f fibre	by a	weal	solu	tion			
of causti	c pota	sh	-	-	-	-	-	$2 \cdot 62$	$3 \cdot 020$	
Albumen		-	-	-	-	-	-	$4 \cdot 44$	5 · 119	
Casein -	-	-	-	-	-	-	-	0.80	$0 \cdot 922$	
Sugar and	extrac	tive 1	matte	r -	_	-	-	13.80	15.911	
Water -	-	-	-	-	-	-	-	$12 \cdot 22$		
								98·94 S	. 99•984	

In the above analysis, the oil and zein were obtained only from the fibre.

2. Small blue corn, used for parching.

After drying at 180°, it gave

Starch -	- ·	-	-	•	-	-	-	42.56
Oil -	-	-	-	-	-	-	-	5.30
Sugar and ex	tracti	ive m	atter	-	-	-	-	$15 \cdot 32$
Dextrine or g	um	-	-	•	-	-	-	7.52
Fibre after d	igesti	ing se	veral	ly in	alcoh	ol and	a	
weak solut	ion of	caus	stic po	tash	-	-	-	$8 \cdot 56$
Matter taken	up fr	om fil	bre by	a we	ak so	lution	of	
caustic pot	ash	-	•	-	•	•	-	$8 \cdot 54$
Albumen	-	-	-	•	-	-	-	5.00
Casein -	-	-	-	-	-	•	~	$2 \cdot 04$
Zein or glute	n	•	•	-	•	•	-	4.78
•								00.04.0
								99·64 S.

3. Early 8-rowed White-flint, raised by Mr. Bement of Albany.

									Calculated without the water.
Starch	-	-	-	•	-	-	-	$57 \cdot 47$	$67 \cdot 573$
Oil -	•	-	-	-	-	-	•	2.55	$2 \cdot 909$
Fibre	-	-	•	•	-	-	-	$6 \cdot 07$	$6 \cdot 925$
Dextrine	or gu	ım	-	-	•	•	-	4.01	$4 \cdot 575$
Sugar ar	-		e ma	tter	-	-	-	13.21	15.072
Albumer		-	-	-	-	-	-	2.27	2.590
Casein	-	-	-	-	-	-	-	0.39	$0 \cdot 444$
Zein or g	luten	-	-	-	-		-	1.67	1.905
Water	_	_	-	-	-	-	-	11.46	
									
								99·10 S.	$99 \cdot 793$

On the changes which have taken place in the relative proportions of the inorganic matter at different periods of growth. That certain changes take place in the relative amounts of the inorganic matter, appears highly probable from the phenomena of vegetation in general. If, however, we collect the facts as established by the foregoing proportions (see pp. 160 – 192 inclusive), it seems established that certain elements vary in their respective amounts, when the young is compared with the old or mature plant. As silica, or silicic acid, is one of the elements essential to the composition of the plant, it will be found to have changed its relative proportions. In the leaves, for example, it increases very regularly from the earliest stage till its maturity is acquired. In the sheaths, the same fact is established: thus, in July, the percentage of silica was 11.60; in October, 51.25. In the stalks, it was 2.50 in August; in October, 12.85. The phosphates also change places; but instead of there being an increase in the leaves as they increase in age, it diminishes. Thus the phosphates of the sheaths, in July, amount to 10.40 per centum; they decrease till about the 2d of August, when they amount to 8.9; they increase again to 12.70 per

centum, when they decrease to 9.75 per centum. In the stalks, however, analogous changes occur, an increase followed by a decrease. The alkalies of the sheaths, especially soda, diminish from 34.48 to 22.44 per centum. The alkalies of the stalks diminish from 35 in the case of soda, to 17.79; and in potash, from 13.35 to 11.11.

As the most important elements of the fruit or grain are alkalies and phosphates, they seem to be first elaborated in the leafy parts, or perhaps in connexion with the organic matter, the fluids, vitalized and fitted to fulfil the most important offices in the most important parts. A direct transference from the soil to the seed would seem to put them in this important part unprepared, and hence they pass into the foliage, and there undergo those changes which especially fit them for the office of continuing the species. Hence, too, in the ripening of seed, the source of nutriment is the foliage, and hence too the most important matters may appear to have diminished in amount. The silica, however, being required in seed for the purpose of protecting it, is rarely if ever in such abundance as to diminish the amount of it in leaves and stalks. Hence the proportion is relatively greater in the ripe than in the unripe stem; and so too as silica is the element of support and strength, it is the more necessary to the foliage when exposed to winds and rains until it attains its full size and growth: it is required here; but in the fruit, silica would be a damage, except as a thin covering to protect the soft and delicate embryo.

The foregoing facts are a clue to the true and proper time for cutting corn as a fodder; inasmuch as it is evident it ought to be done before the phosphates and alkalies pass to the seed and fruit, and before the silica predominates so far as to reach its maximum quantity. Just before and at the period of blossoming, analysis shows that they should be cut. Then too the stalks and leaves are less woody: there is less fibre and more nutritious matter in the same weight. The leaf particularly becomes woody and siliceous. The stalk, when of moderate size, is more valuable than the leaf.

ON THE COMPARATIVE VALUE OF CORN FOLIAGE FOR FODDER.

Having stated at considerable length the proportions of the parts composing the maize plant, and also the composition of its organic and inorganic matters, I propose now to consider some of those properties which indicate its relative value as a cereal, and its importance as one of the nutriments designed for man and animals.

On this last point, I may remark here that the importance of this grain is indicated in the great extent of territory over which it is susceptible of cultivation. To the north, it is not forced from the soil by cold until we reach a latitude of near 50 degrees. So its cultivation is feasible as far south of the equator; and under the equator, it is grown at an elevation of 7500 feet. In the latitude of Albany, it nearly reaches its vertical limit of cultivation at the height of 12 or 1500 feet. The extent of its cultivation is of course

greatly increased in consequence of its annual growth, and its requiring only a season to perfect its grain.

Another fact equally strong in itself, is its disposition to form varieties, which adapt themselves to the climate or the length of the season. In this constitutional power, it exceeds I believe all the other cereals. It has been observed that from 75 to 100 varieties are known. Some of these varieties attain maturity with great rapidity; others more slowly, in which case they usually attain a great height, sometimes that of small trees. The growth of the latter is as rapid as the former, but their size is not completed till after the lapse of ninety days.

Of the comparative value of maize. In estimating the value of an article of food, it is essential that we take into account the amount produced on a given surface, and the extent to which it exhausts the soil. I will attempt to ascertain these amounts in this place, inasmuch as they are both important and valuable elements with which the farmer should become familiar.

Amount obtained in an ordinary crop of maize. As all parts of the plant are more or less valuable, and inasmuch too as they take from the soil valuable elements, it will be instructive to state how much is thus removed in the crop, in the leaves, stalks, silks, tassels, husks, cobs and grains. An acre of land may be estimated as producing 4900 hills, with three perfect plants, or stalks as they are usually called. They may be supposed to stand in rows three feet apart, and to range in true lines in two directions. Taking the White-flint corn for our estimate, it will be found by reference to page 167, that a plant of this variety gives the following weights of its several parts:

						Green.	Dry.	Ash.
Tassels	-	-	-	-	-	130 · 7 grs.	57 grs.	4·37 grs.
Stalks	-	-	-	-	-	5766.9	880.1	$35 \cdot 75$
Leaves	-	-	-	-		$2797 \cdot$	763.8	81.35
Sheaths	-	-	-	-	-	2536 •	$494 \cdot$	$40 \cdot 46$
Husks	-	-	-	-	-	$3534 \cdot$	802	$29 \cdot 53$
Silks	-	-	-	-	-	57·	36 •	1.72
Cobs	-	-	-	-	•	$2935 \cdot 5$	-864	18.
Kernels	-	-	-	-	-	4170.5	$1633 \cdot$	32.7
						${22007 \cdot 6}$	${4022 \cdot 9}$	243.88

In a single plant there are 4023 grains of dry matter, in which there is dissolved 243.88 grains of inorganic matter, consisting, as will be observed in the analyses, of phosphates, soluble silica, potash, soda, etc. A field of an acre contains, as has been indicated, 14700 of such plants, weighing probably some more and some less, but whose average amounts will not differ materially from the weight of the plants already determined and referred to. So it is probable that more than three plants will be found in a hill; but assuming this as the number, it will be easy for any one to compute the variations which may arise from a greater or less number of plants in the field.

The number of plants in an acre amounts to 14700; and they will remove from the soil 3585036 grains, or 7468.82 oz. = 466.80 lbs. of inorganic matter. Of the 3585036 grains, the parts of the plant will contain, in the

Tassels	-	-	-	-	-	•/	-	64239 grs.
Stalks	-	-	-	-	-	-	-	525525
Sheaths	-	•	-	-	-	-	-	594962
Leaves	•	•	-	-	-	-	-	1195845
Silks -	-	-			-	-	-	25284
Husks	ca.	-	-	-	-	-	-	434091
\mathbf{Cobs}	•	-	-	-	-	-	-	264600
Kernels	-	-	-	-	-	-	-	480690
								3583236

From the foregoing results, we find that the amount of elements removed in the entire crop of one acre will stand as follows:

The tassels and stalks will contain

							Tassels.	Starks.
Silica	•	-	•	-	-	-	5·105 lbs.	8.789 lbs.
Earthy phosphates	; -	-	-	-	-	-	0.822	10.362
Lime	-	•	-	-	-	-	0.196	1.928
Magnesia -	-	-	-	-	_		0.053	0.640
Potash	•	_		•	-	-	0.573	11.087
Soda	-	•	-	_	-	-	0.744	$17 \cdot 094$
Chlorine -	-	-	•	-	-	-	0.198	$7 \cdot 491$
Sulphuric acid	-	49	_	-	-	-	0.335	7.382
•								
							8.023	$64 \cdot 773$

The leaves and sheaths,

								Leaves.	Sheaths.
Silica -		-	-		-	-	-	S2.6S1 lbs.	39.667 lbs.
Earthy pho	sphat	es		•	•	-	-	$29 \cdot 272$	7.546
Lime -	•		-	-	-	-	-	$9 \cdot 406$	1:581
Magnesia	-	-	•	-	-	-	-	1.910	0.589
Potash -	-	_		•	-	-	-	19.704	$5 \cdot 571$
Soda -	_	-	-	-	-	-	-	$13 \cdot 142$	$9 \cdot 262$
Chlorine	***			-	-	-	-	$15 \cdot 072$	2.202
Sulphuric a	cid	•	-	•		-	-	$6 \cdot 461$	8.928
~									
								$177 \cdot 648$	75 •356

T.	he	husks

								Husks.
Silica -	-	-	-	-	-	-	-	26.922 lbs.
Earthy pho	sphat	es	-	-	-	-	-	14.831
Lime -		-	-	-	-	-	-	0.254
Magnesia	-	-	-	-	-	_	-	0.040
Potash -	-	-	-	-	-	-	-	1.984
Soda -	-	-	-	-	-	-	-	$5 \cdot 555$
Chlorine	-	-	-	-	-	-	-	$3 \cdot 141$
Sulphuric a	cid	-	-	•	-	-	-	3.770
								$56 \cdot 497$

The cobs and kernels,

			,							Cobs.	Kernels.	
í	Silica	-	-		-	-	-	-	-	4.678 lbs.	5.939 lbs.	
]	Earthy	pho	sphat	es	-	-	-	-	-	8.229	$22 \cdot 187$	
]	$_{ m Lime}$	-	-	-	-	-	-	-	-	0.103	0.100	
]	Magne	sia	-	-	-	-	-	-	-	0.309	1.506	
	Potash		-	-	-	-	-	-	-	$12 \cdot 315$	14.950	
j	Soda	-	-	-	-	-	-	-	-	$2 \cdot 034$	14.118	
(Chlorin	ne	-	-	-	-	-	-	-	0.045	0.309	
	Sulphu	iric a	acid	-	-	-	-	-	-	0.118	2.740	
										27.831	61.849	

The several amounts, according to the foregoing results, will stand as follows:

									lbs.	oz.
Silica	-	-	-	-	-	-	-	-	173	$12 \cdot 496$
Earthy phospha	tes	-	-	-	-	-	-	-	93	3.984
Lime	-	-	-	-	-	-	-	-	13	9.248
Magnesia -	-	-	-	-	-	-	-	-	5	0.752
Potash -	-	-	-	-	-	-	-	-	66	2.944
Soda	-	-	-	-	-	-	-	-	61	$15 \cdot 184$
Chlorine -	-	-	-	-	-	-	-	-	28	$7 \cdot 328$
Sulphuric acid	-	-	-	-	-	-	-	-	29	11.696
									471	15.632

Maize must be ranked then among the most exhausting crops. It is evident that poor soils will scarcely repay the farmer for its cultivation. It is evident, too, that unlike other cereals, there is little danger of using too much manure in its cultivation. It is not liable to run to foliage, and thereby fail to produce grain; neither will it lodge, or fall down by its own excessive disproportion of organic to its inorganic matter.

There is one more important view which may be taken of the analyses: they will dispel the notion so frequently inculcated by writers and lecturers, that bone earth is the main

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manure for a maize crop; or, rather, that it is enough to furnish a manure which consists of elements found in the grain or fruit. For the perfection of any crop, it is as necessary that the leaves, silks and tassels be furnished with their appropriate food, as the grain itself; for it can not be doubted but that the grain itself depends upon the full development of the parts which precede it. Supply them with matter suitable for their increase and perfection, and the grain is supplied also. It must not be forgotten that these parts very frequently contain elements which are not found, except in very small proportions, in the seed or grain; yet it is plain enough, that in some way or other, these elements are quite essential to their perfection.

The value of the maize crop does not depend solely upon its grain. The foliage, including the stalk, is freely eaten by cattle and horses; and, if properly dried, it is by no means an inferior kind of fodder. The amount of green foliage raised on an acre furnishes two or three tons of dry fodder. By reference to the analyses of the foliage, pp. 225-6-7, the leaves, stalks, sheaths, tassels, etc. contain albumen, casein, dextrine, oil, wax and sugar; substances necessary to the sustenance of the animal body, and which are required to supply the waste of the system. These are both nitrogenous and heat-producing bodies, or contribute to the growth of muscle, the generation of heat, and the formation of fat; and it will be seen that the amount of these elements is very considerable, from the great amount of foliage produced.

For a more satisfactory solution of the problem respecting the value of the maize plant as a nutriment, it may be compared with several species of grasses. Nutriment for domestic animals consists of bodies which have been described in the first part of this report; they are mainly albuminous. Casein is found in milk, and exists in most vegetables in a modified form. Starch, dextrine, resin, oil and wax exist in all vegetables which are eaten by cattle. The leaves and leafy parts furnish a mere trace of starch. Upon their surface there exists a green coloring matter called *chlorophyl*, which is insoluble in water, and which is always associated with wax. The wax protects the surface from the action of rain and dew; and when the plant is cut for food, it supplies in the system the place of oil, and aids in maintaining the temperature of the body. The chlorophyl and wax cover the surfaces of fruits, and serve the same purposes as in the leaf, and impart that glossy character to the outside which is always admired in red and high colored fruits. Wax and chlorophyl, as may be observed in the foregoing analyses, have always been obtained from the stalks and foliage of corn (see pp. 224 and onward).

The analyses of the young maize plant, very dry, cut July 5, gave the following results:

Fibre or insoluble matter	-	-		-	$59 \cdot 00$
Sediment from the solution in	wate	er	-	-	3.72
Albumen and casein -	-	-	•		7.96
Dextrine, sugar and extract	-	-	-	-	$26 \cdot 40$
Water			•		3.67
					$100 \cdot 25$

The wax was not obtained: it is a product, however, which varies less than most others.

The same kind of corn, cut July 16,	gave
-------------------------------------	------

Fibre or	in	solub	le ma	tter	-	-		-	$49 \cdot 44$
Sedime	nt,	most	ly fib	re -	-	-	-	-	8.82
Extract					-	•	-	-	18.00
Albume	n, י	with a	dher	ing chl	orophy	yl and	some	fibre,	$4 \cdot 00$
Resin	-	-	-	-	-	-	-	-	1.30
$\mathbf{Chlorop}$	hyl	and	wax,	mostly	wax	-	•	-	2.82
Water	-	-	-	-	-	-	-	-	8.76
Casein	-	-	-	-	-	-	-	-	1.70
									94.86

The foliage of the ripe plant, cut September first, gave

\mathbf{Fibre}	-	•	-	-	•	-	-	-	$61 \cdot 24$
Fibre in	the	form of	of se	dimer	it -	-	-	•	5.76
Extract	and	sugar,	mos	stly e	xtract	-	-	-	$19 \cdot 00$
Albume		-		-	•	-	-	-	0.23
Casein	-	-	-	-	-	-	-	-	1.60
Water	-	-		-	-	-	-	-	10.17
									$97 \cdot 90$

The watery extract, when dry, has the flavor and taste of dry bread or rusk. This is the fact with all the sediments obtained by water from hay or the dried grasses. This flavor, with the slightly bitter principle of the extract, renders hay an agreeable and pleasant food to the ruminants. The amount of matter dissolved in cold water aided by macerating several days, and after pressing the insoluble fibre in a linen bag, furnishes from the maize foliage 40 to 60 per centum. All the soluble parts are taken up; and the fibre is left, after being dried, in the state in which it passes from the intestinal canal. This matter, which has been removed by water, is not entirely soluble: it appears, after standing, that the chlorophyl and wax are suspended in the menstruum, and by rest are slowly deposited. So some of the resinous matter is obtained in the watery solution. The solution, when its temperature is raised to $140-180^{\circ}$, gives invariably flakes of a coagulum which is undoubtedly albumen. So a precipitate is given afterwards by acetic acid.

These several products are more or less colored by, and still furnish the characteristics of, the elements to which they are referred. The chlorophyl does not appear to be assimilated, inasmuch as it appears in the excrements, and gives them in ruminants their green color. The wax, however, with which it is associated, disappears in part at least, and is undoubtedly, along with the resinous matters, comprised among the calorifient bodies.

The albuminous matters (including casein) which are destined to supply the waste in the animal system, in the maize plant cut July 5, amount to 7.96; in that cut in July 6, to 3.80; and in the leaves of the ripe maize cut September 1, to 1.83 per centum. In

July, therefore, there is more nutritive matter than in the earlier or later stage of the plant. This was to be expected. In the ripe plant, the leaf as well as the stalk becomes more insoluble.

If the nutriment of the foliage of maize is compared with that of the common grasses, it will be found that the latter are generally more nutritive than the former: thus, timothy hay furnished, in the dried state,

Fibre	-	-	-	-	-	-	-	-	68.14
Wax ar	nd chi	loroph	ıyl	-	-	-	-	-	2.80
Albume	en	•	-	•	-	-	-	-	1.89
Casein	-	-	-	-	-	-	-	-	$2 \cdot 34$
Sugar,	extra	ict an	d dex	trine	-	-	-	-	8.20
Water	-	•	-	•	-	-	-	-	$12 \cdot 30$
									95.67

The albuminous matters amount to 4.23, and the calorifient to 11.00. I suppose that the extractive matter belongs to the calorifient bodies. Dextrine seems to replace starch in the grasses. When there is a pith, as in the maize stalk, we find a trace of starch.

Red-top, another common and favorite hay, gave the following results:

\mathbf{Fibre}	-	-	-	-	-	-	-	-	65.00
Albume	n	-	-	-	-	-	-	-	1.49
Casein	-	-	-	-	-	-	-	-	1.80
Chlorop	hyl	and wa	X	-	-	-	-	-	11.62
Resin	•	-		-	-	-	-	-	3.08
Extract	and	sugar	-	-	-	-	-	-	9.00
Water	-	-	-	-	-	-	-	-	10.00
									${101 \cdot 99}$

The sediment, which is allowed to subside for twelve hours, is filtered from the solution before it is treated for albumen; and it is believed that considerable albumen goes down with this sediment, inasmuch as it is adhesive, and bears the characters of an albuminous body, although it is mixed very largely with fibre, chlorophyl and wax; but it is all reckoned with the fibre, and the albumen is only obtained by heat, and the casein by acetic acid. In the red-top, 3·29 is set down as albuminous; but 11·62 per centum of wax is rather a remarkable result: the wax forms the greatest part of this amount. The calorifient products are larger than in timothy, amounting to 23·70. It is probable the albuminous matters are stated too low, and that the deficiency arises from the subsidence of from one to two per centum with the sediment referred to above. If not, they exist in about the same proportion as in the corn of the middle of July.

The red clover contains less albuminous matter than the grasses, as will be seen from the following analysis:

Fibre -		•	•	-	-	-	-	-	$64 \cdot 19$
Albumen			-	-	-	-	-		1.57
Casein -			-	-	-	-	-	-	trace.
Wax and	chlor	ophy.	l	-	-	-	-	-	4.20
Resin -		•	-	-	-	-	-	-	2.60
Extract ar	d sug	gar	-	-	-	-	-	-	18.80
Water .			•	-	-	•	-	-	9.60
									99.96

In the clover, the calorifient and fatty matters are in greater proportions than in the grasses or maize, while the matters required by muscle and sinew exist only in a small amount.

The white daisy (Chrysanthemum leucanthemum), so common in many meadows, and often thoroughly hated by the farmer, gave, in a partial analysis, the following results:

Chlorophyl and was	x -	-	-	-	-	-	3.60
Resin	-	-	-	-	-	-	$1 \cdot 44$
Extract and sugar	-	-	-	-	-	-	8.96

The albuminous matters were not obtained. The calorifient properties equal the grasses; and I believe cows, when they become fond of it, give a rich milk. One hundred grains gave dry matter 22:09, ash 1:65.

A carex (one of the swamp grasses), but which was collected before it blossomed, gave results as follows:

Fibre	-	-	-	-	-	$86 \cdot 20$
Chlorophyl and wax	-	-	-	-	-	2.00
Albumen	-	-	-	-	-	2.84
Casein	-	-	-	-	-	trace.
Resin	-	-	-	-	-	0.47
Extract and sugar -	-	-	-	-	-	6.60
						$98 \cdot 11$

The carex is therefore not deficient in albuminous matter. The waxy matter and sugar exist in a small amount only, while by far the greater part of the grass is fibre.

The Glyceria fluitans, which is an aquatic grass, the seeds of which are sweet and nourishing, is not much more valuable in its dry state than the carex. It consists of

Fibre	-	-	-	-	-	$79 \cdot 96$
Albumen	-	-	-	-	-	$2 \cdot 40$
Casein	-	-	-	-	-	trace.
Extract and sugar -	-	-	-	-	-	7.00
Water	-	-	-	-	-	10.00
Chlorophyl and wax	-	-	-	-		1.20
						100+56

This grass is freely eaten by cattle, and it is sweet when green. Its watery extract, however, is not more sensibly sweet than that of the red-top.

The foregoing analyses of the grasses were made for the purpose of comparing their proximate composition with that of the maize plant. There is quite a striking similarity between them in the amount of wax and chlorophyl, while they differ in the quantity of woody fibre. This increases, however, with the age of the plant; but the samples employed were not too far advanced for the purposes in view.

The amount of water in the thoroughly sun-dried grasses is usually about 10 or 12 per centum, which seems to be about the quantity required constitutionally. The solutions were all obtained by cold water: the chlorophyl and wax were obtained by boiling in alcohol, the product of which was redissolved by ether, to separate the wax and chlorophyl. The remainder was steeped in water, which dissolved out the sugar and some extract, leaving a small quantity of resin, soluble only in alcohol.

The nutrient powers of the grain or kernel of maize varies considerably with the variety, as may be seen in the foregoing analyses. Starch, as a general rule, is the element in the greatest abundance; but it has been shown, that in one variety it is small, amounting to 11.60 per centum only. In this variety, the albuminous matters, including casein, are increased to 20.14 per centum. The gluten, which is a nitrogenous body, also, when counted with the albumen, gives us the additional quantity of 4.62 per centum of matter fitted to supply the waste of the system; or, in other words, to create nerve and muscle. It is the most nutritive of the varieties of maize. If the dextrine is added to the starch, this variety cannot be regarded as deficient in calorifient matter, and hence it is fully equal in calorifient powers to beans. In another point of view, it must be regarded as superior to beans, viz. its fattening properties; inasmuch as it contains 5.11 per centum of oil and sugar. Other varieties contain more than 50 per centum of starch, and less albuminous matters than the sweet variety. A reference to the analysis will probably furnish the reader with all the information he may wish.

It is probable that modes of cultivation will materially affect the composition of this grain. A high cultivation will increase the relative amount of certain elements: in one kind of management, the albumen may be formed in greater quantity; in others, the starchy. This idea, however, requires proof by experiment; and however it may prove, it is evident that maize is very remarkable in its varied composition, as it exists in its numerous varieties.

To what extent the nutrient powers of meal and flour are changed by cooking, has never been actually determined: their solutions are more or less altered. Thus the gluten of flour, which is soluble in alcohol, ceases to be so in bread: starch is probably changed into dextrine. The following is an imperfect analysis of bread, but may be stated for some of the facts which it contains:

Soluble in alcohol	l -		-	-	-	-	7.40
Glutinous matter	-	-	-	-	-	-	5.00
Matter insoluble i	n cold	wate	er, and	l mix	ed wit	h a	
little starch -	-	-	-	-	-	-	21.32
Albumen -		-	-	-	-	-	1.00
Casein	-	-	-	-	-	-	1.28
Dextrine -	-	-	-	-	-	-	8.21
Sugar and extrac	t -	-	-	-	-	-	$13 \cdot 50$
Water	-		-	-	-	-	$40 \cdot 00$
							98.61

Fresh bread still contains 40 per centum of water.

DESCRIPTION OF SEVERAL VARIETIES OF MAIZE.

I have already alluded to the fact that maize, in the course of the period of its cultivation, has run into a great number of varieties. These are not only well characterized by their form and markings, but they seem to be as permanent as the species. The Sweet corn is a variety whose origin goes so far back that its history is lost. It was found in possession of the Indians of this country, and has not as yet materially changed in its features. The Tuscarora is also an old variety. The following descriptions of a few of the most common varieties cultivated in New-York, may be acceptable to those who have an interest in the production of kinds.

- 1. Rocky-mountain corn (Pl. 26, fig. 2 & 3). The principal feature of this kind is its tunic, or husk investing its kernel. This is long and pointed, and completely envelopes the kernel. From this interesting fact, this kind has been regarded by a few naturalists as a distinct species of maize. At first, the existence of such an envelope may appear to favor this view; but when it is known that all kernels of maize are originally furnished with a tunic in their young state, and also that this tunic disappears under cultivation, it seems to cast considerable doubt over the correctness of this opinion. In its present state, this is not a valuable variety, but is much like a wild plant that requires an exfoliation of its coarser coverings. This actually takes place, and the nutriment which formerly supported the husk is diverted to the grain. It improves, therefore, by cultivation, and probably may originate some valuable varieties. The size of the ear differs only a trifle from that of the Tuscarora, but it is loose and spongy. It is cultivated at present only as a curiosity.
- 2. Tuscarora corn (Pl. 26, fig. 1 & 6). Kernel large, white and flat, starchy; chit reddish, the corneous portion thin. Form trapezoidal, wide, with grooves on the upper side; the central one is a deep excavation extending from the chit to the outside or corneous part; lateral one narrow, and half the length of the central one. Inferior side convex.

The kernel is inserted upon a thick shortish reddish cob. Ear 8-rowed. It gives a white, nutritious flour. Specific gravity 1.110. Specimen furnished by Mr. N. Salisbury, Cortland county.

3. Calico corn. Color remarkably variegated with bright hues and intermixed with white. Cuticle thin, investing apparently a mass of starch, with a slightly reddish tinge. Ear eight-rowed. Cob chaffy. Specific gravity 1.066. Furnished by Mr. Jewett, of Addison county, Vermont.

This is beautiful on the ear: it is cultivated mostly as a curiosity, or for popping. The kernel contains a large amount of starch, and is deficient in gluten. It is the lightest of all the varieties.

4. White-flint corn (fig. 5). Color white. Corneous portion well developed, and investing the starchy portion. Ear long and slender, eight-rowed passing into twelve. Specific gravity of that raised last year, 1.266; of the specimen raised this year, 1.301. This is among the heaviest.

This variety (the small White-flint) is early, and hence must be regarded as one of the best kinds of maize. It is rich in oil, and gluten and starch, and hence for food it must rank very high; and as its meal is whiter than that of the yellow corn, it may be preferred for bread, especially when it is intended to mix it with wheat flour.

- 5. Middle-sized eight-rowed yellow corn (fig. 11). Yellow. Kernel large, oblong, flattish. Corneous portion well developed, investing a thick layer of starch. Groove single, wide. Ear cylindrical. Cob comparatively small, from nine to ten inches long; rows 8, and arranged at the base in twos. Specific gravity 1:316, and ranks with the 8-rowed white-flint in value, and seems to possess a higher specific gravity.
- 6. Golden Sioux corn (fig. 12.) Yellow. Ear large. Kernel thick and pointed; width subequal, sometimes higher than wide. Corneous portion well developed. Rows 12, and arranged in twos, or may be irregular; the number increasing to 14, and even 16 at the base, when they become crowded and irregular. Specific gravity, 1·299. The cob is large at the base, and hence is liable to mouldiness. It ripens late; and from these two defects, it is not so much esteemed as the smaller and earlier kinds. Its foliage is much greater, and the exhausting powers increased. The meal is as rich as that of either of the preceding kinds, provided it has ripened well. It is the Dutton corn of the late Judge Buel, as I am informed by Mr. Howard. It was introduced from the Indian country, first into the eastern part of Massachusetts, and afterwards into Vermont. One ear only is usually perfected on a single stalk.
- 7. Small eight-rowed yellow corn (fig. 15). Yellow, 8-rowed. It resembles the large 8-rowed, only it is smaller: it passes into that variety by cultivation, and can hardly be distinguished from it. That which was cut August 11, had a specific gravity of 1.241. Probably, when thoroughly dried, its weight is proportionally increased. It is a fine variety: being early, its foliage is smaller than that of the white-flint, and it frequently produces three ears upon a stalk.
 - 8. Twelve-rowed Canada corn. Yellow, and frequently brownish. Corneous portion

well developed. Kernels very hard, and set close upon the axis or cob; pointed, higher than wide. Rows irregular at their base. Corneous portion thick; groove single and wide. Specific gravity, 1.289. The cob of this variety is small, and the kind is early. It is esteemed by the Vermont farmers.

- 9. Canada pop-corn, Egyptian corn (fig. 4). Yellow. Color blue and dark purplish-red. Ear small, firm and hard. Cob small and reddish. There are several varieties of the yellow and the dark colored, all of them early and small, and used only for popping. Specific gravity of the dark-purplish Indian Canada pop-corn, 1.298.
- 10. Rice corn. White and yellow. Kernel pointed; mucronate, or the back terminates in a point where the silk was inserted. Rows 12-16, and irregular at base. Ear 4 to 6 inches long, and tapering. It is a beautiful variety, and probably one of the richest and sweetest. It bears from three to four ears on a stalk, but is rather unproductive. It is used principally for popping.
- 11. Illinois or Lady-finger corn. Pale yellow. Ear small, slender and tapering. Rows 12. Kernels small, pointed, rounded upon the back. It is an unproductive kind, bearing sometimes four ears upon a stalk, but the stalk is from 7 to 8 feet high. It is a late kind, remarkable for the great amount of stalk and leaves, or what might be termed offal.

The indented kinds are also quite numerous.

- 12. Large Ohio Dent corn (fig. 14). Ear large, cylindrical. Yellowish-white. Corneous portion placed upon the side; back indented. Starch extends to the investing membrane. Size remarkably large, and ears from 12 to 14 inches long. It has 868 kernels upon an ear, in 12 to 14 rows. Cob reddish and chaffy. One hundred and seventy bushels of this corn have been raised upon an acre.
 - 13. Virginia Golden corn (fig. 9). Reddish-yellow, polished.
- 14. Kentucky corn (fig. 8). Kernel white, somewhat shrivelled, elongated and pointed. Both these varieties are starchy and light, and are known in market as Ohio or Southern corn.

The large white-flint is a later variety of white corn, but is cultivated here occasionally. Its ear and kernel is large and white, with a specific gravity of 1.151.

DISEASES OF MAIZE.

THE MAIZE BRAND (Uredo maydis, Decandolle; U. zeæ, Chevalier).

Plate LVII. Fig. 1, 2.

All the species of brand, more or less, cause decisive injury to the organs of the plants which they infest; but the maize brand, among all the kinds of brand found in our cultivated grasses, produces the greatest and most extensive local transformations. It attacks

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all the parenchymatous organs of the maize plant, and more or less completely destroys them. The stalk, however, and the female and male blossoms, are the parts which it most especially affects. The leaves no longer furnish the great parenchymatous masses necessary for their development; and usually it seizes merely on their lowest parts, or also only on the husk-bearer; but its development here is already imperfect, and it forms on the leaf-organs only brand-bladders of the size of a poppy seed to a pea. In all the parenchymatous organs, however, it developes itself in the form of masses; and in good soil, and in actual cultivation of the maize, I have seen brand-bladders of the size of a child's head. Its development is a peculiar one, as it forces out great masses of cellular tissue, formed from the tissue of the mother plant, and similar in formation to the latter.

Some parts of the organs affected by the brand, swell and become white. The green color and compact formation of the outer skin gradually passes into a soft watery tissue of a silky lustre, the skin of which allows the large cellular formation to be seen through it by the naked eye. If we more closely examine this pathological product, we find that it consists of tolerably large tender-walled substance, the cells of which, like that of the normal vegetable tissue, contain sap, and possess a large slimy cellular kernel sticking on the side. In each of these cells, at a later period, is secreted a slimy grannlous substance, which is yellowish, and afterwards brownish, in which still later the brand is developed. Prof. Meyen examined this brand formation very critically, and we may here be allowed to repeat his investigations:

At first is seen in the large and juicy cells of the maize plant, or especially in the pathological cellular substance, the above mentioned little deposites of slime, which are produced on the inner surface of the cellular walls. From these, at first wholly irregularly formed, almost transparent deposites, proceed fibrous, dismembered and branching structures, which already exhibit a plant-like form, and which by their later changes more clearly evidence the same. These truly parasitic formations are in the beginning colorless, almost entirely transparent, and only under a strong magnifying power exhibit a fine-grained organised structure in their tender slimy substance; but soon it is observed that particular boughs of this little plant are branched out; and in individual cases, yet more developed, branches and twigs stand closely crowded together. At the same time with this branching, the fibres are already partially separated into small globular bodies, sometimes at the base, and sometimes at the point of the fibres; but for the most part their little side branches first separate off themselves. Many fibres are wholly changed into little branches in a wreathed form, which still hang together. They are originally ellipsoidal, and then become more or less globular; are at first of a yellowish and afterwards of a brownish color, and at last brown. But they likewise separate themselves from the branches producing them, and often before they have reached their normal size, which follows after their separation as it were by a sort of after ripening. By and by all the fibres fall away into such spores or grains of brand; by and by, too, the cells of the diseased vegetable substance are destroyed; and if we carefully cut through lengthwise (Fig. 1, b) the brand-bladders not yet opened or sprung apart (Fig. 1, a), we find that the white

cellular substance appears to be interwoven with irregular masses of brand, partially isolated and in the form of cells: the cellular substance, which still remains standing, forms white sheath-walls and cells, or, better described, deficiencies, the hollow space of which is filled with the dark brown brand. By and by this remains of the cellular tissue, constituting sheath-walls, becomes absorbed, and only the outer skin of the brand-bladder continues standing; but it begins likewise to be colored reddish or smutty, to become wrinkled or in folds, to dry up, and finally to tear open, by which the substance of the brand-spores is emptied and as it were sown out.

This species of brand causes manifold degenerations of particular parts and organs of the mother plants. On the stalk it forms irregularly rounded brand-bladders, very greatly differing in size. On the female blossoms, it never attacks all the blossoms (blüthen) of an ear; the blossoms on the top of the ear are for the most part more exposed to the brand than those at the base. Often only those fruit-buds that stand at the very tip, and frequently only the basilar ones, are diseased. Here the brand attacks only the fruit-knot, and changes it directly into a brand-bladder; so that indeed a person may find on the latter still the remains of the wasted pistil. But the rachis (midrib?) itself I have never found entirely gone. More frequently it seizes on the husk-leaves, and then changes the whole ear, or the fruit-bearing branch, into an organ not unlike a pine apple: it thickens all the leaves, and forms them similar to the scales of a fir cone. But in the male blossoms (blüthen) the brand seizes on the receptacle and the anthers, more rarely the petals, and changes all these organs into white, curled-up, easily-bent brand-bladders, one to three lines thick and often two or three inches long, which are likewise white, and of a beautiful silky lustre, sligtly tinged with red at the tip and on the side springing open to let out the spores.

The spores (Fig. 2) in their normal state are globular, but they are very frequently likewise somewhat ellipsoidal: in a ripe state, they are brown. The spore-skin is covered with little warts; and on many spores may be observed a dark point in the middle, the little opening (feusterchen? hilum) by which they were fastened to the fibrous bearer. Their diameter varies from 0.000320 to 0.000340 Paris inch.

This species always impairs some blossoms, as soon as it is seated in the ear, while the other blossoms standing near bear good ripe kernels. The brand-bladders can be very easily removed from the living plants by cutting them out; only this must be done as timely as possible, in order that in cutting them out, the bladders may not scatter their powder, and thus a future crop of brand not be prevented. For seed only, kernels should be selected from plants which have remained wholly free from the brand. This kind of brand is, by the structure of its spores, different from all others, and only related to the wheat brand.

Explanation of the Illustrations.

Fig. 1, brand-bladders: a, on the stalk of maize of the natural size; b, such a brand-bladder cut through lengthwise. Fig. 2, spores strongly magnified.

Note. The smut of the maize may often become a serious injury to a crop. It forms its black smutty vesicles upon all parts, except perhaps the leaves. The ears, however, suffer the most, inasmuch as this is the valuable part of the plant. Probably the period when the ear is forming is the most favorable for the propagation and growth of this fungus. There can be but little doubt but that the spores of the fungus are taken up by the roots, and conveyed in the sap to all parts of the plant. Hence it should be a point to remove the smutted parts the moment they are discovered, and to burn them. In a small field of maize, where last year (1847) there were only a few infected stalks, they were multiplied this year more than ten times, and I believe solely from the neglect to destroy the fungus.

VII. BROOM CORN.

Broom corn, being closely allied at least to the cereals, may be placed with propriety in this connection. Observations, as in the case of maize, have been made upon its growth during the entire season. I did not deem it necessary, however, to follow them up with numerous analyses, although it is extensively cultivated in this State. The proportions which have been determined, answer the double purpose of showing the progress of the increase of the plant, and the amount of mineral matter which it contains at different stages and periods of growth. We can determine, from the same data, the exhausting powers of the crop.

L	-								
				PRO	PORT	IONS.			
тіме. 1848.	PARTS.							QUANTITY.	PER CENTUM.
June 27.	Whole plant	•	-	-	-	-	-	100.00 grs.	
	Dry -	-	-	-	-	-	-	14.04	$14 \cdot 09$
	Ash -	-	•	-	-	-	-	$1 \cdot 44$	$1 \cdot 44$
	Calculated d	ry	-	•	-	-	-		10.25
July 5.	Height 16 ir	ches.	Firs	st jo	int ju	ıst app	pear	ing.	
	Leaf-blades	-	-	-	-	•	•	52.24 grs.	60.886
	Leaf-sheaths	3 -	-	^	-	-	-	33.56	$39 \cdot 114$
	Weight of v	vhole	plant	-	-	•	-	85.80	100.000
	1. Leaf-sheaths	-	-	-	-	-	-	$33 \cdot 56$	
	gave of Water	-	•	-	-	-	-	$30 \cdot 52$	90.912
	Dry matte	er -	-	-	-	-	-	$3 \cdot 04$	9.088
	Ash [deci	ded ta	aste of	pot	ash]	-	-	0.48	$1 \cdot 427$
	Percentage of			~	_	-	-		$15 \cdot 702$
	Percentage	of org	anic n	natte	er -	-	-		$84 \cdot 298$

ANALYSES OF BROOM CORN.

TIME. JULY 5.	PARTS. 2. Leaf-blades	quantity. $52 \cdot 24$ grs.	PER CENTUM.
JULY O.	Water	45.78	83.805
		8.46	16 · 195
	Dry matter	5.40	10, 199
	Ash [has a decided taste of soda and	0.01	1.740
	potash]	0.91	1.742
	Percentage of ash calculated on dry matter,		10.756
	Percentage of organic matter		89.244
July 16.	1. Stalks	325.50 grs.	
	Water		
	Dry matter		
	Dry	$265 \cdot 00$	8.140
	Ash	1.75	0.537
	Calculated dry		6.603
	2. Sheaths	251.400	
	Water	239 · 390	
	Dry	12.010	
	Ash	3.615	1.430
	Calculated dry	0 010	11.670
	•	005.00	22 010
	3. Leaves	265.50	
	Water	232.50	
	Dry	33.40	1 900
	Ash	$3 \cdot 44$	1.329
	Calculated dry		10.990
July 29.	1. Leaves	508.00 grs.	
	Water	412.00	
	Dry	96.00	
	Ash	$10 \cdot 25$	2.010
	Ash calculated dry		10.680
	Dry matter		19.090
	2. Sheaths	$383 \cdot 560$	
	Water	$336 \cdot 560$	
	Dry	$47 \cdot 000$	
	Ash	$5 \cdot 375$	1.401
	Ash calculated dry		11.436
	Dry matter		$12 \cdot 230$
	3. Stalk	892.20	
	Water	817.70	
	Dry	74.50	
	Ash	7.00	0.800
	Ash calculated dry	, 00	9.580
	Ash calculated ally		9 900

TIME.		D + D/D()							OT A NUTRICE	DED GENTUM
August 4.	1.	Stalk -	_				_	-	QUANTITY. 1277.00 grs.	PER CENTUM.
1,000,000	•••	Water	-	•	-	-	_	-	1141.50	
		Dry -		-	-	-	-	_	135.50	
		Ash -	-	-	-	-	-	-	8.60	0.673
		Ash calcula	ited o	drv			-	_		$6 \cdot 340$
		Dry matter		-	-	-	-			10.610
	9	Sheaths -	_	_	_	_	_		577·00	
	~.	Water	_	_	_	_	-	_	493.60	
		Dry -	_	_	_	_	_	_	83.40	
		Ash -	_			-	_	_	8.00	1.380
		Calculated	drv		_	-		_	0 00	9.597
		Dry matter		_	_	-	_			$14 \cdot 450$
	9								002.600	11 190
	ن.	Leaves -	-	-	-	•	-	-	293.600	
		Dry -	-	-	-	•	•	-	57·700	
		Ash -	- 	•	-	•	•	-	$5 \cdot 615 \\ 9 \cdot 731$	
		Calculated Dry matter		-	-	-	-		19.600	
		Per centum		-	-	•	•	-	19.000	1.812
		1 er centun	ı	•	-	-	•	•		1.012
August 11.	1.	Sheaths	-	-		-	_	-	670.00 grs.	
		Water		-	-	-	-	-		
		Dry -	-	-	-	-	_	-		
		Ash -	-	-	-	-	-	-	10.72	1.600
	2.	Leaves -	_				_	_	508.00	
	~.	Water	-		-	-		-	347.00	
		Dry -	-	_	-	-	-	-	161.00	
		Ash -	-	-	-	-	-		13.11	2.580
		Calculated	dry	-	-	-	-	-		$8 \cdot 142$
		Dry matter	-	-	-	-	-	-		31.690
	3.	Stalks -	_	_	_		-	_	$2270 \cdot 00$	
		Water		-	-	-	-	-	$1864 \cdot 00$	
		Dry -		-	-	-	-	-	$406 \cdot 00$	
		Ash -	-	-		-	-	-	$21 \cdot 25$	0.936
		Calculated	dry	-	-	-	-	-		$5 \cdot 234$
		Dry matter		-	-	-	-	-		17.880
	1	Brush -	_	_	_	_	_	_	$463 \cdot 00$	
	4.	Water	_		-	-	-		$273 \cdot 00$	58.980
		Dry -			-	•			190.00	<i>50 0</i> 00
		Ash -	_		-			_	$7 \cdot 09$	1.531
		Calculated	dry					-	. 50	3.731
		Dry matter			-	_		_		41.290
		L'y mancel								1. 200

TIME.		PARTS.							QUANTITY.	PER CENTUM.
August 22.	.1.	Stalk -	-	-	-	-	-	-	1130 · 000 grs.	
		Water	-	•	-	-	-	-	817.000	$71 \cdot 940$
		Dry -	-	-	-	-	-	-	313.000	1 000
		Ash -	-	-	-	-	-	-	$15 \cdot 375$	1.360
		Calculated		-	-	-	-	•		4.912
		Dry matter	•	-	-	-	-	-		27.600
	2.	Sheaths	-	-	-	•	-	-	$292 \cdot 50$	
		\mathbf{W} ater	-	-	-	-	-	-	$193 \cdot 55$	$66 \cdot 150$
		Dry -	-	-	-	-	-	•	$99 \cdot 00$	
		Ash -		-	~	-	-	-	$6 \cdot 10$	$2 \cdot 092$
		Calculated	dry	-	_	-	-	•		$6 \cdot 161$
		Dry matter	•	-		-	-			33.910
	3	Brush -	_	_	_	_	_	_	580.50	
	0.	Water	_	_			_	_	$457 \cdot 50$	78.810
		Dry -	_	_	_	_	_	_	123.00	70 010
		Ash -					_		5.52	0.937
		Calculated	drv				_	•	5 5 3	4.487
		Dry matter		_		_	_	_		21.010
		-1 <i>j</i>								21 010
September 1	. 1.	Leaves -	-	-	-	-	-	-	193.50 grs.	
		Water	-	-	-	-	-	-	$82\cdot 50$	
		Dry -	-	-	-	-	-	-	111.00	
		Ash -	-	-	-	-	-	-	$5 \cdot 12$	2.652
		Calculated	dry	-	-	-	-	-		4.612
		Dry matter		-	-	-	-	-		57.510
	2.	Sheaths	_	-	-	-	-	-	$285 \cdot 50$	
		Water	_	-	-	-	-	_	$162 \cdot 50$	
		Dry -	_	-	_	-	-	-	123.00	$3 \cdot 212$
		Ash -	-	-	-	•	_	-	9.18	
		Ash calcula	ated	dry	-	-	-	-		$7 \cdot 544$
		Dry matter		-		-	_	-		43.080
	3	Stalk -		_	_	_		_	1526 · 00	
	ο.	Water	_	-	_	-	-	_	998.00	
		Dry -	_	_	_	-	-	-	528·00	
		Ash -	_	-	_	_	_	_	20.00	1.310
		Calculated		_	-	_	_	_	20 00	3.787
		Dry matter		_	_	_	_	_		34.600
	А	•			•		_	_	005.00	94 000
	4.	Brush -	-	-	-	-	-	-	895.00	
		Water	-	-	-	-	-	-	533.00	
		Dry -	-	-	-	-	-	-	362.00	
		Ash -	- J	-	-	-	-	-	$14 \cdot 75$	1.647
		Calculated		-	-	-	-	-		4.072
		Dry matter	-	-	-	-	-	-		40.430

ANAYSES OF THE PARTS BELONGING TO BROOM CORN.

1. Analysis of the stalks.

								in a ton of stalks
Silex	-	-	-	-	-	-	$6 \cdot 24$	1.828 lbs.
Earthy phosphates	-	`-	-	-	-	-	$16 \cdot 66$	4.881
Lime	-	-	-	-	-	-	6.25	1.831
Magnesia -	-	•	-	-	-	•	3.74	$1 \cdot 095$
Potash	•	-	-	-	-	-	$30 \cdot 40$	8.907
Soda	-	-	-	-	-	-	$15 \cdot 46$	$4 \cdot 529$
Sulphuric acid	-	-	-	-	-	-	9.07	2.657
Chlorine -	-	-	-	-	-	-	$2 \cdot 14$	0.627
Peroxide of iron	-	-	-	-	-	-	2.61	0.764
Organic matter and	l mag	gnesia	ı -	-	-	-	$6 \cdot 24$	1.828
-								
							98.81	$28 \cdot 947$

2. Analysis of the sheaths of the Broom corn.

										Removed in a ton.
Silica	-	-	-	-	-	-	•	-	$40 \cdot 20$	28.903 lbs.
Earthy p	hospl	hates	-	-	-	-	-	-	15.00	10.785
Lime -	-	-	-	-	-	-	-	-	$3 \cdot 00$	2.157
Magnesi	a	-	-	-	-	-	-	-	$3 \cdot 24$	$2 \cdot 329$
Potash ·	-	•	-	-	-	•	-	-	$26 \cdot 56$	19.096
Soda -	•	-	-	-	-	-	-	-	$7 \cdot 33$	5.270
Sulphuri	ic aci	d	-	-	-	-	-	-	$3 \cdot 57$	$2 \cdot 566$
Chlorine	:	-	-	-	-	-	-	-	1.72	1.236
									100.00	
									$100 \cdot 62$	$72 \cdot 342$

3. Analysis of the ripe Broom corn brush, with the seeds.

•	, ,						,		
-	-	-	-	-	-	-	$32 \cdot 50$	11.960	lbs.
spha	tes	-	•	-	-	•	$36 \cdot 15$	13.303	
-	-	-	-	-	-	-	$0 \cdot 40$	0.147	
-	-	-	-	-	-	•	0.10	0.036	
-	-	-	•	-	-	-	$27 \cdot 32$	10.053	
-	-	-	-	-	-	•	$2 \cdot 37$	0.870	
-	-	-	-	-	-	-	2.50	0.846	
acid	-	-	-	-		unde	etermine d.		
							$101 \cdot 14$	$37 \cdot 215$	
								sphates 32·50 sphates 36·15 0·40 0·10 27·32 2.50 acid undetermined.	Removed from ton of brush sphates 32·50 11·960 13·303 0·40 0·147 0·10 0·036 27·32 10·053 2.50 0·846 acid 2.50 undetermined.

4. Composition of the ash of Broom corn seed.

	•	_						
Carbonic ac	cid	•	-	-	•	-	not dete	ermined.
Silicic acid	-	-	-	-	-	-	-	41.975
Sulphuric ac	cid	-	-	-	-	-	not det	ermined.
Phosphoric	acid		•	-	-	-	•	28.760
Phosphate of	of perc	oxide	of ire	n	-	-	-	0.525
Lime -	•	-	-	-	-	-	-	0.845
Magnesia	-	-	-	-	-	-	•	3.010
Potash -	-	•	•	-	-	-	-	3.920
Soda -	•	-	-	-	-	-	-	$7 \cdot 247$
Chlorine	•	-	-	-	-	-	•	0.245
Organic aci	ds	-	-	-	-	-	-	$4 \cdot 200$
8								
								90·727 S.
			PROPO	RTIONS.				
Water -	•	-	-	-	-	-	-	$12 \cdot 22$
Dry matter		-	-	-	-	-	-	87.78
Ash -	•	•	•	-	-	-	-	3.00
Percentage	of ash	calc	ulated	on th	e dry	ma	tter,	3·417 S

VIII. BUCKWHEAT.

In its classification as a plant, buckwheat belongs to a family far removed from the cereals; but in the composition and properties of its seed, it approximates to them closely, and hence it is placed here. A few analyses only have been made of it. The two following will show the composition of the ash of the seed. They also show, in the amount of earthy phosphates and phosphoric acid, a remarkable similarity to the grains of the cereals. Its specific gravity is 1.081:

Silica	-	-	-	-	-	-	7.06
Earthy phosphat	es -	-	-	-	-	-	57 ·60
Lime	-	-	-	-	-	-	0.14
Magnesia -	-	-	-	-	-	-	2.66
Potash	-	-	-	-	-	-	$23 \cdot 33$
Soda	-	-	-	-	-	-	$2 \cdot 04$
Sulphuric acid	-	-	-	-	-	-	7.30
Chlorine -	•	-	-	-	-	•	0.20
							100.23

The amount of silica may have been increased from want of attention to the foreign matter upon the seed: its well known grittiness, when not removed by a mill, renders the supposition probable.

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Composition of the ash of buckwheat.

98·24 S.

CHAPTER V.

THE LEGUMINOUS PLANTS.

GENERAL REMARKS ON THIS ORDER OR FAMILY OF PLANTS. THE PROPORTIONS IN WHICH THE ELEMENTS EXIST IN THE BEAN AND PEA: COMPOSITION OF THE ASH, PROXIMATE ORGANIC COMPOSITION OF THE BEAN AND PEA.

The Bean and Pea are regarded as among the important products of the farm and garden. Containing as they do a large amount of nutriment, and being also susceptible of preservation in a dry state for a long time, and under circumstances which would injure the cereals, they supply an important food where it is very much needed, as at sea, and for laborers who are engaged in work which continually wastes the muscles. The bean and pea particularly supply the place of meat. Horses and cattle, however, are not partial to beans; and, indeed, it is not safe to feed beans to horses. Sheep are safely fed with beans: they are regarded as extremely valuable for this kind of stock, and as acting favorably in the development of a large and heavy fleece.

Clover, which belongs to this family of plants, has been spoken of in a former chapter. It has been impossible to give that time to the examination of this family, which I desired. I have been obliged to confine my attention to a few varieties of beans and peas. This partial and limited account of the composition of a highly interesting family of plants could not well be avoided. What is accomplished, however, may be of some moment, and form an inducement to prosecute researches more extensively in this order.

OF BEANS AND THEIR FOLIAGE.

AUGUST 22.

PROPORTIONS.

					PROF	ORTIO	NS.					
1.	Windsor b	ean	leave.	s and	l stem	s.						
	Quantit	y	-	-	-	-	-	-		960.00	grs.	
	Dry	•	-	-	-	-	-	-	-	$284 \cdot 00$	J	
	Water		-	-	-	-	-	-	-	$676 \cdot 00$		
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	-	43.87		
	Calcula	ted d	lry	-	-	-	-	-	-	$148 \cdot 00$		
	Percen	tage	of dr	y ma	tter	•	-	•	-		29	•50
2.	Windsor b	ean.										
	Quantit	٦,					-			586.00	are	
	Dry	. <i>.</i>	_	_	_		_	-	_	24.00	810.	
	Water	_		_	-	_	_	_	_	562.00		
	Ash	_		-	-					9.22		
	Ash cal	culat	ed dr	v -	-	-	•	-		22.51		
	Percent				itter	-	•	-	-		4	.09
	•	, b		•								
					AUG	JUST	6.					
3.	Small roun	nd fi	eld b	ean s	tems	(Plan	its on	e-fou	rth gr	own).		
	Quantit	y	•	•	-	-	-	-	-	100.00	grs.	
	Water	•	-	-	•	•	-	-	•	80.75		
	Dry	•	•	-	•	-	•	•	-	$19 \cdot 25$		
	Ash	-	•	-	•	•	-	•	-	1.90		
	Ash cal	culat	ed dr	y -	•	-	•	-	-	9.87		
4.	The leaves	of t	he sa	me ı	ariety	y•						
	Quantit	v	-	-			-	-		100.00	grs.	
	Water	•	•	_	-		-	•	•	83.00	5	
	Dry	-	-	_	-		-	-	-	15.56		
	Ash	•		-				•	•	3.08		
	Ash cal	culat	ed dr	v -	-	•	-	-		18.599		

ANALYSIS OF BEANS.

5. Stem	s and lead	ves of	• the	Wind	lsor b	ean.				
	os with real	000 9	0.00	,, ,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,				_		ed in a ton of bean straw.
Silica -	-	-	•	•	•	•	-	$35 \cdot 10$		5.907 lbs.
Earthy pl	hosphates	-	-	•	•	•	-	13.65	1	$3 \cdot 452$
Lime -	•	•	•	•	•	•	•	$28 \cdot 125$	2	8.771
Magnesia	a -	•	•	•	-	•	•	trace.		
Potash -		•	•	•	-	-	•	16.030	1	6.398
Soda -		-	•	-	•	•	•	trace.		,
Sulphuri	ic acid	•	-	•	•	-	•	1.005		1.028
Chlorine	•	•	•	-	-	•	•	2.315		$2 \cdot 368$
								95.725	9	7 · 924
6. Pod	of the W	indson	· bea	n.						
	ilica -	•	•	•	-			•	4.500	
	hosphates	-				•	-		26.000	
	ime -		•			•	•		22.500	
	Iagnesia		-	•			-	•	2.420	
	otash -	_	•	•		•		•	19.440	
	oda -	•	_						11.260	
	ulphuric a	icid		-	•			-	2.600	
	hlorine	-	•	•	-			•	1.160	
	rganic ma	atter	_		-	•	_	-	2.300	
	arbonic a			-		-	•	•	5 ·453	
									97.506	
7. Whi	te Kidney	bean bean	•							
	Silica -	-	-	-	-	-	•	-	0.80	
E	Carthy and	alka	line	phosp!	hates	•	•	•	30.95	
	ime -	-	-	•	-	-	-	•	0.35	
\mathbf{M}	Iagnesia	-	-	•	•	-	-	•	0.45	
	otash -	-	-	•	-	-	-	•	$35 \cdot 04$	
S	loda -	-	-	-	-	-	-	•	$2 \cdot 42$	
	ulphuric :	acid	-	-	•	-	-	• •	4.98	
	hlorine	-	-	-	•	-	•	-	0.52	
	rganic m	atter	-	•	•	-		•	6.60	
	arbonic a		•	•	•	-	•	•	18.85	
									100.00	
									100.96	

Bean ash always contains more or less of carbonic acid. As it effervesces briskly with acids; so, the amount of organic matter seems to be greater than in other kinds of ash. The bean burns without difficulty, and furnishes a white ash in lumps of the size and

shape of the bean itself; and it is rather remarkable that, where so much potash exists, the consumption should take place with very little danger of fusion; while in the cereals, the greatest difficulty is encountered in combustion, and the ash will fuse at a low temperature. This fact is probably owing to the state of combination of the elements, and to the presence of a large amount of alkaline phosphates which fuse very readily.

8. A large white bean, the large white English or Butterfield bean.

Silica	-	-	-	-	-	-	0.275
Еактну Рнозрна	TES:						
Lime -	-	•	-	•	-	-	$3 \cdot 145$
$\mathbf{Magnesia}$	•	•	•	-	-	-	6.620
ALKALINE PHOSPI	HATES	:					
Potash -	•	-	•	-	-	-	0.095
Soda and ph	osph	oric a	cid	-	-	-	18.865
Lime	-	-	-	-	•	-	0.042
Magnesia -	-	-	-	-	•	•	trace.
Potash	-	-	-	•	-	•	$39 \cdot 946$
Soda	-	-	-	-	-	-	6.866
Phosphate of perc	oxide	of iro	n -	-	-	-	0.525
Chloride of sodi	um	-	-	-	•	-	0.265
Sulphuric acid	•	-	-	-	-	•	$2 \cdot 955$
Carbonic acid	•	-	-	-	•	•	5.400
Organic matter	-	-	-	-	-	unde	termined.
							84·825 S.
							Ox Oze D.
		7		7 . 1			
9. Po	ods oj	f the	same	kind	of b	ean.	20.005
Silica	•	f the -	same -	kind •	of b	ean. -	28.925
Silica Earthy Phospha	TES:	•	•	•	of b	ean. -	
Silica EARTHY PHOSPHA Phosphate of	TES:	•	•	•	of b	ean. -	0.075
Silica EARTHY PHOSPHA Phosphate of Lime -	TES:	•	•	•	of b	ean. - -	0·075 0·360
Silica EARTHY PHOSPHA Phosphate of Lime - Magnesia	TES:	•	•	•	of b	ean. - - -	0·075 0·360 8·770
Silica EARTHY PHOSPHA Phosphate of Lime - Magnesia Silicic acid	TES:	•	•	•	of b	ean. - - -	0·075 0·360 8·770 0·125
Silica EARTHY PHOSPHA Phosphate of Lime - Magnesia	TES:	•	•	•	of b	ean.	0·075 0·360 8·770 0·125 14·135
Silica EARTHY PHOSPHA Phosphate of Lime - Magnesia Silicic acid	TES:	•	•	•	of b	ean.	0·075 0·360 8·770 0·125 14·135 13·975
Silica EARTHY PHOSPHA Phosphate of Lime - Magnesia Silicic acid Phosphoric a	TES:	•	•	•	of b	ean. - - - -	0·075 0·360 8·770 0·125 14·135 13·975 0·815
Silica EARTHY PHOSPHA Phosphate of Lime - Magnesia Silicic acid Phosphoric a	TES:	•	•	•	of b	ean.	0·075 0·360 8·770 0·125 14·135 13·975 0·815 19·975
Silica EARTHY PHOSPHA Phosphate of Lime - Magnesia Silicic acid Phosphoric a Lime Magnesia -	TES:	•	•	•	of b	ean.	0·075 0·360 8·770 0·125 14·135 13·975 0·815
Silica EARTHY PHOSPHA Phosphate of Lime - Magnesia Silicic acid Phosphoric a Lime - Magnesia - Potash	TES: f percentage acid	•	•	•	of b	ean.	0·075 0·360 8·770 0·125 14·135 13·975 0·815 19·975
Silica EARTHY PHOSPHA Phosphate of Lime - Magnesia Silicic acid Phosphoric a Lime Magnesia - Potash Soda	TES: f percentage acid	•	•	•	of b	ean.	0·075 0·360 8·770 0·125 14·135 13·975 0·815 19·975 4·007
Silica EARTHY PHOSPHA Phosphate of Lime - Magnesia Silicic acid Phosphoric a Lime - Magnesia - Potash - Soda Chloride of sodi	TES: f percentage acid	•	•	•	of b		0·075 0·360 8·770 0·125 14·135 13·975 0·815 19·975 4·007 0·580 0·034 6·200
Silica EARTHY PHOSPHA Phosphate of Lime - Magnesia Silicic acid Phosphoric a Lime - Magnesia - Potash - Soda - Chloride of sodi Sulphuric acid	TES: f percentage acid	•	•	•	of b		0·075 0·360 8·770 0·125 14·135 13·975 0·815 19·975 4·007 0·580 0·034
Silica EARTHY PHOSPHA Phosphate of Lime - Magnesia Silicic acid Phosphoric a Lime Magnesia - Potash Soda Chloride of sodi Sulphuric acid Carbonic acid	TES: f percentage acid	•	•	•	of b		0·075 0·360 8·770 0·125 14·135 13·975 0·815 19·975 4·007 0·580 0·034 6·200

10. Vine.	s of t	he sar	ne va	riety	of b	$\epsilon an.$	
Silica				•	-	-	21.575
EARTHY PHOSPHATE	s:						
Lime -	-	•	-	-	•	-	5.825
Magnesia -	-	-	-	-	-	-	$3 \cdot 286$
Phosphate of	peroxi	de of	iron	•	-	-	3.825
Phosphoric ac	id	-	-	-	-	-	$9 \cdot 935$
Silicic acid	-	-	•	-	•	-	trace.
Lime	-	-	-	-	-	•	$5 \cdot 062$
Magnesia -	-	-	-	-	-	-	0.055
Potash	-	-	-		-	-	$7 \cdot 053$
Soda	-	-	•	-	-	-	$36 \cdot 194$
Chloride of sodium	-	-	-		-	-	0.118
Sulphuric acid -	-	-	-	-	-	-	0.464
Carbonic acid -	-	-	-	-	-	-	4.450
Organic matter	•	-	-	-	-	und	ermined.
							97·944 S.

ANALYSES OF THE ASH OF SEVEN VARIETIES OF BEANS.

1	. Whit	e Kid	ney	bean.							
	Quant	titv		-		-	-	-	-	400.00 grs.	Per centum.
	-	•	_	•		-	-		-	347.00	
	Wate	r -	-	•	-	-	-		_	53.00	$13 \cdot 250$
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-		-	-		-	-	12.09	$3 \cdot 020$
	Ash o	alcul	ated	dry	-		-	-	-		13.930
				•							
2.	. Refug	gee, or	100	0-to-1	bean.						
	Quant	tity	-	-	-	-	-	-	-	400.000 grs.	
	Dry	-	-	•	•	-	-	-	-	354.000	
	Wate	r -	•	-	-	-	-	-	-	$46 \cdot 000$	11.500
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-		14.625	3.650
	Ash o	calcul	ated	dry	•	-	-	• .	-	-	$16 \cdot 490$
3	. Field	l-pea l	bean.		-					n.	
	Quan	•	-	-	-	-	_	-	-	400.00 grs.	
	Dry	-	-	•	-	-	-		-	348.00	
	Wate	r -	-	-	-	-	•		-	$52 \cdot 00$	$13 \cdot 000$
	Ash	•			-		-	-	-	$12 \cdot 63$	$3 \cdot 157$
	Ash o	calcula	ated	dry	-	-	•	-	-		

	4. Turtle-sou	p bea	n.							Per centum.
	Quantity	•	-	-	-		-		400.00 grs.	
F enter	Dry -	•	-	-	-		•		298.40	
	Water -	-	-	-	-	-	-		101.60	$25 \cdot 400$
	Ash -	-	-	-	-	-	•	-	12.35	3.087
	Ash calcula	ated d	ry -	•	•	-	-	•		16.650
	5. Horticultu	rist b	ean.							
	Quantity	-	-	-	-				400.00 grs.	
	Dry -	-	-	-	•	-	-	-	355 · 50	
	Water -	-	•	-	-	-	•	-	44.50	$11 \cdot 125$
	Ash -	-	-	-	-	-	-	-	$15 \cdot 74$	3.935
	Ash calcula	ated d	lry	•	-	•	-	-		17.710
	6. Early Chi	na bed	zn.							
	Quantity	•	-	•	-		-	-	400.00 grs.	
	Dry -	-	-	•	-	•	-	-	359.00	
	Water -	-	•	-	-	-	-	-	41.00	
	Ash -	-	-	•	-	-	-		$16 \cdot 59$	1.025
	Ash calcula	ited d	ry -	-	•	-	•	•		4.147
	7. Early Moh	nawk	bean.							
	Quantity	•	-		•	-			400.00 grs.	
	Dry -	-		•	-	-		-	347.00	
	Water -	-	-	-	•	-	-		$53 \cdot 00$	14.250
	Ash -	•	•	•	•		-		12.69	$3 \cdot 172$
	Ash calcula	ted d	ry -	-	•	•	-	-		
			•							

Composition of the inorganic matter of the seven foregoing varieties of beans.

The ash of all the varieties contains carbonic acid.

1. White kidney bean.

i. vv nite kiar	ueu oe	an.							
Silica -		•					_	1.100	Calculated without carbonic acid or organic matter. 1.222
	1 .					_			
Earthy pho	sphat	es	-	•	•	•	•	40.850	$45 \cdot 384$
Lime -	•	•	•	•	-	•	-	0.021	$0 \cdot 023$
Magnesia	•	•	•	•	•	•	•	0.010	0.011
Potash -		-	•	•	•	•	•	31.625	$35 \cdot 135$
Soda -	•	-	•	•	•	•	-	$11 \cdot 135$	$12 \cdot 320$
Sulphuric a	acid	-	-	•	•	•	-	3.785	$4 \cdot 205$
Chlorine	-	-	-	-	-	•	-	trace.	
Carbonic ac	cid	•	-	-		· -	•	8.401	
Organic ma	atter	-	-	-	-	-	-	2.620	
								99.637	98·350

Cult.							0.400	Calculated without carbon
Silica	-		-	-	-	-	2.400	2.476
Earthy phosphate	es	-	-	-	-	-	45.700	$47 \cdot 162$
Lime -	-	-	-	-	-	-	trace.	0.000
Magnesia -	-	-	-	-	-	-	0.020	0.020
Potash	-	-	-	-	-	-	33.270	34.334
Soda	-	•	-	-	-	-	10·S00	11.145
Sulphuric acid	-	-	-	•	-	-	3.510	$3 \cdot 622$
Chlorine -	-	-	-	-	-	-	trace.	
Carbonic acid	-	-	-	•	-	-	$3 \cdot 210$	
							98.890	98.759
3. Field-pea bean.								
								Calculated without carbon acid or organic matter.
Silica	-	-	•	•	-	-	$2 \cdot 125$	$2 \cdot 454$
Earthy phosphate	es	•	-	-	-	-	30.500	$35 \cdot 227$
Alkaline phospha	ates	1	-	-	-	-	$4 \cdot 400$	$5\!\cdot\!0 ext{S}2$
Lime	-		-	-	-	-	trace.	
Magnesia -	-	-	-	-	•	•	none.	
Potash	-	-	-	-		-	$29 \cdot 285$	$33 \cdot 824$
Soda	-			-	-	-	$13 \cdot 105$	$15 \cdot 136$
Sulphuric acid	-	-	-	-	-	-	6.535	$7 \cdot 547$
Chlorine -	-	-	-	-	•	-	trace.	
Carbonic acid	-	-	-	-	-	-	$10 \cdot 235$	
Organic matter	-	•	-	-	-	-	5.312	
							$\overline{97\cdot 107}$	$\frac{-}{99 \cdot 270}$
4. Turtle-soup beam	·.							
			Specin	c gravi	ty, 1°2	242.		Calculated without carbon acid or organic matter.
Silica	-	-	-	-	-	-	2.900	$3 \cdot 204$
Earthy phosphate		-	-	-	-	-	$27 \cdot 950$	$30 \cdot SS4$
Alkaline phospha	ites	-	-	-	-	-	$9 \cdot 100$	$10 \cdot 055$
Lime	-	-	-	-	-	-	0.200	0.221
Magnesia -	-	-	-	-	-	-	trace.	
Potash	-	-	-	-	-	-	19.610	21.669
Soda	-	-	-	-	-	-	$24 \cdot 250$	$26 \cdot 769$
Sulphuric acid -	-		-	-	-		5.500	$6 \cdot 077$
Chlorine -	-	-	-	-	-	-	none.	
Carbonic acid	-	-	-	-	-	-	6.571	
Organic matter	•	-	-	-	-	-	2.000	
-0								

2

The Turtle-soup bean is particularly excellent for soups, having the rich flavor of the green turtle soup. Whether this quality is connected with the composition of the ash, is not easily determined: it however contains much soda and less potash than other varieties. It is worthy of notice, too, that the soda of beans is not in combination with chlorine, but probably with sulphuric acid.

y with sulphuric acid.							
5. Horticulturist bean.							Calculated without carbonic acid or organic matter.
Silica	-	-	-	-	-	0.200	$0.\overline{2}19$
Earthy phosphates	-		-	-	-	31.875	$34 \cdot 935$
Alkaline phosphates	-	-	-	-	-	3.750	$4 \cdot 110$
Lime	•	-	-	-	-	trace.	
Magnesia	-	-	-	-	-	trace.	
Potash	-	-	-	-	-	$38 \cdot 080$	41.735
Soda	-	-	-	-	-	$11 \cdot 475$	12.574
Sulphuric acid -	-	-	-	-	-	$4 \cdot 340$	4.756
Chlorine	-	-	-	-	-	0.010	0.010
Carbonic acid -	-	-	-	-	-	$8 \cdot 400$	
Organic matter -	-	-	•	-	-	1.200	
						96 • 130	$98 \cdot 339$
6. Early China bean.							Calculated without carbonic acid.
Silica	-	-	-	-	-	1.655	1.820
Earthy phosphates -	-	-	-	-	-	$32 \cdot 956$	$36 \cdot 251$
Alkaline phosphates	-	-	-	-	-	2.756	3.031
Lime and magnesia	-	-	-	-	-	traces.	
Potash	_	-	_	-	-	$37 \cdot 421$	$41 \cdot 163$
Soda	_	-	-	-	-	10.256	11.281
Sulphuric acid -	-	-	_	-	-	$5 \cdot 470$	$6 \cdot 017$
Chlorine	-		-	-		traces.	
Carbonic acid -	-		-		-	8.410	
Organic matter -	-	-	-	_		1.620	
018						100.544	99.563
~ 7 7 7 7 7 1 1 1 1						100 944	Calculated without organic
7. Early Mohawk bean.						0.000	matter or carbonic acid.
Silica	-	•	•	-	-	0.200	0.230
Earthy phosphates -	•	•	•	-	-	33.800	38.870
Alkaline phosphates	-	-	•	-	-	2.500	1.875
Lime and magnesia	-	-	-	-	-	traces.	
Potash	-	-	-	-	-	34.050	$39 \cdot 157$
Soda	-	-	-	-	-	8.845	10.171
Sulphuric acid -	-	-	-	-	-	4.725	$5 \cdot 433$
Chlorine	-	-	-	-	-	0.430	0.494
Carbonic acid -	-	-	-	-	-	13.860	
Organic matter -		-	-	-	-	1.210	
						${99.620}$	${97 \cdot 230}$

It appears from the foregoing analyses, that the ashes of the several varieties of bean differ somewhat; the greatest deviation from the majority of them being in the Turtlesonp bean. The ash varied somewhat in the degree of causticity, or in the amount of carbonic acid. The ash, in each instance, was perfectly burned and white. There is a great resemblance between the composition of the ash of beans and that of the cereals, in respect to their richness in earths, alkalies and phosphates.

The exhausting power of a crop of beans may be determined by the following calculations. A bushel of beans weighs about 60 lbs. There will be removed in every ten bushels of beans:

					Turtle-soup bean.		Horticulturist bean.
						lbs. ez.	lbs. oz.
Silica	-	•	-	-	-	0 8.584	0 0.755
Earthy phosphates	-	-	-		-	2 2.732	7 8.360
Alkaline phosphat	es	-	-		-	1 10.936	$0 14 \cdot 160$
Lime	-	-	-	•	-	0 0.592	
Magnesia -		-	-	-			
Potash	-	•	-	-	-	$3 \ 10.045$	8 15.790
Soda	-	-	-	-	-	4 7.780	2 11.329
Sulphuric acid	-	•	-	-	-	1 0.280	1 0.387
Chlorine -	-	-	-	•	-		0 0.037
						16 8.949	${21} \ 2.818$

There will be removed in a ton of foliage:

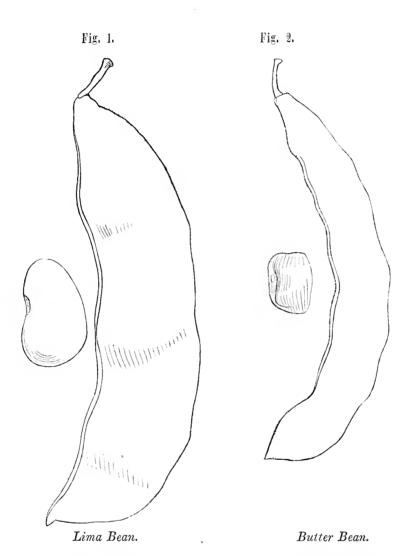
								lbs.	oz.
Silica	•	-	-	-	-	-	-	14	15.341
Earthy p	hosph	ates	-	-	•	•	-	15	12.861
Lime	-	-	•	-	-	-	-	3	7.965
Magnesia	a -	-	-	-	-	•	-	0	0.608
Potash	-	-	-	-	-	-	-	4	13.977
Soda	-	-	-	•		-	-	25	0.160
Sulphuric	c acid	-	-	-	-	-	-6	0	$5 \cdot 129$
Chlorine	•	-	•	-	-	-	-	0	1.304
									0.045
								64	$9 \cdot 345$

Beans, although put upon what is regarded the barren soils, still make shift to supply themselves with the richest and most valuable constituents of the soil; and hence they may be regarded as an exhausting crop. The plan of starving the bean is, after all, the true mode of treating it, otherwise it would become strong and unpalatable.

PROXIMATE ORGANIC ANALYSES OF THREE VARIETIES OF BEANS.

1. Fi	eld pea	-bean.								
S	Starch	-	-	_	-	-	-	•	•	14.68
I	egumi	ne		-	-	-	-	-	-	20.00
	lbumei		-	-	-	-	-	-	-	0.97
C	asein	-	-	-	-	-		-	-	$2 \cdot 21$
F	`ibre	-	-	-	-	-	-	-	-	$30 \cdot 71$
Б	Resinous	s matt	er		-	-	-	-	-	4.50
F	Extracti	ve ma	tter	-	-	-		-	-	$12 \cdot 20$
	Vater	-	-	-	-	-	-	-	-	13.00
										98.27
2.	White .	Kidne	y bea	n.						
S	Starch	-	-	-	-	-	-	-	•	$36 \cdot 74$
L	egumi	ne	-	-	-	-	-	-	-	18.60
A	Jbumer	and	case	in			-	-	-	$9 \cdot 92$
F	ibre		-	-	-	-	-	-	-	15.42
S	ugar a	nd ext	tract	-	-	-	-	-	-	$7 \cdot 20$
	Vater	-	-	-	-	-	-	-	•	$13 \cdot 25$
										101.13
3. 1	Refugee	e, or 1	000-t	o-1 be	an.					
S	tarch	-	-	-	-	-	-	_	-	30.20
\mathbf{L}	egumir	ne			-	-	-	-	-	$17 \cdot 44$
	lbumen		casein	-	-	_	-	-	-	8.40
F	ibre	-	-	-	-	-	-	-	-	$21 \cdot 22$
S	ugar ai	nd ext	ract,	mostly	y extr	act	-	-	-	6.75
	Vater	-	-	-	•	-	-	-	-	11.50
			,							95.51

The starch is separated from the bean meal by washing and subsidence; albumen by heat, and casein by acetic acid, which, however, does not exist except in small quantities. The remainder is evaporated to dryness, and the dry matter washed with a large quantity of alcohol, which dissolves the resinous, extractive matter and sugar. The remainder, which is insoluble in alcohol, is set down as legumine after drying in the water bath.



These Lima and Butter beans were premium specimens of those varieties, which were exhibited at the Horticultural Exhibition of the Albany and Rensselaer Counties Society. The first was a very fine and perfect sample of the variety.

The Butter bean is a beautiful bush bean, having a fine blue color, and, as its name implies, is very rich and nutritive. It is not extensively cultivated, but well deserves to replace some of the older varieties. The plant, it will be observed, breaks up into numerous varieties, each of which is marked by distinct characters. In fact the varieties of bean are more distinct than those of other plants.

ANALYSES OF THE PEA (PL. 48).

PROPORTIONS.

1. Early June pea vine.

					•	,	-				
1 D.u.	of	11.0 0							Quantity.		er centum.
1. Botto Dry	т ој	ine v	ппе	•	-	-	-	•	$89.00 \\ 22.00$	grs.	
Wate:		•	-	••	•	-	•	•	67.00		
Ash	r -	•	•	•	-		-	-	2.28		0.55
	-	-	•	-	-	•	-	•			$2 \cdot 55$
2. Top	-	•	-	-	-	-	-	-	100.00		
Dry	-	•	-	-	-	-	-	-	22.30		
Water	ı -	-	-	-	-	-	•	-	77.70		1 00
Ash	-	-	-	-	-	-	-	-	1.20		1.20
3. Leaf	-	-	-	•	-	-	-	-	100.00		
Dry	-	•	-	-	-	-	-	•	24.60		
Water	-	-	-	-	-	-	-	-	75.40		
Ash	-	-	-	-	-	•	-	-	2.15		$2 \cdot 15$
4. Pods	-	-	-	-	-	-	-	-	$100 \cdot 00$		
Dry	-	-	-	-	-	-	-	-	$10 \cdot 11$		
Water	-	-	-	-	-	-	-	-	89.89		
Ash	-	-	-	-	-	-	-	-	$1 \cdot 43$		1.43
5. The 1	Pea	-	-	-	-	-	-	-	100.00		
\mathbf{Dry}	-	-	-	-	-	-	-	-	30.70		
$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	-	$1 \cdot 12$		$1 \cdot 12$
		0	77.	, ,			11	,	c		
		2.	Vet	ch vii		or sm	iall i	veagej	form pea.		
Quant	ity	-	-	-	-	-	-	•	100.00	grs.	
Dry	-	-	-	-	-	-	-	•	25.00		
Water	r -	•	-	-	-	-	-	-	75.00		
Ash	-	-	-	-	-	-	-	-	1.45		1.45
	9	<i>a.</i>	1	. £ 17			47	11			
			uysis	oj u	ie asi	i oj i	ine si	nauva	vedge form		
	Silie		- ,	•	-	-	•	•		4.30	
		hy ph	ospha	ates	-	-	•	-		38.40	
	Lim		-	-	-	-	-	-		0.10	
		gnesia		-	-	•		-		0.96	
	Pota		-	-	-	•	•	•		27.95	
	Sod			-	-	-	-	-		3.57	
		phuric			-	•	-	-		8.72	
		bonic			-	-	•	-		13.39	
	Org	anic	matte	r	•	-	-	-		3.50	
										100.59	

This variety of small pea is raised for fodder in some parts of western New-York, and is there held in esteem.

4. Proximate organic analysis of the green June pea.

Quantity	/ -	-	-	-		-	-	-	$150 \cdot 00$	grs.
Starch	-	-	-	-	-	-	-	-	11.78	
Albume	n	-	-	-	-	-	-	-	3.57	
Casein	-	-	-	-	-	-	-	-	0.18	
\mathbf{Fibre}	-	-	-	-	-	-	-	-	$9 \cdot 34$	
Residue	conta	ining	legun	nine,	sugar	and	extrac	et,	$20 \cdot 00$	
Water	-	-		-	•	-	-	-	$105 \cdot 00$	

The starch exists in the pea when it is green. When ripened, its elements are changed; the starch disappears mostly, and legumine may be said to take its place. The above analysis is quite imperfect, and is taken from the notes mainly to show the amount of starch when the plant is in its green state.

CHAPTER VI.

THE ESCULENT VEGETABLES.

I. THE CABBAGE TRIBE.

This natural tribe of plants are distinguished botanically by their tetradynamous flowers, and chemically by containing a large amount of nitrogen, which, during their decomposition, is manifested by exhaling an odor closely resembling decaying animal matters. Medicinally they possess what is called antiscorbutic and stimulant qualities. So strongly stimulant are some of these plants, or parts of them, that unless this character is obtunded by mucilage, they become useless as food for man. In the cabbage, turnip, cress and radish, this stimulating principle is shielded by mucilage. The seeds are remarkable, too, for the great amount of sulphur which they contain, amounting in some cases to nearly 30 per centum. Oil is one of their most valuable products: that of the rape is well known, and is in general use, both as a manure, and for feeding cattle.

The analyses which have been made in the laboratory for the Survey, have been confined to the cabbage mostly: they were undertaken by Mr. Salisbury, and have been carefully executed. A statement of these analyses has been given in the American Journal of Agriculture and Science, and it is from this source that the following determinations are compiled. For a full account, the reader may consult the work just referred to.

ANALYSES AND EXAMINATION OF FIVE VARIETIES OF CABBAGE.

1. The Scotch, Strasburgh or Drum-head cabbage.

Head compact, yellowish-green without and blanched within. Furnished by Mr. J. B. Hutson of Albany.

			PRO	PORTIC	NS.			
								100.00 parts gave
Water -	-	-	-	-	-	-	-	SS·665
Dry matter	-	-	-	-	-	-	-	11.335
Ash -	-	-	-	-	-	-	-	0.790
Ash calcula	ted	dry	-	-	-	-	-	$6 \cdot 969$

The proportions of its proximate organic elements are as fellows:

Sugar and	extrac	tive r	natter	, acco	mpan	ied w	rith		Calculated without the water.
the subs					_				
cabbage (-	-	•	-	٠_	-	$9 \cdot 32$	$55 \cdot 977$
Oil and wa	x -	-	-	-	•	-	-	0.16	$\cdot 962$
Fibre with	a littl	e star	ch an	d chlo	orophy	yl -	-	0.94	5.646
Fibre -	-	-	-	-	-	-	-	$2 \cdot 16$	$12 \cdot 974$
Matter disse	olved	by po	tash 1	from 1	the fib	ore (al	bu-		
men) -	-	-	-	-	-	-	-	0.88	5.281
${f Albumen}$	•	-	-	-	-	-	-	1.74	$10 \cdot 461$
Casein -	-	•	-	-	-	-	•	0.36	2.157
Dextrine	-	-	•	-	-	-	•	1.09	$6 \cdot 542$
Water -	-	-	•	-	-	-	-	$83 \cdot 42$	
								100.07	98·190 S.

The produce of an acre may be estimated by taking 15 lbs. as the average weight of each plant, and assuming that it will occupy a square yard. An acre will produce, according to the estimate, 36 tons, containing 30 tons of water and 6 tons of dry matter. The nitrogenous matter of an acre, according to the foregoing calculation, will be 357.6 lbs. in the green plant, and 2148 lbs. in the dry. A ton of the green plant contains 15.8 lbs. of inorganic matter, and 139.38 lbs. in the dry plant. There will be removed from an acre, 569 lbs. of inorganic matter. A ton of the green plant contains 1768 lbs. to 232 of dry matter.

2. Analysis of the ash of the Drum-head cabbage, with a statement relative to its exhausting

		power.		
	Per centum.	Elements in a ton.	Elements removed from an acre.	Calculated without wa- ter, carbonic acid and organic matter.
Silica	0.550	0.550 lbs.	2.958 lbs.	0.767
Phosphoric acid	11.870	1.876	$63 \cdot 784$	16.544
Lime	2.698	0.426	$14 \cdot 484$	4.662
Magnesia	$3 \cdot 345$	0.529	17.986	1.533
Phosphate of perox. of i	ron, 1·100	$0 \cdot 174$	5.916	3.761
Potash	20.078	$4 \cdot 172$	$142 \cdot 448$	27.985
Soda	$30 \cdot 116$	4·75 8	$161 \cdot 772$	41.976
Sulphuric acid	10.449	1.651	$56 \cdot 134$	14.564
Chlorine	0.741	0.117	3.978	1.033
Organic acids	4.805			
Carbonic acid	$13 \cdot 500$			
				
	99·252 S	14.253	$469 \cdot 460$	$112 \cdot 825$

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3. Analysis of the large yellow Savoy cabbage.

Gardeners distinguish 3 sub-varieties: the Large yellow, the Dwarf, and the Green. Leaves crisped; stalks short and thick; weight from 4 to 10 lbs. Furnished by Mr. Huston.

				PRO	PORTIO	NS.			100.000 parts gave
Water	-	-	-	-	-	-	-	-	86.525
Dry	-	-	-	-	-	-	-	-	$13 \cdot 475$
$\mathbf{A}\mathbf{s}\mathbf{h}$	•	-	-	- ,		-	-	-	0.890
Ash cal	culat	ed dry	7 -	-	-	-	-	-	6.605 S.

The proximate organic elements obtained from this cabbage were as follows:

									Calculated without water.
Sugar, and t	heo	dorous	matte	er of c	abbag	ge -	-	8.60	46.386
Oil and wax	c -	-	-	-	-	-	-	0.26	1.402
Fibre, starch	h and	d chlor	ophyl	-	-	-	-	$2 \cdot 47$	$13 \cdot 323$
Pure fibre	-	-	•	-	-	-	-	$2 \cdot 41$	$12 \cdot 999$
Matter solu	ble	in pot	ash	-	-	-	-	1.31	$7 \cdot 066$
Albumen	-	•	-	-	-	-	-	$2 \cdot 05$	11.057
Casein -	-	-	-	-	-	-	-	0.40	$2 \cdot 158$
Dextrine	-	•	-	-	-	-	-	1.04	$5 \cdot 609$
Water -	-	-	-	-	-	-	-	81.23	
								99.77	100·000 S.

Composition of the ash of the Savoy cabbage.

									Amount of elements in a ton of dry cabbage.
Silica -	-	-	-	•	-	-	-	1.100	1.453 lbs.
Phosphoric ac	id	-		-	-	-	-	20.825	$27 \cdot 510$
Lime -		-	-	-	-	-	_	0.906	$1 \cdot 197$
Magnesia	•	-	•	-	-	-	-	4.760	6.288
Phosphate of	per	oxide	of ir	on	-	-	-	$1 \cdot 050$	1.387
Potash -	•	-	-	-	-	-	-	$17 \cdot 525$	$23 \cdot 151$
Soda -		-	-	-	-	-	-	$23 \cdot 366$	30.867
Sulphuric acid	1	-	-	-	-	-	-	15.811	20.886
Chlorine	-	-	-	-	-	-	-	1.974	2.608
Organic matte	er	-	-	-	-	-	-	$3 \cdot 260$	
J									
								90·577 S.	. 115:347

4. Analyses of the Red cabbage (Sub-var. rubra).

Gardeners also distinguish 3 sub-varieties: the Large red or Red Dutch, the Dwarf, and the Aberdeen red.

				PRO	PORTIO	NS.				
								1	00.000	parts gave
Water		-	-	-	•	-	-	- 8	87.915	. 0
Dry	-	-	-	-	-	-	-	-	12.085	
Ash	-	-	-	-	-	-	-	-	0.930	
Calcul	ated	dry	-	-	-	-	-	-	7.695	S.
				AN	ALYSI	s.				
									Calcul	lated without water.
and ex	tract	ive n	atter	-	-	-	-	8.70	8	$51 \cdot 479$
d wax		-	-	-	-	-	-	0.12		0.710
starch	and	color	ing m	atter	-	-	-	1.20		$7 \cdot 101$
ibre	•	-	-	-	-	-	-	2.98]	17 · 134
ved by	potas	sh (al	bumer	1)		-	-	0.19		$1 \cdot 124$
nen	-	•	-	-	-	-	•	1.83	1	10.828
ı •	-	-	•	-	-	-	-	0.72		4.260
ne	-	•	-	-	-	-	-	1.16		6.864
	•	-	-	-	-	-	-	83.35		
								100.25	S. 9	99.500
	Dry Ash Calcul and ex d wax starch	Ash - Calculated and extract d wax starch and fibre - ved by potas nen - in ine -	Dry Ash Calculated dry and extractive mad wax - starch and color fibre ved by potash (almen ine	Dry	Water	Water	Dry	Water	Water -	100 · 000 Water

With the same assumptions as in the first variety noticed, the produce of an acre will be 24.2 tons, allowing the heads to average 10 lbs. Of this quantity, 20.2 tons will be water, and 4 dry cabbage, of which a ton will contain 324 lbs. of nitrogenous matter, and only 54.8 lbs. in the green state. A ton of the green plant will contain 18.6 lbs. of inorganic matter; the dry, 153.9 lbs.; and an acre will be deprived of 446 lbs. according to these calculations.

The ash contains the following amount of elements in 100 parts:

								Elements in a ton of dry cabbage.
Silica	-	-	-	-	-	-	3.350	5·156 lbs.
Phosphoric acid	-	-	•	-	•	-	10.110	15.559
Lime	-	-	-	-	-	-	2.869	$4 \cdot 415$
Magnesia .	•	•	•	-	-	-	3.350	$5 \cdot 433$
Phosphate of pe	roxide	of i	ron	•	-	-	3.600	5.541
Potash	•	-	•	-	-	-	$20 \cdot 203$	31.093
Soda	-	-	-	-	-	-	$26 \cdot 937$	$41 \cdot 456$
Sulphuric acid	-	-	-	-	-	-	14.849	22.853
Chlorine -	-	-	•	-	-	-	0.987	1.519
Organic matter	-	-	-	-	-	-	5.050	
Carbonic acid	-	-	-	-	-	-	8.600	
							$100 \cdot 185$	113·025 S.

5. Analysis of the Cauliflower cabbage (Sub-var. botrytis).

From the garden of Mr. Douw of Greenbush.

PROPORTIONS.

								100.000 parts gave
Water	-	-	-	-	-	-	-	85.700
Dry -	-	-	-	-	-	-	-	$14 \cdot 300$
Ash -	-	-	-	-	-	-	-	1.520
Calculated of	lry	-	-	-	-	-	-	10·629 S.

Head consisting of flower buds and pedicels.

Sugar and	extrac	tive r	natter	, and	the	odoroi	us	
matter o	f the s	pecies		-	-	-	-	8.08
Fibre and	starch	-	-	-	-	•	-	0.63
Fibre -	-	-	-	-	-	-	-	1.76
Wax and c	hloropl	nyl	-	-	•	-	-	$0 \cdot 34$
Soluble in	potash	•	-	-	-	-	-	$2 \cdot 19$
Albumen	-	-	-	-	-	-	-	0.94
Casein -	-	-	•	-	-	-	-	0.05
Dextrine	-	-	_	-	-	-	-	1.74
Water -	-	-	-	-	-	•	-	85.70
								101·43 S.

Composition of the ash.

								A ton of dry cabbage contains.
Silica	-	-	-	•	-	-	1.25	2.652 lbs.
Phosphoric acid	and per	roxide	e of ir	on	-	-	$28 \cdot 65$	$60 \cdot 904$
Lime	-	-	-	-	-	-	0.73	$1\cdot 542$
Magnesia -	-	-	-	-	-	-	2.16	$4 \cdot 592$
Potash	-	-	-	-	-	-	18.41	$39 \cdot 136$
Soda	-	-	-	-	-	-	$33 \cdot 44$	71.087
Sodium	-	-		-	-	-	0.20	0.425
Sulphuric acid	-	-	-	-	-	-	$9 \cdot 79$	20.812
Chlorine -	-	-	-	-	-	-	0.30	0.638
Organic matter	-	-	-	-	-	-	$3 \cdot 05$	
Carbonic acid -	-	-	-	-	-	•	1.85	
							99.83	201·788 S.

6. Examination of the Turnip-rooted cabbage (Sub-var. caulo-rapa). Furnished by Mr. Douw.

			PRO	PORTION	is.			
								100.000 parts gave
Water -	-	-	-	-	-	-	-	$91 \cdot 140$
Dry -	-	_	-	-	-	-	-	8.860
Ash -	-	-	-	-	-	-	-	0.805
Calculated d	lry	-	-	-	-	-	-	9·086 S.
			AN	ALYSI	s.			
Sugar and e	extrac	tive	matte	er, w	ith th	e odor	ous	
matter of	cabba	age	-	-	-	•	-	4.880
Starch and f	ibre	-	-	-	-	-	-	0.285
Fibre -	-	-	-	-	-	-	-	1.560
Soluble in p	otash	-	-	-	-		-	0.650
Albumen	-	-	-	-	-	-	-	0.655
Casein -	-	-	-	-	-	-	-	0.570
Dextrine	-		-	_	-	-	-	1.190
Water -	-	-	-	-	-	-	-	$91 \cdot 140$
								100·930 S.

Composition of the ash.

		- 0	pocc	200.00	0.00			
							El	ements in a ton of dry cabbage.
Silica	-	-	-	-	-	-	0.675	1.228 lbs.
Phosphoric acid	and p	eroxid	le of	iron	-	-	16.925	30.802
Lime	•	-	-	-	-	-	3.380	$6 \cdot 156$
Magnesia -	-	-	-	-	-	-	1.875	3.410
Potash	-	-	-	-	-	-	21.575	$39 \cdot 266$
Soda	-	-	-	-	-	-	27.840	53.668
Sodium -	-	-	-	-	-	-	1.395	2.538
Sulphuric acid	-	-	-	-	-	-	9.970	18.145
Chlorine -	-	-	-	-	-	-	$2 \cdot 120$	$2 \cdot 538$
Organic matter	-	-	-	-	-	-	$3 \cdot 525$	
Carbonic acid	-	-	-	-	-	-	8.850	
							$98 \cdot 130$	157·751 S.

I commenced an examination of the turnip, its top or leaves and root, but it remains unfinished from want of time. The leaves contain in 500 parts 85.92 of dry matter and 415.08 water, or 17.18 per centum of dry matter; ash 13.69, or 2.73 per centum of inorganic matter. Of this 13.69, 3.69 is silica, 2.35 earthy phosphates, 1.25 carbonate of lime, .615 magnesia, 1.26 potash, and 3.73 soda. The composition was influenced by an unmanured soil, and by a small quantity of adherent foreign matter.

7. A complete analysis of the ash of turnip tops, according to Mr. R. M'CALMONT, furnishes the following result:

Silica	-	-	-	•	_	-	1.39
Phosphoric acid	-	•	-	-	-	-	$9 \cdot 22$
Lime	-	-	-	-	-	-	$22 \cdot 98$
Magnesia -	-	-	-	-	-	-	2.87
Oxide of iron -	-	-	-	-	-	-	0.42
Potash	-	-	-	-	-	-	$16 \cdot 42$
Chloride of potassi	um	-	-	-	-	~	9.61
Chloride of sodium		-	-	-	-	-	$6 \cdot 65$
Sulphuric acid	-	-		-	-	_	16.31
Carbonic acid	-	-	-	-	-	-	13.18
							$99 \cdot 05$

Turnip tops furnish a food for horses and cattle, superior to the root. Whether the tops give an unpleasant flavor to the milk of cows, I am not particularly informed.

To the husbandry of New-York, turnips are not so essential as they are in England: that they constitute a valuable addition to the resources of the farmer is well established, even if it were for no other purpose than to furnish a variety to the food of stock.

The turnip root gave the following proportions:

Dry	_				_		_	_	1000.00 parts gave 97.30
•	-	-	•	-	-	_	-	-	
$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	-	$7 \cdot 50$
		8.	Anal	ysis q	f the	Suga	ar bee	t.	
									1000.00 parts gave
Dry	-	-	-	-	-	-	-	-	132.71
$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	•	-	-	-	•	$5 \cdot 00 *$

^{*}Journal of Agriculture, and the Transactions of the Highland and Agricultural Society of Scotland, for October, 1847.

II. THE SQUASH TRIBE.

1. Analyses of two varieties of Squash and one of Pumpkin.

						PR	PORTION	īs.				Don continu
1.	Vegetable	Mari	row,	Boston	Mar	row	(Pl. 4,	fig.	1),	1000 · 000	grs.	Per centum
	Dry	-	-	-	-	-	•	•	-	$124 \cdot 905$	_	$12 \cdot 490$
	Water	-	-	-	-	-	-	-	-	$875 \cdot 095$		
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	•	•	-	-		7.220		0.722
2.	Custard s	quash	(Pl.	6, fig.	. 1)	-	-	-	•	1000.000	grs.	
	Dry	-	-	•	-	-	-	-	-	$76 \cdot 760$		
	Water		-	-	-	-	-	-	-	$923 \cdot 240$		$92 \cdot 320$
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-		-	-	-	-	-	-	4.875		0.487
				-49						¥		
3.	Pumpkin		-	-	-	-	-	-	-	1000.000	grs.	
	Dry	-	-	-	-	•	-	-	-	70.100		
	Water	-	•	•	-	-	-	-	-	$929 \cdot 900$		$92 \cdot 990$
	$\mathbf{A}\mathbf{s}\mathbf{h}$		-	-		-	-	-	-	$5 \cdot 490$		0.549

The amount of water in the three foregoing vegetables, as will be seen, is large, a fact with which all who observe are familiar. The amount of ash is small; and its constitution is not very unlike that of the cereals and of the edible tuberous plants, except that soda is in a great excess over the potash, if a single analysis is sufficient for establishing a fact.

The squash and the pumpkin, like all other domesticated plants, form numerous varieties, which differ in external form and color; and what is far more interesting, is their differences in composition also; differences, which, though they may not separate the varieties far from each other, yet are so considerable that analysis readily discovers the fact. Each seems to possess its own powers of appropriating nutriment, in a manner which has been already pointed out in regard to varieties of maize growing upon the same ear. This fact may have been acknowledged in some instances, yet I am not aware that it has been very distinctly asserted before, and upon a sufficient authority, or based upon well ascertained facts. Neither has it been stated that a great similarity exists between the edible tubers, the cereals, and the esculents, those which are so highly esteemed and in almost universal use.

2. Analysis of the ash of the Vegetable Marrow squash.

Silica	-	-	-	•	-	-	2.000
Coal	-	-	-	-	•	-	0.020
Earthy phosphate	es -	-	-	-	-	-	31.200
Alkaline phosph	ates	-	-	-	-	-	3.300
Lime	-	-	-	-	-	-	0.200
Magnesia -	-	-	-	-	-	-	0.001
Potash	-	-	-	-	-	-	8.980
Soda	-	-	-	-	-	-	$33 \cdot 140$
Sulphuric acid	-	-	-	-	-	-	5.510
Chlorine -	-	-	-	-	-	-	$6 \cdot 160$
Organic matter	-	-	-	-	-	•	$3 \cdot 140$
Carbonic acid	-	-	-	-	-	-	3.780
							0.00
							$97 \cdot 620$

3. Proximate organic analysis of the Vegetable Marrow squash.

Quantity		-	-	-	-	-	-	-	200.000 grs.
Starch -		-	-	-	-	-	-	-	1.594
Fat -		-	-	-	-	-	-	-	0.206
Fibre -		-	-	-	-	-	-	-	6.855
Albumen			-	-	-	-	-	-	2.730
Dextrine		-	-	-	-	-	-	-	$2 \cdot 390$
Sugar -	٠	-	•	-	-	-	-	-	$15 \cdot 400$
Water -		-	-	-	-	-	-	-	$175 \cdot 020$
									199 • 191

4. Proximate organic analysis of the Butter squash.

Starch -		-	-	-	-	-	-	none.
Fat -	-	-	-	-	-	-	unde	termined.
Fibre -	-	-		-	-	-		6.065
Albumen	-	-	-	-		_	-	2.215
Dextrine	-	-	-	-	-	-		0.250
Sugar and	extract	-	-	-	-	-	-	23.380
Water -	-	-	-	-	-	-	-	$189 \cdot 660$
								200 520
								200.520

In the product sugar and extract, it was sensible to the taste that the Butter squash was less sweet than the Vegetable Marrow. In cooking, and in analysis, the former is superior to the latter.

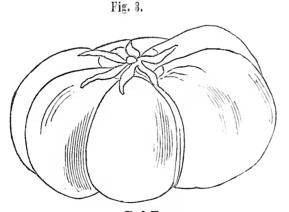
5. Proximate organic analysis of the fruit of the Egg plant (Pl. 7).

Sweet matter, extract, and a small quantity of											
a bitter p	rincip	le	-	-	-	-	-	$3 \cdot 040$			
Starch with	a littl	e ch	lorop	hyl	-	•	-	0.365			
Albumen	-	•	•	-	-	-	-	0.255			
Casein -	-	-	-	-	-	-	-	9.200			
Dextrine	-	-	-	-	-	-	-	0.370			
Fibre -	-	-	-	-	•	-	-	0.760			
Matter disso	lved o	ut of	fibre	byav	veak	solut	ion				
of caustic	potasł	ı -	-	-	-	-	-	0.975			
Water -	•	-	-	-	-	-	-	$91 \cdot 353$			
								97·318 S.			
			PROP	ORTION	s.						
Water -	-	-	-	-	-	-	-	$91 \cdot 353$			
Dry matter	-	-	-	-	-	-	-	8.647			
Ash -	-	-	-	-	-	-	-	0.604			
Ash calculat	ed on	the	dry n	natter	-	-	•	6.981 S.			

The Egg plant (Solanum melongena) is a valuable species, bearing a large purple fruit upon a prickly stem. It is fleshy, though rather watery; yet when fried in butter, it resembles very tender meat in flavor. It is a delicacy which is far less common than might be expected. A single berry sells for one shilling. The plant is an annual, herbaceous and branching: it grows two feet high. It may be prepared for food in various ways, and is considered as wholesome and delicious. Like the tomato, it is cultivated from the seed, which must be sown early in a warm dry mellow soil.

The Tomato (Solanum lycopersicum) has been noticed already in connection with the potato. This species has increased greatly in esteem during the last ten years. It is now

38

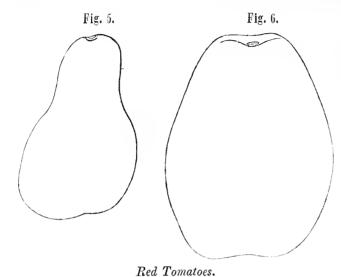


Red Tomato.
[AGRICULTURAL REPORT — Vol. II.]



Fig. 4.

Yellow Tomato.

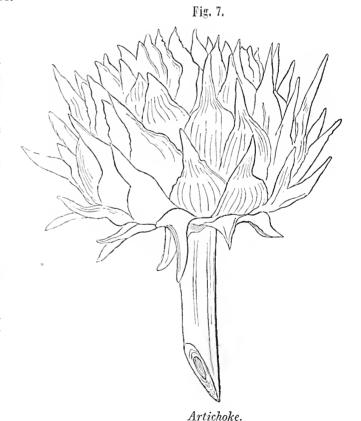


constant in the markets, and sells for a high price. The plant bears a large amount of fruit when well manured. The fruit is at first of a green color, and becomes red or yellow when ripe. It runs into numerous varieties by cultivation: the large red-lobed or torulose variety is one of the best. Some of the varieties are figured as in figs. 3, 4, 5, 6. The small round yellow fig. 4 is a beautiful berry, and is less acid than the large fig. 3.

The tomato is eaten in various ways, as raw and mixed with sugar, or fried, etc. Added to soup in pro-

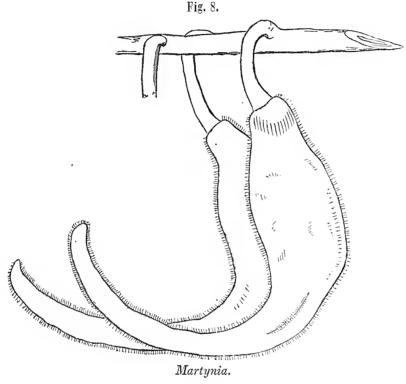
per quantities, it greatly improves it. It is also pickled. It may be dried in the sun, or in an oven, and kept for winter's use.

The Artichoke, fig. 7 (Cynara colymus), takes its name from the stiff hard spines of the head (which is much like a thistle), and which resemble the teeth of a dog. As spiny as the head appears, it is still regarded with favor, and is prepared for the table. The fleshy receptacle and interior part of the plant, and the upper part of the stalk, on being boiled and well buttered, furnish a very savory dish. It is a native of Europe, but has become naturalized here. It is cultivated from suckers planted in rows, and about three feet apart. It flowers in September or late in August.



MARTYNIA. 299

The Martynia (fig. 8) is a strong-smelling vegetable, the capsules and calyx of which are prolonged into thick and fleshy hooks. In its green state, these parts are esteemed for pickles: when too old, or when the seed has ripened, it becomes woody, tough, and useless. It serves to increase the variety of condiments, and may become of some importance, inasmuch as the plant is very easy to cultivate, and is frequently found growing spontaneously by roadsides.

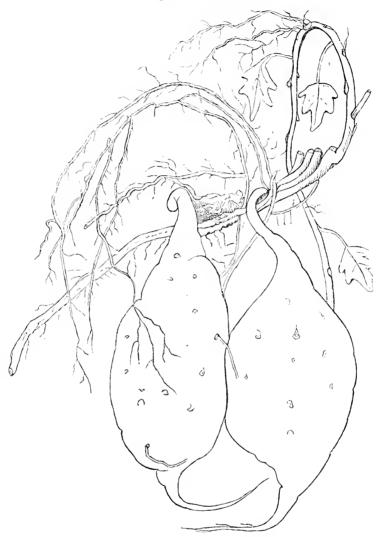


III. THE SWEET POTATO.

The family to which the Sweet potato belongs is more or less poisonous, from the presence of an acrid or milky juice. When this is dissipated by heat or drying, the fleshy tuberous part is not only harmless, but constitutes an agreeable and nutritious food, in which starch is one of the most abundant constituents. Jalap and scammony, drastic purgatives, belong to the same genus as the Sweet potato. The most valuable species of this family all belong to the tropics; and none, I believe, are found within the frigid zone. With care and proper management, the Sweet potato may be grown for the table in the vicinity of Albany. In New-Jersey they are cultivated with great success, and supply the demands in a great measure of the markets of New-York, and the cities and villages upon the Hudson river.

The foliage of the Sweet potato is a slender, leafy, climbing vine: the roots or tubers appear large, and out of proportion to the foliage.





ANALYSES OF THE SWEET POTATO.

Cut when ripe, and furnished by Mr. SEARS, of the North American Phalanx, N. Y. Soil sandy and light.

	Percentage.										
1.	\mathbf{T} OP	-	-	-	-	-	-	-	-	1065.400 grs.	9
	\mathbf{Dry}	-	-	-	to.	=	ta		-	335.000	
	Water	-	-	-	•	-	-	-	-		$31 \cdot 443$
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-		-	-	6.840	2.042

											Percentag	c.
2. Ste	M	-	-	-	-	-	•	•	-	100.000	grs.	
D	ry	•	-	-	-	-	-	-	•	21.800		
	ater		_			-	-	-	-	78.200		
	sh	-	-	-	-		•		-	1.735		
3. Lea		•	-	-	-	-		-	•	100.000		
	ry	-	-	-	-	•	-	-	-	20.400		
W	Tater	-	-		-	-	-	-	-	79.600		
. A	sh	•	-	•	-	-	-	•	-	2.630	2.63	3 0
4. Tu	7F76 •											
	putsid									658.000		
		е	•	•	۰	•	-			175.200		1
	ry	•		-	-	•	•					
	ater	-	-	-	-	-	-	-	-	482.800		
A	sh	•	-	-	-	-	-	•	-	3.380	1.93	4
2. I	nside	_		_	-	-	-	-	-	878.000		
\mathbf{D}_{1}	rv	-	_		_	-	-	-	-	$264 \cdot 000$		
	ater	-		-	-			-	-	$614 \cdot 000$		
	sh	_	_	-		-	-	10	-	3.650	0.41	5
		lculat	ed dr	v			•		-		1.37	
	ry ma		•	<i>J</i>	-		_		-		30 · 12	
	•										35 2.5	
	Iiddle	e	-	-	-	-	-	-	-	1950.000		
	ry	-	-	-			-	•	-	604.000		
W	ater	•	-	-	-	-	•	•	-	$1346 \cdot 000$		
\mathbf{A}	sh	•	-	-	-	-	-	-	-	8.650	0.44	.3
\mathbf{A}	sh ca	lculat	ed dr	y		•	-	•	-		1.43	2
D	ry ma	atter	-	_	-	-	-	-	-		$30 \cdot 97$	4
4 7	·									2150 000		
	Botton	n	-	-	-	•	-	-	-	2158.000		
	ry	-	-	-	•	•	•	**	-	504.000		
	ater	•	-	-	-	-	•	•	-	1654.000		
As		•	•		-	-	-	-	-	13.220	0.61	
		lculat	ed d	ry	-	•	-	-	~		$2 \cdot 62$	
\mathbf{D}_{1}	ry ma	ıtter	-	-	-	-	-	•	-		$23 \cdot 35$	4
				1. A	lnalys	sis of	the l	eaves	and	stems.		
		Silica		-	-	-	-	-	-		23.600	
		Earth	y ph	ospha	tes	-	-	_	-		28 · 575	
		Carbo									15.000	
		Magn					-		_	•	none.	
		Potas			-	-	-		_		18·515	
		Soda			_	_		_	_	-	9.460	
		Sulph	uric	acid	_	_	-	_	_	· -	2.785	
		Chlor			_	_	_	_	_	_	2.090	
		J.1101		-	-	-		-		_	~ 000	
										1	$00 \cdot 025$	

2. Analysis of the tubers.

Silica	-	•	•	-		-	1.850
Earthy phosph	ates	-	-	-	-	-	21.100
Carbonate of li	me -	-	-	-	-	-	0.600
Magnesia -	-	-	-	-	-	-	0.500
Potash	-	-	-	-	-	-	$48 \cdot 365$
Soda	-	-	-	•	-	-	$5 \cdot 025$
Sulphuric acid	-	-	-	-	-	-	1.200
Chlorine -	-	-	-		-	-	4.090
Organic matter	-		-	-	-	-	$2 \cdot 456$
Carbonic acid	-	-	-	-	•	-	15.725
							00.011
							98.911

It will be seen, by comparing the amount of potash in the Sweet potato with that in the common potato, that a very close resemblance exists between them in the amount of potash. There is also the same close resemblance in the percentage of the ash; but it is proper to remark that the percentage of ash may be rather low, on account of its easy fusibility and its liability to causticity.

3. Proximate organic analysis of the Sweet potato.

Starch -	-	-	-	-	•	-	-	$19 \cdot 955$	
Sugar and	extrac	tive n	natte	r, mos	stly su	gar	-	5 ·800	
Dextrine	-	-	-	-	-	-	-	0.750	
Fibre after	boiling	g in a	wea	k solu	ition o	f cau	stic		
potash	-	-	-	-	-	-	-	1.850	
Matter dis	solved	out of	fibre	e by a	weak	solu	tion		
of caust	ic potas	sh	-	-	-	-	-	2.100	
Albumen	-	-	-	-	-	-	-	5.900	
Casein -	-	-	-	-	-	°=	-	1.050	
A body the	at rese	mbles	bal	sam	-	-	-	0.225	
Water -	-	-	-	-	-	-	-	$69 \cdot 515$	
								101.835	S.
			PRO	PORTIC	NS.			•	
								Per centum.	
Water -	-	-	-	-	-	•	-	$69 \cdot 515$	
Dry matter	r -	-	•	-	-	-	-	$30 \cdot 485$	
Ash -	-	-	-	-	•	-	-	1.090	
Ash calcul	ated or	the o	dry n	natter	-	-	-	$3 \cdot 247$	S.

CHAPTER VII.

MISCELLANEOUS ANALYSES.

ANALYSES OF FLAX, FLAXSEED, AND HEMP.

				PRO	PORTION	ls.			
1. Sun-da	ried	flax	-	_	-	-	-	-	100 · 00 grs.
Dry	-	-	-	-	-	-	-	-	$91 \cdot 42$
Water	-	-	-	-	-	-	-	-	8.58
Ash	-	-	-	-	-	-	-	-	1.80
2. Flaxs	eed	-	<u>:</u>	-	-	-	-	-	100.00
\mathbf{Dry}	-	-	-	-	-	-	-	-	$92 \cdot 72$
Water	-	-	-	-	-	-	-	-	$7 \cdot 28$
Ash	-	-	-	-	-	-	-	-	3.70

1. Analysis of the straw and fibre.

Soil based upon Taconic slate, and formed of round and slaty gravel and clay loam.

•	•					• 0	•	
						Г	Removed in a ton of straw.	
Silica	-	-	-	-	-	7.500	$2 \cdot 272 \text{ lbs.}$	
Earthy phosphates		-	-	-	-	$23 \cdot 225$	$7 \cdot 037$	
Carbonate of lime	-	-	-	-	-	$25 \cdot 400$	$7 \cdot 696$	
Magnesia	-	-	-	-	-	2.680	0.812	
Potash	-	-	-	10	-	25.870	7.838	
Soda	-	-	-	-	-	7.835	$2 \cdot 374$	
Sulphuric acid -	-	-	-	-	-	$3 \cdot 440$	1.042	
Chlorine	-	-	-	-	-	0.440	0.133	
Organic matter -	_	-	-	-	-	2.130		
3								
						98.530	$29 \cdot 204$	
	2.	$\mathcal{A}nal$	ysis c	of flas	cseed.			
		·				Ren	noved in every 100 lbs. of see	eđ.
Silica	-	-	-	-	-	8.00	0.666	
Earthy phosphates	-	-	-	-	-	47.50	1.757	
Carbonate of lime	-	•	-	-	-	0.20	0.007	
Magnesia	-	-	-	-	-	0.10	0.003	
Potash	-	-	-	-	-	18.09	0.669	
Chloride of sodium	-	-	-	-	-	20.09	$0 \cdot 743$	
Sulphuric acid -	-	-	-	-	-	7.56	0.279	
Chlorine	-	-	-	-	-	trace.		

101.54

 $4 \cdot 124$

Flax and hemp are exhausting crops: the earthy phosphates, lime and the alkalies, form important elements in the plants.

3. Analysis of hemp.

Silica -	-	-	-	-	-	-	-	$12 \cdot 00$
Earthy phos	phates	-	-	-	-	-	-	36.50
Carbonate o	f lime	-	-	-	-	-	-	$25 \cdot 90$
Magnesia	-	-	-	-	-	-	-	4.50
Chloride of	sodiu	n	-	-	-	-	-	1.00
Potash -	-	-	-	-	-	-	-	15.00
Soda -	-	-	-	-	-	-	-	$3 \cdot 30$
Sulphuric ac	cid	-	-	-	-	-	-	1.50
_								
								99.70

From the foregoing analyses, it is evident that both flax and hemp are exhausting crops. They have too the disadvantage in cultivation, that usually nothing is returned to the soil. The ash of the straw might, however, be scattered over the soil, though it is rarely if ever done.

ANALYSIS OF SPEARMINT.

Stem and leaves nearly dry.

			PRO	PORTION	s.			
Dry plan	t -	-	-	-	-	-	-	$100 \cdot 00$
Dry -	-	-	-	-	-	-	-	98.00
Ash -	-	-	-	-	-	-	-	$2 \cdot 72$
			AN	ALYSI	s.			
Silica -	-	-	-	-	-	-	-	$13 \cdot 00$
Earthy p	hosphate	es	-	-	-	-	-	$24 \cdot 15$
Carbonat	e of lim	ie .	-	-	-	-	-	$23 \cdot 50$
Magnesia		-	-	-	•	-	-	0.62
Potash -	-	-	-	-	-	-	-	$24 \cdot 20$
Soda -	-	-	-	-	-	-	-	$3 \cdot 40$
Chlorine	-	-	-	-	-	-	-	$4 \cdot 20$
Sulphuric	c acid	-	-	-	-	-	-	$3 \cdot 40$
Organic i	matter	-	-	-	-	-	-	1.50
								97.97

The herbage of spearmint and peppermint, as is well known, is used for the oil they contain. The latter is cultivated extensively in some parts of the country. The spontaneous growth of the former is depended upon mostly for its oil. Both are exhausting crops.

EXAMINATION OF THE YELLOW DOCK (Rumex crispus).

This plant, which is regarded as a troublesome weed, has some valuable properties which make it worthy of a passing notice in this place. My assistant, Mr. Salisbury, has given a very full account of its properties and composition in the American Journal of Agriculture and Science, which the reader may consult.

The plant, as usual, was divided into several parts; the leaves, seed, stem and root.

The organic composition of the leaves, when in their mature state, is as follows:

Starch -	-	•	-	-	-	-	-	none.
Fibre with a	little	chlore	phyl	•	-		-	1.755
Albumen	-	•	•	-	•	-	-	0.030
Casein -	-	-	-	-	-	-		0.200
Dextrine	-	-	-	-	-	-	-	0.920
Bitter extract	-	-	-	-	-	-	-	$3 \cdot 570$
Lignin contai	ning	ash 8	$\cdot 053$	•	-	-	-	$9 \cdot 685$
Dry matter	•	•	-	-	-	-	-	$16 \cdot 160$
Water -	•	-	-	•	-	-	-	83.680
								99·680 S.

The petioles contain nearly one per centum of free oxalic acid.

The ash of the leaves is composed of the following elements:

						,		
Carbonic ac	cid	-	-	-	•	-	-	$12 \cdot 400$
Silica 🖫	-	-	-	-	-	•	•	3.900
Earthy pho	sphat	es	-	-	-	-	-	24.000
Lime -	•	-	-	•	•	-	-	1.633
Magnesia	-	-	-	-	-	-	-	0.880
Potash -	-	-	-	-	-	-	-	10.613
Soda -	-	-	-	-	-	-	-	22.880
Sodium	-	-	-	-	-	-	-	3.889
Chlorine	-	-	-	-	-		•	5.920
Sulphuric a	icid	-	-	-	-	-	-	$1 \cdot 477$
Organic aci	id	-	•	-	-	-		8.500
_								
								96·092 S.

The	stalks	gave	the	following	result:
-----	--------	------	-----	-----------	---------

g				•					
	Carbonic aci	id	a	-	-	-	-	-	12.800
	Silica -	-	-	-	-	-	-	-	2.800
	Earthy phosphates (containing phosphoric acid								
	$14 \cdot 747)$	-	_	-	-	-	-	-	21.700
	Lime -	•	-	-	-	-	-	-	2.590
	Magnesia		-	-	-	-	-	-	$2 \cdot 040$
	Potash -	-	•	-	-	-	-	•	$12 \cdot 430$
	Soda -	-			-	-	-		$19 \cdot 753$
	Sodium	•	-	-	-	-	-	-	$3 \cdot 368$
	Chlorine		-	-	-	-	-	-	5.085
	Organic aci	ds		-		-	-	-	9.200
									96·896 S.
	1	Proport	tions	of wa	iter,	etc. in	the s	talk.	
	Water -	_	-	•		-	-	•	83.800
	Dry -	-		-	-	•	-	-	$16 \cdot 200$
	Ash -			-		-	-	-	1.480
	Calculated of	lrv	_	-		-	-	~	9.512
		•							
		oxima	te or	ganie	ana	lysis e	of the	e root.	
	Starch -	-	-	-	-	-	-	-	5.987
	Albumen	• `	•	-	-	-	-	-	1.431
	Casein -	•	-	•	-	-	-	-	0.226
	Dextrine	•	•	•	-	•	-	-	$2 \cdot 024$
	Bitter extra	ct (con	taini	ng ox	alic a	nd ta	nnic a	acids)	$5 \cdot 184$
	Fibre (containing ash 1.998)								$14 \cdot 228$
	Total of dry matter								29.080
	Water -			-	•	-		-	70.320
	** 4,01								
									99·400 S.
The ash was f	ound to con	tain :							
	Carbonic ac	eid	-	-	-	-	-	-	11.000
	Silica -	-	-	-	-	-	-	•	0.500
Earthy phosphates (containing 17.38 phosphoric									
	acid)	-	-	•	-	•	-	-	$26 \cdot 200$
	Lime -	-	-	•	-	-	-	-	3.827
	Magnesia	•	•	-	-	-	-	-	7.620
	Potash -	-	•	•	-	-	•	-	$9 \cdot 723$
	Soda -	-	-	•		-	-	-	18.480
	Sulphuric a	cid	-	-	-	-	-	-	4.502
	Chlorine		-	-	•	-		-	3.898
	Organic ma	tter	-	-	•	-	•	-	11.600
	J								0
									97·460 S.

ANALYSIS OF THE ASH OF THE COFFEE BEAN, AND OF THE LEAVES OF GREEN TEA.

٠										Coffee.	Tea.
Ť	Silica	-	-	-	-	-	-	-	-	15.625	16.380
	Earthy	pho	spha	tes	-	-	-	-	-	33.700	$42 \cdot 912$
	Carbon	nate	of lir	ne	-	-	-	-	-	3.300	1.280
	Magne	sia	-	-	-	-	-	-	-	0.750	0.582
	Potash	٠-	-	-	-	-	-	-	-	$35 \cdot 800$	22.850
	Soda	-	-	-	-	-	-	-	-	$4 \cdot 445$	8.112
	Sulphy	aric a	icid	-	-	-	-	-	-	5.650	$5 \cdot 275$
	Chlori	ne	-	_	-	-	-	-	-	none.	trace.
	Organi	ic ma	atter	-	-	-	-	-	-	0.350	1.768
										·	-
										99.580	$98 \cdot 159$
					•	P	ROPORT	ions.			
										Coffee.	Tea.
	Dry	-	-	-	-	-	-	-	-	800.00	$250 \cdot 000$
	$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	-	-	-	-	-	25.88	$12 \cdot 125$

Coffee and tea appear to be exhausting crops, both the bean and leaf being exported, and hence entirely lost to the soil upon which they grew.

CHAPTER VIII.

THE FRUIT AND FOREST TREES.

OF THE COMPOSITION OF THE ASH OF FRUIT AND FOREST TREES, WITH REMARKS ON THE DISTRIBUTION OF THE ELEMENTS IN THE PARTS AND ORGANS COMPOSING THEM.

The fact ascertained by Saussure, that plants require a certain amount of inorganic matter for the perfection of their foliage and seed, has made the subject of inquiries into the chemical constitution of vegetables extremely interesting and important. This fact increases in interest in consequence of another fact well ascertained, namely, that plants not only require a certain amount of matter usually denominated inorganic, but that this amount differs essentially with the different families, and even with the different species of plants. So it appears too that different parts and organs of the same plant require different amounts of the elements composing the ash. The seed and fruit, the root and trunk, the leaves and branches, each secure for themselves different quantities of what we may call the solid food. Even the seasons of the year are marked in the life of the vegetable by different proportions of inorganic matter. The vegetable system goes on accumulating inorganic matter slowly, until it has collected a store of food for future consumption.

I shall not attempt to post up what is known upon the subject before us; my object being to state generally the results of my investigations, and those which have been made by my assistant during the Survey. The analysis of the ash of forest and fruit trees has been prosecuted during a part of the last three years: more than a year of constant labor by myself and assistant has been thus bestowed. The analyses, it is proper to say, have not been made with the sole object of ascertaining the elements, or the proportion of elements, which enter into the constitution of plants; but a higher object has been in view, namely, the determination of the law of distribution of the elements. That a law should control their distribution appeared to be highly probable from analogy; and it may be inferred that the ends of existence are better subserved under its control, than if their distribution had been left to accident.

I now proceed to state that it was early conceived that the inorganic matter, in the growth of plants, was constantly determined by two movements: first, an upward movement, which really constitutes the circulation; and, secondly, an outward movement, by which the matters are transferred from the centre to the circumference, which movement results in a diminution of the inorganic matter of the interior, an effect which constantly

carries outwards certain elements which are of greater importance than others. I allude here particularly to the phosphates, which, it is well known, are of primary importance in the fruit and seeds. The analyses have been conducted in part with reference to this supposed law of outward movement. The plan adopted, and which has been generally followed, was to divide the tree into the following parts: 1, the inside wood, or heart wood; 2, the outside wood; 3, the bark; 4, the wood of the twigs; 5, the bark of the twigs; 6, the leaves; 7, the seed, and the fruit with its envelopes. This plan of proceeding appeared to exhaust the subject, unless indeed we extend it so far as to inquire what changes take place in the amount of the elements in the different seasons. A few analyses have been made under this inquiry. The subject is of sufficient importance to merit a careful investigation.

The proper preparation of the ash is an operation that has been attended with some trouble. It is very liable to become caustic, or sub-caustic, and hence it happens that sufficient carbonic acid is never obtained to saturate the bases. This is not a matter of so much consequence as might at first appear: carbonic acid is not regarded as a constituent of the wood, but as a secondary product formed from the organic acid by ignition; and the object in determining the amount of carbonic acid, is to test the accuracy of the analysis. The ash should be prepared at as low a temperature as possible, which shall at the same time secure a perfect combustion of the organic matter.

STATEMENT OF SOME RESULTS OF THE FOLLOWING ANALYSES.

I. It appears as a general result of the following analyses, that solubility controls in part the distribution of the inorganic matter of vegetables. Water never accumulates in the interior, or in the heart wood of a tree, but is carried upwards and ontwards, and hence is found in a greater proportion in the leaves and in the outside wood. Being itself the great medium through which the solid matters are conveyed and distributed throughout the organs of vegetables, it is agreeable to what we might deduce a priori in regard to the distribution of the several proportional amounts of the inorganic matter. Moving as it is found to do, in a larger quantity to the periphery of the head and trunk, it must necessarily carry to the periphery a greater amount of solids than to the interior.

II. The distribution is not wholly controlled by the present movements of the water or sap which enters into the circulation of a tree. Some of the materials which form the wood of the interior is also absorbed and carried outwards to the periphery, or to the trunk. This statement is borne out by the results which have been so often obtained. The amount of ash of green and living wood is almost uniformly greater in the outside wood than in the heart wood. These two kinds of wood being weighed immediately after the tree is cut down, and equal portions of each taken, notwithstanding the water of the outside usually surpasses that of the inside, its ash exceeds that of the inside.

III. The bark of trees, as well as the outside covering of seeds, differs essentially in

constitution, as will be shown in the succeeding analyses, from the wood. In this part, the phosphates are deficient: although they are always present, still there is a great diminution in their amount, and to this statement an exception is not known. The bark, in its chemical constitution, is composed mainly of lime, probably an organic salt of lime, which by ignition is converted into a carbonate. There is also a great deficiency of potash and soda: they are equally small in amount with the phosphates. One or two curious facts, however, are observed in regard to the alkalies: it was found, for instance, that the outside and apparently lifeless part of the bark (the corky part of elm bark, for example) was richer in potash than the inside bark. This may be, and probably is, an isolated fact. It is well known to common observers that the elm is particularly rich in potash, which pervades all its parts and organs more generally than in other trees.

The final cause of the accumulation of lime in the bark, is to serve as an outward defence of the internal or young wood: it secures a firmer covering than can be provided by the other elements. In this respect there is a very close analogy with the formation of the outer covering of the lower animals, which, as is well known, consists mainly of carbonate of lime, as in the crustacea and annulose animals. As in the animal tissues, outward defences are set up, so in the vegetable they are not left wanting. Here too a provision is furnished, by which this great amount of lime is returned to the soil. The bark is annually detached, and its store of soluble or partially soluble salts of lime is rapidly returned to the soil. An aged oak may have regenerated its bark repeatedly from its own debris: its outward covering, which is partially renewed every year, may derive a portion of its supply from what previously formed part of its own organism.

Upon this general fact is founded the law of distribution, the inorganic matters respectively being determined to the periphery of the head and trunk. The few exceptions which have been found, and which militate against this law, can hardly be regarded as sufficiently numerous and important to overthrow it.

Of the substances which appear to be more steadily and uniformly determined to the outside wood, I may state the following, leaving out of the account water, which acts as the medium through which every thing is conveyed into and throughout the tree.

The phosphates abound more in the periphery than in the centre: they are found also in larger quantities in the young wood of the twigs; therefore both the outside of the tree, and its growing branches, store up for the time being a larger amount of these essential elements than the interior. It is by this mode of distribution that the seeds and fruit receive their needful supply of phosphates: at least the supply is derived more immediately and directly, than if the contrary mode of distribution prevailed.

The final cause of this distribution of the phosphates, is to restore again to the soil those important elements. They pass along the outside wood of the trunk, obtaining probably an accession from the interior by the outward or lateral movement of the sap: they then pass to the seed and fruit, through the small branches, twigs and leaves. All the surplus amount of the phosphates in the leaves is restored to the earth by their fall and decay; and, besides, in a multitude of instances, the seed and fruit also return by their maceration

and decay, or in consequence of their constituting the food of animals. The phosphates, in this case, after performing their several offices in the system of animals, return to the soil from whence they were originally derived.

Another explanation, which may appear more plausible, is, that after a certain period in the life of the oak, or any tree, a greater amount of inorganic matter may accumulate, than at any previous period, in consequence of the spreading of the roots, and the subsequent increased supply of this kind of nutriment from the original sources. In its full maturity and strength, it takes up from the soil a greater amount of inorganic matter, which it is then more capable of assimilating or converting into wood.

The quantity, however, of inorganic matter may vary with the season. There may be an accumulation of the phosphates and some other substances in autumn, which is designed to go on slowly during the winter, by which a sufficiency of these elements is secured for the immediate use of the tree in the spring when the leaves are about to be developed. This undoubtedly would take place in the periphery mainly, or newer wood. From some of the analyses which were made of winter cut wood, there appears a larger amount of phosphates in the wood than in that which was cut after the leaves were fully formed.

Whatever view we may adopt, it is evident that the analysis of the ash, taken without reference to the organ or part, does not determine all we wish to know of the chemistry of vegetation. The truth of this remark is sustained by all the analyses of the organs of plants, as leaves, fruit and seed; and I have no doubt it is equally true of the parts of the trunk, the heart and alburnum, and the smaller branches.

PROPORTIONS AND ANALYSES OF THE ASH OF FOREST AND FRUIT TREES.

- I. PROPORTIONS OF THE PROXIMATE ELEMENTS.
 - I. SOFT WOODS, OR CONIFEROUS TREES.
- 1. WHITE PINE (Ash of seasoned bark).

Water -	•	•	-	-	-	-	•	$6 \cdot 10$
Dry bark	-	-	-	-	-	-	-	$93 \cdot 90$
Ash -	-		-	-	-	-	-	0.22

2. Yellow Pine (Pinus rigida).

			Sap wood.	Bark of twigs.	Heart wood.	Leaves.	Small limbs.
Water -	-	-	$37 \cdot 00$	40.32	$22 \cdot 50$	$54 \cdot 55$	47.00
Dry wood	-	-	63.00	50.6 8	77.50	$45 \cdot 45$	$53 \cdot 00$
Ash -	-	-	0.15	0.64	$0 \cdot 15$	0.50	0.25

0	n	~	(T ·			
ರ.	KED	CEDAR	(Juniperus	vir	ouniana	١.

							Sap wood.	Heart wood.	Bark. *
Water -	-	-	-	-	-	-	41.94	17.50	$20 \cdot 90$
Dry wood	-	-	-	-	-	-	58.06	$82 \cdot 50$	$79 \cdot 10$
Ash -	-	-	-	-	-	-	$0 \cdot 15$	$0 \cdot 04$	
Organic ma	itter	-	-	-	•	-	$57 \cdot 91$	$82 \cdot 46$	8.42

4. Hemlock (Abies canadensis), seasoned.

Water -	-	-	•	-	•	-		18.00
Dry wood		-	-	-	-	-	-	82.00
Ash -	-	ca ca	-					0.61

II. THE HARD WOODS.

1. WHITE OAK (Quercus alba).

Water		-	Sap wood. 35 · 440	Heart wood. $30\cdot 900$	Bark of the trunk. $27 \cdot 71$	Bark of twigs. $40\cdot30$	Small limbs. $35\cdot000$
Dry	-	-	$64 \cdot 460$	$69 \cdot 100$	$72 \cdot 29$	59.70	$65 \cdot 000$
$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	0.640	0.180	11.30	4.72	0.550
Organic	matt	er,	$63 \cdot 920$	68.920	$69 \cdot 39$	$54 \cdot 98$	$64 \cdot 450$
Calculat	ed dr	у,	0.991	0.261	15.63	7.90	0.846

2. Black Walnut (Juglans nigra).

Water	-	_	Sap wood. 38.900	Heart wood. $45\cdot050$	Bark of the trunk. $48 \cdot 75$	Bark of the twigs. $49 \cdot 260$	Wood of the twigs. 39.65
Dry	-	-	$61 \cdot 100$	$54 \cdot 950$	51.25	50.800	$60 \cdot 35$
\mathbf{Ash}	-	-	0.540	0.360	$3 \cdot 99$	3.715	1.20
Organic	matt	er,	$59 \cdot 936$	$47 \cdot 285$			
Calculat	ed dr	y,	0.950	0.919			

Tap root of the young tree.

										Bark of the root,
Water	-	-	-	-	-	•	-	-	$56 \cdot 17$	58.60
Dry -	-	-	-	-	-	-	-	-	43.83	$41 \cdot 40$
Ash	-	-	-	•		-	•	-	0.85	1.10

3. Common Butternut (Juglans cinerea). Branch one inch in diameter.

							Wood.	Bark.
Water	-	•	-	-	-	-	$38 \cdot 50$	40.00
Dry	-	-	-	-	-	•	61.50	60.00
Ash	-	•	-	-	-	-	0.37	2.80

^{*} Seasoned stick 32 years old.

4. Beech wood (Fagus sylvestris).

		`			Sap wood.	Heart wood.	Woodsof twigs.	Bark of twigs.
Water	-	-	-	-	$40 \cdot 45$		$37 \cdot 40$	35.61
Dry	-	-	-	-	$59 \cdot 55$		$62 \cdot 50$	$64 \cdot 39$
Ash			-	-	0.85	0.56	0.47	$5 \cdot 53$

The percentage of charcoal in beech wood amounts to 17.16. Deducting 0.85 for inorganic matter, it leaves 16.94 from which all volatile matter has been expelled by ignition in a close vessel.

5. Iron-wood (Ostrya virginica).

					Sap wood.	Heart wood	Wood of twigs.	Bark of twigs.
Water	-	-	-	-	30.00	$36 \cdot 96$	21.45	23.83
Dry	-		-		$65 \cdot 00$	$63 \cdot 04$	$78 \cdot 55$	76 · 17
Ash	-	-	-	-	0.196	0.301	0.64	$9 \cdot 321$
Organic	matte	er	-	-	64.805	$62 \cdot 739$	$77 \cdot 91$	$69 \cdot 07$
Calculat	ed dr	y		-	0.300	0.475	0.815	9.321

The iron-wood, when ignited in a close vessel, gives 16.21 per centum of charcoal, or matter free from volatile substances: deducting 0.30 for inorganic matter, it leaves 15.91 as pure charcoal.

Seasoned wood of the Ostrya virginica, or Iron-wood.

					Heart wood.	Sap wood.	Bark.
Water -		-	•	-	14.80	19.06	$14 \cdot 30$
Dry		•	-		$85 \cdot 20$	80.94	85.70
Ash -		-	-	-	0.40	0.28	S:06
Organic matte	er -	-	•	-	84.80	$80\cdot66$	77.64
Calculated dry	y -	-	-	-	0.467	0.344	$9 \cdot 406$
	,						-

The Iron-wood is one of the instances in which the percentage of ash is greater in the heart wood than in the sap wood. The tree, as is well known, grows slowly, and never attains a large size; and the one from which the wood was taken for experiment, was about eight inches in diameter. The seasoned wood was from a small tree about the same size as the preceding, but was one hundred years old.

6. Broad-leaved Laurel (Kalmia latifolia).

•				Wo	od of the trunk.	Wood of the root.	Bark of trunk.	Leaves.
Water	-	-	-		30.30	$36 \cdot 30$	18.73	$49 \cdot 19$
Dry	-	-	-	-	70.00	63.70	81.27	50.81
Ash		-		-	0.22	0.10	0.70	1.46

The percentage of charcoal of this very compact wood amounts to only 7.30; and deducting the inorganic matter, 6.60 is pure charcoal.

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7. Horse Chestnut (Æsculus hippocastanum).

	Sap wood near the bark.	Sap wood near	Heart wood.	Wood of limbs.	Bark of limbs.	Outside bark	Inside bark
Water,	50.00	the heart. 47.50	$58 \cdot 05$	$23 \cdot 70$	$36 \cdot 25$	of the trunk. $17 \cdot 35$	of the trunk. $44\!\cdot\!35$
Dry -	$50 \cdot 00$	$52 \cdot 50$	41.95	76.30	$63 \cdot 75$	$83 \cdot 65$	$55 \cdot 65$
Ash -	0.35	0.62	1.50	1.15	$3 \cdot 50$	$10 \cdot 00$	$5 \cdot 00$

Limbs of the Æsculus hippocastanum (same tree), at different stages.

1. April 26. Limbs taken about two inches in diameter. Buds just bursting.

							Bark.	Wood.
Water	-	-	-	-	-	-	51.00	$56 \cdot 13$
Dry matte	r	-	-	-	-	-	$49 \cdot 00$	43.87
Ash	-	-	-	-		-	$5 \cdot 00$	0.51
Ash calcu	lated	l on th	ie dry	matt	ter		$10 \cdot 204$	1·162 S.

2. May 29. Limbs about two inches in diameter. Flowers just fallen.

						Wood.	Bark.
Water -	-	-	-	•	-	48.03	$53 \cdot 50$
Dry matter	-	-	-	-	-	51.97	$46 \cdot 50$
Ash -	-	-	-	-	-	0.58	$4 \cdot 29$
Ash calculated	on th	ie dry	matt	er	-	1.116	9·226 S.

The two following analyses exhibit a difference in the amount of ash and dry matter in the leaves of two Horse Chestnut trees growing on the same soil, and subjected as nearly as they could well be to the same conditions. There was also much difference in the appearance of the bark, flowers, and shape of the leaves in the two trees. The leaves were gathered May 18th, when the trees were in full bloom; and the middle leaflet of equally vigorous leaves from each tree was selected for analysis.

No. 1. The middle leaflet weighed 57.5 grains, and gave

					A	ctual quantities.	Per centum.
Water -	-	-	-	-	-	$41 \cdot 19$	71.633
Dry matter	-	-	-	-	-	16.31	$28 \cdot 367$
Ash	-	-	-	-	-	1.33	$2 \cdot 313$
Ash calculated	on th	ie drv	matt	er	-		8·157 S.

No. 2. The middle leaflet weighed 77.25 grains, and gave

					A	etual quantities.	Per centum.
Water -	-	-	-	-	-	$55 \cdot 40$	71.715
Dry matter	-	-	-	-	-	21.75	$28 \cdot 285$
Ash	-	-	-	-	-	1.58	$2 \cdot 045$
Ash calculated	on th	e dry	matte	er	-		7·264 S.

The leaves of No. 2 are much larger (the leaflets being wider and longer) than those of No. 1.

8. White Elm (U'mus americana).

		Sap wood.	Heart wood. Out	side bark of trunk.	Inside ditto.	Bark of limbs.	Wood of limbs.
Water		$34 \cdot 65$	$49 \cdot 50$	$19 \cdot 00$	47.50	42.60	$36 \cdot 50$
Dry	-	$65 \cdot 35$	$50 \cdot 50$	81.00	$52 \cdot 50$	<i>5</i> 7·40	63.50
Ash	-	0.80	0.35	$8 \cdot 25$	$7 \cdot 25$	7.50	0.45

The elm gave, on ignition in a close vessel, 15.84 per centum of charcoal, or matter free from volatile elements: subtracting the inorganic matter, it leaves 15.04 per centum of pure charcoal. The outside bark of this elm consists of alternate layers of common ligneous matter and cork, which, though thin, is fine and elastic.

9. Chestnut (Castanea vesca).

							W	ood of the trank.
Water	-	-	-		-			$42 \cdot 35$
Dry	5	-	-	-	-	-	_	57.65
Ash		•	•	•	-		-	$0 \cdot 48$

The percentage of charcoal in this wood amounts to 9.75 only: ash being deducted, it leaves 9.27.

10. White Maple (Acer dasycarpum).

				Sap wood.	Heart wood.	Bark of the trunk.	Bark of limbs.	Wood of limbs.
Water		-	-	35.50	$37 \cdot 50$	40.00	$41 \cdot 45$	31.00
\mathbf{Dry}^{-1}	-	-	-	$74 \cdot 50$	$62 \cdot 50$	60.00	$58 \cdot 45$	69.00
$\mathbf{A}\mathbf{s}\mathbf{h}$	-	-	-	0.25	0.20	$3 \cdot 25$	2.75	0.35

11. WILLOW. Trunk ten inches in diameter.

							Sap wood.	Heart wood.	Bark of the trunk.
Water		-	-	-	-	-	$59 \cdot 55$	$37 \cdot 45$	41.10
Dry	-	-	-	-	-	-	$40 \cdot 65$	$62 \cdot 55$	58.90
Ash	-	-	-	-	-	-	0.28	0.25	$6 \cdot 26$

12. GRAPE VINE (Vitis labrusca).

								Wood.
Water	-	-	-	-	-	-	•	$40 \cdot 26$
\mathbf{Dry}	•	-	-	-	-	-	-	$59 \cdot 74$
Ash	-		-	-	-		-	0.98

13. Black Ash. Seasoned wood.

						Sap wood.	Bark.
Ash	-	-	-	-	-	0.34	8 · 19

14.	Bass-w	000	(Til	ia an	nerica	na).				
			`			,			Sap wood.	Bark.
	Water	•	-	-	•	•	e	•	$51 \cdot 30$	$46 \cdot 32$
	\mathbf{Dry}	•	•	•	•	9	٠	-	48.70	$53 \cdot 68$
	Ash	-	-	•	-		•	-	0.28	3.57
15.	Вьаск	Bir	есн ()	Retuli	a erre	olea).				
10.	DIACK	DIN	(1	Jeiui	i ciic	isuj.			Sap wood.	Heart wood.
	Water	•	-	•	-	-	•	•	$38 \cdot 90$	34.61
	Dry		-	-	-	-	•	-	61.10	$65 \cdot 39$
	Ash	•		-				-	0.05	0.26

The black birch gives 16.01 per centum of charcoal: deducting the inorganic matter, it leaves 15.96 per centum from which all volatile matter is expelled by ignition in close vessels.

16. St.	AFF-T	REE	(Cere	astrus	scan	dens)	٠		
	69		`			,		Bark.	Wood.
W	ater	-		•	•	•		39.091	46.094
Dr	y ma	tter	a		•	•	-	$60 \cdot 909$	53.906
As	sh	-	-	•	-	-	-	4.608	0.523
As	sh cal	culat	ed on	the d	lry ma	atter	•	$7 \cdot 582$	0·971 S.

III. FRUIT TREES.

1 Pr	ATR '	TRE	7.						
1. 12	1111	110111	••				Sap wood.	Heart wood.	Bark of the trunk.
Water	-	-	•			-	48.80	$22 \cdot 05$	63.70
Dry	-	-	•	•			$37 \cdot 20$	77.95	$30 \cdot 30$
Ash	-	-	-	-	•	-	0.20	0.10	1.99
	Ro	ot of	the .	Pear	tree.				
								Wood.	Bark.
W	Tater	•	-	-	•	•		$22 \cdot 33$	58.80
Dı	ry	-	•	•	•	••		$79 \cdot 67$	$46 \cdot 20$
As	sh	-	•	-	-	-		0.40	3.26

The wood of the pear gives 9.79 per centum of charcoal. The wood of the pear is soft, close grained and easily wrought, and hence is sometimes substituted for box in large wood engravings.

2. Sweet Apple tree.

						Sap wood.	Heart wood.	Bark of the trunk.
Water		27	-	•	•	$39 \cdot 10$	$33 \cdot 35$	$59 \cdot 00$
Dry	-		•	•	-	60.90	$66 \cdot 65$	41.00
Ash	•	-	•	9	•	0.35	0.16	4.55

5	3. Sour	App	LE TR	EE.					
							Sap wood.	Heart wood.	Bark of the limbs.
	Water	-	-	-	•	-	$39 \cdot 13$	46.30	$45 \cdot 10$
	\mathbf{Dry}	-	-	•	-	•	60.87	53.70	$54 \cdot 90$
	Ach	_	_	_	_	_	0.95	0.90	2.22

The apple-tree wood gives a heavy compact coal amounting to 15.90 per centum, or 15.70 abstracting the ash.

гас	ung ine a	SII.									
4	SMALL R	ED (ая анг	v. S	easoli	ed.					
1.			711121616		Cuson	icu.				Wood	
	Water		•	-	•	-	-	-	•	10.00	
	Dry	•	-		-	-	-	-	-	90.00	
	Ash	•	-	•	-	•	-	-	•	0 · 17	
5.	Реасн тв	REE.	The	leave	es (A	ugus	t 7).			•	
	Water	_							Leaves. 67.00		ust formed.
	Dry		•	-			-				3 · 93
	Ash	_	_					6	4.62		, 00
	ASII	_	-						4 02		
	Bark	of t	he twi	gs (N	Iay 1	0).					
	Water -	-		-	-	-	-	-	•	60.8	
	Dry -	-	-	-	-	-	-	•	-	$39 \cdot 4$	
	Ash -	-	•	-	-	-	-	-	~	5.66	
	Healt	hy I	Peach l	eaves,	fron	ı Ne	w-Jei	sey.			
	Water	•		-	-	-	-	-	-	69.00	
	Dry -	-		-	-	-	-	-	-	31.00	
	Ash -	-	-	-	•	-	•	-	-	1.75	
	Peach	lear	ves dise	eased ·	with i	the u	ellows				
	Water	•				-	•		-	64.00	
	Dry -		-					-	-	36.00	
	Ash •	•	-	-	-	-	-	•	•	1.82	
	Healt	hy I	Peach l	eaves	(Mal	acot	oon),	from	New-	Jersey.	
	Water	•		-	`-		•		-	$63 \cdot 10$	
	Dry -	-	_	-	•	-		-	•	36.90	
	Ash -	•	-	-	•	-	-	-	-	3.00	
	Leave	s dis	seased 1	with t	he vei	llows				3	
	Water									$54 \cdot 46$	
	Dry -			-		-			-	45.54	
	Ash -	_		4				- -		3.87	
	*****	_		-	-	_	_	_	-	5 01	

6. Leaves of the	CATA	WBA	GRAI	e, p	icked	June	· 2.	Full size.
Water	-	-	-	-	-	-	-	$72 \cdot 388$
Dry matter	•	-	-	-	•	-	-	27.612
Ash -	_	-	_	-	-	_	-	9 · 138

Ash calculated on the dry matter - - 7.746 S.

7. Leaves and petioles of the Virgalieu Pear-tree, picked May 23. Flowers just fallen.

Water -	-	-	-	-	-	-	-	$73 \cdot 082$
Dry matter	-	-	•	-	-	-	-	$26 \cdot 918$
Ash -	-	•	-	-	-	-	-	1.299
Ash calcula	ted o	n the	dry n	natter	-	•	-	4·213 S.

8. Leaves and petioles of the Ox-HEART CHERRY, picked May 23. Flowers just fallen.

Water -	-	-	•	•		-	-	$74 \cdot 429$
Dry matter		•	-	-	•	-		$25 \cdot 571$
Ash -	-	•	-	•	-	-	-	1.569
Ash calcula	ated o	n the	dry n	natter		•	-	6·135 S.

9. Leaves and petioles of the LADY APPLE, picked May 23. Flowers just fallen.

								-		-
\mathbf{W}_{a}	iter ·	-	•	•	-	-	-	-	•	$70 \cdot 261$
Dry	mat	ter	-	-	-		-	•	•	$29 \cdot 739$
Asl	1	•	-	•	•	-	-	•	•	$1 \cdot 412$
Asl	h cal	eulate	ed on	the	dry	matter	-	•	-	4.747 S.

The following proportions of water, dry wood and ash, were determined by my assistant Mr. Salisbury. The specimens of the trees were obtained in May.

TABLE I.

	The F	lose.	Xanthoxylui	n american.	Laurus s	assafras.	Populus tremuloides.	
	Wood.	Bark.	Wood.	Bark.	Wood.	Bark.	Wood.	Bark.
Percentage water, Percentage dry - Percentage ash -	32·67 67.33 0·33	36·92 63·08 2·41	24·00 76·00 0·49	22·05 77·95 7·88	30·38 69·62 0.38	46·26 53·74 2·17	40·52 59·48 0·33	50·70 49·30 3·95

TABLE II.

	R	hus typhina		Wild	Plum.	Cornus alternifolia.		
	Bark.	Sap wood.	Heartwood	Wood.	Bark.	Wood.	Bark.	
Percentage water, Percentage dry - Percentage ash -	35·74 64·26 2·89	25·24 74·76 0·28	21·06 78·94 0·43	18.80 81.20 0.37	26·08 63·92 6·05	34.03 65.97 0.20	54·15 45·85 2·41	

11. ANALYSES OF THE ASH.

1. FOREST TREES.

1. Hemlock (Pinus canadensis).

Tree sound. Diameter five feet from the base, 25 inches. Average thickness of the bark about one inch. Half of the diameter was produced during the last fifty years growth: these 50 outside layers were taken for outside wood; and the remaining layers, 154 in number, for inside wood. Whole number of layers five feet from the ground, 204: average thickness of each layer, 0.05637 of an inch.

•				0	•	0			
					Bark.	Outside wood.	Inside wood.	Bark of twigs.	Wood of twigs.
Potash	-	•	•	-	2.86	$19 \cdot 23$	$1 \cdot 64$	1.575	0.506
Soda	-	-	•	-	$3 \cdot 47$	8.46	0.54	1.333	$5 \cdot 31$
Chloride	of sod	ium	-	-	0.03	$0 \cdot 10$	0.03	0.994	0.42
Sulphuri	ic acid	-	-	•	$3 \cdot 48$	3.03	$11 \cdot 96$	$4 \cdot 47$	3.213
Carbonio	e acid	•		-	$24 \cdot 33$	7.81	7.63	24.00	28.83
Lime	-	-	-	-	31.48	$10 \cdot 11$	9.88	31.05	$25 \cdot 20$
Magnesi	ia	•	-	•	0.01	2.48	$3 \cdot 24$	0.30	2.88
Phospha	ite of p	eroxi	de of	iron,	1.49	$2 \cdot 72$	0.29	1.55)
Phospha	ate of \overline{l}	ime	-	-	$16 \cdot 45$	16.66	11.38	18.87	20.40
Phospha			esia	•	$5 \cdot 17$	9.89	$26 \cdot 45$	1.28)
Organic	matter		-		$3 \cdot 48$	1.71	2.40	$4 \cdot 10$	0.212
Insoluble	e silica	-	-	-	$13 \cdot 40$	$5 \cdot 28$	1.05	0.40	0.70
Coal	-	•	-	-	$1 \cdot 22$	$1 \cdot 24$	0.85	0.48	1.51
					106·87 S.	88·72 S.	77·31 S.	90.402	89.231

2. Sugar Maple (Acer saccharinum).

Tree sound. Diameter three feet from the ground, 28 inches; do. twelve feet from the ground, $21\frac{1}{2}$ inches. From the base to the limbs, 62 feet. Whole length of the tree, 107 feet. Average thickness of the bark, $\frac{3}{4}$ inch. Age 224 years. At twelve feet from the base, the 100 outside layers were taken for outside wood, making a thickness of $4\frac{5}{8}$ inches; the remaining layers were taken for inside wood. Growth very uniform. Average thickness of each layer, 0.04464 of an inch.

Potash	_		_	$\overset{\mathrm{Bark.}}{0\cdot\mathrm{SS}}$	Outside wood 8.77	Heart wood. $4 \cdot 21$
Soda			-	7.75	0.964	1 21
Chloride of sodium	-	-	-	0.08		0.08
Sulphuric acid -		•	•	1.497	1.171	1.03
Carbonic acid -	-		-	$37 \cdot 12$	$37 \cdot 247$	$33 \cdot 33$
Lime	-	•	•	$49 \cdot 33$	31.86	$43 \cdot 14$
Magnesia	-	-		$3 \cdot 64$	8.40	$7 \cdot 24$
Phosphate of peroxic	de of i	ron	-	0.35	0.70	1.34
Phosphate of lime	-	-	-	$3 \cdot 13$	$5 \cdot 70$	5.09
Phosphate of magne	sia	•	-	0.05	1.80	0.55
Organic matter -	-	-	-	1.50	$2\cdot 40$	1.93
Insoluble silica -	-	-	-	$0 \cdot 15$	0.50	$0 \cdot 55$
	•			105·417 S	. 100.512	98·16 S.

3. Chestn	ur (Casta	nea ı	esca)				
	`			,		Bark.	Outside wood.	Inside wood.
Potash -	-	-	-	-	-	$1 \cdot 36$	$4 \cdot 56$	$2 \cdot 73$
Soda -	•	-	-	-	•	0.319	1.41	1.98
Chloride of	sodi	um	•	-	-	trace.		
Sulphuric a	cid	-	-	-	-	0.312	0.50	
Carbonic ac	id	•	-	-		39.90	23.842	$29 \cdot 52$
Lime -		-	-	-	-	51.60	40.76	$38 \cdot 20$
Magnesia	-	-		-	-	0.60	$5 \cdot 77$	0.513
Phosphate of	f pe	roxide	of ir	on	-	0.20	1.30	0.30
Phosphate o	f Iin	ne	-	-	-	2.90	$17 \cdot 44$	8.60
Phosphate of	of ma	agnesi	a -	-	-			
Organic ma	tter	-	-	-	-	$5 \cdot 00$	1.74	3.20
Silica -	-	-	-	-	-	1.20	$1 \cdot 43$	1.71
Coal -	-	•	-	-	-	1.00	0.914	1.76
Moisture	-	-	-	-	-	3.00		$2 \cdot 13$
						107:391	99.666	$90 \cdot 143$

4. Horse Chestnut (Æsculus hippocastanum.)

Tree in full bloom May 26, at which time the young sprouts (of about five weeks growth), petioles, leaves, peduncles and flowers were gathered for analysis; but the root analyzed was cut from its tree in April

	Bark of root.	Wood of root.	Bark of sprouts.	Wood of sprouts	s. Petioles.	Leaf-blades.	Peduncles and flowers.
Potash -	10.88	16.83 (17.97	$26 \cdot 33$	30.95	7.58	30.71
Soda -	$15 \cdot 71$	3.27	1. 0.	20 00	00 00	7 90	00 11
Chloride sodiun	n, 0·76	0.36	$2 \cdot 27$	$4 \cdot 47$	$4 \cdot 12$		
Sulphuric acid,	$7 \cdot 32$	6.39	$26 \cdot 45$	$28 \cdot 46$	30.93	21.94	$30 \cdot 55$
Carbonic acid.	$8 \cdot 47$		$5 \cdot 71$	$0 \cdot 04$	$2 \cdot 27$	$18 \cdot 29$	$0 \cdot 13$
Lime -	36.83	$5 \cdot 19$	$7 \cdot 39$	0.06	2.93	$2 \cdot 91$	0.17
Magnesia -	0.90	$5 \cdot 30$	0.25	0.05	$0 \cdot 10$	$3 \cdot 00$	0.01
Phos. perox. iro	n, 1·40	1.10	0.30	0.30	0.20	$2 \cdot 50$	4.20
Phos. lime,	$10 \cdot 15$	$45 \cdot 64$	$15 \cdot 90$	22.70	$17 \cdot 95$	31.75	$25 \cdot 65$
Phos. magnesia	0.25	0.01	0.50	0.02	0.75	0.75	3.75
Organic matter	5.60	$3 \cdot 60$	$22 \cdot 46$	17.09	9.60	$6 \cdot 24$	$3 \cdot 20$
Insoluble silica,	$2 \cdot 00$	3.30	0.80	0.50	0.20	$5 \cdot 50$	2.00
Coal -	0.80	11.00					
	101·07 S.	101·99 S.	100·00 S.	100·00 S.	100·00 S.	100·46 S.	100·37 S.
			PROPORTI	ONS.			
Per cent. water	57.95	$50 \cdot 14$	82.90	$88 \cdot 10$	$87 \cdot 94$	74.80	$85 \cdot 24$
Per cent. dry,	$42 \cdot 05$	$49 \cdot 86$	$17 \cdot 10$	11.90	$12 \cdot 06$	$25 \cdot 20$	14.76
Per cent. ash,	3.88 S.	1.04 S.	1.63 S.	1·15 S.	1·56 S.	1·68 S.	1·83 S.

5. SWAMP WHITE OAK (Quercus bicolor).

						Bark.	Outside wood.	Inside wood.
Potash -	-	-	-	-	-	0.459	$20 \cdot 49$	14.79
Soda -	-	-	-	-	-	trace.	$3 \cdot 15$	3.69
Chloride of	sodiu	ım	-	-	-			
Sulphuric ac	id	-	-	-	-	0.295		
Carbonic aci	d	-	-	-	-	$40 \cdot 335$	$32 \cdot 919$	34.61
Lime -	-	•	-	-	-	$52 \cdot 26$	30.23	35.87
Magnesia		-	-	-	-	0.25	0.50	0.51
Phosphates	-	-	-	-	-	3.50	5.20	6.30
Organic mat	ter	-	-	-	-	2.13		2.70
Silica - ·	-	-	-	-	-	$2 \cdot 00$	1.50	0:50
Coal -	-	-	-	-	-	$2 \cdot 50$	$4 \cdot 00$	1.60
Moisture	-	~	-	•	-			2.60
٠						103.729	97.999	103 · 17

6. White Oak (Quercus alba).

Ash obtained from the green wood.

			Sap woo	d. Heart wood.	Wood of twigs.	Bark of trunk,	Bark of twigs.
Potash -	-		- 13.41	9.68	$9 \cdot 74$	0.25	1.27
Soda	-	•	- 0.52	5.03	6.89	2.57	$4 \cdot 05$
Sodium -	-	-	2.78	0.39	0.16	0.08	0.08
Chlorine -	•	-	4 • 24	0.47	0.25	. 0.12	0.13
Sulphuric acid	-	-	- 0 · 12	0.26	0.08	0.03	trace.
Phosphate of per	roxi	de of iron	n,)			0.60	
Phosphate of lin	ne	-	$\cdot \begin{cases} 32 \cdot 25 \end{cases}$	13.30	$23 \cdot 60$	10.10	$14 \cdot 15$
Phosphate of ma	igne	esia	.)				
Carbonic acid	-		8.95	$19 \cdot 29$	$17 \cdot 55$	29.80	$30 \cdot 33$
Lime - ·	-		30.85	43.21	34.10	54.89	47.72
Magnesia -	-		0.36	0.25	0.50	0.50	0.50
Silica	-		0.21	0.88	0.55	0.25	0.65
Soluble silica		•	. 0.80	0.30	0.60	0.25	0.65
Organic matter		-	5.70	7.10	5.90	1.16	1.52
			100.18	S. 100·06 S.	. 99·99 S.	100·05 S.	100·00 S.

The oak grew in the immediate neighborhood of Albany, upon a stiff clay known as the Albany clay.

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7. White Elm (Ulmus americana).

Tree sound. Diameter three feet from the base, 28 inches; ditto fourteen feet from the base, 25½ inches. Mean thickness of the bark, 3/4 inch. Whole length of the tree, 111 feet. Number of layers fourteen feet from the base, 208. Average thickness of each layer, 0.05769 of an inch. Thickness of layers quite uniform. From 80 to 85 outside layers taken for outside wood; the remaining layers, for inside wood.

1.					
FIRST SPECIMEN.		Bark of trunk.	Outside wood.	Bark of twigs.	Wood of twigs.
Potash	-	3.79		5.82	9.61
Soda	-	1.65		.6.53	18.41
Chloride of sodium -	-	trace.		0.07	
Sulphuric acid -	-	0.14	$12 \cdot 02$	2.80	$3 \cdot 98$
Carbonic acid -	-	$39 \cdot 44$		24.72	$26 \cdot 07$
Lime	-	27.46		16.92	14.77
Magnesia	-	13.10		$3 \cdot 00$	$2 \cdot 40$
Phosphate of peroxide o	f iron	n ,)			•
Phosphate of lime -	-	3.40		$24 \cdot 50$	$22 \cdot 35$
Phosphate of magnesia	-)			
Organic matter -	•	2.00		1.50	
Insoluble silica -	-	1.75			0.50
Coal	-				0.30
Moisture	-	$3 \cdot 10$			
		85.86	12.02	95.82	97.49
SECOND SPECIMEN.		Sap wood.	Heart wood.	Outside bark.	Inside bark.
Potash	-	15.85	8.64	$5 \cdot 32$	1.17
Soda	-	7.64	$20\cdot 49$	$3 \cdot 22$	$2 \cdot 17$
Chlorine	-	0.74	0.08	1.21	0.05
Sulphuric acid -	-	0.12	0.14	0.10	0.04
Phosphate of peroxide	of iro	on, 1.82	1.05	$4 \cdot 00$	
Phosphate of lime -	-	} ₁₄ ·53	$2 \cdot 75$	$19 \cdot 55$	3.775
Phosphate of magnesia	-	14.99	2 10	13 99	0 119
Carbonic acid	-	29.51	$28 \cdot 225$	$13 \cdot 26$	42.515
Lime	-	20.08	$22 \cdot 635$	$30 \cdot 26$	$42 \cdot 495$
Magnesia	-	4.72	$10 \cdot 08$	4.84	8.16
Silica	•	2.00	$3 \cdot 25$	$12 \cdot 15$	1.25
Soluble silica	-			1.60	
Organic matter -	•	1.45	1.80	4.12	0.40
		98·46 S.	99·115 S	5. 99·59 S.	102·025 S.

8. Red Elm (Ulmus fulva).

Tree sound. Average diameter four feet from the base, 20 inches. Average thickness of bark, $\frac{3}{8}$ inch. Average thickness of each layer, 0.0564 of an inch. Between 20 and 25 layers were taken for outside wood, thickness $2\frac{3}{4}$ to 3 inches; the remaining layers were taken for inside wood. Growth quite uniform.

uniform.					Bark.	Outside wood.	Heart wood.	Bark of twigs.	Wood of twigs.
Potash -		-	-	•	0.11	$13 \cdot 43$	$7 \cdot 34$	3.79	5.82
Soda -		•	-	•		16.96	7.89	7.87	19.74
Chloride of s	odi	um	•	•		0.02	0.05	trace.	0.20
Chloride of p	ota	ıssiuı	m	-	0.06				
Sulphuric aci		-	-	-	$5 \cdot 36$	0.81	$4 \cdot 57$	5.79	8.94
Carbonic acid	ł	-	-	-	$34 \cdot 41$	16.96	26.89	$33 \cdot 12$	12.69
Lime -		-	•	-	44.64	31.00	$34 \cdot 79$	$32 \cdot 12$	$17 \cdot 72$
Magnesia -		-	-	•	3.09	$5 \cdot 24$	$2 \cdot 20$	1.68	4.80
Phosphate of	· pe	eroxi	de of	iron,	0.04	0.65	1.35	0.20	0.40
Phosphate of	liı	me	•	•	} 6·36 {	$12 \cdot 97$	11.28	6.00	25.80
Phosphate of	m	agne	sia	•	} 0.30 {	$2 \cdot 93$	2.18		0.20
Organic matt	ter		-	-	2.10	1.93	1.60		10.84
Silica -		-	-	•	2.81	1.31	0.55	$6 \cdot 50$	$0 \cdot 40$
Coal -		-	-	-				$3 \cdot 50$	
					. 8.98 S.	104·24 S.	100·69 S.	100.57	107·45 S.

9. Cork	\mathbf{E}_{LM}	(Ulmus)	racemosa)	١.
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	•			•			Wood.	Bark.
Potash	•_	-	-	-	-	•	$25 \cdot 93$	8.884
Soda		-	-	-	•	-	1.70	0.498
Chlorine		-	-	-	-	•	0.30	0.560
Sulphuric	c acid	ł	-	•	-	•	$2 \cdot 57$	$4 \cdot 485$
Phosphat	e of	perox	ide o	f iron	-	-)	
Phosphat	e of	lime	-	•	•	-	313.77	$5 \cdot 605$
Phosphat	e of	magr	esia	•	-	-)	
Carbonic	acid	•		•	•	-	17.70	$19\cdot 568$
Lime	-	-	-	•	•	-	$22 \cdot 83$	46.912
Magnesia	ı	-	-	•		•	8.20	1.557
Silica	-	-	-	•	-	-	3.57	$11 \cdot 214$
Soluble s	ilica		•	•	-		1.67	1 · 121
Organic	matt	er	-	-		•	not dete	rmined.
							98·24 S.	99·S07 S.

The structure of this wood is singular and beautiful, and is probably readily distinguished from any other wood by a transverse section (See Pl. viii. fig. 2). In this figure, the

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wavy light bands mark the direction of the pores through which the sap ascends, and the circular zone of pores shows the commencement of a new annual layer. The last are always formed in the spring, when the first growth of wood begins; the former belong to successive periods during its summer growth. The Elm, together with the allied genus, the *Celtis*, exhibits this peculiar wavy arrangement of its summer pores. There are variations, in the different species, in the arrangement of these pores; yet the plan, as a whole, is the same.

For magnificent specimens of the Elm, the vallies of the Genesee, and the Black river in Jefferson county, are surpassed by no other parts of the world. Hundreds of elms may be seen in either of these sections of country, exceeding by far the famous Pittsfield Elm in Berkshire, Mass. The elms of the Mohawk valley belong generally to the pendulous variety, and do not excel in size, but many may be seen profusely decked with slender branches, and give great beauty to the landscape. This variety is figured in the first volume of Agriculture, Pl. ii. The elm has been sacrificed in all parts of the State, during its first settlement, for the sake of its ash. It furnishes not only a large percentage, but it is also rich in alkalies.

The elm is a favorite on both continents. It is highly ornamental, and the wood is useful for many purposes. Like domesticated animals, it seems to have been designed for man in his more civilized state; and, like them too, it breaks out into numerous varieties, in some of which we have the majestic trunk adorned with a towering, upright or spreading head; or a trunk profusely decked with slender limbs, as if covered with twining ivy, and a head profuse in long pendant and waving plumes.

In some parts of New-York, especially where meadows skirt a sluggish stream, the White Elm stands unrivalled in height and girth. Its hundred feet to a branch, a girth numbering a score feet or more, mark but a common size. The rich alluvial bottoms are best adapted to the wants of this tree: it there finds an abundance of food and water; and as it continues to grow for centuries, it here attains its maximum strength and size, and well deserves the name of the forest monarch. For transplanting, it is recommended by its easy culture and rapid growth, and its ability to adapt itself to any soil or location.

The Elm has a wide range of growth. The Saskatchawan, or in Canada at 48° 20′ north, to Georgia south, constitute its extreme limits, and so it extends itself from the Atlantic far west. The north and south boundaries of New-York embrace probably the most favorable zone for this tree.

The wood of the elm rarely if ever splits very free; its fibres being too much twisted or interwove with each other, to admit of an easy separation. In consequence, however, of this very fault, it is better adapted to planks for stabling, for ox yokes, and all purposes where lightness and strength are required, and where an easy separation of its fibres would be a damage, as wagon hubs, and large ship blocks.

10. Hickory (Carya alba).

The wood had been seasoned during one summer and fall, and grew in the valley of the Mohawk.

					Outside sap wood.	Inside sap wood.	Heart wood.	Bark.
Potash ·	•	-	-	-	$7 \cdot 472$	$20 \cdot 187$	$12 \cdot 210$	2.340
Soda -		-	•	-	0.084	0.085	0.055	0.125
Chlorine .		-	-	-	0.096	0.085	0.065	$0\cdot 145$
Sulphuric a	icid	-	-	•	0.892	4.640	5.260	1.925
Phosphate	of pe	rox	de of	iron)			
Phosphate	of li	me	•	-	§ 14·440	$11 \cdot 450$	$6 \cdot 340$	$5 \cdot 000$
Phosphate	of m	agn	esia	-)			
Carbonic ac	cid	•		-	$29 \cdot 576$	$21 \cdot 405$	33.630	33.995
Lime -		-	•	-	$38 \cdot 264$	$27 \cdot 695$	$43 \cdot 520$	$51 \cdot 105$
Magnesia -	•		-	-	6.200	8.600	$4 \cdot 000$	0.820
Silica -		-	•	-	$4 \cdot 200$	$6 \cdot 150$	$1 \cdot 300$	$4 \cdot 550$
Soluble sili	ca	•		-	0.280	0.010	trace.	0.250
Organic ma	itter	•	•	•	undete	ermined.		
					101·504 S.	100·331 S.	106·390 S.	100·255 S.

11. Iron wood (Ostrya virginica).

				Sap wood.	Heart wood.	Wood of twigs.	Bark of trunk.	Bark of twigs.
Potash -	-	-	-	1.581	14.549	20.76	0.696	$2 \cdot 78$
Soda	-0"	-	-	0.025	0.086	2.97	0.053	0.405
Chlorine -	-	-	-	0.049	0.098	0.25	0.04	$0 \cdot 15$
Sulphuric acid	-	-	-	0.086	0.378	0.64	0.086	0.52
Phosphate of p	eroxid	le of	iron,)				
Phosphate of li	me	-	-	5.65	$23 \cdot 10$	$35 \cdot 4$	$5 \cdot 10$	10.55
Phosphate of m	agne							
Carbonic acid	•	-	-	$36 \cdot 159$	$20 \cdot 139$	$12 \cdot 22$	33.853	33.975
Lime	-	-	-	48.791	$27 \cdot 461$	20.98	57.932	$48 \cdot 225$
Magnesia -	-	-	-	$4 \cdot 20$	$4 \cdot 40$	$5\cdot 6$	1.20	1.00
Silica	-	-	-	0.50	0.40	0.4	0.25	2.30
Soluble silica	-	-	-				0.276	
Organic matter	•	-	-	2.853	undeter	mined.		
				99·577 S.	90·611 S	. 99·21 S.	99·456 S.	99·905 S.

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12. Black Birch (Betula lenta).

Tree slightly hollow at the base. Diameter eight feet from the base, $17\frac{3}{4}$ inches. Mean thickness of the bark, $\frac{1}{2}$ inch. Number of layers eight feet from the base, 235. Average thickness of each layer, 0.03564 of an inch. Growth quite uniform. From 80 to 85 of the outside layers were taken for outside wood: thickness $3\frac{1}{4}$ inches.

		•					Bark.	Outside wood.	Heart wood.
Potash	-	-		•	-	-	$2 \cdot 59$	7.34	$8 \cdot 25$
Soda -	-		•	-	-	-	1.87	$12 \cdot 93$	10.30
Chloride of	sodi	um			-	-		0.10	$0 \cdot 12$
Sulphuric a	cid			-		•	$9 \cdot 537$	2.61	$4 \cdot 12$
Carbonic ac	eid	-	-	-	-	-	$37 \cdot 156$	$24 \cdot 06$	$26 \cdot 44$
Lime -	-	-	-	-	-	-	$41 \cdot 446$	$31 \cdot 14$	34.86
Magnesia	-	-	-	-	-	-	$2 \cdot 32$	8.0	$2 \cdot 64 -$
Phosphate	of pe	roxid	e of in	on		-	0.20	1.4	1.40
Phosphate of	of lin	ne	-	-	-	-	$10 \cdot 20$	$16 \cdot 4$	19.89
Phosphate	of m	agnes	sia		-	-	none.	0.6	0.51
Organic ma	atter	-	•		-	-	0.44	$3 \cdot 2$	$4 \cdot 39$
Insoluble si	lica	•	-	-	-	-	0.40	1 ° 6	1.90
Coal -	-	•	-	-	-	-	0.30		
							${106 \cdot 459}$	109·68 S.	114·82 S.

13. YELLOW BIRCH (Betula excelsa).

Tree sound. Diameter four feet from the base, $21\frac{1}{2}$ inches. Average thickness of the bark, $\frac{1}{2}$ inch. Number of layers four feet from the base, 184. Average thickness of each layer, 0.0557 of an inch.

					Bark.	Outside wood.	Heart wood.
Potash	•	-	-	-	0.917	$4 \cdot 237$	$5 \cdot 60$
Soda	-	-	•	•	0.543	$12 \cdot 36$	$9 \cdot 54$
Chloride of sodium	-	-	-	-	$2 \cdot 369$	$0 \cdot 12$	
Sulphuric acid -	•	-	-	-	$3 \cdot 021$	6.87	1.92
Carbonic acid -	-	• •	-	-	$33 \cdot 25$	18.857	36.69
Lime	•	-	•	•	$48 \cdot 22$	27.50	$22 \cdot 33$
Magnesia	-	-	-	-	$2 \cdot 40$	8.96	$9 \cdot 76$
Phosphate of peroxi	de of	iron	,)	•		$2 \cdot 30$	1.41
Phosphate of lime	-	-	}	•	9.70 }	18.50	$15 \cdot 35$
Phosphate of magne	esia	-)		\$	10 00	10 00
Organic matter -	-	-	-	-	1.50	$5 \cdot 00$	0.80
Insoluble silica -	-	•				1.60	0.90
Coal	-	-	-	e		$1 \cdot 00$	0.30
					101.94	107.304	104.60

14. Bass-wood (Tilia americana).

Tree sound. Mean diameter four feet from the ground, 22 inches. Average thickness of the bark, 1 inch. Age 182 years. About 60 of the outside layers were taken for outside wood, thickness $4\frac{1}{2}$ inches; the remaining layers taken for inside wood. Growth uniform. Average thickness of each layer, 0.0549

of an inch.		Bark.	Outside wood.	Heart wood.	Bark of twigs.	Wood of twigs.
Potash	-	1.26	$10 \cdot 12$	$4 \cdot 05$	1.90	$14 \cdot 55$
Soda	-	12.77	2.88	$10 \cdot 41$	$9 \cdot 14$	
Chloride of sodium	-	0.24	0.50	0.52	0.15	$0 \cdot 10$
Sulphuric acid -	•	0.72	. 0.88	0.27	$4 \cdot 19$	$13 \cdot 34$
Carbonic acid -	-	25.38	$16 \cdot 64$	17.96	$22 \cdot 84$	$3 \cdot 94$
Lime	-	41.92	$38 \cdot 36$	$45 \cdot 24$	$29 \cdot 56$	11.56
Magnesia	-	$2 \cdot 24$	$7 \cdot 36$	7.44	$3 \cdot 00$	$7 \cdot 44$
Phosphate of peroxide	of iron,	0.20	1.20	$1 \cdot 30$	0.31	0.60
Phosphate of lime -	•	8.50	$17 \cdot 95$	8.96	$24 \cdot 77$	38.92
Phosphate of magnesia	a -	0.30	2.60	0.04	$0 \cdot 72$	1.28
Organic matter -	-	1.70	$2 \cdot 53$	$2 \cdot 00$	$2 \cdot 40$	9.61
Insoluble silica -	-	$4 \cdot 60$	$2 \cdot 10$	1.40	0.40	$0 \cdot 10$
Coal	-				0.80	
		99·83 S.	103·12 S.	99·59 S.	100·18 S.	101·44 S.

15. Butternut (Juglans cinerea).

Tree sound. Diameter three feet from the ground, 2 feet 8 inches; eleven feet from the ground, 1 foot 8 inches. The section for analysis was taken eleven feet from the ground. Average thickness of the bark, 3 inch. Age 146 years. Between 65 and 70 outside layers were taken for outside wood, thickness $4\frac{1}{2}$ inches; the remaining inside layers taken for inside wood. Growth of tree more rapid when

young.			Bark.	Outside wood.	Heart wood.	Bark of twigs.	Wood of twigs.
Potash	-		1.00	4.42	1.00	0.63	3·28
Soda	-	-	11.27	$5 \cdot 61$	14.82	11.24	14.59
Chloride of sodiur	n	-	0.15	0.16	0.13	0.03	0.03
Sulphuric acid		-	0.74	$13 \cdot 33$	21.43	$5 \cdot 33$	$5 \cdot 36$
Carbonic acid	•	-	$32 \cdot 12$	$20 \cdot 02$	$4 \cdot 48$	18.92	$7 \cdot 02$
Lime	-	-	37.68	$38 \cdot 98$	$42 \cdot 02$	$24 \cdot 48$	9.08
Magnesia -	-	-	10.08	$3 \cdot 52$	$4 \cdot 00$	$2 \cdot 22$	$5 \cdot 34$
Phosphate of perc	xide o	firon	, 0.30	$3 \cdot 40$	$3 \cdot 41$	0.41	0.50
Phosphate of lime		-	$2 \cdot 25$	$2 \cdot 20$	0.59	$29 \cdot 25$	$40 \cdot 39$
Phosphate of mag	nesia	-	$0 \cdot 15$	0.06	0.28	$1 \cdot 04$	1.61
Organic matter	•-	-	$2 \cdot 80$	$3 \cdot 40$	$3 \cdot 20$	$4 \cdot 41$	$5 \cdot 20$
Insoluble silica	•	-	0.30	4.80	$5 \cdot 40$	0.40	$0\cdot 32$
Coal	•	-				0.80	1.21
Water	-	•					$3 \cdot 41$
•			98·84 S.	100·20 S.	100·76 S.	99·16 S.	97·34 S.

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16. Red Beech (Fagus ferruginea).

Tree a little hollow at the base. Diameter three feet from the ground, 28 inches; fourteen feet from the ground, 22 inches, sound. Average thickness of the bark, \(\frac{1}{4} \) inch. Section for analysis taken fourteen feet from the ground. Age 240 years. Growth quite uniform. Average thickness of each layer, 0.0453 of an inch. Between 60 and 65 outside layers taken for outside wood; the remaining layers for inside wood.

for inside wood.		Bark.	Cutside wood.	Heart wood.	Bark of twigs.	Wood of twigs.
Potash	-	0.13	12.13	4·04	0.63	11.00
Soda	-		15.58	$25 \cdot 53$	4.63	11.79
Chloride of sodium	-		0.05	0.24	$0 \cdot 14$	0.16
Sulphuric acid -	-		0.47	0.62	$4 \cdot 54$	$14 \cdot 68$
Carbonic acid -	-	$40 \cdot 41$	$24 \cdot 39$	$24 \cdot 59$	18.18	1.79
Lime	-	$52 \cdot 29$	$31 \cdot 56$	31.82	$23 \cdot 52$	$2 \cdot 31$
Magnesia	-	0.32	$5 \cdot 44$	$1 \cdot 44$	$3\cdot 41$	$6 \cdot 08$
Phosphate of peroxide of	firon) (0.85	0.40	0.41	0.80
Phosphate of lime	-	{ 1.96 }	$17 \cdot 23$	$22 \cdot 04$	18.89	$35 \cdot 60$
Phosphate of magnesia	-) (0.93	0.02	$0 \cdot 10$	10.89
Organic matter -	-		1.86	2.80	$3 \cdot 01$	$10 \cdot 50$
Insoluble silica -	-	$3 \cdot 30$	$1 \cdot 45$. 1.60	$29 \cdot 00$	0.92
Coal	-	$1 \cdot 50$				$0 \cdot 31$
		99·91 S.	111·99 S.	115·14 S.	106·46 S.	106·83 S.

17. BLACK CHERRY (Cerasus serotina).

Tree sound. Diameter three feet from the ground, 36 inches; twelve feet from the ground, 30½ inches; seventy-two feet from the ground, 24 inches. From base to limbs, 72 feet; whole height, 104 feet. Average thickness of bark, ¾ inch. Number of layers twelve feet from the ground, 227. Average thickness of each layer, 0.06387 of an inch. The tree grew most rapidly during the first 100 years of its life; for the last 60 years, growth very slow.

				Bark.	Outside wood.	Heart wood.	Wood of twigs.
Potash -	-	-	-	0.07	$4 \cdot 93$	$5 \cdot 94$	$10 \cdot 12$
Soda -	-	-	-	8.86	$14 \cdot 39$	$12 \cdot 28$	$13 \cdot 77$
Chloride of sodi	um	-	-	0.08	$0 \cdot 12$	0.12	0.26
Sulphuric acid	-	-	-	1.91	$19 \cdot 25$	$2 \cdot 01$	$9 \cdot 76$
Carbonic acid	-	-	-	$26 \cdot 31$	$6 \cdot 75$	22.78	$10 \cdot 33$
Lime -	-	-	-	$46 \cdot 99$	21.67	$19 \cdot 12$	$8 \cdot 19$
Magnesia -	-	-	-	$1 \cdot 16$	$2 \cdot 24$	0.21	$4 \cdot 24$
Phosphate of pe	roxid	e of i	ron,	1.01	1.61	1.90	0.80
Phosphate of lin		-	-	5.85	12.70	$9 \cdot 94$	$25 \cdot 25$
Phosphate of ma	agnes	sia	-	0.84	$2 \cdot 44$	2.76	1.20
Organic matter	•	-	-	4.01	3.60	6.01	$7 \cdot 20$
Insoluble silica	-	-	-	$1 \cdot 20$	5.80	10.70	1.10
Coal	-	-	-	1.80	1.70	$5 \cdot 20$	$3 \cdot 20$
Moisture -	-	-	-		$3 \cdot 40$		3.30
				100·09 S.	100·60 S.	98·97 S.	$\frac{-}{98 \cdot 72}$

18. MISCELLANEOUS ANALYSES.

1. Leaves of the Iron-wood, collected September 30. Yard of the State House.

Silica -	-	-	-	-	-	-	10.500
EARTHY PHOSPHA	ATES:						
Phosphate of	of perc	oxide (of iro	n -		$5 \cdot 575$	
. Phosphate of	f lime	e -	-	-		0.719	
Phosphate o	of mag	gnesia			-	0.125	
Phosphate of	of silic	a -	-	-		$7 \cdot 325$	
Phosphoric	acid	-	-		-	11.581	
•							$25 \cdot 325$
Lime	-	-	-	-	-	-	$37 \cdot 484$
Magnesia -	-	-	-	-	-	-	0.075
Potash -	-	-	-	-	-	-	7.369
Soda	-	•	-	-	-		5.902
Chloride of sodi	um	-	-	-	-	-	1.900
Sulphuric acid	-	-	-	-	-	-	0.223
Carbonic acid	-	-	-	-	-	-	$10 \cdot 400$
Organic matter		-	-	-	-	-	2.850
							102·028 S.
		PROP	ORTIO	vs.			
Water -	-	•	-	-	-	-	54.680
Dry matter -	-	-	-	-	-	-	45.320
Ash	- '	-	-	-	-		4.260
Ash calculated d	lry-	-	-	-	-	-	9·399 S.

2. A species of foreign Rose-wood. Color dark, and much used in ornamental cabinet work.

Silica	-	-	-	-	-	-	$9 \cdot 475$
Earthy phospha	ates	-	-	-	-	-	$4 \cdot 150$
Lime		-	-	-	-	-	$61 \cdot 189$
Magnesia -	-	-	-	-	-	-	0.527
Potash -	-	-	-	-	-	e -	0.791
Soda	-	-	-	-	-	-	$4 \cdot 185$
Chloride of soc	dium	-	-	-	-	-	0.276
Sulphuric acid	-	-	-	-	-	-	0.051
Carbonic acid	-	-	-	-	-	-	$20 \cdot 205$
Organic matter		-	-	-	-	-	. 3.175
-							
							100·924 S

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3. Bark of Maple twigs.

Silica	_	-	-			-	1.25
Earthy phosphate	S	-		• /	•	-	18.00
Carbonate of lime	:	•				-	$53 \cdot 35$
Magnesia -	-	-	-	-	-	-	1.65
Potash -	-	•	-	•	•	-	$9 \cdot 45$
Soda	•	•	-	•	-	-	5.71
Sulphuric acid	-	-	-	-		-	$4 \cdot 15$
Chlorine -	-		-	-	-	-	trace.
Organic matter	-	-	-	-	•	-	1.80
							95.36

II. FRUIT TREES.

1. Plum (Prunus domestica).

Tree cut first of May.

PROPORTIONS.

Percentage water	-	-	Bark of root. 48 • 51	Wood of root. 44.64	Bark of limbs. 27 • 50	Wood of limbs. 20.33
Percentage dry wood	-	-	$51 \cdot 49$	$55 \cdot 36$	$72 \cdot 50$	$79 \cdot 67$
Percentage ash -	-	•	3·12 S.	0·24 S.	4·37 S.	0·38 S.
			ANATOR	e		

/		ANALYSIS,			
	Plum pits*.		Wood of root.	Bark of limbs.	Wood of limbs.
Potash	- 13·92	9.86	$\{40.31\}$	8.59	11.63
Soda	- 10.08	$6 \cdot 63$	1 -0 01 1	$19 \cdot 49$	
Chloride of sodium	- 2.25	$4 \cdot 22$	0.103	1.03	0.18
Sulphuric acid -	- 6.11	5.22	4.64	4.09	20.34
Carbonic acid (not deterr	nined).				
Lime	- 23.30	$22 \cdot 74$	$0 \cdot 17$	$39 \cdot 42$	$8 \cdot 12$
Magnesia	- 4.80	0.98	$0 \cdot 20$	$3 \cdot 76$	$6 \cdot 56$
Phosphate of peroxide of	iron,) (6.90	1.20	$2 \cdot 30$	0.60
Phosphate of lime -	- {8.00}	$7 \cdot 62$	31.98	7.50	$24 \cdot 99$
Phosphate of magnesia	_) ($3 \cdot 28$	$17 \cdot 12$	trace.	1.16
Organic matter -	- 6.65	1.76	$2 \cdot 50$	$1 \cdot 40$	4.60
Insoluble silica -	- 27.20	$21 \cdot 40$	1.80	8.40	0.70
Coal	-	$3 \cdot 60$	0.90		1.60
	102·31 S.	94·21 S.	100·923 S	. 95·98 S.	80·53 S.

^{*} This analysis was made with two grains of ash.

2. Peach (Amygdala persica).

Small seedling peach. Age of the tree, 23 years. Mean diameter, $3\frac{1}{2}$ inches. Thickness of bark, $\frac{1}{7}$ inch. Growth rather slow. Average thickness of each layer, 0.0699 of an inch.

	Bark of trunk.	Wood of trunk.	Bark of root.	Wood of root.	Leaves.	Pits*.	Bark of limbst.	Wood of limbst.
Potash	1.20	7.11	$3 \cdot 162$	8.58	$12 \cdot 41$	18.47	8.85	19.21
Soda	\$ 1.20	11.15	1.92	$15 \cdot 92$		5.21		8.11
Chloride sodium,	0.04	0.16	0.33	$5 \cdot 60$		2.70	0.28	0.24
Chloride potassium	,				0.36			
Sulphuric acid	$4 \cdot 19$	1.51	$3 \cdot 44$	0.58	$12 \cdot 12$	$15 \cdot 12$	$6 \cdot 18$	8.07
Carbonic acid (not	determin	ed).						
Lime	$42 \cdot 17$	$23 \cdot 26$	$38 \cdot 48$	0.11	14.77	16.80	31.98	$24 \cdot 64$
Magnesia -	$2 \cdot 16$	$6 \cdot 40$	2.91	0.01	8.00	$1 \cdot 33$	6.00	9.76
Phosph. perox. iron	, 0.45	0.32) (1.02	$2 \cdot 47$	$1 \cdot 33$	1.60	0.60
Phosphate lime,	18.79	$29 \cdot 19$	{ 10.40 }	$18 \cdot 10$	$10 \cdot 44$	17.98	8.50	$13 \cdot 20$
Phosphate magnes	ia, 0·01	1.34) (30.00	$3 \cdot 15$	0.05	0.20	0.20
Organic matter,	$3 \cdot 30$	$5 \cdot 20$	3.60	2.55	0.86	6.61	5.00	8.40
Insoluble silica,	$4 \cdot 15$	1.35	$9 \cdot 40$	$6 \cdot 46$	$6 \cdot 42$	$10 \cdot 00$	4.30	1.00
Coal			1.40		4.48		1.00	1.20
	109·04 S.	104·97 S.	$\overline{104.562}$	89·02 S.	86·85 S.	128·77 S.	99·03 S.	104·99 S.

Leaves of the Peach tree, July 22.

Carbonic acid	•	•	-	•	•	•	$13 \cdot 300$
Silicic acid	•	•	•	-	-	•	0.600
Phosphates	-	•	-	•	•	-	9.600
Lime	•	•	-	-	•	•	$16 \cdot 220$
Magnesia -	•	•		-	-	-	5.900
Potash -	•	-	•	-	-	-	14.280
Soda -	• ,		•	•	-	-	21.220
Chlorine -	•	-	• -	-	-	-	$5 \cdot 120$
Sulphuric acid	-	•		-	•	•	$4 \cdot 420$
Organic acids	•	-	-	-		•	7.900
							98·560 S.

^{*} Analysis made with two grains of ash.

[†] Peach limbs half an inch in diameter.

	Leave	s affe	cted	with	the	yellows	·	
Carbonic ac	id	-	-	•	-	•	•	$13 \cdot 200$
Silicic acid		•	-	•	-	-	•	0.800
Sulphuric ac	id	-	•	-		-	-	4.430
Phosphates		•	•	-		-	-	11.600
Lime -	-	-	•	-	-	-	-	$14 \cdot 300$
Magnesia	-	-	-	-	-	-	-	5.300
Potash	-	-	•	-	-	-		14.440
Soda -	-	-		-	-	-	-	$22 \cdot 280$
Chlorine	-	-	-	-	-	-	- /	4.740
Organic acid	s	-	-	-	-	-	-	4.300
								99·390 S.

3. Apple (Pyrus malus).

Sweet apple. Age of the tree, 19 years. Diameter of section taken for analysis, 6 inches. Thickness of the bark, \(\frac{1}{4}\) inch. Average thickness of each layer, 0.1447 of an inch.

Potash	-				Bark. 0 • 44	Outside wood. 3.288	Heart wood. $2\cdot 75$	Bark of root. 0.66	Wood of root. 15.07
Soda	-	-	-	-	$1 \cdot 53$	$3 \cdot 33$	1.62	11.38	21.99
Chloride	of s	sodiui	m	-	0.30	0.33	0.51	0.10	0.11
Sulphuric	ac:	id	-		$38 \cdot 39$	12.21	22.17	30.83	1.84
Carbonic	acio	d	-	-	$49 \cdot 56$	15.79	38.98		
Lime	-	-	-	-	1.86	15.56	2.66	1.00	11.64
Magnesia	ι	-		•	2.56	3.52 -	2.93	8.72	$0 \cdot 16$
Phosphat	e of	pero	xide o	of iro	n, 1		(0.72	0.91
Phosphat	e of	ˈ lime	•	-	3.60	$37 \cdot 50$	$24 \cdot 40 \left\langle \right.$	$6 \cdot 39$	13.96
Phosphat	e of	mag	nesia	-)		(31.35
Organic r	nati	ter		-	$3 \cdot 35$	$3 \cdot 20$	3.60	1.80	1.20
Insoluble	sili	ca	-	-	1.26	0.45	0.50	2.86	1.46
Coal	-	-	-	-	1.26	0.35	0.01	0.72	
					${104 \cdot 21}$	95.528	99.83	65·18 S.	99·69 S.

4. Rose Tree.

T. IUSE IREE.								
*					-		Bark.	Wood.
Potash	-	-	-	-	•	-	$5 \cdot 12$	11.60
Soda	-	-	- 1	-	-	-	8.52	5.99
Chloride of sodium	n	-	-	-	-	~	$3\cdot 20$	3.00
Sulphuric acid	-	-	í	-	•	-	$5 \cdot 00$	6.00
Carbonic acid -	-	-	-	-	-	-	$28 \cdot 79$	15.87
Lime	-	-	-	-	-	-	$22 \!\cdot\! 56$	10.46
Magnesia -	-	-	-	-	-	-	2.86	3.80
Phosphates -	-	-	-	-		-	15.30	31.00
Phosphate of iron	-	-	-	-	-	-	1.60	
Organic matter	-	•	-	-	-	-	$3 \cdot 30$	3.00
Silica	-	-	-	-	-	-	$2 \cdot 40$	$2 \cdot 30$
Coal	-	-	-	-	-	-	1.00	0.50
Moisture -	-	-	-	-			$2 \cdot 00$	1.00

5. MISCELLANEOUS ANALYSES.

1. Leaves of the	Lady Apple,	picked May 23.	Flowers just fallen.
------------------	-------------	----------------	----------------------

Carbonic acid	-	•	-	-	-	-	-	$8 \cdot 55$
Silicic acid	-	-	-	-	-	-		$4 \cdot 65$
Sulphuric acid		-	•	-	-	-	-	$10 \cdot 12$
Phosphates	•	•	-	-	-	-	-	26.60
Lime	-	-	-	-	-		-	$3 \cdot 38$
Magnesia	-	•	-	•	-	-	-	$2 \cdot 74$
Potash -	-	-	-	-	-	-	-	$27 \cdot 17$
Soda -	-	-	-	-	-	-	-	11.83
Chlorine	-	-	-	-	•	•	-	0.79
Organic acids		-	-	-	-	-		$3 \cdot 60$
								00 40 C

99·43 S.

2. Leaves of the Pear-tree, picked May 23. Flowers just fallen.

Carbonic ac	id	-	•	-	-	-	-	11.560
Silicic acid	-	-	-	-	-	-	-	1.750
Phosphates	-	-	-	-	•	-	-	$25 \cdot 050$
Lime -	-	-	-	-	-	•	-	4.715
Magnesia	•		-	-	-	-	-	4.500
Potash -	-	•	-		-	-	-	18.950
Soda -		•	-	-	-	-	-	$15 \cdot 190$
Sulphuric ac	cid, c	hlori	ne an	d orga	anic a	acids	not	
determine	d.			_				
								81.715 S

3. Leaves of the Oxheart Cherry, picked May 23. Flowers just fallen.

Carbonic aci	d	-	•	-	-	-	-	$11 \cdot 450$
Silicic acid	-	-	-	-	-	-	-	1.850
Phosphates	-	-	-	•	-	-	-	$26 \cdot 650$
Lime -	-	-	-		-	-	-	3.941
Magnesia	-	-	-	•	•	-	•	3.465
Potash -	-	-	-	-	•	-	-	$23 \cdot 757$
Soda -	-	-	-	-	-	-	-	$12 \cdot 365$
Sulphuric ac	id,	chlorin	ne an	d org	anic a	icids	not	
determine	d.			_				

83·478 S.

4.	Leaves o	f the	Earl	у На	rvest	Apple,	col	lected	Septem	ber 30.	Bearing fre
	Silica -	•	•	-	-	•	-	•	-	5.77	5
	EARTHY	PHOS	PHAT:	ES:							
	\mathbf{P} ho	spha	te of	pero	xide (of iron	1	-	4.875	i	
	Pho	spha	te of	lime				•	1.416		-
	\mathbf{Pho}	spha	te of	mag	nesia	-		-	trace.		
		ca -	-						5.125		
	Pho	spho	ric a	cid .	_		,	-	5.359)	
		•								16.77	5
	Lime -	,	-	-	-		-	•	-	36.39	8
	Magnesi	а			•		-			0.07	
	Potash -							_		13.17	
	Soda -		_	_	_	_	_	_	_	11.61	
	Chloride	of s	- odiur	n	_		_	_	_	0.06	
				11	•	-	•	•	•		
	Sulphuri			•	•	•	•	•	•	0.13	
٠	Carbonic			•	•	•	-	•	•	15.20	
	Organic	matt	er	•	•	•	-	•	-	2.85	0
									•	101.06	- 5 S
					PROP	RTIONS.					
	Water -		•	-	-	•	-	•	•	54.34	
	Dry -		-	•	-	•	-	•	•	45.65	
	Ash - Calculate		•	-	•	•	-	•	•	4·19 9·16	
٠.	Leaves of Silica -	•	•	•	•	-	•	•	-	4.25	ng fruit. O
					rida a	of iron			4.600		
								•			
		-	te of			•		•	7.559		
		-	te oi	mag	nesia	•		•	0.660		
	Silie			., -	•	•		•	0.450		
	Pho	spho	ric ac	id •	•	•		•	3.781		•
								-		16.55	
	Lime -		•	•	•	•	-	•	•	39.85	
	Magnesi		-	-	-	•	-	•	•	5.920	
	Potash -		•	•	•	-	-	•	•	8.79	3
	Soda -		-	-	•	•	-	•	-		
	Chlorine		•	-	•	-	-	•	-	0.55	4
•	Sulphuri	c aci	d	-	-	•	-	•	-	$4 \cdot 46$	4
	Carbonic			-	•	-	•	. ,	-	17:12	5
	Organic	matt	er	-	-	-	•	•		3.00	0
										100 · 509	- 9 S.
	Water -		_	_	PROPO	RTIONS.		_		56 · 138	3
	Dry -		-	-	•		-	_		43.86	
	Ash -			•	-	-			-	$3 \cdot 260$	O
	Calculate	ed dr	y	-	-	6	-	•		7.514	

6.	Leaves	of the	Larg	e Y ell	low	Spanish	h C	herry,	collecte	d Septemb	er 30.
	Silica				_	-				$4 \cdot 225$	
	Phosph	ates	-	-	-	-			-	37 · 175	
	-		-	-	-	-			-	21.957	
	Magnes	sia			-	_			-	$3 \cdot 195$	
	Potash		-		-	•		-	-	13.948	
			-			-		-	-	1.657	
	Chlorid	e of s	odiun	α		-		-	•	0.410	
	Sulphu			•	-	-	_	•	-	10.260	
	Organi				-	-			-	7.650	
	-6										
										100.577	S.
					PRO	PORTIONS.					
	Water	•	•	•	•	•	-	-	•	58.628	
	\mathbf{Dry}	•	•	•	-	-	-	•	-	41.372	
	Ash	•	•	•	-	•	-	-	•	$3 \cdot 434$	
	Calcula	ted dr	y	•	•	•	•	•	-	8.300	S.
7.	Leaves Silica	of the	Cata	wba G	rap	e, colle	cted	l Septe	mbe r 30.	Fruit al	oundant.
	EARTHY	T DUIGE			•	•	•	•	•	20 100	
		ospha						_	6.750	1	
		iospha							11.648		
		iospha iospha			2001			_	0.150		
		lica lica		mag.	1031	-			4.050		
		iospho						-	$6 \cdot 152$		
		iospno	iic ac	nu -		-		-	0 102	· 28·750	
	Lime		•						•	26.258	
	Magnes	sia				-			-	5.330	
	Potash					_		•	•	1.710	
	Soda		_			-				2.983	
	Chlorid	e of s	odiun	n	_		_	-		0.305	
	Sulphu			•					-	1.426	
	Carbon				_	-	_		-	8.900	
	Organi				-	-			-	$3 \cdot 450$	
	0										
										102.262	S.
					PRO	PORTIONS.		•			
	Water		•	•	-	-	•	•	-	11.169	
	Dry	•	-	-	-	-	-	•	-	28.831	
	Ash	•	•	-	-	-	-	•	-	2.282	
	Calcula	ited di	У	-	-	-	•	•	•	7 ·915	S.

8.	Leaves of	the	Catawba	Grape.	picked June 2.	Nearly full grown.	

Carbonic acid		-	-	-	-	-	3.050
Silicic acid	-	-	-	-	-	•	$29 \cdot 650$
Sulphuric acid	•	-	-	-	-		$2 \cdot 062$
Phosphates -	•	•	-	-	-	•	$32 \cdot 950$
Lime -	-	-	-	-	-	-	4.391
Magnesia -	-	-	_	-	-	-	1.740
Potash -	-	-	-	-	-	-	$13 \cdot 394$
Soda	-	-	-	-	-		9.698
Chlorine -	. •	-	-	-	-	-	0.741
Organic acids	`.	-	-	-	-		$2 \cdot 250$
3							
							99·926 S.

9. Fruit of the Black Walnut.

			Rind.	Shell.	Meat.	
Silica	-	-	$1 \cdot 35$	0.40	1.85	
Earthy phosphates	-	-	15.60	18.50	$40 \cdot 95$	
Carbonate of lime	-	-	23.75	5.60	0.10	,
Magnesia	•	-	$1 \cdot 55$	0.10	trace.	
Potash	-	•	$41 \cdot 43$	47.00	$22 \cdot 99$	
Soda	-	-	$7 \cdot 12$	10.21	4.98	
Sulphuric acid -	-	-	2.65	9.84	11.05	
Chlorine	-	-	1.60	$2 \cdot 15$	trace.	
Organic matter -	•	-	1.30	$5 \cdot 40$	5.00	
Alkaline phosphates	•	-			$9 \cdot 10$	
			96.35	$99 \cdot 20$	96.02	

10. Currant leaves and flowers.

Two hundred grains of the leaves gave 4.00 grs. of ash, and the same weight of the flowers gave 2.95 of ash; but the analysis was not finished. The leaves are particularly rich in soda and the phosphates. The analysis of the flowers was undertaken for the purpose of determining the amount of silica, an element which I have found rather abundant in floral organs, particularly the petals. In the ash of the blossom of the currant I found 9 per centum of silica, and 28 per centum of potash, a result which indicates the predominance of potash rather than soda in these organs.

N. B. It is due me to state, that in the following analyses having my initial, in the two preceding sheets, the carbonic acid was not obtained, viz: Analyses of the Hemlock and Sugar Maple, page 319; those of the Red Elm, p. 323; those of the Hickory, p. 325; those of the Black Birch, p. 326; and those of the Red Birch; p. 328. These are the only cases, in my analysis, where the carbonic acid is inserted without being actually determined: it was retained in these instances through mistake. In the following analyses having my initial, in the present sheet, the carbonic acid was not determined, and is not inserted, viz: Analyses of the Plum, p. 330; those of the Peach, top of p. 331; and of the root of the Apple tree, p. 332.

J. H. Salibbury.

ANALYSES OF THE ASH OF THE LEAVES, BARK AND WOOD OF THE CORNUS FLORIDA.

Made by	Dr.	GEORGE	HAND	SMITH,	of	Rochester.
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					Leaves.	Bark.	Wood.
Carbonic acid -	-	-	-	-	$17 \cdot 250$	$17 \cdot 479$	$15 \cdot 396$
Silicic acid -	-	-	-	-	4.850	4.7S5	$6 \cdot 200$
Phosphates -		-	-	-	24.850	18.500	31.850
Sulphuric acid	-	-	•	-	$3 \cdot 050$	1.750	1.925
Lime	-	-	•	-	33.493 =	$44 \cdot 470$	21.391
Magnesia -	-	-	-	-	1.240	$1 \cdot 150$	0.450
Potash	-	-	-	-	5.561	1.390	11.373
Soda	-	-	-	-	6.820	4.370	$5 \cdot 116$
Chlorine -	-	-	-	-	0.627	0.246	0.444
Organic acids -		-	-	-	$2 \cdot 150$	4.860	4.835
			1		99.892	${99.000}$	98.980

OBSERVATIONS ON THE PRECEDING ANALYSES.

It is obvious from the foregoing analyses of the ash of fruit and forest trees, that a large supply of inorganic matter is essential to a vigorous growth. In fruit trees this is particularly the case. These enjoy, in a far less degree, the power to recruit themselves from a supply by their own waste or debris. In a forest, the leaves and bark fall to the ground and decay, and in due time return to that state which fits them for food. In an orchard, however, the usual mode of managing the grounds prevents an accumulation of food in this way. Hence they are placed in the same position as other crops, as to their effect in removing the nutriment from the soil; and hence it is essential to their vigor, that a regular supply be furnished them. The substances which trees require are evidently calcareous and alkaline elements for the wood, and phosphatic and alkaline elements for the fruit. Lime exists in a large percentage in the bark. A compost of peat, lime and wood ashes, or the ordinary barnyard manures, will be always useful; and a large outlay in these matters will repay the expense in the quality and size of the fruit. Ample experience proves the great utility of this mode of treating fruit trees; and fruit trees which are neglected, and left to shift for themselves, will in time cease to grow, and will moreover be preyed upon by lichens and fungi.

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THE BEST TIME FOR CUTTING TIMBER, &c.

Experience has proved that trees for timber, if cut at one season of the year, are far more durable than if cut at another. Various reasons have been suggested why this is so, and it is not perhaps yet fully determined; still, as the time which experience has pointed out as the best for durability is during the autumn, it is generally supposed that this property is modified by the amount of sap in the trunk, and the maturity of the wood itself. In the spring, or at any early period of it, the trunk of most trees is pressed with the ascending sap. The leaves as yet are still folded in the bud, and the surfaces for exhalation are only sufficient to carry off very slowly the watery part of the sap. Even after the leaves have expanded, or until mid-summer has arrived, the tree abounds in juices. When, however, the dry and sultry summer has arrived, and the new wood and buds have been matured and formed, the watery part of the sap is mostly exhaled, and probably too the circulation is less active as the leaves become sere.

It is stated by Mr. Emerson, author of the valuable report on the trees and shrubs of Massachusetts, that the soft maple cut in September is three times more lasting than ash or walnut cut in the winter; and from numerous inquiries which he has made in various quarters, and from information obtained from reliable sources, it seems he has established the fact that autumn is the time for cutting timber. When it is determined to cut timber, it is of considerable importance to strip off the bark in the spring, that the body of the tree may dry during the summer. When, however, it is an object to reproduce a forest from the remaining stumps, the winter, or the very first of spring, is much more favorable to the growth of sprouts.

There are then two seasons for cutting wood: if it is expected to last, it must be cut the last of summer, or during the early part of autumn; if it is wished to clothe the surface with a new growth of trees, the cutting must be made late in winter. It is, however, possible to modify these arrangements: if, for example, the wood is designed for timber, if it is deprived of its bark in the spring, it may be allowed to stand and season till winter arrives, which is a period when farmers have less to do than in the summer or autumn.

In seasoning, wood retains an amount of water which may be regarded as its constitutional supply. This constitutional water is very important; for, upon its presence some of the most valuable properties of the wood depend. I refer to elasticity and strength. If wood, for example, is dried in a water bath at 212° till it ceases to lose weight, its elasticity and strength is very much diminished. Hickory, when dried in this way becomes as brittle as pine. In ordinary seasoning, or in steaming, I believe the strength of wood is not diminished. This observation may not be of much practical importance, as this last plan of seasoning is rarely followed. The amount of water varies, as will be observed, in different species of trees, as well as in herbaceous plants.

In another point of view, the amount of water is important to be known, for the difference between taking green and dry wood to market, as well as in its consuming, is very great; and so also, as ample experience proves, there is a material difference in burning green and dry wood. The quantity of water in the wood varies from 20 to 50 per centum, and probably the average amount will not differ much from 35 or 40 per centum. This water is not only of no use to the fire-wood, but it is prejudicial, as it must be dissipated by heat, in which act heat or caloric becomes latent and lost, especially if the wood is consumed upon a hearth or in a stove.

In addition to the effect of water in diminishing the combustibility of wood, the alkalies have also considerable influence of this kind. Elm, which is a potash wood, burns with less freedom than hickory, which contains much lime. It is, however, possible that the size of the pores of wood may modify its combustibility. Black oak is a notable instance of a slow and drizzling combustion: the pores are large and numerous, from which the watery sap continually oozes.

ADDITION TO CHAPTER VIII.

Analysis of several specimens of Salina salt, made for the purpose of comparing its purity with that of foreign salt; together with an analysis of the waste of the sediment, for the purpose of determining its value for manure.

1. Coarse solar salt obtained from a store in .	Albany.
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Chloride of sodium (pu	ılt)	-	-	-	$92 \cdot 980$	
Carbonate of lime -	-	-		-	-	0.010
Sulphate of lime -	-	-	-	_	-	1.315
Sulphate of magnesia	-	-	-	-	-	0.035
Sulphuric acid -	-	-	-	-		0.669
_						-
						95.019

2. Refined salt for the use of the dairy.

1. W. H. PORTER's brand.

Chloride of	sodiu	m	-	-	-	-	-	$97 \cdot 466$
Silica -	•	-	•	•		-	-	. 0.010
Sulphate of	lime	-	-	-	-	-	-	1.799
Lime -	-	-	-	-	**	-	-	0.058
Magnesia	-	-		-	-	-	-	0.085
								${99.415}$
								00 410

2. J. P. HASKIN'S brand.

Chloride of	sodiu	m	-	-	-	-		95.819
Silica -	-	-		•		-	-	0.020
Sulphate of	lime	6	•	-		•	•	1.753
Lime -	-	-	-	-	-	-	-	0.043
Magnesia	-			-	-	-	-	0.074
Water -		-	-	•			-	$1 \cdot 190$
								
								98.899

3. Brand of H. GREENMAN & Co., and of H. W. Nolton & Co. which is substantially the same.

Chloride of	sodiu	m	-	-	-	-	-	$95 \cdot 113$
Sulphate of	lime	-	-	-	-	•	-	0.402
Magnesia	-	-	-	-	-	-	-	0.120
Water -	-	-	-	-	-	-	-	1.500
								$97 \cdot 135$

4. Smith's refined dairy salt.

Chloride of sodiu	m	•	-	-	10	-	98.886
Chloride of magn	nesiui	m	-	-	-	-	$0 \cdot 139$
Sulphate of lime	-	•		-	-	-	0.375
Insoluble matter		-	•	-	-	-	trace.
Water		-	-	-	-	- (0.500
		-					99.900

This is probably as pure a manufacture as can be obtained. Taking out the water, there is but a trifle over one half per centum of what may be regarded as impurities; which amount, in reality, can have no prejudicial influence as a preserver of organic bodies. The main or most important quality is dryness, inasmuch as the power of salt to take from meat a part of its constitutional water depends upon this quality.

3. Analysis of the waste called Pan-scale.

Chloride of sodium	-	-				73.922
Chloride of lime -	•	-		-	•	$7 \cdot 469$
Chloride of magnesia	-	-	-	-	-	1.683
Sulphate of lime -	-	-		-	-	$12 \cdot 369$
Silica	-	-	-	-		0.200
Organic matter -	-	-	-	•	-	1.500
						074 4 4 0
						$97 \cdot 143$

It is evident from this analysis, that the waste of the salt-works, if this specimen is a fair example, is very valuable as a fertilizer, and hence ought to be saved. Indeed it seems that it might be redissolved, and the pure salt extracted. If used for a manure, it can be transported farther than gypsum, inasmuch as its composition is more valuable.

It is a matter of great importance to the farmers of New-York, to know that their own salt-works furnish a material equal to any in market, for curing meat and butter, and for other domestic purposes.

It is evident from the foregoing analyses, which were undertaken at the request of the Secretary of the New-York State Agricultural Society, that the salt of Salina contains no substance injurious to dairying purposes. The only source of danger is that the salt may be damp. Wet salt is entirely unsuitable for preserving animal substances; inasmuch as the principal operation of salt, as a preserver, is due to its power of absorbing water from the material to be preserved. Hence salt should be always thoroughly dried. If this rule is observed, in every case where it is doubtful as it regards the Onondaga salt of the brands given above, let it be dried. I may state that an individual has used this salt for 30 years in packing meat, without having in any case a spoilt barrel; and being a merchant, extensive experience in this way is certainly sufficient testimony to place the Onondaga salt with the best in market, for private and public purposes.

CONCLUSION.

The first volume relating to the agriculture of New-York treats of the soils and the constitution of the rocks, and their relations to each other. It was designed to prepare the way for the present volume. Although it is perhaps as full as could be expected, still much at this time might be done to improve it and render it more useful, or at least better adapted to the wants of farmers.

The statistics of agriculture are mostly omitted; and when introduced, it was for the purpose of comparing premium crops grown upon different soils, and in different parts of the State. This volume contains too much, perhaps, which relates to geology proper, but which was introduced for the purpose of giving an epitome of what was known upon the subject, this information being spread over the four volumes prepared by the gentlemen who had charge of this department of the Survey.

In the part relating mostly to geology, I introduced an extended account of the Taconic System. I felt justified in this course, for the reason that in this State these rocks present a highly important feature in its geology. It is not an inconsiderable part of the State over which these rocks prevail, and hence as a geological fact the existence of this system could not be passed over by one who was well satisfied as to his position; and although this position might be discredited, and a reputation for sagacity might be hazarded by a committal as to the fact, still I should not have felt justified in the mere statement of a conjecture or possibility. It is, however, proper to state, for the benefit of those persons who are unacquainted with the controversy upon this subject, that geologists are still divided upon the fact. Among those who have been the most strenuous in opposition to it, is my colleague Mr. James Hall; as he has made it the subject of remark in his volume of Palæontology, and has opposed the views which I maintained in the volume referred to, on the ground that the fossils which I there described were those which are common to the Hudson-river group, I deem it a duty I owe to myself to say that a committee of the American Association of Geologists and Naturalists has reported adversely to Mr. Hall's position. This removes thus far the supposed or theoretical objections to my views, and favors the position I had taken as it regards the reality of this ancient system of sedimentary rocks. Other assumptions in the volume of Palaeontology would be noticed here, were it not that I propose to resume the subject at a future time.

This volume, it is hoped, contains matter of a practical kind, in a form which has not been before accessible to the farmers of this country. The analyses, though numerous, have been conducted with as much care as it was possible to bestow; and I may state that all the weighings of products, without exception, have been to the hundredth of a grain.

conclusion. 343

The volume contains original analyses; and the composition of most of these substances had not been previously investigated in this country, except to a very limited extent. In preparing the work for publication, I could not feel satisfied with giving the results of others; although they may have been obtained by chemists before whom I could lay no claim to distinction, still our own soils and grains, our own products, were the ones which required this kind of labor. This view of the subject has always been entertained by myself; and though I felt an interest in the analyses which have been made in Europe, still they do not, as it appears to me, give that full information which we want. I have been, as will be seen, influenced by this view of the subject in taking up for analysis so many kinds of food, and other productions of the earth, and attempting thereby to furnish that kind of information which seemed so important respecting their composition. I have not attempted to deduce, from these analyses, all the inferences which may clearly be drawn from them. Hence they stand more as facts, which may be used in various ways for illustrating the economy of the vegetable kingdom.

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APPENDIX.

TEMPERATURE OF THE SOIL,

AS INDICATED BY A SERIES OF OBSERVATIONS AT ALBANY, N. Y., FOR TWENTY MONTHS, BEGINNING WITH MAY, 1847, AND ENDING WITH DECEMBER 1848.

THE importance of determining terrestrial temperature is felt more and more as advances are made in scientific agriculture and gardening. We have no space here for general observations on the utility of the subject.

Albany is in latitude 42 deg. 39 min. The observed mean temperature of the place is 48 deg. 47 min., at an elevation of 130 feet above tide. The elevation of the place where the following observations were made is about 100 feet above tide, in the open space in the rear of the old State House, where the morning sun shines early, and continues till the after part of the day, when the spot is shaded by the walls upon the west side of the area.

		MAY.		
•	OBSERVATIONS O	ON ATMOSPHERE, DE	CW, RAIN, AND WINI).
	5 o'clock a. m.	12 o'elock m.	3 o'elock p. m.	7 o'clock p. m.
DAY OF MONTH.	Air 4 feet above surface. Air 4 inches above surface, naked soil. Air 4 inches above surface, grass land. Dew. Rain.	Alr, 4 feet above surface. Alr, 4 inches above surface, naked soil. Alr, 4 inches above surface, grass land. Rain. Wind.	Air, 4 feet above surface. Air, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Rain.	Air, 4 feet above surface. Air, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Rain.
1 2 3 4 5 6 7	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	56 47 N. 45 46 S. 53 46 N. W 58 47 W. 62 48 W.	S. N. W 53 47 N. W 60 49 N. W 67 49 N. W	53 50 S. 46 46 S. 48 48 N. 56 50 N. 61 51 N. 62 53 N. 66 55 N.
ean,ighest,	45 48	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	56 ¹ / ₄ 50 55 46

Observations.—1. Morning clear. Afternoon few thin clouds. 2. Cloudy. Rainy. Moderate breeze. 3. Morning rainy. Afternoon thick clouds. Stiff n. w. breeze. 4. Heavy dew. Afternoon few clouds. Stiff n. w. breeze. 5. Heavy dew. Few clouds. Slight breeze. 6. Morning clear. Heavy dew. Afternoon few clouds. Moderate breeze. 7. Morning clear. Smoky. Heavy dew. Afternoon few clouds. Slight breeze.

		5 (o'clo	ck a	m.			12 (o'elo	ck m			3 o'c	elock	р. 1	m.		7 o'c	elock	p. n	a.
DAY OF MONTH.	Air, 4 feet above surface.	ches nake	Air, 4 inches above surface, grass land.	Dew.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Rain.	Wind.
8 9 10 11 12 13	50 54 52 58 56 56 53	54 53½ 55½ 57 55 55 55	53 53 55 57 57 57 59	57 55 49		N. N. N. W. S. W. S. N.		$ \begin{array}{r} 56\frac{1}{2} \\ 58 \\ 62 \\ 62 \\ 62\frac{1}{2} \\ 63 \end{array} $	54 58 58		N. W. S.	74	57 65 65 62 65 65	551		N. W. N. W. S. W. S. E. N.	60 68 68 69 66 64	57 63 62 64 63 62	54 57 59 62 59 ¹ / ₂ 60		N. V

Observations.—8. Cloudy. Commence raining at 5 a.m., and continued till 11 a.m. Afternoon cloudy. Moderate breeze. 9. Few clouds. Soil in good working order. Horse chesnut leaves 2 inches long. 10. Few thin clouds. Warm. Currants in flowers. 11. Few clouds. Moderate breeze. Heavy dew. 12. Heavy dew. Few clouds. Strong breeze. Variable. 13. Few thin clouds. Light dew. Moderate breeze.

										М	AY.		
	ов	SER	VAT	HOL	is o	n s	OIL	of	GR	ASS	LAN	D.	
	5 0'0	elock	a.m	12 c	cloc	k m.	7 0'0	elock	p.m	300	elock	p.m	
DAY OF MONTH.	Surfaee.	4 inches below surface.	9 inches below surface.	Surfaee.	1	9 inches below surface.	Surfaee.		9 inches below surface.	Surface.		9 inches below surface.	OBSERVATIONS.
14 15 16 17 18 19 20 21	42 46 41 53½ 52½ 47 50 53	53 53 53 49 55 55 53 54 55	55 55 55 53 56 55 55 56 56	80 72 74 76 66 72 80	64 62 66 65 60 60 64 57	57 57 58 57 57 56 59 56	79 80 75 88 68 75 78	68 70 67 73 62 61 77 77	57 60 62 58 58 57 58 58	64 65 62 66 58 64 64 64 60 63	63 63 62 65 58 58 58 58 52 55	563	Forenoon eldy Rainy Afternoon few elds Soil wet 1 in d'
Mean, Highest, . Lowest,	53	53½ 55 49	55 56 53	85 80 64	82 ₄ 66 62	57 59 56	773 86 68	$69\frac{3}{8}$ 77 61	593 62 58	623 66 58	61 65 58	573 59 57	

	OBSERVATIONS ON A	rmosphere, dew,	RAIN, AND WIND.	
	5 o'clock a.m.	12 o'eloek m.	3 o'clock p. m.	7 o'clock p. m.
DAY OF MONTH.	Air, 4 feet above surface. Air, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Dew. Rain.	Air, 4 feet above surface. Air, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Rain.	Air, 4 fect above surface. Air, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Rain.	Air, 4 feet above surface. Air, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Rain.
14 15 16 17 18 19 20 21 22	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{vmatrix} 74 & 74 & 71 & \dots & \text{s.w.} \\ 70 & 71 & 70 & \dots & \text{s.} \\ 75 & 65 & \dots & \dots & \text{n.} \end{vmatrix} $	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$
Mean, Highest, Lowest,	56 53 56	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	65 62½ 61 70 68 64 56 56 58

				_						I	MA	Υ.	
	ов	SER	VA	CION	is o	N S	OIL	OF	GRA	SS I	LAN	D.	
DAY OF	5 o'c	lock	a.m	12 oʻ	cloc	k m	3 o'e	lock	p m	7 o 'c	lock	рт	
MONTH.	Surface.	inches surfa	9 inches below surface.	urface.	4 inches below surface.	9 inches below surface.	urface.	inches surfa	9 inches below surface.	ırface.	inches surfa	9 inches below surface.	OBSERVATIONS.
23 24 25 26 27 28 29 30 31	57 59 58 61 43 41 533 54 48	56 58 58 61 49 54 57 55 52	56 57 57 60 54 56 58 58 58	65 72 88 59 65 74 78 57 58	59 64 69 58 58 63 66 57 56	57 58 66 58 56 58 62 58 55	64½ 60 86 70 81 91 58½ 54	64 70 61 71 72	58 60 62 58 58 62 58 54	63 62 66 60 70 70 55 52	61	58 60 62 58 58 63 63 58 54	Cloudy. S. breeze. Morn. slight shower, Soil wet 1 in deep Cloudy. Slight showers. Leaves and twigs of apple tree commence dying at Syracuse. Morning cloudy. Slight showers. Afternoon few heavy clouds Morning cloudy. Slight showers. Afternoon few clouds. Heavy dew. Few clouds. Slight breeze. Heavy dew. Few clouds. Slight breeze. Heavy dew. Few clouds. Soil in good working order. Morning cloudy. N. breeze. Afternoon few thin clouds. Morning few clouds. Afternoon cloudy. Rainy. Soil we
Mean, Highest, Lowest.	52½ 61 43	$ \begin{array}{r} 55\frac{1}{2} \\ 61 \\ 49 \end{array} $	563 60 54	68½ 88 57	61 ² 65 56	$68\frac{2}{3}$ 66 54	705 86 55	$64\frac{5}{8}$ 70 55	583 62 54	$62\frac{1}{3}$ $70\frac{1}{2}$ 52	$ \begin{array}{r} 61_{8}^{1} \\ 68 \\ 54 \end{array} $	59½ 63 54	at 7 p. m. 2 inches deep.

	OBS	ERV	ATI	ONS	ON	AT	MO	SPH1	ERE	, DE	w,	RAI	N, 1	AND	WI	ND.					
		5 0'	clock	A.	Μ.			12 o'	clock	м.			3 o'e	lock	p. m	1.	′	7 o'c	lock	р. п	1.
DAY OF MONTH.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Dew.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 fect above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above. surface, grass land.	Rain.	Wind.
23 24 25 26 27 28 29 30	59 63 58 64 44 50 53 49 48	59 63 58 63 42 48 53 50 46	$ \begin{vmatrix} 57 \\ 62 \\ 58 \\ 62 \\ 42 \\ 47 \\ 52 \\ 48 \\ 46 \\ 8 \end{vmatrix} $	40 46 52 48 45	57 62 57 58	N	70 76 54 65 74 78 55	64 70 78 56 64 74 78 55 56	64 70 62 54 64 72 77 55 56		S. S. E. S.	75 66 79 80	64 67 77 60 70 80 56 52	64 67 72 60 685 78 56 51		S.E. S.W S.W S.E. S.E.	71 53 62 74 74 56	64 62 66 51 60 69 70 54 51	62 62 62 65 58 67 68 58 58	52	S.: S.: S.: N S.:
Mean, Highest, Lowest,	. 543 . 64 . 44	53 <u>1</u> 63 46	62				65; 78 54	78 78 54	$\begin{array}{c c} 64\frac{1}{3} \\ 77 \\ 54 \end{array}$			66: 80 54		645 78 51			63 74 52	603 70 51	$\begin{bmatrix} 61\frac{1}{3} \\ 67 \\ 53 \end{bmatrix}$		

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ļ	ō oʻc	l'k a	. m.	12 oʻ	clock	m.	3 o'cl	'n p.	m.	7 o'c.	l'k p	. m.	5 o'c	l'k a	. m.	12 oʻ	clocl	ζm.	3 o'o	el'k p	. m.	7 o'c	ıl'k p	. m.
DAY OF MONTH.	ırface.	4 inches below the surface.	9 inches below surface.	Surface.		9 inches below surface.		4 inches below sur- face.	g inches below sur- face.		4 inches below sur- face.	9 inches below sur- face.		4 inches below surface.	9 inches below sur- face.			9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.
1 2 3 4 5 6 7	52 57 52 60 53 54 54	54 58 57 60 57 58 59	55 58 56 60 58 59 60	69 81 74 73 77 72 75	63 65 67 64 62 66 63	57 58 60 60 60 60 60	67 76 66 76 74 75 75	63 70 65 68 70 66 72	58 62 60 60 61 60 64	63 62 64 62 61 64 64	63 64 64 63 64 65 68	58 61 61 60 61 60 65	52 56 51 59 50 52 50	53 56 54 58 54 54 54 55	54 56 56 58 58 57 57	68 74 74 73 69 71 76	69 64 64 62 59 66 62	55 58 58 59 58 59 59 61	68 80 66 80 77 75 80	63 69 63 67 66 67 70	57 59 58 60 60 59 64	63 60 63 62 58 64 64	62 64 62 62 62 64 63	58 60 59 60 59 59 59 62
Mean, Highest Lowest,	545 60 52	57\\\ 60\\ 54	58 60 55	$ \begin{array}{c c} 74\frac{3}{8} \\ 81 \\ 69 \end{array} $	64.1 67 62	594 60 57	723 76 66	673 72 63	60^{3}_{4} 64 58	$62\frac{7}{8}$ 64 61	64 ³ 68 63	60 ² 65 58	52 ₈ 59 50	54 7 58 53	565 58 54	72½ 76 68	633 69 59	58.1 61 55	75 80 66	663 70 63	$59\frac{5}{64}$ 57	62 64 58	623 64 62	598 62 58

	5	oʻcloc	ka.ı	m.		-	l2 o'	cloc	kт.		3	o'e	lock	p. m	1.		7 o'c	lock	p. n	1.
	Air, 4 fect above surface. Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Dew.	Rain.	J.	face.	4 menes face, nake	Air, 4 inches above surface, grass land.	Rain.	i	4 feet abor	enes nake	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	2° 4	Air, 4 inches above surface, grass land.	Rain.	Wind.
1 2 3 4 5 6 7	52 52 58 57 52 49 60 60 50 49 53 52 52 49	49 61	49	58 S.	E. V. V. W. W. J. W.	64 68 71 70 64 68 73	65 70 70 69 63 68 74	64 69 70 69 63 68 74		S.	68 69 68 70 65 74 74	68 70 66 70 66 72 76	69 66 70 65		S. W. S. W. W. W. N. W	66 62 56	64 60 65 62 55 64 64	64 58 64 62 55 62 64		S. S. N. W
Mean, Highest, Lowest,	60 60	52 ³ / ₈ 61 48				$66\frac{7}{8}$ 73 64	67 74 63	$66\frac{5}{8}$ 74 63		· · · · · · · · · · · · · · · · · · ·	$ \begin{array}{c c} 69\frac{3}{4} \\ 74 \\ 65 \end{array} $	$ \begin{array}{c c} 69^{3}_{4} \\ 76 \\ 66 \end{array} $	$687 \\ 74 \\ 65$			$64\frac{1}{8}$ 68 56	62 65 55	$61\frac{1}{4}$ 64 55		

Observations.—1. Morning cloudy. Slight shower. Soil wet 4 inches deep. Occasional sunshine during day. Slight wind. 2. Morning cloudy. A fine thunder shower last night. Soil very wet. At 7 p. m. clear moderate breeze. 3. Morning few clouds. Afternoon cloudy. Moderate wind. 4. Morning cloudy. Rainy. Soil wet 9 inches deep. Afternoon few clouds. Warm. The leaves of the Button wood (Platanus occidentalis) begin to die. 5. Few floating clouds. Slight breeze. Pleasant. 6. Few floating clouds. Moderate breeze. Pleasant. Dew. The leaves of the Thorn bush (Cratægus coccinea) begin to die. 7. Clear. Slight breeze. Warm. Light dew.

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	5 oʻ	cloc	k a.ın	12 o	'cloc	k m	3 0'0	elock	a m	70'0	clock	p.m	5 0'0	lock	p.m	12 o	'cloc	k m.	3 0'0	elock	p.m	70'0	elock	рт
DAY OF MONTH.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.
8 9 10 11 12 13 14	58 58 68 66 60 56 62	60 59 65 66 62 58 62	61 62 64 65 62 59 63	74 68 65 68 64 77 73	63 64 71 66 62 70 65	66 64 67 64 62 66 63	65 72 83 69 73 78 69	66 69 72 66 66 70 67	65 65 68 65 62 66 63	64 68 72 67 62 66 60	65 68 72 67 64 66 65	64 68 64 63 64 63	56 57 68 65 58 52 61	59 58 63 65 58 55 60	58 58 64 64 60 58 60	74 67 85 68 63 77 71	62 62 74 66 60 67 64	59 69 70 64 60 63 61	67 76 84 69 73 78 70	64 64 73 66 66 68 66	61 61 70 63 60 63 62	64 68 72 67 59 66 60	64 66 72 66 63 64 64	61 62 66 63 62 61 62
Mcan, Highest Lowest,	68	60 66 58	$ \begin{array}{r} 60\frac{1}{2} \\ 65 \\ 59 \end{array} $	69 77 65	66 71 62	$\begin{array}{r} 64\frac{1}{2} \\ 67 \\ 62 \end{array}$	698 83 65	65 72 66	$ \begin{array}{r} \\ 627 \\ 66 \\ 62 \end{array} $	67 72 62	$ \begin{array}{r} 65\frac{1}{3} \\ 72 \\ 64 \end{array} $		595 68 52	593 65 55	60 64 58	72 85 63	65 74 60	635 70 59	737 84 67	78 73 64	$\begin{array}{c} 62\frac{7}{8} \\ 70 \\ 60 \end{array}$	65 72 60	65 ₈ 72 63	$62\frac{3}{66}$ 66 61

	OBSERVATIONS ON A	TMOSPHERE, DEW,	RAIN, AND WIND.	
	5 o'clock a. m.	12 o'clock m.	3 o'clock p. m.	7 o'clock p. m.
DAY OF MONTH.	Ali, 4 feet above surface. Ali, 4 inches above surface, nalted soil. Ali, 4 inches above surface, grass land. Dew. Rain.	Air, 4 feet above surface. Air, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Rain.	su 44 jac	Air, 4 feet above surface. Air, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Rain.
8 9 10 11 12 13 14	60 56 58 58 57 s. 58 58 57 s. 71 70 70 s. s. 68 68 68 s. s. 59 58 58 s. s.w 54 54 54 53 s. 62 61 61 s.e.	74	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$
Mean, Highest, Lowest,	71 70 70	69 70 703 82 84 82 59 59 59	$ \begin{array}{c c c c c c c c c c c c c c c c c c c $	65 66 65 71 76 71 56 56 56

Observations.—8. Cloudy in forenoon. Afternoon showery. South wind. 9. Night rainy. Forenoon cloudy. Strong S. wind. Afternoon few clouds. Moderate breeze. The leaves of the Iron wood (Ostrya virginica) and Black Walnut (Juglans nigra) begin to die near Albany. 10. Few clouds. Strong S. breeze. 11. Cloudy. Strong S. breeze. Towards night showery. The leaves and small twigs of Apple, Pear and Quince begin to die near Albany. 12. Forenoon cloudy. Towards night clear. Moderate breeze. 13. Morning clear. Moderate dew. Afternoon cloudy at 7 p. m. Slight shower. 14. Showery during night. Heavy thunder shower about 2 p. m. At 7 p. m. thick clouds. Strong W. breeze. Cool.

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		BSE)		T10	NS (ON S	01L	OF	GR.	ASS	LAI	VD.		0	BSEI	RVA	T10	ns (ON I	NAK	ED	SOII	۱.	
	5 o'c	lk a	. m.	12 oʻ	cloc	kт.	7 oʻc	lk p	. m.	3 o'	elk p	. m.	5 o'c	ılk a	. m.	12 oʻ	cloc	km.	3 0,0	elk p	. m.	7 o'c	lk p	. m.
DAY OF			sar-		sar-	sur-	1	- 1	sur-		sar-	sar-		. 1	sar-		sur-	sar-		sur-	sur-		sur-	sur-
MONTH.		s below face.	s below face.		below face.	s below face.	•	below faee.	below face.		below face.	below face.		below face.	below face.		face.	below face.		below facc.	below face.		below face.	below face.
	Surface.		9 inches		4 inches	9 inches		اتنه	9 inches		4 inches	9 inches below face.	Surface.		9 inches	t	4 inches	9 inches below face.	- 1	-	9 inches	Surface.	4 inches	9 inches
15 16	56.	57 55	60 57	56. 76	58 64	59 58	56	58	58	55 58	57 62	58 60	50 50	54 53	59 56	55 74	55 60	56 56	53	55	57	52 57	53 60	56 58
17 18	55 56	57 60	59 61	78: 74:	64 66	61 62	74 78	70 72	62 65	66	67	65	52 53	54 56	56 59	74 71	60 68	58 59	80 83	73 75	59 60	65	66	63
19 20	61 62	62 60	62 60	62 68	62 63	61	61 66	64 64	61 62	59 64	60 64	61 62	60	60 58	60 58	61 68	60 64	60 60	60 66	60 64	60 60	58 64	59 64	60 61
21 22	62 64	62 64	61 62	70	68	64	73 82	68 72	62 63	68 70	67 70	64 66	61 64	61 62	60 62	70	64	62	73 82	70 73	64 65	68 69	68 70	63 64
Mean, Highest Lowest,	64	59 ² / ₃ 64 55	601 62 57	60½ 78 56	555 68 58	533 64 58	$61\frac{1}{4}$ 82 56	58½ 72 58	54½ 65 58	70	55 70 57	$ \begin{array}{r} 54\frac{1}{2} \\ 66 \\ 58 \end{array} $	$ \begin{array}{r} 56_{4}^{1} \\ 64 \\ 50 \end{array} $	57 ¹ / ₄ 62 53	58 ³ / ₄ 62 56	59½ 74 55	537 68 55	513 62 56	$62\frac{1}{8}$ 83 53	883 75 55	53½ 65 57	$ \begin{array}{r} 54\frac{1}{8} \\ 69 \\ 52 \end{array} $	55 70 53	$ \begin{array}{r} 53\frac{1}{8} \\ 64 \\ 56 \end{array} $

		clock a. m.			12 o'c	lock n	n.		3 0,610	ock p. 1	n.	ĺ	7 o'c	lock	p. n	a.
DAY OF MONTH.	1 . •	Air, 4 inches above surface, grass land. Dew.	1 75	Air, 4 feet above surface.	face, nake	۱ ا	Ġ.	, 4 feet surface.	Air, 4 inches above surface, naked soil. Air, 4 inches above	surface, grass land. Rain.	Wind.		4 inches ace, nake	Air, 4 inches above surface, grass land.	Rain.	Wind.
15 16 17 18 19 20 21 22	48 48 53 53 54 53 54 52 61 60 62 61 62 62 64 64	48 53 53 .52 53 .50 60 61 61 62 62 61 64	W. W. N. N. S.	52 66 70 74 62	$\begin{array}{c c c} 52 & 52 \\ 67 & 66 \\ 72 & 73 \\ 74 & 73 \\ 61 & 66 \\ 66 & 66 \end{array}$	2 5 1 6 73	W. S. W. N. E. S. E.	52 72 76 63	52 5 73 78 62 66 66 66 66 66 66 6	74 76 66 52 57 66 73	W. N. E. S. S. E.	60 67	51 59 66 59 65 67 66	51 59 66 59 65 67		W.S.S.S.S.

OBSERVATIONS.—15. Cloudy. Strong W. wind. Cool. 16. Morning few clouds. Strong W. breeze. Afternoon clear. Moderate breeze. 17. Morning clear. Light dew. Afternoon few clouds. Moderate breeze. 18. Morning clear. Afternoon few clouds. Light dew. Slight breeze. 19. Cloudy. Showery. Strong wind. 20. Cloudy. Showery. Strong breeze. 21. Cloudy. Showery. Thunder shower at 12 m. 22. Floating clouds. Soil very wet. Warm.

											JU	NE.												
	ОВ	SER	VA'I	TIOI	1S O	N S	OIL	OF (GRA	SS I	LAN	D.		0.	BSEI	RVA	TIO	NS (ON I	NAK	ED	soI	L.	
	5 0'0	elock	a.m	12 o	cloc.	km.	3 0'0	lock	p.m	7 o'c	elock	рm	5 o'c	lock	a.m	12 o	cloc	k m	3 0'0	lock	p.m	7 0'0	elock	p,n
DAY OF MONTH.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	inches surfac	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	inches surfac	9 inches below surface.
23 24 25 26 27 28 29 30	64 62 64 69 68 67 69 64	64 64 63 69 68 68 70 66	64 64 66 68 66 68 69 67	74 81 84 82 78 87 68 72	69 68 74 71 70 76 68 70	65 66 68 68 68 71 68 68	82 84 85 78 86 93 67 84	74 76 76 75 78 80 68 74	67 66 70 71 72 74 68 68	72 72 76 72 74 76 66 68	72 72 74 70 74 76 67 70	66 68 69 68 70 72 68 68	62 60 62 68 66 67 68 61	62 62 64 67 66 67 68 64	62 63 65 66 64 67 68 66	71 81 85 83 80 88 68 72	66 66 73 72 70 74 68 70	63 64 67 69 66 69 67 66	86 85 88 78 88 94 67 93	76 75 80 75 84 30 68 76	64 64 69 69 67 73 67 66	71 72 76 72 74 77 66 68	71 72 76 72 76 76 76 76 67 71	66 69 68 69 71 67 68
Mean, . Highest Lowest,	65_{8}^{7} 69 44	$ \begin{array}{r} 66\frac{1}{2} \\ 70 \\ 63 \end{array} $	66½ 69 64	78 87 68	$ \begin{array}{r} 70\frac{3}{4} \\ 76 \\ 68 \end{array} $	673 71 65	823 93 67	75½ 80 68	69½ 74 66	72 76 66	$71\frac{7}{8}$ 76 67	$ \begin{array}{r} 68\frac{5}{8} \\ 72 \\ 66 \end{array} $	441 68 60	65 68 62	$ \begin{array}{r} 65\frac{1}{8} \\ 68 \\ 62 \end{array} $	$78\frac{1}{2}$ 66 68	698 74 66	66 ³ 69 63	843 94 67	76 84 68	$ \begin{array}{r} 67\frac{3}{8} \\ 73 \\ 64 \end{array} $	78 77 66	$72\frac{5}{8}$ 76 71	68 71 66

	OBSERVATIONS	ON ATMOSPHERE, D	EW, RAIN, AND WIN	ID.
	5 o'clock a. m.	12 o'clock m.	3 o'clock p. m.	7 o'clock p. m.
	Air, 4 feet above surface. Air, 4 inches above surface, nalced soil. Air, 4 inches above surface, grass land. Dew. Rain.	Air, 4 feet above surface. Air, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Rain. Wind.	Air, 4 feet above surface. Air, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Rain. Wind.	Air, 4 feet above surface. Surface. Air, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Rain.
23 24 25 26 27 28 29 30	62 62 62 62 N. 63 60 60 58 N. E. 64 63 63 62 N. 70 69 69 68 S. W. 72 70 70 66 W. 68 68 68 66 N. 60 61 61 61 N. W.	84 84 84 84 85 84 82 83 85 86 82 80 80 80 80 84 84 84 84 84 66 66 66 64 N	. 82 82 82 S. W. 83 84 83 W.	78 79 77 S. W. 75 74 73 S. 78 74 73 S. W. 77 75 75 S. W.
Mean, Highest, Lowest,	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

											JU	LY.												
		OI	BSER	VA	TIG	NS C	N G	RAS	SS L.	ΛNΙ	ο,			0	BSE	RVA	TIC	NS	ON	NΛΙ	KED	soi	L.	
	5 oʻ	elk. a	a. m.	12 o	'eloc	ek m.	30'0	elk.	p. m.	7 o'	elk I	o. m.	5 o'	elk a	ı. m.	120	cloc	k m.	3 0'0	elk p	. m	700	ılk p	. m.
DAY OF MONTH.	Surface.		9 inches below surface.		-	9 inches below surface.	Surface	4	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.		9 inches below sur- face.	Surface.	4 inches below surface.	9 inches below surface.
$egin{array}{c} 1 \\ 2 \\ 3 \\ 4 \\ 5 \\ 6 \\ 7 \\ \end{array}$	62 64 65 67 64 66 68	64 65 66 68 67 69 68	66 66 68 67 68 68	73 79 79 78 85 81 80	68 72 71 70 72 73 72	69 68 69 68 70 69 69	74 78 84 82 88 80 84	72 73 77 76 78 73 75	68 68 70 69 70 70 70	69 70 76 75 75 72 75	69 71 75 72 73 73 74	67 68 69 68 70 70 70	58 60 65 67 62 67 66	62 64 65 67 66 68 68	65 65 65 68 66 68 67	73 82 79 78 86 85 86	65 72 71 69 73 70 74	66 68 69 68 71 69 70	87 85 96 99 99 95 98	72 74 89 75 76 74 77	68 69 69 69 71 71 70	69 72 76 78 78 78 73 78	70 71 76 74 74 74 75	67 68 69 68 70 70 70
Mcan, Highest Lowest	65 67 63	$ \begin{array}{r} 66\frac{5}{8} \\ 69 \\ 64 \end{array} $	67 68 66	79.1 85 73	$71\frac{1}{8}$ 73 68	685 70 68	70 88 74	75 77 72	693 70 68	73 75 69	70 75 69	69 70 67	64 67 58	55 68 62	66 68 65	79 ³ 86 73	70½ 78 65	68§ 71 66	793 99 87	763 89 72	695 71 68	75 78 69	73 76 70	69 70 67

		5 o	'clo	ek a.	m.			12 o	cloc	kт.			3 0'0	clock	p. n	1.		7 oʻc	lock	p. n	1.
	4 feet ab surface.	4 inches a ace, nake	Air, 4 inches above surface, grass land.	Dew.	Rain.	Wind.	4 feet abourface.	4 inches ace, nake	Air, 4 inches above surface, grass land.	Rain.	Wind.	4 <u>1</u>	4 inches ace, nake	Air, 4 inches above surface, grass land.	Rain.	÷.	Air, 4 feet above surface.	ches	Air, 4 inches above surface, grass land.	Rain.	Wind.
1 2 3 4 5 6 7	59 65 66 68 62 67 67	58 64 66 68 62 67 67	58 64 64 66 66 66	56 62 62 67 60 66 66		N. N.E. N.E. N. N.	76 80 80 83 84 84 87	76 80 80 81 85 84 86	76 78 89 81 84 83 84		N. N. E N. E. N. E. N.	78 80 85 85 88 86 87	78 80 86 87 90 86 86	76 78 86 85 89 85 84		N. N. S. N. N.	72 73 77 78 78 78 74 79	69 70 76 76 77 73 77	68 69 76 75 75 72 75		I
Mean, Highest, Lowest,	68	70 68 58	65 66 58	$62\frac{5}{8}$ 67 56			82 87 76	815 86 76	803 84 76			84 87 78	825 90 78	83 ¹ 89 76			75 79 72	74 77 69	$72\frac{5}{8}$ 76 68		

Observations.—1. Few clouds. N. breeze. Leaves of Ostrya virginica and Acsculus hippocastanum continuc to die. 2. Very few thin clouds. Moderate breeze. Light dew. 3. Few thin clouds. Moderate breeze. Light dew. 4. Morning clear. Afternoon few thin clouds. Light breeze. Light dew. 5. Very few thin clouds. Light breeze. Light dew. Warm. 6. Clear. N. breeze. Light dew. 7. Morning clear. Afternoon very few thin clouds. N. breeze. Light dew.

											JU	LY.												
	C	BSE	RV	TIC	NS	ON :	soli	OF	' GR	ASS	LA	ND.		0	BSE	RV	TIC	NS	ON	NAF	ED	soi	L.	
	5 o'	elk a	a. m.	12 o	cloe	k m.	3,00	elk p	. m.	7 o	elk p). m.	5 o'	elk a	a. m.	12 o	'eloc	km.	3 o	elk p	. m.	7 0'0	elk p	. m
D. W. OF		sur-	Sur-		sur-	-Jns		-ins	sur-		sar-	sur-		sur-	sur-		sur-	sur-		sur-	sur-		-Jns	sur-
DAY OF MONTH.	Surface.	4 inches below face.	9 inches below face.	Surface.	4 inches below face.	9 inches below face.	Surface.	4 inches below face.	9 inches below face.	Surface.	4 inches below face.	9 inches below face.	Surface.	4 inches below face.	9 inches below face.	Surface.	4 inches below face.	9 inches below face.	Surface.	4 inches below face.	9 inches below face.	Surface.	4 inches below face.	9 inches below face.
8 9 10 11 12 13	68 69 70 74 75 70	69 70 72 74 74 74 72	69 70 72 73 73 73 72 70	90 89 86 96 92 90	73 78 75 78 78 78 80	70 71 73 76 76 76	82 84 94 89 90 78 86	76 78 80 80 79 76	70 72 75 76 76 73	77 78 82 81 78 74	76 76 76 79 78 76	70 72 75 76 76 76 73	65 69 68 74 76 69	68 69 72 74 75 71	68 69 71 73 74 72	94 104 91 103 98 100	77 82 76 85 84 84	69 71 73 77 77 76	105 94 104 100 100 78	76 98 92 86 86 76	70 72 76 77 77 77 73	84 81 86 85 80 74	76 80 86 85 84 75 76	70 73 76 77 77 73
14 15	64 63	69 68	70	90 90	77 75	71 71	81	79 78	74 72	73 73	76 76	72 72	64 58	67 65	69 69	91 96	80 78	71 72	92 88	84 82	74 74	72 73	76	72 72
Mean, Highest Lowest,	$69\frac{1}{8}$ 63	$\begin{vmatrix} 71 \\ 74 \\ 68 \end{vmatrix}$	$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	$91\frac{3}{8}$ 96 86	76_{3}^{3} 80 73	73 76 70	$85\frac{1}{2}$ 94 78	78^{1}_{4} 80 76	$73\frac{1}{2}$ 76 70	77 82 73	76§ 79 76	$ \begin{array}{r} 731 \\ 76 \\ 70 \end{array} $	68 76 58	$ \begin{array}{r} 70\frac{1}{8} \\ 75 \\ 65 \end{array} $	$70\frac{5}{8}$	$97\frac{1}{8}$ 104 91	$ \begin{array}{r} 80^{3}_{4} \\ 85 \\ 76 \end{array} $		94 105 78	85 98 76	$74\frac{1}{8}$ 77 70	69_{4}^{1} 86 72	793 86 75	735 77 70

10 68 66 66 66 S. E. 82 83 82 S. E. 86 88 88 S. 78 77 11 73 74 74 E. 85 89 87 S. 84 84 82 S. E. 80 80 79			12 o'clock m.	1	3 o'clocl	k p.	m.		7 0 0	elock p.	m.
8 67 65 64 64 N. 87 89 88 S. 89 90 87 S. E. 84 83 80 89 69 69 68 67 N. 88 90 88 S. 88 88 87 S. 81 80 78 10 68 66 66 66 66 S. E. 82 83 82 S. E. 86 88 88 S. 78 77 77 78 77 78 77 78 7	4 feet surface	A the other	ches nake ches grass	A foot	4 feet a surface. 4 inches acc, nake 4 inches acc, grass acc, grass acc, grass		Wind.	, 4 feet surface.	, 4 inches a face, naked	l inches	Wind.
$\begin{array}{cccccccccccccccccccccccccccccccccccc$	E. 87 88 82 85 83 83 76		89 88 S. 90 88 S. 83 82 S. E 89 87 S. 89 88 E. 87 87 N. W. 77 76 N.		89 90 87 88 88 87 86 88 88 84 84 82 84 86 86 74 73 73 76 78 78	73	S. E. S. S. S. E. S. N. W.	84 81 78 80 74 74 70	83 80 78 80 74 74 74 70	80 78 77 79 73 74 73 69	S. I S. I S. I

OBSERVATIONS.—8. Clear. Smoky. Slight dcw. Moderate breeze. 9. Morning clear. Light dcw. Afternoon few hazy clouds. Smoky. Light breeze. 10. Light dcw. Few hazy clouds. Smoky. Light breeze. 11. Morning floating clouds. Light breeze. Warm 7 p. m. Cloudy. Stiff'S. E. breeze. 12. Forenoon moderately cloudy. Slight breeze. Warm. 7 p. m. cloudy. Signs of rain. 13. Cloudy. Shower last night. Wet soil 1 inch dccp. Heavy thunder shower between 1 and 2 p. m. Wet soil 8 inches dccp. Moderate breeze. 14. Clear N. breeze. Soil wet from 8 to 10 inches dccp. Shower during night. Few hazy clouds. Light dcw. Moderate breeze. Smoky.

											JU	LY.												
	OI	BSEF	RVA'	TIOI	NS C)N S	OIL	OF	GRA	SS	LAN	D.		О	BSE	RVA	TIO	NS (ON :	NAK	ED	so11	և.	
	50'0	clock	a.m	12 o	cloc	km.	3 0 '0	elock	p.m	70'0	clock	рт	500	elock	a.m	12 o	`cloc	k m	300	clock	p.m	70'0	elock	p,m
DAY OF MONTH.	Surface.	4 inches below surface.	9 inehes below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surfaee.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surfaec.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.
16 17 18 19 20 21 22 23	65 69 73 74 75 74 73 69	68 71 74 75 76 75 74 71	70 72 73 74 76 75 74 73	88 90 90 99 102 80 84 92	66 78 77 82 83 75 78 79	72 73 74 76 77 75 75 75	87 90 86 86 84 76 80 92	77 80 80 84 86 76 78 82	73 75 75 77 79 75 76 76	77 81 81 82 76 75 76 76	76 79 80 80 83 75 76 78	73 75 76 77 79 75 75 76	62 67 73 72 74 73 73 65	67 70 74 74 75 74 72 68	69 71 73 74 75 74 72 72	100 104 97 106 106 80 85 104	82 78 86 92 76 81	73 74 74 76 77 74 74 74	97 99 90 86 86 76 78 102	84 88 84 94 86 76 78 84	73 76 75 77 79 74 75 75	78 84 82 80 76 74 74 76	82 82 80 82 82 74 76 80	73 76 76 76 77 79 74 74 75
Mean, Highest Lowest,	71‡ 75 65	73 76 68	733 76 70	905 102 83	783 88 66	$\begin{vmatrix} 74\frac{1}{8} \\ 77 \\ 72\frac{1}{2} \end{vmatrix}$	92	803 86 76	753 79 75	78 82 75	783 83 75	75 3 79 73	$ \begin{array}{r} 69_{8}^{7} \\ 74 \\ 62 \end{array} $	713 75 67		973 106 80	83 <u>5</u> 92 76	$74\frac{1}{2}$ 77 73	89.1 102 76	84 ¹ 94 76	$75\frac{1}{2}$ 79 73	78 84 74	$\begin{array}{ c c }\hline 79\frac{1}{2} \\ 82 \\ 74 \\ \end{array}$	$75\frac{1}{2}$ 79 73

					1	12 () C10	ck m	•		3 O C.	lock	р. т			7 o'c	clock	(P.	aı.
MONTH.	1,00	Air, 4 inches above surface, grass land.	Dew. Rain.	Wind.	, 4 feet al surfaee.	4 inches ace, nake	Air, 4 inches above surface, grass land.	Rain.	Wind.	4 s	4 in	surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Rain.	Wind.
16 17 18 19 20 21 22 23	$\begin{array}{c cccc} 62 & 60 \\ 67 & 67 \\ 72 & 72 \\ 75 & 74 \\ 76 & 76 \\ 72 & 72 \\ 75 & 75 \\ 66 & 66 \\ \end{array}$	60 72 74 75	59	N.E. S.E. S. S.E. S.E. S.	81 84 89 90 78 82 80	83 88 87 92 96 78 82 84	82 87 86 91 95 78 82 83	70	S. E S. E S. S. S. S. S. W.	83 86 81 81 76 74 77 81	84 87 82 81 77 74 78 84	86 81 79 76 74 77	79 77 70 74	s. s.w. s. s. s. s.	77 80 80 84 75 74 73 73	76 80 80 81 75 74 73 73	74 79 79 84 75 74 73 73	75	S. S. E S. E S. S. S. S. N

Observations.—16. Clear, Smoky, Light breeze. 17. Few heavy clouds. Strong breeze. 18. Morning, few heavy clouds. Afternoon cloudy. Thunder shower. 19. Morning few hazy clouds. Afternoon cloudy. Showery. Soil wet $1\frac{1}{4}$ inches deep. 20. Morning, floating clouds. Thunder shower between 2 and 3 p. m. 7 p. m. showery. 21. Cloudy. Showery. Soil wet $\frac{3}{4}$ inches deep. Strong breeze. 22. Morning, thick clouds, strong S. breeze. Afternoon showery. 23. Few clouds. Smoky, north breeze. Warm.

											JU1	LY.												
	01	BSEI	RVA	TIO	NS O	N S	OIL	oF	GRA	ss :	LAN	D.		0]	BSEI	RVA	TIO	NS C)N N	IAK	ED :	SOIL	٠.	
	5 0'0	el'k a	. m.	12 o	cloc	кm.	3 o'c	l'k p	. m.	7 o'c	el'k p	o. m.	5 0'0	l'k a	. m.	12 oʻ	c1oc	k m.	3 0'0	:1'k p). m.	7 o'c	1'k p	. m.
DAY OF MONTH.	Surface.	4 inches below the surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.		9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.		9 inches below surface.
24 25 26 27 28 29 30 31	68 72 74 62 59 59 62 68	71 74 73 66 66 62 64 66	72 74 73 68 68 66 66 68 68	94 80 74 68 82 84 80 82	78 75 73 72 70 76 73 72	74 74 73 68 69 70 70 68	88 80 69 84 87 82 70 86	83 76 73 75 77 77 72 77	76 74 74 73 72 72 70 69	80 76 64 66 63 72 68 82	78 76 71 73 70 76 71 77	76 74 72 73 71 73 70 69	66 72 74 56 56 57 61 68	70 73 72 62 63 64 63 66	72 74 72 67 68 68 68 68	106 83 74 90 93 92 82 85	80 76 74 73 71 80 75 75	74 74 73 68 69 71 71 69	96 82 69 92 92 88 70 88	85 77 72 78 80 81 73 80	77 74 73 73 72 73 71 70	82 76 64 65 64 74 68 82	80 76 70 72 72 76 72 78	77 74 72 72 73 74 71 70
Mean, . Highest Lowest,		67 ² 74 62	$ \begin{array}{r} 69\frac{5}{8} \\ 74 \\ 66 \end{array} $	$ \begin{array}{r} 80\frac{1}{2} \\ 94 \\ 68 \end{array} $	$73\frac{5}{8}$ 78 70	70^{3}_{4} 74 68	80 ₄ ³ 88 69	$76\frac{1}{2}$ 83 72	72½ 76 69	712 82 63	74 78 70	72 ¹ 76 69	63 ³ 74 56	$66\frac{5}{8}$ 73 62	69_{8}^{5} 74 67	$ \begin{array}{r} 88\frac{1}{8} \\ 106 \\ 74 \end{array} $	75\\ 80 71	$71\frac{1}{8}$ 74 68	84 96 69	784 85 72	73 77 70	72 82 64	$\begin{array}{ c c } 74\frac{1}{2} \\ 80 \\ 70 \\ \end{array}$	73 77 70

		5 o	cloc	k a.	m.			12 o'	cloc	cm.			3 oʻc	lock	p. m	۱.	1	7 o'c	lock	p. n	1.
DAY OF MONTH.	su	4 inches ace, nake	Air, 4 inches above surface, grass land.	Dew.	Rain.	d.	Air, 4 fect above surface.		Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above. surface, grass land.	Rain.	Wind
24 25 26 27 28 29 30	66 72 75 57 56 58 68 69	66 72 75 57 56 58 68 69	66 71 75 57 56 55 68 69	62 53		N. S. S. N. W. S. E. S. S.		87 78 68 70 74 70 74 76	86 78 68 68 73 75 73 75		S. S. N.W N.W S. W	7 69 7 72	87 79 65 72 74 76 66 81	86 78 65 70 73 75 66 60	62 64	S. S. N. N. S. E.	80 76 62 63 64 69 65 74	78 76 62 62 62 69 65 74	78 75 62 61 64 68 65 74	73	

Observations.—24. Morning clear. Light dew. Afternoon few clouds. Moderate breeze. 25. Forenoon cloudy, Strong south breeze. Afternoon slight showers. 26. Cloudy, showery, moderate breeze. 27. Few clouds, moderate breeze, cool morning. 28. Few heavy clouds. Morning cool. Moderate breeze. 29. Morning clear. Heavy dew. Cool. Afternoon cloudy, showery. Moderate breeze. 30. Morning cloudy, showery. Soil wet 12—14 inches deep. Afternoon few clouds.

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	ов	SER	VAT	TION	s o	n sc)IL	OF (- GRA	ss I	AN	D.		0	BSE	RVA	TIO	ns (ON I	NAK	ED s	SOII	٠.	
	5 0 0	elock	a m	12 o	cloc	k m	3 o'c	lock	a m	7 o'c	lock	p.m	5 o'c	loek	рm	12 o	cloc	kт.	3 0 '0	lock	p.m	7 o'c	lock	p m
DAY OF MONTH.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.		9 inches below surface.
1 2 3 4 5 6 7	72 67 68 66 68 67 69	70 67 69 68 69 68 70	70 70 70 70 70 70 70 70	86 86 86 90 86 84 69	76 77 74 77 75 75 75	72 72 71 72 71 72 71 72 71	87 82 82 86 83 79 69	76 77 76 79 76 76 76 76	72 72 72 73 73 73 73 71	74 72 73 74 74 75 69	75 74 75 76 75 76 76 70	72 72 72 73 73 73 73 71	72 64 60 64 65 63 68	70 66 67 67 68 67 69	70 69 68 69 70 70 70	86 87 88 92 90 88 68	76 78 76 80 77 77 69	72 72 70 73 72 73 70	88 86 82 88 84 82 68	77 79 78 81 78 78 69	72 72 73 74 74 74 84 70	74 70 72 74 73 76 68	75 74 74 76 75 77 69	72 72 73 74 74 74 74 70
Mean, Highest Lowest,	72	$68\frac{1}{70}$ 67	70 71 71	84 90 69	75 77 70	71 72 71	81 87 69	$75\frac{1}{2}$ 72 70	72 73 71	73 74 69	74 76 74	72 73 71	$63\frac{1}{72}$ 64	69 70 66	69 70 68	85 92 68	76 80 89	72 73 70	82 88 68	77 81 69	74 84 72	$72\frac{1}{2}$ 72 68	74 75 69	72 74 72

		5 o'c	lock a	a. m.			12 c	o'clo	ck m	١.		3 o'c	lock	p. n	1.		7 o'c	lock	р. п	a.
DAY OF MONTH.	Air, 4 feet above surface.	face, nake	surface, grass land. Dew.	Rain.	÷	, 4 feet surface.	face, nake	Air, 4 inches above surface, grass land.	Rain.	d.	, 4 feet surface.	face, nake	Air, 4 inches above surface, grass land.	Rain.	ď.	, 4 fect surface.	4 inches acc, nake	Air, 4 inches above surface, grass land.	Rain.	Wind.
1 2 3 4 5 6 7	66 65 65 64 64	72 7 66 6 65 6 65 6 64 6 64 6 67 6	$ \begin{array}{c cccc} 1 & & & \\ 5 & 62 \\ 62 & 62 \\ 4 & 62 \\ 4 & 62 \\ & & & \\ & & & \\ & & & \\ \end{array} $		S. N.W. W. N.E. N.E. N.W.	80 76 76 80 78 78 78 68	82 80 78 82 82 80 68	81 76 78 81 81 79 68	66	S. W. E. N.E. N.E. N.W.	81 77 76 80 80	83 79 77 82 81 77 68	82 78 77 81 80 76 68	68	S. W. E. N.E. N.E. N.W.	74 69 72 72 72 72	74 68 70 70 70 70 74 67	74 68 70 70 70 74 67	67	S.I W S.
Iean, Iighest, owest,	72	$ \begin{array}{c cccc} 67 & 6 \\ 72 & 7 \\ 64 & 6 \end{array} $	1	1		76 80 68	79 82 68	77½ 81 68			77 81 68	78 83 68	77 82 68			71 74 67	62 74 67	52 74 67		

Observations.—Floating clouds. Slight sprinkling between 7 and 8 a.m. 2. Few floating clouds. Moderate dew. 3. Few heavy clouds near the horizon. Heavy dew. 4. Few floating clouds. Heavy dews. 5. During forenoon few heavy clouds. Afternoon thick clouds. Moderate dew. 6. Cloudy in morning, heavy thin clouds at noon. Thick floating clouds at night. 7. Cloudy. Slight sprinkling of rain in morning. Afternoon stormy.

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	OB	SER	VAT	ION	is o	n s	OIL	of	GRA	ASS :	LAN	D.		О	BSE	RVA	TIO	NS (ON I	NAK	ED :	soii	٠.	
	5 0'0	elock	a.m	12 o	cloc	k m.	3 0 %	elock	p.m	7 o'c	lock	р.т	5 o'	el'k a	a. m.	12 o	'cloc	ek m.	30%	el'k j). m.	7 0'0	el'k p	. m.
DAY OF MONTH.	Surface.	4 inches below surface.	9 inches below surface.	Surface.		9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.		9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surfaee.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surfaec.	4 inches below surface.	9 inches below surface.
8 9 10 11 12 13 14 15	68 69 72 74 72 71 72 72 72	68 69 72 74 73 73 73 72 73	69 70 71 73 74 74 73 74	75 78 89 92 86 87 87	72 72 76 81 78 78 78 78 80	70 70 72 74 74 74 75 75	74 76 84 93 86 86 86 88	72 72 78 82 82 82 79 83	70 71 74 76 76 76 76 75 76	72 75 78 78 75 76 73 76	72 72 77 80 78 78 77 78	70 71 74 76 76 76 76 75 76	68 69 72 74 72 69 70 70	67 69 71 74 72 70 72 72	68 69 70 73 73 73 73 73 73	75 79 90 94 88 92 89 92	72 74 80 85 79 80 80 81	70 70 73 74 74 74 75 75	74 77 85 96 90 90 88 93	72 74 82 84 81 84 83 86	70 72 75 76 76 77 76 77	72 76 78 78 74 78 78 78 76	72 74 78 80 78 79 78 78	70 72 75 76 76 76 76 76 76
Mean, . Highest Lowest,	71.1 74 68	713 74 68	721 74 69	85± 92 75	768 81 72	713 75 70	84½ 93 74	78 ³ 83 72	74.1 76 70	765 78 78 72	76½ 80 72	$74\frac{1}{76}$ 76 70	$70\frac{1}{2}$ 74 68	$ \begin{array}{r} 807 \\ 74 \\ 67 \end{array} $	$71\frac{1}{2}$ 73 68	87 ³ / ₈ 91 75	787 85 72	$73\frac{1}{8}$ 75 70	81§ 96 74	$81\frac{1}{8}$ 86 72	75 77 70	76^{1}_{4} 78 72	$77\frac{1}{8}$ 80 72	745 76 70

		5 c	'elo	ck a.	m.			12 o	'cloc	ek m			3 o' e!	lock	p. n	١.		7 o'c	loek	p. m	۱.
DAY OF MONTH.	2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2. 2				Wind.	4 feet above	4 inches ace, nake	Air, 4 inehes above surface, grass land.	Rain.	Wind.	4 feet abor face.	4 inches ace, nake	All, 4 inches above surface, grass land.	Rain.	Wind.	4 feet abor face.	4 inches ace, nake	Air, 4 inches above surface, grass land.	Rain.	Wind.	
8 9 10 11 12 13 14 15	69	67 69	67 69	66	67	N.E. N.E. S.E. S. W. W. N.W.	73 78 84 85 80 82 83 82	74 79 86 88 81 84 86 86	75 78 85 87 80 83 85 85		W. E. S. N. N.E. S. S.	73 77 85 87 80 84 82 84	73 77 85 89 82 84 84 86	73 76 84 88 81 84 83 85		N.E. S. E. S. W. E. S. E.	72 76 78 73 74 74 76 75	72 76 78 73 72 74 75 75	72 75 77 73 72 74 74 74	73	N.E S.E S. N. I W. N.E S. S. E
Iean, Iighest,	75	70 ³ / ₈ 75 67	704 74 67				807 89 73	83 88 74	824 87 75			$81\frac{1}{2}$ 87 73	82½ 89 73	813 88 73			74^{3}_{78} 78 72	$74\frac{3}{8}$ 78 72	737 77 72		

Observations.—8. Cloudy. Moderate breeze. 9. Cloudy. Moderate breeze. 10. Few thick floating clouds. Stiff south breeze in afternoon. 11. Few thin clouds in the forenoon. Clouds increased in the afternoon. Light shower at 5 o'clock p. m. Light breeze. 12. Few thin clouds, moderate breeze. Pleasant. 13. Heavy dew. Clear. ight breeze. 14. Few heavy clouds. Light breeze. 15. Few heavy clouds. Light breeze.

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DAY OF MONTH.	Surface.	4 inches below surface.	9 inches below surface.	Surface.		9 inches below surface.	Surface.		9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below
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Mean, Highest, Lowest.	$\begin{bmatrix} 64\frac{1}{4} \\ 72 \\ 60 \end{bmatrix}$	66 ³ 73 63	70 74 67	$ \begin{array}{r} 76\frac{3}{8} \\ 82 \\ 67 \end{array} $	72 <u>5</u> 76 70	$ 70^{3}_{4} $ $ 74 $ $ 69 $	$77\frac{3}{4}$ 85 72	$74\frac{1}{8}$ 77 70	$71\frac{3}{75}$ 70	72 77 68	723 76 69	$72\frac{1}{8}$ 76 69	64 72 60	$ \begin{array}{r} \hline 67\frac{1}{8} \\ 73 \\ 62 \end{array} $	69 73 66	797 86 75	74 77 72	$ \begin{array}{r} 70\frac{5}{8} \\ 74 \\ 69 \end{array} $	783 87 71	$75\frac{5}{80}$ 80 70	72½ 75 70	$70\frac{1}{2}$ 77 64	73½ 78 68	7: 7: 6

21 64 64 64 60 S. 76 76 76 S.W. 76 77 76 S.W. 70 70 69 22 62 62 62 57 S. 75 75 75 S. 73 74 74 S. 70 70 70 23 60 60 58 55 N. 70 72 70 N.W	ONTH			5	oʻelo	oek a	. m.			12 o	'cloo	ek m		3	o'c	lock	p. m	ı.		7 o'e	lock	p. n	1.
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Observations.—16. Morning cloudy. Middle of day few hazy clouds. At 7 p. m. cloudy, moderate breeze. 17. Morning cloudy, rest of day few hazy clouds. Moderate breeze. 18. Thunder shower during night. Thick floating clouds, stiff breeze. 19. Morning cloudy, light shower. Middle of day, few clouds. 7 p. m. clear, moderate breeze. 20. Forepart of day few clouds. At 4 p. m. a moderate shower. Light breeze, heavy dew. 21. Few clouds. Moderate breeze, light dew. 22. Few clouds, light breeze, moderate dew. 23. Morning clear, heavy dew. In afternoon few clouds, moderate breeze. 24. Morning clear, heavy dew. Afternoon few clouds, light breeze. 25. Clear north breeze. From the 25th to the 31st the weather was very moderate, about like the 20th, 21st, 22d and 23d. Most of time clear. Moderate north breeze. No rain of any consequence. Moderate dews. Nights pleasant.

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	0]	BSEF	RVA'	rior	vs o	N S	OIL	OF	GRA	ss i	LAN	D.		0	BSE	RVA	TIO	NS (ON :	NAK	ED	soi	L.	
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DAY OF MONTH.	Surface.	4 inches below the surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.		4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	1. 1	9 inches below surface.
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Mean, Highest Lowest,	66 70 60	664 70 60	673 70 66	803 83 75	73 ⁷ / ₈ 76 71	703 74 69	$46\frac{7}{80}$	$45\frac{7}{8}$ 76 71	$ \begin{array}{r} 44\frac{1}{2} \\ 74 \\ 69 \end{array} $	72 ₄ 76 66	$72\frac{3}{8}$ 76 70	$71\frac{1}{8}$ 74 70	66 70 60	66½ 70 60	67 ¹ / ₄ 70 ⁴ 65	82 ₄ 85 75	$74\frac{1}{8}$ 76 72	$71\frac{1}{74}$ 74 70	463 80 71	46 76 71	443 74 70	72½ 76 70	725 76 70	784 74 70

	5	o'ele	oek a	. m.			12 o	'cloe	ek m			3 0'0	eloek	р. 1	n.	7	o'el	oek 1). m	,
DAY OF MONTH.	Air, 4 feet above surface. Air, 4 inches above surface naled soil	ches ab	Dew.	Rain.	d.	4 fect ab surface.	ન જૂ∣	Air, 4 inches above surface, grass land.	Rain.	Wind.	_	4 inches ace, nake	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 fect above surface.	4 inches	Air, 4 inches above surface, grass land.	Rain.	Wind.
1 2 3 4 5 6 7 8	$ \begin{array}{c cccc} 65 & 65 \\ 64 & 64 \\ 70 & 70 \\ 71 & 71 \\ 74 & 74 \\ 66 & 66 \\ 61 & 61 \\ 64 & 64 \\ \end{array} $	64 70 70 74 66 61			N. N. W. S. N. S. N. N. S. S.	78 76 79 76 80 70 74 73	77 88 80 78 80 70 70 73	76 77 80 77 80 70 70 74 73		N. N. S. N. S. N. E.	76 80 71 74 71	76 80 71 75 71	76 80 71 74 71		N. S. N. E. S.	74 73 78 72 76 67 70 70	74 73 78 72 76 67 70 70	78 72 76 67 70		N S N S V E S
Mean, Highest, Lowest,	74 74	74				75 ³ 80 70	77 80 70	757 80 70			$46\frac{1}{2}$ 80 71	$46\frac{5}{80}$	$ \begin{array}{r} 46\frac{1}{2} \\ 80 \\ 71 \end{array} $			$ \begin{array}{c c} 72\frac{1}{2} \\ 78 \\ 70 \end{array} $	$72\frac{1}{2}$ 78 70	78		

OBSERVATIONS—1. Clear N. breeze. 2. Clear N. breeze. Smoky. 3. Morning cloudy. Afternoon clear. Moderate S. breeze. Warm. 4. Few hazy clouds. N. breeze. Smoky. 5. Thick floating clouds. Strong S. breeze. 6. Rained during night. Thick floating clouds. Moderate breeze. 7. Morning clear. Afternoon few clouds. Moderate breeze. 8. Cloudy. Moderate S. breeze.

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	ов	SER	VAT	TIOI	is o	N S	OIL	OF (GRA	SS I	LAN	D.		0	BSE	$\mathbf{RV}I$	TIC	NS (ON I	NAK	ED	SOII	٠.	
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DAY OF MONTH.	Surfaee.	4 inches below surface.	9 inches below surfaee.	Surface.	inches surfae	9 inches below surfaec.	urface.	surfac	9 inches below surface.		4 inches below surface.	9 inches below surface.		4 inches below surface.	y inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inehes below surface.	9 inches below surface.
9 10 11 12 13 14 15 16	73 61 60 62 54 59 52 48	70 63 60 62 63 61 53 51	70 65 63 64 65 63 59 59	74 67 69 66 73 62 68 69	73 68 66 66 70 63 63 65	70 67 64 64 65 64 62 61	70 67 68 67 71 59 60 70	72 68 61 66 70 60 60 67	70 67 66 65 65 62 61 63	65 62 64 65 66 56 55 58	69 64 64 64 66 59 58 60	70 66 66 65 67 61 60 63	73 61 60 62 64 59 52 48	70 62 60 62 63 61 53 51	70 64 62 64 64 63 58 58	75 67 69 67 73 62 69 70	74 68 66 66 70 63 63 65	70 66 64 64 65 64 62 62	70 67 66 67 70 59 60 71	72 60 66 66 70 60 60 68	70 66 65 65 65 62 61 64	65 62 64 65 66 56 55 58	69 64 64 64 66 59 58 60	70 65 65 65 66 61 60 63
Mean, . Highest Lowest,	58 ₈ 73 48	$60\frac{3}{8}$ 70 51	64 70 59	$ \begin{array}{ c c c c c c c c c c c c c c c c c c c$	663 73 63	$ \begin{array}{r} $	$66\frac{1}{2}$ 71 59	$65\frac{1}{2}$ 72 60	$64\frac{5}{8}$ 70 61	$61\frac{5}{8}$ 66 55	63 69 58	$\begin{array}{r} 64_8^5 \\ 70 \\ 60 \end{array}$	598 73 48	603 70 51	627 70 58	69 75 62	667 74 63	$\begin{array}{c} 64\frac{5}{8} \\ 70 \\ 62 \end{array}$	66 71 60	66 72 60	643 70 61	61 66 55	63 69 58	64 70 60

4		5 (o'elo	ck a	. m.			12 c	o'clo	ck m		;	3 o'c	lock	р. п	١.	,	7 o'c	lock	p. r	n.
DAY OF MONTH.	, 4 feet a surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Dew.	Rain.	d.	4 feet a surface.	4 inches ace, nake	Air, 4 inches above surface, grass land.	Rain.	નું	4 fect al surface.	4 inehes acc, nake	Air, 4 inches above surface, grass land.	Rain.	٠;	Air, 4 feet above surface.	4 fac	Air, 4 inches above surface, grass land.	Rain.	Wind.
9 10 11 12 13 14 15 16	76 62 60 62 64 58 52 49	76 62 60 62 64 58 52 49	76 62 60 62 64 58 52 49	46 44		S. N. N. N. N. W. N. S.	70 66 68 66 70 61 60 66	70 66 68 66 70 61 60 66	70 66 68 66 70 61 60 66		S.W. N. N. N. N. W. W. W.	65 66 68 67 68 60 62 67	65 66 68 67 68 60 62 67	65 66 68 67 68 60 62 67	65	S.W. N. N. N. N. W. W. W.	62 64 64 65 66 54 57 60	62 64 64 65 66 54 57 60	62 64 64 65 66 54 57 60	60	N. N. N. W. W. W. N. V.
Mean, Lighest, Lowest,	76	$ \begin{array}{r} 603 \\ 76 \\ 49 \end{array} $	$ \begin{array}{r} 60\frac{3}{8} \\ 76 \\ 49 \end{array} $				$ \begin{array}{r} 65\frac{7}{8} \\ 70 \\ 60 \end{array} $	65 ⁵ / ₈ 70 60	657 70 60			65 ³ / ₈ 68 60	653 68 60	65 ³ / ₆₈ 68			$ \begin{array}{r} $	$ \begin{array}{r} 68\frac{1}{2} \\ 66 \\ 54 \end{array} $	$61\frac{1}{66}$ 54		

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	OH	BSER	VAT	TION	is o	N S	OIL	OF (GRA	ss i	AN	D.		O	BSE	RVA	TIO	NS (ON I	NAK	ED	son	L.	
	5 0'0	elk a	. m.	12 oʻ	cloc	k m.	3 0'0	elk p	. m.	7 o	elk 1). m.	5 o'	elk a	ı. m.	12 o	'cloc	km.	3 0%	elk p). m.	7 o'	elk p	. m.
DAY OF MONTH.	Surface.	4	9 inches below surface.	Surfaee.		9 inches below surface.			9 inches below surface.	Surface.	4 inehcs below surface.	9 inche	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches	Surface.	4 inches below surface.	9 inches below surface.	Surface.		9 inches below surface.
17 18 19 20 21 22 23	50 56 60 54 59 52 52	51 57 59 56 58 64 53	57 61 61 59 60 59 58	80 70 62 68 71 79 74	63 61 62 64 66 66 70	61 62 62 62 62 64 63	70 68 61 65 72 79 66	66 67 61 62 66 70 65	61 64 60 62 62 65 64	61 62 62 62 62 62 62 62	61 65 62 62 63 65 63	62 64 62 62 62 64 63	50 56 60 54 59 52 52	51 57 58 56 58 54 53	57 60 60 58 60 58 58 58	82 71 65 69 72 81 80	64 62 63 65 66 67 71	62 62 62 62 62 64 63	71 68 62 65 72 81 66	66 67 61 62 65 71 64	62 64 60 62 62 65 64	61 62 62 62 62 61 62	61 64 62 62 63 64 63	62 64 62 62 62 63 63
Mean, Highest Lowest	54 ³ / ₄ 60 50	55 8 59 51	$ \begin{array}{c c} 59\frac{1}{8} \\ 61 \\ 57 \end{array} $	72 80 62	645 70 61	62^{1}_{4} 64 61	$\begin{array}{r} 68\frac{5}{79} \\ 79 \\ 61 \end{array}$	$\begin{array}{c} 65 {}^{1}_{4} \\ 70 \\ 61 \end{array}$	$62\frac{5}{65}$ 65 60	613 62 61	63 65 61	$\begin{array}{c} 62\frac{5}{8} \\ 64 \\ 62 \end{array}$	$ \begin{array}{r} 54\frac{5}{8} \\ 60 \\ 52 \end{array} $	55 ₄ 58 51	585 58 57	74 ¹ / ₄ 82 65	$\begin{array}{c c} 65\frac{3}{8} \\ 71 \\ 62 \end{array}$	$\begin{array}{c} 62\frac{3}{8} \\ 64 \\ 62 \end{array}$	69 ¹ / ₄ 81 62	$65\frac{1}{8}$ 71 61	$ \begin{array}{r} 65\frac{5}{8} \\ 65 \\ 60 \end{array} $	$\begin{array}{c} 61\frac{5}{8} \\ 62 \\ 61 \end{array}$	$\begin{array}{r} 62\frac{5}{8} \\ 64 \\ 61 \end{array}$	$62\frac{5}{64}$ 62

		5 0	`clo	ck a.	m.			12 o	cloc	kт.		:	3 o'c	lock	р. п	١.	7	o'cl	ock	p. m	•
DAY OF MONTH.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Dew.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	4 inches icc, nake	Air, 4 inches above surface, grass land.	Rain.	Wind.
17 18 19 20 21 22 23	50 58 62 57 62 56 48	50 58 62 57 62 56 48	50 57 62 57 62 56 48	45 56 50 48		N. N. S. E. N. E. W. N.	67 69 64 64 68 69 70	68 69 64 64 68 71 74	67 69 62 64 68 70 72		NE N. N.	69 68 62 63 68 69 71	69 68 62 63 68 72 69	69 68 61 63 68 70 68		S. NE N. W.	64 66 63 62 64 66 66	64 66 63 62 63 66 66	63 65 62 62 63 64 61	••••	S N
Mean,	62	$ \begin{array}{r} 56\frac{1}{8} \\ 62 \\ 50 \end{array} $	56 62 48				$ \begin{array}{r} 673 \\ 70 \\ 64 \end{array} $	$68\frac{3}{8}$ 74 64	67 8 72 63			66 71 62	674 72 62	665 70 61			00	$ \begin{array}{r} $	623 65 61		• • •

Observations—17. Clear. Fresh N. breeze. 18. Few thin clouds. Moderate breeze. 19. Morning few thin clouds. Afternoon cloudy. Moderate breeze. 20. Morning, few thin clouds. Afternoon cloudy. Moderate breeze. 21. Morning few thin clouds. Afternoon thick clouds. Strong N. breeze. 22. Clear. Moderate breeze. Heavy dew. 23. Morning foggy. Clear. Heavy dew. 7 p. m. few clouds. S. breeze.

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	О	BSEI	RVA	TIO	NS C	N S	OIL	OF	GRA	ss I	LAN.	D.		0	BSE	RVA	.TIO	NS (ON I	NAK	ED S	SOIL		
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MONTH.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.
24 25 26 27 28 29 30	59 57 54 56 60 53 53	59 57 55 56 60 54 54	61 60 57 58 58 58 58	64 60 59 71 77 69 58	61 60 57 63 68 61 58	60 60 58 58 60 60 58	60 60 58 68 77 69 58	62 60 58 63 71 63 58	62 60 58 58 61 61 58	62 56 57 60 64 60 55	62 59 58 63 68 61 56	62 60 58 58 61 61 59	59 57 54 56 69 53 52	59 57 55 56 60 54 53	59 57 58 58 58 58 57	68 60 59 72 80 71 58	63 60 57 63 70 61 58	62 60 58 58 60 59 58	62 60 58 68 80 71 58	62 60 58 63 73 63 58	60 60 58 58 61 61 58	62 56 57 60 64 60 55	62 58 58 63 67 61 56	60 59 58 58 61 61 58
Mean, . Highest Lowest,	56 60 53	$ \begin{array}{r} 56\frac{3}{8} \\ 60 \\ 54 \end{array} $	585 61 57	65 3 77 58	61½ 68 57	59½ 60 58	643 77 58	$\begin{array}{c} 62\frac{1}{8} \\ 71 \\ 58 \end{array}$	59 <u>5</u> 62 58	59½ 64 55	61 68 56	593 62 58	57½ 69 52	56 ³ 60 53	58½ 60 57	663 80 58	615 70 58	59 <u>8</u> 62 58	65 ³ / ₇₁ 58	62 3 73 58	593 61 58	59½ 64 55	60 <u>.5</u> 68 58	593 61 58

	OBSERVATIONS	ON ATMOSPHERE, D	EW, RAIN, AND WIN	D,
	5 o'clock a. m.	12 o'clock m.	3 o'clock p. m.	7 o'clock p. m.
DAY OF MONTH.	Air, 4 feet above surface. Air, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Dew. Rain.	Air, 4 feet above surface. Alr, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Rain.	Air, 4 feet above surface. Air, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Rain. Wind.	Air, 4 feet above surface. Air, 4 inches above surface, naked soil. Air, 4 inches above surface, grass land. Rain.
24 25 26 27 28 29 30	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	63 64 63 W.	60 59 58 W.
ean, ighest, owest,	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

Observations.—24. Morning light dew, hazy clouds. Afternoon cloudy. 5 p.m. commenced raining. Moderate breeze. 25. Cloudy. Showery. Moderate N. breeze. 26. Morning cloudy. Rainy. Rained steady since 5 p.m. of the 24th. Afternoon cloudy. N. breeze. 27. Cloudy. Damp air. Commenced raining 4 p.m. Moderate breeze. 28. Few thin clouds. Moderate breeze. 29. Morning clear. Heavy dew. Foggy. Afternoon few floating clouds. 30. Morning cloudy. Light showers between 11 a.m. and 2 p.m. Moderate breeze.

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	0	BSEI	RVA'	TIO	NS C	N S	OIL	oF	GRA	SS :	LAN	D.		0]	BSEI	RVA	TIO	NS C	N N	IAK.	ED S	OIL	•	_
	5 0	el'k a	ı. m.	12 oʻ	cloc	k m.	3 0.0	ı'k p	. m.	7 oʻc	el'k p	 o. m.	5 0'0	el'k a	ı. m.	12 o	cloc	k m.	3 oʻc	l'k p	. m.	7 o'c	el'k p	. m.
DAY OF MONTH.	Surface.	4 inches below the surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	W O	9 inches below surface.	Surfaee.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.
1 2 3 4 5 6 7	50 49 49 50 52 53 53	52 51 53 52 52 53 53	57 57 56 56 56 55 56 57	59 65 70 76 75 68 69	56 58 64 62 58 60 61	57 57 59 58 57 57 57	57 67 72 78 77 67 63	57 63 67 62 66 61 61	57 58 60 57 60 58 58	54 56 59 54 62 60 60	56 60 63 56 64 61 60	67 68 60 58 60 58 58	50 49 49 50 52 53 53	52 51 52 52 52 52 53 53	56 56 54 54 54 55 56	59 66 73 78 77 68 70	56 58 64 62 59 60 61	59 57 58 57 56 57 58	57 67 74 78 79 67 63	57 63 67 62 66 60 61	57 58 60 58 60 58 58	54 56 59 54 61 59 59	55 60 62 56 63 60 59	56 58 60 58 60 57 57 58
Mean, Highest Lowest.	53	52 ³ / ₈ 53 51	56 ³ 57 55	687 76 59	597 64 56	57§ 59 57	685 78 57	$62\frac{3}{8}$ 67 57	583 60 58	577 62 54	60 64 56	583 60 58	523 53 49	52½ 53 51	55 56 54	$70\frac{1}{8}$ 78 59	60 64 56	57 ³ / ₈ 59 56	69 <u>3</u> 79 57	623 67 57	$ \begin{array}{r} $	57\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	593 63 55	58½ 60 56

		5 (o'clo	ck a	. m.		1	12	o'clo	ock n	1.		3 0.0	elock	7 p. r	n.		7 0'0	clock	тр. 1	n.
DAY OF MONTH.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Dew.	Rain.	Wind.	, 4 feet a surface.	face, nake	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	, 4 inches a face, naked	Air, 4 inches above surface, grass land.	Rain.	- - -	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Rain.	Wind.
1 2 3 4 5 6 7	49 48 47 50 48 53 53	49 48 47 50 48 53 53	49 48 47 50 48 53 53	48 48 47 48 48 52 52		s. N. N. S. S.	61 59 63 63 62 64 64	61 59 63 64 63 64 66	61 59 63 63 62 64 66	56	S. N. N. S. S. S.E.	60 62 65 64 68 64 63	60 62 66 64 67 64 63	60 62 65 64 69 64 63		N. N. N. S. S.	57 59 58 60 62 60 61	56 58 56 60 61 59 60	55 57 56 60 60 58 59		N N N S S E

Observations.—1. Light dew. Cloudy. Damp air. Light showers from 11 a.m. to 1 p.m. 2. Heavy dew. Floating clouds. Moderate N. breeze. 3. Heavy dew. Few thin clouds. Moderate breeze. 4. Heavy dew. Clear. N. breeze. Leaves begin to fall. 5. Heavy dew. Clear. Moderate S. breeze. 6. Heavy dew. Thin clouds. Moderate breeze. 7. Heavy dew. Morning few thin clouds. Afternoon cloudy. Moderate breeze.

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		01	BSER	lVA'	TIO1	ns o	N G	RAS	S L	ND	•			01	BSE	RVA	TIO	NS C)N I	NAK	ED	son	L.	
	5 0'0	elk.	ı. m.	12 o	cloc	km.	3 o'e	lk. p	. m.	7 oʻc	lk p	. m.	5 o'e	lk a.	. m.	12 o'c	eloek	m.	3 0'0	lk p	. m	7 o'e	Ik p	. m
DAY OF		sar-	sur-		sur-	sur-		sur-	sar-		sur-	sar-		sur-	-ins		sur-	sur-		sur-	sar-		sur-	-Ins
MON I H.	Surface.	4 inches below face.	9 inches below face.	Surface.	4 inches below face.	9 inches below face.			9 inches below face.	ırface.	ું કુકા જુ	9 inches below face.	ŀ	inches	y inches below face.	1	4 inches below face.	y inches below face.	Surface.	4 inches below face.	9 inches below face.	Surface.	inches	9 inches below
8 9	55 53	55	57 57	62 62	59 60	58 57	62 55	61 57	58 57	59 53	58 56	58 57	55 53	55 56	57 56	62 62	59 60	58 57	62 55	61 57	58 57	59 52	58 55	58 57
10	50	52	56	62	61	58	60	61	58	55 50	56 53	58 56	50 46	52	55	63	61 57	58 56	60	61	58	54	56	58
11 12	46 42	48 46	55 53	58 50	57 50	56 53	56 50	57 50	56 53	50	50 50	53	40 42	48 46	54 52	58 50	50	52	56 50	56 50	56 52	50 50	52 50	50 52
13	52	52	53	58	56	55	54	56	55	53	54	55	52	52	53	59	56	55	54	56	55	52	53	5
14 15	42 40	45 44	52 51	56 50	56 51	53 52	56 50	56 51	53 52	50 55	52 59	53 52	41 39	44 43	51 50	56 50	56 60	53 50	55 49	56 50	53 50	49 44	51 48	55
Mean, Highest Lowest,	55	493 55 44	544 57 51	$57\frac{1}{4}$ 62 50	564 61 51	55 ₄ 58 52	55 ³ / ₈ 62 50	$ \begin{array}{r} 56\frac{1}{8} \\ 61 \\ 50 \end{array} $	55 ₄ 58 52	517 59 45	53½ 58 49	55‡ 58 52	47 ₄ 55 39	49½ 56 43	53½ 57 50	$ \begin{array}{r} 51\frac{1}{2} \\ 62 \\ 50 \end{array} $	57 ³ / ₈ 61 56	547 58 50	55½ 62 49	55 7 61 50	547 58 50	51 ₄ 59 44	527 58 48	54 58 58

		5 o'elo	oek a	. m.			12 c	eloe	k m	•		3 0'	clock	ср. 1	n.		7 oʻ	eloel	тр. 1	n.
DAY OF MONTH.	Air, 4 feet above surface.	face, nake 4 inches face, gras		Rain.	Wind.	4 feet abourface.	4 inches a ace, naked	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	4 inches ace, nake	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	4,8	Air, 4 inches above surface, grass land.	Rain.	Wind.
8 9 10 11 12 13 14 15	55 5 54 5 51 47 47 47 52 42	55 55 54 54 51 51 47 47 42 42 52 52 42 42 36 35	48 45		N. N. W. W. S. W. W.	62 59 60 56 50 57 54 48	62 59 60 56 50 57 54 48	62 59 60 56 50 57 54 48		S. E. W. W. N.W. S. W. W. N.W.	60 58 58 56 51 54 53	58 58 56 51 54 53 48	60 58 58 56 51		S.E. N. W. W. S. W. W.	60 56 53 50 50 48 47	60 55 52 50 50 47 46 42	50 50 50 50 47 46 42	59	S. E N. N. W. S. W. W.
Iean, Iighest, Iowest,	55	48½ 48 55 57 36 35			:	553 62 48	55 ² / ₄ 62 48	553 62 48			55 ³ / ₈ 60 48	$ \begin{array}{r} \\ 55\frac{1}{2} \\ 60 \\ 48 \end{array} $	$ \begin{array}{r} 54\frac{3}{4} \\ 60 \\ 48 \end{array} $			50 ₈ 60 43	50 ¹ / ₄ 60 42	50 60 42		

Observations.—8. Cloudy. Afternoon slight showers. 9. Few thin clouds. Fresh W. breeze. 10. Few floating clouds. Stiff W. breeze. 7 P. M. clear. Light dew. 11. Heavy dew. Morning clear. Afternoon few clouds. Strong W. breeze. 12. Morning cool. Light dew. Thin clouds. Afternoon cloudy. Strong S. breeze. 13. Few thin clouds. Moderate breeze. Smoky. 14. Morning clear. Light frost. Afternoon few clouds. W. breeze. 15. Few clouds. Heavy frost. Moderate wind.

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	OI	BSER	VAT	CION	is o	N S	OIL	OF	GRA	.ss 1	LAN	D.		0	BSE	RV A	TIO	NS	ON	NAK	ED	soi	L.	-
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DAY OF MONTH.	Surface.	4 inches below surface.	9 inches below surface.	Surface.		9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.		9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.	Surface.	4 inches below surface.	9 inches below surface.
16 17 18 19 20 21 22 23	38 45 47 46 41 52 48	44 46 47 47 45 52 50	50 50 52 52 51 54 52	52 60 60	52 52 57 53	51 52 55	52 60 64 60 57 58	52 55 58 47 56 54	51 53 56 55 55 55 54	54 59 54 52 54 56 52	51 53 56 54 55 54 55 53	51 53 56 54 55 54 54 54 53	37 45 46 45 40 52 48	43 45 46 46 41 52 50	48 49 51 51 49 53 52	52 61 60 55	52 52 57 57	50 51 55 53	52 61 64 60 57 58	52 55 58 58 57 56 54	50 53 56 55 55 55 53	48 54 57 54 52 54 56 52	52 54 58 54 53 54 55 53	50 53 56 54 55 54 55 54 53
Mean, Highest Lowcst,		474 52 44	50½ 54 50	$ \begin{array}{r} 56^{3} \\ 60 \\ 52 \end{array} $	53 ¹ 57 52	523 55 51	$ \begin{array}{r} 59\frac{1}{2} \\ 64 \\ 52 \end{array} $	$53\frac{2}{3}$ 58 54	55 56 51	535 58 49	543 58 52	533 56 51	$ \begin{array}{r} \hline 44\frac{5}{8} \\ 52 \\ 37 \end{array} $	$46\frac{3}{4}$ 52 43	503 57 48	57 61 52	53 57 52	52 ₄ 55 50	57 ² / ₃ 63 52	55 58 52	54 56 50	553 57 43	541 58 52	53 ± 56 50

		5	o'ele	ek a	. m.			12 c	'clo	ck m			3 0 0	elock	р. 1	n.		7 o	clocl	ср. г	n.
DAY OF MONTH.	Air, 4 feet above sur-	25 26	Air, 4 inches above surface, grass land.	Dew.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.		Wind.
16 17 18 19	36 45 45	35 45 45	35 45 45	32 45		s. s. s.	54 60	54 60	54 60		s. s.	55 60 68	54 60 68	54 60 68		s. s. s.	46 56 63 54	46 56 63 54	46 55 62 54		s. s. w
20 21 22 23	46 40 52 46	46 40 52 46	46 40 52 46	42 38	50	N. W. S. W.	60 52	60 52	60 52		N. 	60 57 62	60 57 62	60 57 62		N. N. S.	52 54 62 51	52 53 62 51	51 53 62 50		N. N. S. W

Observations.—16. Clear. Heavy frost. Ground frozen slightly. Moderate breeze. 17. Morning clear. Light dew. Afternoon few thin clouds. S. breeze. 18. Heavy dew. Very few thin clouds. S. breeze. 19. Cloudy. S. breeze. 20. Heavy dew. Very few thin clouds. N. breeze. 21. Heavy dew. Few clouds. 22. Cloudy. S. breeze. Morning rainy. 23. Morning cloudy. Stiff W. breeze. 7 P. M. clear.

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	C	BSE	RVA	\TI(NS	ON i	SOIT	OF	GR	ASS	·LA]	ND.		. 0	BSE	RVA	TIC)NS	ON	NAl	ED	SOI	L.	
	5 oʻ	elk a	. m.	12 o	`eloc	k m.	3 0	elk p	. m.	7 o'	clk p	. m.	5 o'	elk a	ı. m.	12 o	'cloc	k m.	3 о'	clk i). m.	7 0	elk p	. m
		sur-	sur-	-	sur-	sur-	-	sur-	sur-		sar-	sur-		sur-	ms		sur-	sur-		sur-	sur-		-ins	sur-
DAY OF MONTH.	Surface.	l inches below face.	inehes below face.	Surface.	l inches below face.	9 inches below face.	Surface.	4 inches below face.	9 inches below face.	Surface.	l inches below face.	dinches below face.	Surface.	l inches below face.	dinches below face.	Surface.	l inches below face.	9 inches below face.	Surfaee.	finches below face.	dinehes below face.	Surface.	4 inches below face.	9 inches below faee.
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31 Mean,	$\frac{39}{321}$	355	$\frac{43}{39}$				59	54	50				39 315	353	43 383				56	54	50			
Highest Lowest,	54	52 ⁸	52 43										54	52 38	52 ⁸									

		5	oʻclo	ck a	m.		1	12	o'clo	ock n	1.		3 o'	clock	ъ.	m.		7 oʻ	clock	р. 1	n.
DAY OF MONTH.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Dew.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.		Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Rain.	Wind.	Air, 4 feet above surface.	Air, 4 inches above surface, naked soil.	Air, 4 inches above surface, grass land.	Rain.	Wind.
24 25 26	54	44	54		54	s. w						60	60	60		w.			· · · · ·		
26 27 28 29 30 31	33 33 32 32 32 44	33 33 32 32 44	33 33 32 32 32 44	32 32 32 32 32 44	••••	S. S.						53 57	53 57				40	38	38		N.
Mean, Highest,	$\frac{321}{54}$			$ \begin{array}{r} 32\frac{1}{2} \\ 54 \\ 32 \end{array} $																	

Observations.—25. Cloudy. Morning showery. 27. Heavy frost. Ice froze in vessels exposed 4 inch thick. 28. Ice froze 4 inch thick. N. breeze. 29. Heavy frost. Clear S. breeze. 30. Heavy frost. Very few clouds. 31. Heavy dew. Few clouds. S. breeze.

Surface [11, bed.] 9 in, bed. 2 ft, bed. 4 ft, bed. 4 ft, bed. 5 ft, bed. 5 ft, bed. 4 ft, bed. 5 f			OBSE	RVA'	OBSERVATIONS	ON	SOIL.		-	ALBANY BLUE CLAY.	ANY	BLUI					1	VEA	WEATHER.	.:			
8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	DAY OF MONTH.	Surfaces			in. be	1. 2 ft	bel.	4 ft. t surfac			- 1. - 2.	ft. be	-i .		Air.		Dir	e'tion wind.		ree o		Degr	ree C
3.8 3.8 <th></th> <th></th> <th>8 a.m</th> <th></th> <th></th> <th></th> <th></th> <th></th> <th></th> <th>8 3 ·m p.</th> <th>m 3.</th> <th>3 m</th> <th>6 m_a.1</th> <th>8 m</th> <th>m p.1</th> <th>9 m</th> <th></th> <th></th> <th>a.m</th> <th>p.m</th> <th>1</th> <th></th> <th>E</th>			8 a.m							8 3 ·m p.	m 3.	3 m	6 m_a.1	8 m	m p.1	9 m			a.m	p.m	1		E
31.3 34.87 38.47 41.44 46.64 43.11 46.33 35.67 37.66 39.44		4	######################################									<u> </u>											4
	Semi-monthly mean, Monthly mean,	31.3 35.6	34.	92	38.47	1	11.44	**	- 1	- භ	- 1	- 90	. m				2	2		-	:	5	_

			JA	.NUAR	RY, 184	8.					
DAY OF MONTH.	OBSERV ON S		OBSERV ON TE				WEA	THER.			
	2 feet below surface.	4 feet below surface.	Black Walnut.	Horse Chestnut.	Ai	r.	Direction of wind.	Force of wind.		Degree of cloudiness.	Inches
	8 3 a. m p.m	8 3 a. m p.m	8 3 a. m p. m	8 3 a. m p.n	6 8 a. m a. m	3 9 p.m p.m	a. mp.m	a. mp.m	Mean a. m	Mean Mean	
1 23 3 4 5 6 7 8 9 10 11 12 13 14 15	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	43 43 43 43 44 42 42 42 41 41 40 40 42 42 42 42 42 42 42 42 42 42 42 42 42				56 42 40 32 21 25 28 4 30 30 32 38 42	S S N N N N N N N N N N N N N N N N N N	1 3 1 1 1 4 2 7 5 4 2 2 3 2 6 1 1 1 1 1 1 1 1 1	$ \begin{vmatrix} 2 & 3 \\ 1 & 0 \\ 2 & 3 \\ 4 & 4 \\ 4 & 1 \\ 2 & 0 \\ 2 & 3 \\ 4 & 2 \\ 1 & 0 \\ 1 & 2 \\ 1 & 4 \\ 1 & 5 \\ 2 & 5 $	$ \begin{vmatrix} 3 & 3 \\ 0 & 0 \\ 2 & 1 \\ 5 & 4 \\ 1 & 1 \\ 3 & 3 \\ 2 & 2 \\ 1 & 0 \\ 0 & 0 \\ 3 & 2 \\ 4 & 4 \\ 6 & 5 $	12
mi-monthly mean,	39.23	41.72			•	28			2.18		46
16 17 18 19 20 • 21 22 23 24 25 26 27 28 29 30	$ \begin{bmatrix} 35 & 35 \\ 35 & 35 \\ 35 & 35 \end{bmatrix} $ $ 35 & 35 \\ 34 & 34 \\ 34 & 35 \\ 35 & 35 \\ 35 & 35 \\ 35 & 35 \\ 34 & 34$	$ \begin{vmatrix} 37\frac{1}{2} & 38 \\ 38 & 38 \\ 38 & 38 \\ 38 & 37\frac{1}{2} & 37\frac{1}{2} & 37\frac{1}{2} & 38\\ 37\frac{1}{2} & 38 \\ 37\frac{1}{2} & 37\frac{1}{2} $	26 25 26 32 32 32 31 31 30 31 30 32 32 33 34 34 35½ 38 32 35 32 35 32 35 32 35	$ \begin{vmatrix} 30 & 29 \\ 27 & 27 \\ 28 & 24 \\ 26 & 29 \\ 29 \\ 1 & 30 \\ 31 & 31 \\ 30 & 30 \\ 31 & 30 \end{vmatrix} $	$egin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$. N N N N N N S	3 3	$egin{array}{cccccccccccccccccccccccccccccccccccc$	1 3 0 1 2 3 1 0 3 5 6 1 4 2	$egin{array}{cccccccccccccccccccccccccccccccccccc$

OBSERVATIONS.—1. Hazy. Warm S. breeze. 3. Heavy frost in morning. Warm day. 5. Snow $\frac{1}{2}$ inch deep. High wind. 6. Cold N. Wind. 8. Light fall of snow. 9. Snow fell last night 8 inches deep. Strong W. breeze. 10. Very cold. 11. Very cold. 14. Light showers of rain. 15. Light showers oceasionally. 16. Snow has nearly disappeared. Soil froze 6 inches deep. 17. Light frost during night. 20. Cold S. wind. 26. Smoky. Light showers during the afternoon. 27. Morning foggy. Smoky. Rain from 10 a. m. to 4 p. m. 28. Morning foggy and smoky. 30. Light fall of snow. White frost during the night. 31. Soil froze $\frac{1}{2}$ inch deep during night.

			FE	BRUA	RY, 1848.			
DAY OF MONTH.		VATIONS SOIL.		ATIONS REES.		WE	ATHER.	
v	2 feet below surface.	4 fect below surface.	Horse Chestnut.	Black Walnut.	Air	Direction of Wind.	Force of Wind.	Degree of Cloudiness.
	8 3 a.mp.ir	8 3 a. m p. m	8 3 a. m p.m	8 3 a. m p.m	a. ma. mp.mp	1	Mcan, m.d m.	Mean m.d m.a
1 2 3 4 5 6 6 7 8 9 10 11 12 13 14 15	34 34 34 34 34 34 34 34 34 34 34 34 34 3	37 36 37 36 37 36 36 36	31 31 30 30 30 30 30 30 30 30 30 27 28 18 23 23 23	$\begin{vmatrix} 31 & \dots \\ 30 & 30 \\ 27 & 28 \end{vmatrix}$	22. 24 37 13 14 31 32 28 31 38 33 27 29 26 24 28 12 14 28 1 7 33 26 24 2 26 24 2 26 21 18 3 26 24 2 26 2 26	36 N W 20 N N 30 N S 35 N SE 29 N N 22 N N 10 N N 24 N N 44 N N 44 N N 116 N N 117 N N 117 N N	1 2 2 1 1 1 1 1 1 1 1 1 1 1 1 4 2 4 4 4 4 1 1 1 1 1 5 3 2 2 2 2 1 1 1 1 1 1 1 1 1 3 2	$ \begin{vmatrix} 1 & 1 & 1 & 0 & 0 & 0 & 0 & 0 & 0 & 0 &$
emi-monthly mean, 16 17 18 19 20 21 22 23 23 24 25 26 27 28	34\\ 33\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	36 36 36 36 36 36 36 36 36 36 36 36 36 3	25 27 23 26 23 25 20 26 30 30 31 30\frac{1}{2} 31 30\frac{1}{2} 30 29 30 29 30 28 30 29 26 27 28	25 29 17 28 23 27 22 29 33 33 33 34 33 38 32 32 28 31 26 31 30 32 28 28 30 30	24 26 40 2 10 12 32 2 8 13 38 2 12 10 36 3 36 363	3 N N N N N N N N N N N N N N N N N N N	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$

Observations.—1. Snow fell 6 inches deep last night. Thaws slowly in sun. 2. Hard frost last night. Thaws slowly in sun. 3. Thaws slightly in sun. Freezes in shade, 4. Light snow storm in morning. P. m. warm, thaws fast. 5. About 4 inches of snow fell last night. Commenced snowing again at 12 m. 6. Snow 10 inches deep. Snow has been falling almost constantly since 12 m. yesterday. Snow damp. 7. Snow 12 inches deep. Drifts some. 8. Thaws slightly in sun. Good sleighing. 9. Clear. Smoky. Thaws slightly in sun. 10. Smoky. Pleasant. 11. Clear and cold. 12. Cold, very. 14. Snow 10 inches deep. 15. Smoky. 16. Smoky. P. m. warm. 17. Smoky. Cold morning. 18. Smoky. Thaws fast in sun. 19. Smoky. Snow is disappearing rapidly. 20. Rain and hail early a. m. to 2 p. m. Temperature 36°. 21. Snow nearly gone. 22. Rain and snow 12 m. afternoon and evening. 23. Snow disappearing rapidly. 27. Smoky. 28. Snow p. m. 29. Snow a. m. Snow ½ inch deep.

							MA	RC	Η, .	1348	3.										
DAY OF MONTH.	OBS	SERV ON S	ATIO	NS		SERV								WEA	THE	R.					
	2 feet below	surface.	4 feet below	surrace.	Horse	Chestnut.	Black	Walnut.		Λ	ir.			ot wind.		Force of wind.			Degree of cloudiness.		Rain
	8 a. m	3 p.m	8	3 5. m	8 a. m	3 p.m	8 a. m	3 p . m	6 a. m	8 a. m	3 p.m	9 p.m	a. m	p . m	a. m		Mean.	a.m	p.m	Mean.	
1 22 3 4 5 6 6 7 8 9 10 11 12 13 14 15	4 33 33 35 33 6 33 33 33 34 35 36 36 36 36 36 36 3					24 23 23 26 24 29 31 32 30 30 30 23	27 23 24 25 23 ³ 28 32 35 30 29 30 23	28 26 26 28 27 29 36 36 31 32 30 26	16 11 29 20 23 14 26 36 37 21 14 9 36 12	$\begin{array}{c} 18\\ 16\\ 20\\ 21\\ 23\\ 20\\ 28\frac{1}{2}\\ 32\\ 36\\ 24\\ 24\\ 16\\ 32\\ 16\\ 6\\ \end{array}$	30 32 38 34 26 12	19 24 27 30 18 33 42 50 29 23 24 40 25 11	N W N N S S S S S N W N W W	N N S W S S S N W N S W W W W	$ \begin{array}{c} 1 \\ 1 \\ 4 \\ 0 \\ \dots \\ 3 \\ 2 \\ 0 \\ 4 \\ 4 \\ 1 \\ \dots \\ 0 \\ 1 \\ 0 \end{array} $	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	$\begin{array}{c} 3^{\frac{1}{2}} \\ 1^{\frac{1}{2}} \\ 3 \\ 2^{\frac{1}{2}} \\ 1^{\frac{1}{2}} \\ 1^{\frac{1}{2}} \\ 4 \\ 1^{\frac{1}{2}} \\ 1 \\ 1^{\frac{1}{2}} \\ 1 \\ 1 \end{array}$	1 1 4 0 3 2 0 4 4 4 1 0 1 0	$egin{array}{c} 1 \\ 2 \\ 4 \\ 3 \\ 4 \\ 4 \\ 1 \\ 2 \\ 4 \\ 4 \\ 4 \\ 2 \\ 1 \\ \end{bmatrix}$	$egin{array}{c} 1 & 1_{rac{1}{2}} & 4 & 2_{rac{1}{2}} & 2 & 3 & 1_{rac{1}{2}} & 1 & 4 & 1_{rac{1}{4}} & 2 & 1 & 1_{rac{1}{2}} & 2 & 1_{rac{1}{2}}$	0.4
mi-monthly mear				53								25.1									
16 17 18 19 20 21 22 23 24 25 26 27 28 29 30	33 32 ^{1/2} 32 ^{1/2}	32 32 32 32 32 32 32 32 32 32 32 32 32 3	34 15 12 34 15 15 15 15 15 15 15 15 15 15 15 15 15	34 34 34 34 34 34 34 34 34 34 34 34 34 3	38 37	19 21 23 33 34 34 34 36 38 38 38 42 47 57	18 18 25 27 32 35 35 34 32 40 40 40 40 49	25 25 25 25 33 38 38 36 35 41 40 41 44 48 61	1 1 22 18 35 39 34 35 34 28 43 44 40 41 35 50	6 10 25 28 37 42 37 36 32 32 44 41 42 42 39 50	23 32 37 42 43 49 48 48 46 42 44 53 66 66	10 22 27 38 44 44 39 38 35 46 46 43 44 45 50 62	N S W S W N S S S S S S N W S S S S S S	N S S S W W S S S S N W N S S S	$egin{array}{c} 0 \\ 0 \\ 1 \\ 1 \\ 2 \\ 1 \\ 2 \\ 1 \\ 2 \\ 1 \\ 2 \\ 1 \\ 2 \\ 2$	1 1 1 2 3 2 1 2 1 2 1 3 1 4	$egin{array}{c} rac{1}{2} rac{1}{2} \ 1 rac{1}{2} rac{1}{2} rac{1}{2} \ 1 \ 1 rac{1}{2} rac{1}{2} rac{1}{2} \ 1 \ 2 \ 1 \ 2 \ 1 \ 3 \ \end{array}$	0 3 1 4 4 0 4 5	0 0 3 3 4 3 2 1 5 4 6 1 1	$egin{array}{c} 0 \ 0 \ 3 rac{1}{12} rac{1}{12} \ 3 rac{1}{12} \ 3 rac{1}{12} \ 3 rac{1}{12} \ 2  2 rac{1}{12} \ 2 \ 2 rac{1}{12} \ 2 \ 2 \ 2 \ 2 \ 2 \ 2 \ 2 \ 2 \ 2 \$	0.

OBSERVATIONS.—1. Cold wind. 2. Snow 9 a.m. to noon of 3d. 3. Snow 8 inches deep. 4. Snow early a. m. 5. Snow early a. m. 6. Wind shifted at about 12 m. at night. 7. Thaws fast. 8. Froze slightly during night. 9. Rain during night and early a. m. Snow, rain and hail $4\frac{1}{9}$ p. m., till 2 p. m. of the 10th. 11. Thaws fast in sun. 12. Smoky. 13. Smoky. 14. Smoky. 19. Rain and snow 6 p. m. 20. Rain 5 p. m. 21. Strong wind during night. 23. Snow and rain early a. m., occasional showers during the day and night. 26. Rain 2 p. m. to 12 m. at night. 27. Hazy, slight showers. 28. Rain early a. m. till 12 m. at night. 29. Smoky. 30. Smoky.

				APRIL	, 1848	•				
DAY OF MONTH.	OBSERV ON S			ATIONS REES.			WE.	ATHER.		
	2 feet below Surface.	4 feet below surface.	Horse Chestnut.	Black Walnut.	1	Vir.	Direction of Wind.	Force of Wind.	Degree of cloudiness.	Inches of Rain.
	8 3 a.m p.m	8 a.m p.m	$\begin{vmatrix} 8 & 3 \\ a.m & p.m \end{vmatrix}$	8 3 p.m	6 8 a.m a.u	$\begin{vmatrix} 3 & 9 \\ p.m & p.m \end{vmatrix}$	a.m p.m	Mean m.q	Mean m.d	
1 2 3 4 5 6 7 8 9 10 11 12 13 14	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	38 38 38 39 38 40 39 40 40 41 40 42 41 42 41 42 43 42 43 43	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	38 46 41 44 45 49 41 50 43 49 43 53 47 53 50 52 52 52 52 52 50 50	31 50 30 34 39 39 39 48 47 32 40 33 42 32 42 38 38 39 44 43 50 47 44 42 43 36 44	60 42 64 51 75 51 74 63 71 55 59 52 41 39 51 44	W W SW W W W W W W W	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\frac{1}{2}$ 0.25 0.45 $\frac{1}{2}$ 0.01 $\frac{1}{2}$ 0.0 0.01 $\frac{1}{2}$ 0.0 0.00 $\frac{1}{2}$ 0.0 0
emi-monthly mean,	393	391				$46\frac{1}{2}$				
16 17 18 19 20 21 22 23 24 25 26 27 28 29 30	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	44 44 44 45 45 45 45 45 45 45 45 45 45 45	51 57 45 48 40 44 35 46 53 66 52 56 52 46 51 52 52 54 55 55 50 50 56 52 57 50 58 50 59 50 50 50	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	43 4 34 3 30 3 27 3 50 5 50 4 4 40 4 36 4 4 7 4 37 4 53 4 53 5 53 5	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	N W N N N N W S W W W W W N N N S S S S	4 5 4	$egin{array}{c ccccccccccccccccccccccccccccccccccc$	$ \frac{1}{2} $

OBSERVATIONS.—1. Rain. 3. Smoky. 4. Rain. Thermometer 44°. 5. Smoky. 6. Smoky. 7. Smoky. 8. Smoky. 9. Smoky. 10. Smoky. 11. Smoky. 12. Rain early. 13. Rain from 11 a. m. to 6 p. m. 14. Rain and hail p. m. 15. Buds of the Horse Chestnut begin to swell. 16. Morning foggy. Smoky. 17. Smoky. 18. Snow p. m. 20. Light dew. Smoky. 22. Rain p. m. 24. Rain early a. m. 25. Smoky. Leaves begin to show themselves on the Horse Chestnut. 26. Smoky. Lilac leaves \(\frac{3}{4}\) inch long. 29. Rain a. m. Ostrya virginiea in flower. 30 Light dew. Currant and Gooseberry in flower.

											MA	Y, 1	1848	3.											
		овѕ	ERV A	T10	NS ON	I SOI	L•			ERVA	ATION	vs						WE	ATH	ŒR.					
OF MONTH.	4 inc. below	surface.	9 inc. below	surface.	2 feet below	surrace.	4 feet below	surface.	Horse	Chestnut.	Black	walnut.		Aì	r.		Direction of	wind.		Force of Wind.			Degree of cloudiness.		Inches rain.
DAY	8	3 p.m	8	3 p.m	8 a.m	3 p.m	8	3 p.m	8 a.m	3 p.m	8 a. m	3 p.m	6 a.m	8 a.m	3 p.m	9 p.m	a. m	թ.m	a. m	p.m	Mean	a. m	p.m	Mean	
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15	50 51 53 50 56 58 60 58 57 56 56 53 53 53	54 55 54 58 60 60 67 64 58 59 54 56 58 57	55 54	55 55	52 59	$ \begin{array}{c} 48 \\ 48 \\ 48 \\ 48 \\ 50 \\ 2 \end{array} $ $ \begin{array}{c} 52 \\ 54 \\ 2 \\ 53 \\ 53 \\ 52 \\ 52 \\ 52 \\ 51 \end{array} $	46 46 46 47 47 48 49 50 50 50 50 50	47 48 49 49 50 50 50 50 50 50	54 54 50 51 52 50 ¹ / ₂	57 57 55 59 62 67 68 61 57 58 52 52 53 51 54	50 53 53 50 58 $62^{\frac{1}{2}}$ 67 62 61 56 55 49 49 50 50	56½ 56 54 61 62 69 72 67 60 52 51½ 55 51	39 51 51 45 57 62 65 53 54 50 49 46 45 47	51 52 53 51 57 $64\frac{1}{2}$ 63 58 53 $47\frac{1}{2}$ 52 45 52	84 72 62 67 50 60 65 55 69	55 49 57 55 53 61	N S N E S N W S S N W N W W	S SW NE S S S W N S N W S N W S N W	1 1 1 3 1 1 1 2 2 3 1 3 2	4 1 2 1 3 1 2 2 1 4 2 2 3 4 3 3 4 3	$\begin{array}{c} 2^{\frac{1}{2}} \\ 1 \\ 1 \\ 1 \\ 3 \\ 1 \\ 1^{\frac{1}{2}} \\ 1 \\ 3 \\ 2 \\ 2^{\frac{1}{2}} \\ 2^{\frac{1}{2}} \\ 2^{\frac{1}{2}} \\ 2^{\frac{1}{2}} \end{array}$	$ \begin{array}{c} 1 \\ 4 \\ 6 \\ 0 \\ 3 \\ 2 \\ 0 \\ 1 \\ 5 \\ 4 \\ 6 \\ 3 \\ 2 \\ 5 \\ 0 \end{array} $	5 5 3 1 2 1 0 2 3 5 6 4 3 5	$\begin{array}{c} 4\frac{1}{2}\\ 4\frac{1}{2}\\ 2\frac{1}{2}\\ 2\frac{1}{2}\\ 1\frac{1}{2}\\ 0\\ 4\frac{1}{2}\\ 6\\ 3\frac{1}{2}\\ 2\frac{1}{2}\\ \end{array}$	0.3 0.6 0.3 2.3
8. m.	me.,	36.1	3 54	1.78	51	.4	4	s.28							5	$7\frac{2}{3}$									
16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31	55 55 57 61 62 62 62 62 61 63 60 60 62 64 60	59 60 62 64 65 65 64 62 64 65 65 65 65 64 62	55 56 55 58 59 62 61 62 61 62 61 62 61 62 61	62	52 52 53 53 54 56 57 58 58 58 58 58 58 58 58	57 57 58 58 58 58 58	50 50 50 50 50 51 52 52 53 53 54 55 54 54 54	51 52 52 53 53 53 53 53 53 54 54	$ \begin{array}{c ccccccccccccccccccccccccccccccccccc$	57 60 61 62 65 63 62 62 62 63 64 64 65 64 65 64 65 64 65 64 65 64 65 65 64 65 65 64 65 65 65 65 65 65 65 65 65 65 65 65 65	71 68 64 64 63 68 65 65 68 60		57 53 60 65 66 62 63 62 67 57 62 65 69 54	56 56 56 65 67 65 62 63 60 67 63 59 65 67	69 70 84 90 72 82 75 64 79 81 80 78 70 59	60 66 70 77 71 66 63 69 64 71 68 68 70 70 65	S SW S S N E W S S E N S S S W	W W S E S N W S E W E S S W	1 1 1 2 1 1 1 3 2 1 1 1 2 4	1 3 1 1 1 2 1 3 2 1 1 7 2 4	$egin{array}{c} 1_{12} & 2_{11} & 2_{12} & 1_{$	$ \begin{array}{c} 5 \\ 1 \\ 0 \\ 0 \\ 3 \\ 3 \\ 6 \\ 5 \\ 4 \\ 5 \\ 1 \\ 0 \\ 2 \\ 5 \\ 4 \end{array} $	3 1 0 1 5 2 3 5 1 1 1 6 3	$egin{array}{c} 4 \\ 1 \\ 0 \\ 4 \\ 2 \\ 1 \\ 2 \\ 4 \\ 4 \\ 1 \\ 3 \\ 2 \\ 1 \\ 1 \\ 1 \\ 2 \\ 3 \\ 3 \\ 3 \\ 3 \\ 3 \\ 3 \\ 3 \\ 3 \\ 3$	0.4

Observations.—1. Rain 6 to 8 p. m. Heavy dew in morning. 2. Rain 5 p. m. to 10 a. m. of the 3d. Tem. 53°. 3. Horse Chestnut buds at 8 a. m. Tem. 54°. 4. Heavy dew. 5. Rain carly a. m. 6. Flower panicles of the Horse Chestnut begin to appear. Rain from 2 to 4 p. m. thunder shower. 7. Rain early a. m. Thunder shower. 8. Buds of the Black Walnut begin to swell. 9. Rain a. m. Tem. of rain 56°. Young twigs of Horse Chestnut from 9 to 11 inches long, those of the lilac from 10 to 14 inches long. 10. Rain a. m. 0.02. Rain 3 p. m. 11. Rain till late in the evening. Tem. 48°. 12. Lilac in flower. Horse Chestnut begins to flower. 14. Rain early a. m., and p. m. 16. Rain carly a. m. Tem. 55°. 17. Heavy dew. 18. Heavy dew. 19. Rain from 9 to 11 p. m. Thunder shower. 20. Rain 12 m. to 3 p. m. Tem. 71°. Thunder. 21. Rain 2 p. m. to 11 a. m. of 22d. Tem. 65°. Pinguicula vulgaris, Polygala paucifolia, Uvularia sessilifolia and Convalaria bifolia in flower. 22. Rain p. m. 23. About § of the top of the Black Walnut is dead. Horse Chestnut shedding its petals. 24. Erythronium albidum and americanum in flower. 25. Slight shower. Viburnum roseum in flower. 26. Heavy dew. Tem. 66°. 27. Heavy dew. Tem. 58°. 28, Poa pratensis and Triticum repens begin to head. 30. Rain early a. m. to 4 p. m. and from 6 to 7 p. m Tem. of rain p. m. 68°.

										J	UN	E, 1	1848	3.											
		OBS	ERVA	TIO	NS OI	v so	ııs.				ATIO							WEA	тнЕ	R.					
DAY OF MONTH.	4 inc. below	surface.	9 inc. below	surface.	2 feet below	surface.	4 feet below	surface.	Horse	Cuestnut.	Black	Walnut.		Ai	r.		Direction of	Wind.		Force of Wind.		1	Degree of Cloudiness.		Inches rain.
	8 a.m	3 p. m	8 a.m	3 p.m	8 a.m	3 p. m	8	3 p. m	8 a · m	3 p.m	8 a.m	3 p.m	6 a. m	8 1.m	3 p.m	9 p.m	a.m	p.m	a.m	p.m	Mean	ı. m	թ.m	Mean	
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15	56 55 58 63 65 64 60 59 60 62 64 62 59 61 64	60 63 65 71 67 61 63 68 69 72 68 68 65 68	$ 59\frac{1}{2} $ $ 58 $ $ 59 $ $ 62 $ $ 63\frac{1}{2} $ $ 64 $ $ 63$ $ 64 $ $ 63\frac{1}{2} $ $ 63$ $ 63 $ $ 63 $	60 62 63 65 66 62 62 64 64 66 64 63 62 64	$\begin{array}{c} 57\frac{1}{2} \\ 56 \\ 56 \\ 56\frac{1}{2} \\ 59 \\ 60 \\ 59\frac{1}{2} \\ 58\frac{1}{2} \\ 58 \\ 59 \\ 60 \\ 59\frac{1}{2}\frac{1}{2} \\ 58\frac{1}{2} \\ 59 \\ 58\frac{1}{2} \\ 59 \\ \end{array}$	56 56 56 59 60 59 58 58 59 59 59 59 59	56 56 57 57	54 ^{1/2} 54 54 ^{1/2} 54 56 ^{1/2} 56 56 56 56 56 56 56 56 57 57	60 63 65 63 56 57 60 62 63 58 58 54 57 61	58 62 63 64 62 58 59 62 63 59 58 60 62	53 54 62 66 70 66 60 57 60 64 65 60 55 57 63	57 62 66 72 73 76 59 60 68 67 69 61 59 61 68	42 51 62 63 70 62 51 54 57 64 49 46 53 62	48 58 64 70 70 60 53 57 63 68 54 50 58 66	62 78 77 86 80 69 56 70 86 74 72 62 62 62 82	56 70 71 75 65 56 54 63 70 66 65 52 59 62	W S S N S N W N S W W W W N W	W S S S S W NW N S W W W W W W S	3 2 1 2 2 2 2 2 2 2 1 2 2 1 2 2 1 2 1 2	5 3 1 1 2 4 3 2 2 3 3 5 5 3 2	$egin{array}{c} 4 & 2^{rac{1}{10^2-10^2}} \ 1 & 1 & 2 & 2^{rac{1}{10^2-10^2-10^2-10^2-10^2-10^2-10^2-10^2-$		1 1 3 1 2 4 3 1 4 1 1 1 1	1 1 1 1 1	0.53 0.04 0.06 0.06 0.06
S.M.m	ie., 6	$3\frac{1}{3}$	($52\frac{1}{2}$	58	31 8	55 5	93	6	3	,				6	31					$2\frac{2}{3}$			1.8	0.97
16 17 18 19 20 21 22 23 24 25 26 27 28 29 30	67 70 72 71 71 69 68 68 67 65 66 69 70 70	76 77 78 77 72 72 72 74 74 78 70 74	65 68 70 70 69 68 68 68 66 67 69 70	68 71 71 70 72 72 72 71 68 68 68 69 72 70 70		63 66 66 66 64 64 63 66 67 67	60 60 60 61 61 61 61 61 62 62	56 57 58 59 60 60 60 61 61 61 61 62 62	69 71 72 68 67 68 67 64 67 69 72 70 71		69 68 64 68 70 72 70 71	75 77 78 78 75 73 69 67 69 72 73 70 72	70 72 75 77 75 67 64 69 65 60 63 68 74 69 73	79 78 74 74 70 70 70 71 64 66 70 75 72 74		85 82 79 69 69 70 73 65 70 76 78 71 76 73	NW W S S S S W S W W NE S S N	S S S W S S W W	1 1 1 1 1 1 1 2 1 2 1 3 2 1 1	122313255313221	$egin{array}{c} 1 & 1 & 1 & 1 & 1 & 1 & 1 & 1 & 1 & 1 $	$\begin{bmatrix} 0 & 0 & 1 & 3 & 6 & 1 & 1 & 3 & 4 & 4 & 1 & 1 & 3 & 4 & 4 & 6 & 4 & 4 & 4 & 4 & 4 & 4 & 4$	0 1 6 5 1 2 6 1 1 3 2 6 1 4	$\begin{bmatrix} 4 \\ 3 \\ 1 \\ 4 \\ 2 \\ 1 \\ 2 \\ 5 \\ 3 \\ 4 \end{bmatrix}$	$\begin{bmatrix} 1.74 \\ \frac{1}{2} \\ 0.24 \end{bmatrix}$
S.M.n Mo.m				9.3 5.95	72 65	$\frac{.2}{.30}$		9.94 7.47	69 64	$\frac{9.23}{1.52}$		1.96 7.48				$\frac{1.57}{8.95}$					$\frac{1.9}{2.2}$			$\frac{2.6}{2.2}$	$\frac{3.7}{4.7}$

OBSERVATIONS.—1. Cypripidium acaule in flower. 3. Slight sprinkle in the morning. 4. Poa pratensis in flower 5. Rain early a. m. and 4 to 7 p. m. 6. Rain 6 to 7 a. m. 7. Rain 6 to 8 p. m. and 12 m. at night. 8. Rain 9 to 10 a. m., and 9 to 10 p. m. 10. Rain 4 to 5 p. m. and light showers from 7 to 11 p. m. 11. About \(\frac{1}{3} \) of the top of the Black Walnut tree is covered with leaves, the rest was killed by fire last September. 18. Rain 2 to 4 p. m. with thunder, and from 9 to 12 p. m. oecasional showers. Very warm. 19. Slight showers occasionally from 10 a m to 3 p. m. 20. Rain from 10 p. m. to 21st 12 m. 21. Rain 7 p. m. to 5 a. m. of the 22d. Thunder. The Black Walnut begins to send out new shoots. Triticum in flower. 22. Seirpus tenuis in flower. 23. Heavy shower 3 to 5 p. m. and 9 to 10 p. m. 24. Conium maculatum, Verbaseum blattaria, Lilium philadelphicum and Glyceria fluitans in flower just 8. of Albany. The Phleum pratense begins to flower. 28. Heavy rain 3 p. m. to 9 p. m. of the 29th. Thunder. The new shoots of the Black Walnut increase rapidly in size and number. 30. Rain 5 to 7 a. m. and 4 to 6 p. m. Heavy thunder storm.

Winder—North 31. South 13. Fast 1. West 91. Northwest 2. Northwest 1. Southwest 1. Southwest 1. Southwest 1. Southwest 1. Southwest 1.

Winds.—North 3\frac{1}{2}, South 13, East \frac{1}{2}, West 9\frac{1}{2}, Northwest 2, Northeast 1, Southwest \frac{1}{2}, Southeast 0.

Weather.—Rain on 12 days. Warmest day 16th. Coldest day 1st. Highest temperature 94 degrees. Lowest temperature 42 degrees.

Observations—1. Thunder shower 12 to 2 p. m. 3. Rain 7 a. m. to 3 p. m. 6. Chimaphila unbellata and Pyrola secunda in flower. 8. Grasshoppers increase rapidly. Slight shower 6 to 9 a. m. Rain 3 p. m. and during night. 10. The Black Walnut continues to send out new and vigorous shoots, $\frac{2}{3}$ of the top is now covered with thrifty dense foliage. 12. Rain early a. m. 0.04; thunder storm 1 to 4 p. m. 1.50; thunder storm 7 to 9 p. m. 0.28. 14. The potato rot has made its appearance in several parts of the country since the last rains. The Newburyport Herald mentions that there is a slight touch of it in that vicinity, and on Long Island. It has also appeared in Montgomery and Bucks in Pennsylvania. It has not yet made its appearance around Albany. 16. Symphyluna officinale and Lilium canadense in flower. Circium arvense, Asclepias syriaca and Sambucus canadensis flowering, 20. Rain 5 to 6 p. m. 22. Heavy shower 4 to 6 p. m. Tem. rain 70°. Rained from 10 p. m. on the 22d till 8 a. m. 23d. 24. Slight shower during night. 27. Heavy thunder shower 5 to 7 p. m. 28. The Black Walnut continues to grow vigorously. 30. Rain 5 a. m. to 3 p. m. Occasional showers from 3 p. m. till dark. Wind shifted to the north at 9 p. m. force 5 with rain, which continued till 11 p. m. at a rapid rate.

Winds.—North 5 days, East $1\frac{1}{2}$, South 12, West 7, Northeast 2, Southeast $1\frac{1}{2}$, Southwest $\frac{1}{2}$. Northwest $\frac{1}{2}$. Weather.—Clear $\frac{1}{2}$ day, more or less cloudy the rest of the time. Warmest day 21st. Coldest day 7th. Highest temperature of air 88°. Lowest 58°.

										A	LUG	US	т,	184	8.										
Ι.		овѕ	ERV	ATIO	NS C	N S	oil.				ATIO							WE	ATH	ER.					
DAY OF MONTH	4 inc. below	surface.	9 inc. below	surface.	2 feet below	surface.	4 feet below	surface.	Horse	Chestnut.	Black	Walnut.		A	ir.		Direction of	Wind.		Force of Wind.			Degree of cloudiness.		Inches rain.
	8 a. m	3 p.m	8 a. m	3 p. m	8 a.m	3 p.m	8 a.m	3 p.m	8 a. m	3 p. m	8 a.m	3 p.m	6 a.m	8 a.m	3 p.m	9 p.m	8 a.m	$\begin{bmatrix} 3 \\ p \cdot m \end{bmatrix}$	a.ın	p.m	Mean	a.m	p.m	Mean	
1 2 3 4 5 6 6 7 8 9 10 11 12 13 14 15	65 64 64 65 66 ¹ / ₂ 63 64 66 66 66 66 70 69 68 71	68 68 68 68 68 67 68 69 70 73 74 74 75	$ \begin{array}{r} 66 \\ 65 \\ 65 \\ 66 \\ 66 \\ 65 \\ 65 \\ 66 \\ 66 \\ 66 \\ 66 \\ 72 \\ 70 \\ 72 \\ \end{array} $	66 66 66 66 66 66 65 66 68 69 70 70 69 70 2	67 67 67 67 67 67 67 67 67 67 68 68 68 68 69	$ \begin{array}{c} 67 \\ 67 \\ 67 \\ 67 \\ 67 \\ 67 \\ 66 \\ \frac{1}{2} \\ 66 \\ \frac{1}{2} \\ 67 \\ 67 \\ 67 \\ 67 \\ 68 \\ 68 \\ 69 \\ 32 \end{array} $	$64 \\ 64 \\ 64$	64 64 64 64 64 64 64 64 64 63 62 62 62 62 62	$ \begin{array}{c} 68\frac{1}{2} \\ 67 \\ 67 \\ 70 \\ 71 \\ 68 \\ 67 \\ 69 \\ 70 \\ 71 \\ 72 \\ 73 \\ 73 \\ 74 \\ 74 \\ \end{array} $	69 69 70 70 71 70 69 71 73 73 74 75 75 75	66 \(\frac{1}{2}\) 64 \(\frac{1}{2}\) 65 \(\frac{6}{2}\) 68 \(\frac{6}{2}\) 67 \(\frac{64}{2}\) 68 \(\frac{1}{2}\) \(\frac{1}{2}\) \(\frac{6}{2}\) \(\frac{1}{2}\) \(\frac{1}{2}\) \(\frac{6}{2}\) \(\frac{1}{2}\) \(\frac{1}{2}\) \(\frac{1}{2}\) \(\frac{6}{2}\) \(\frac{1}{2}\) \(\frac{1}\) \(\frac{1}\) \(\frac{1}\) \(\f	68 68±2 70 70 69 68 70 70 	64 61 62 68 71 64 60 63 67 65 68 76 70 70	68 69 70 68 73 72 66 69 72 70 70 78 72 74 78	80 80 83 76 82 73 84 90 86 90 93 91 86	70 71 73 74 71 67 73 75 74 76 78 78 78 76 81 80	W NW S S W S S S S S S S	W W S S W W S S S S S S S S S S	1 1 1 2 2 1 1 1 1 1 1 1 2 2	2 1 1 2 2 2 1 1 1 1 1 1 1 2 2	$\begin{bmatrix} 1\frac{1}{2} \\ 1 \\ 1 \\ 2 \\ 2 \\ 1 \\ 1\frac{1}{2} \\ 1 \\ 1 \\ 1 \\ 1 \\ 2 \\ 1.3 \end{bmatrix}$	1 0 1 3 1 1 0 0 1 0 0 2 4 1 1	2 1 4 3 2 0 0 2 0 1 2 1 2	$1\frac{1}{2}$ 1 $3\frac{1}{2}$ 2 $1\frac{1}{2}$ 0 $1\frac{1}{2}$ 3 1 $1\frac{1}{2}$ 3	0.05
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S.m. Mo.	m. 6 mc 6		65 . 66 .			.84	62. 63.							69. 71.						1	.37			2.06 1.69	3.7

Observations.—1. Heavy dew. The potato rot has made its appearance in the vicinity of Boston. 3. Heavy dew the 2d, 3d and 4th. 4. The Rumex crispus begins to ripen its seeds. Large numbers of caterpillars have commenced feeding upon the leaves of the Juglans nigra. 5. Rain 12 m. to 2 a. m. 6. The potato rot has made its appearance in the vicinity of the Helderbergh. 7. The Juglans nigra continues to grow vigorously, the caterpillars have nearly divested it of its leaves. It was cut down this morning. 8th, 9th and 10th smoky. 11. Heavy dews from the 6th to 11th. In Dr. Wendell's, Mr. Prentice's, Mr. Walsh's and other orchards around Albany, the pear, quince and apple trees have commenced dying. The leaves of the grape vine also are dying. 14. The potatoes are rotting in Maine, especially along the sea coast. 15. Warm and dry. The temperature of the air at 8 a. m. and 3 p. m. from the 1st to the 15th is 77.86°, being 3.56° higher than the corn plants, and 6.86° higher than that of the Horse Chestnut. 17. Rain from 5 p. m. to 12 m. with thunder. 18. Rain 12 m. at night to 9 a. m. The potatoes still continue to be affected on Long Island with the rot, especially the less hardy varieties. 21. The potato rot still continue in Maine. 22. The potatoes continue to rot slowly on the Helderberg. 24. The Pittsburgh Gazette says that $\frac{1}{3}$ of the potato crop in that vicinity is affected with the rot. From the 20th to the 22d warm, pleasant, smoky. 28. Rain 12 m. to 12 at night. 29. Rain 12 at night to 6 a. m.

Winds.—North 7 days, East 0. South 11 $\frac{1}{3}$, West 5, Northeast $\frac{1}{3}$, Southeast $\frac{1}{3}$ Southwest $\frac{1}{3}$ Northwest 1. Weather.—Fair $\frac{1}{3}$ days. More or less cloudy $\frac{24}{2}$ days. Rain on 5 days. Warmest day 12th. Coldest 26th. Highest temperature 92°. Lowest 55°.

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DAY OF MONTH	4 inc. below	surface.	9 inc. below	surface.	2 feet below	surface.	4 feet below	suriace.	Horse	Chestnut.	Black	Walnut.		Ai	r.		Direction of	Wind.		Vorce of Wind.		ş	Degree of cloudiness.		Inches rain.
	8 a.m	3 p. m	8 a.m	3 p.m	8 a.m	3 p.m	8 a.m	3 p. m	8 a.m	3 p.m	8 a.m	3 p.n	6 a.n	8 a.n	3 p.n	9 1 p. n	8 na.m	3 p. m	a. m	p.m	Mean	a.m	p.m	Mean	
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15	65 63 63 65 64 63 60 58 62 57 58 59 54 54 58	67 65 65 69 69 66 64 64 65 64 59 58 55	65 64 65 65 66 63 62 63 62 63 58 57 58	$\begin{array}{c} 66 \\ 64\frac{1}{2} \\ 65 \\ 68 \\ 67 \\ 66 \\ 64\frac{1}{2} \\ 65 \\ 62\frac{1}{2} \\ 62 \\ 60 \\ 61 \\ 58 \\ 59 \\ \end{array}$	$ \begin{array}{c} 66 \\ 66 \\ 66 \\ 66 \\ 66 \\ 64 \\ 64 \\ 63\frac{1}{2} \\ 62\frac{1}{2} \\ 61 \\ 60 \end{array} $	63°_{62} 60^{1}_{2} 60°	$\begin{array}{c} 62 \\ 62 \\ 62 \\ 62 \\ \underline{1}_{2} \\ \underline{1}_{$	$\begin{array}{c} 62 \\ 62\frac{1}{2} \\ 62 \\ 61\frac{1}{2} \\ 61 \end{array}$	••				72 66 63 65 62 61 52 50 62 48 56 64 45 45	73 69 66 69 66 58 54 64 56 59 65 51 49 62	82 74 80 88 81 80 79 80 78 81 76 63 68 53 60	69 69 70 72 73 62 60 65 61 63 66 53 52 57	WWWNNWSWWNWS	W W W NW SW N N S W NE S W NE S W	1 1 1 1 1 1 1 2 1 2 1 2 2	2 3 2 1 2 1 2 1 2 1 2 1 2 3 3 3 3 3 3 3	$\begin{array}{c} 1 \\ 2 \\ 1 \\ 1 \\ 1 \\ 1 \\ 1 \\ 1 \\ 1 \\ 1 \\$	1 4 4 3 1 1 0 0 2 0 1 6 0 4 5	2 3 3 4 1 2 1 2 0 3 6 1 5 6	$\begin{array}{c} 1_{\frac{1}{3}\frac{1}{3}\frac{1}{2}\frac{1}{2}}\\ 3_{\frac{1}{2}\frac{1}{2}}\\ 3_{\frac{1}{2}}\\ 1_{\frac{1}{2}}\\ 2_{\frac{1}{2}}\\ 0_{\frac{1}{3}\frac{1}{2}\frac{1}{2}}\\ 4_{\frac{1}{3}\frac{1}{2}\frac{1}{2}}\\ 5_{\frac{1}{3}}\\ \end{array}$	0.06 0.04 0.15 0.43 0.39
.m.r	n. 61	.83	45.		64	.22	62	.13						6	4.48							1.56		2.36	1.1
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Observations.—1. Light dew. Smoky. 3. Light shower at 4 a. m. 4. Light shower early a. m. 5. Heavy dew. Smoky and fair in forenoon. Rain 6 to 10 p. m. 6. The rot is affecting the potato crop considerably in the central and western parts of the State. Many crops are almost entirely destroyed. It has not affected them much in this vicinity. 7. Heavy dew. Smoky. 8. Heavy dew. Cool night. Leaves of trees begin to turn slightly yellow. 9. Very smoky. Warm. 10. Light dew. Smoky. 11. Light dew. Smoky. Rain 6 p. m. to 12 at night. Rain 12 at night to 9 a. m. 13. Light dew. Total eclipse of moon last night. Air cool. 14. Heavy dew. Rain p. m. with thunder. 15. Rain early a. m. 16. Cool air. Smoky. Light dew. 17. Very foggy morning. Light frost on banks of river, not enough to injure vegetation. Rain 5 p. m. to 12 at night. 18. Rain 12 at night to 4 a. m. and at 6. p. m. 19. Smoky. 20. Rain during the day and night with thunder. 21. Smoky. 22. Light dew. Smoky. Rain 2 p. m. and during night. 23. Damp air. Smoky. 24. Rain a. m. Last night rain in Albany. A light snow storm on the Helderbergh. 25. Rain p. m. Smoky. 26. Smoky. Stiff N. breeze. Cool. 27. Light frost. Foggy. Pleasant autumn days. 28. Very smoky. 29. Light shower a. m. 30. Rain early a. m. Smoky. Warm day.

Winds.—North 8 days. East \(\frac{1}{2} \), South 6\(\frac{1}{2} \), West 9\(\frac{1}{2} \), Northeast 1, Southeast 0, Southwest 2, Northwest 1\(\frac{1}{2} \). Weather.—Warmest day, 4th. Coldest 27th. Highest range 88°. Lowest 34°. Rain on 15 days. Light frosts on the mornings of the 17th and 27th. Clear 3\(\frac{1}{2} \). More or less cloudy the remainder of the time.

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4 inc. below	surface.	9 inc. below	surface.	2 feet below	surface.	4 feet below	surface.	Horse	Chestnut.	Black	Walnut.		A	ir.		Direction of	Wind.		Force of Wind.			Degree of cloudiness.		Inches rain.
8 a.m	3 p.m	8 a.m	3 p. m	8 a. m	3 p.m	8 a.m	3 p.m	8 a. m	3 p.m	8 a.m	3 p.m	6 a.m	8 a.m	3 p.m	9 p.m	8 a.m	3 p.m	a.m	p. m	Mean	a.m	p.m	Mean	
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Observations.—1. Smoky. Rain from 3 p. m. to 12 at night. 2. Rain from 12 at night to 12 m. and from 2 p. m. to 12 at night. 3. Rain 12 at night to 6 a. m. and from 3 p. m. to 12 at night. 4. Rain from 12 at night to 8 a. m. The potato crop in this vieinity, except on the Helderbergh, has been injured but little as yet with the rot. The crop in Massaehusetts and Connecticut, especially if late planted, are very good both in quality and quantity. The rot has appeared in Newfoundland, the eastern part of Maine, the northern part of Vermont, and in the Western States generally. In Western and Central New-York, although the vines are stout, yet in many cases but few potatoes are found under them. 5. Very smoky. 6. Very smoky. 7. Heavy dew with a dense fog. 8. Light dew. Rain early a. m. Leaves of Horse Chestnut tree begin to die. Fruit is falling. 9. Heavy frost. 10. Light shower in morning. 11. Light frost. Very smoky. 12. Light frost. Very smoky. 13. Heavy dew. Smoky. 14. Smoky. 15. Heavy dew. Smoky. 16. Smoky. Rain in evening. 17. Light dew. Rain from 9 p. m. till 12 at night. 18. Rain from 12 at night to 12 m. and from 4 p. m. to 6. p. m. 19. Light showers at intervals through the day. 20. Light showers from 12 at night to 8 a. m. 21. Light dew. Rain 3 p. m. and during night. 23. Light frost. 24. Rain from 8 a. m till night. 25. Smoky. 26. Smoky. 27. Heavy dew. Very smoky. 28. Heavy dew. Smoky. Light shower in afternoon.

Winds.—North 12½ days, East 0, South 4½, West 8½, N.E. 1, S.E. ½, S.W. 1, N.W. 2, Monthly force 2.13.

Weather.—Warmest day 17th. Coldest 30. Highest range 68°. Lowest 34°. Rain on 13 days. Clear 3½ days. More or less eloudy the remainder of the time. Frost on the mornings of the 9th, 11th, 12th and 23d.

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DAY OF MONTH	4 inc. below	surface.	9 inc. below	surface.	2 feet below	surface.	4 feet below	surface.	Horse	Chestnut.	Black	Walnut.		A	ir.		Direction of			Force of			Degree of	• 692	Inches rain.
Q	8 a.m	3 p.m	8 a.m	3 p.m	8 a.m	3 p.m	8 a. m	3 p.m	8 a.m	3 p.m	8 a. m	3 p.n	a a r	n 8 n a.m	3 p.n	9 p.n	a, n	ր.ա	a. m	p.n	Mean	a. m	p.n	Mean	
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15	$\begin{array}{c} 43 \\ 42 \\ 43 \\ 44 \\ 52 \\ 40 \\ 38 \\ 36 \\ 32\frac{1}{2} \\ 32\frac{1}{2} \\ 32\frac{1}{2} \\ 32\frac{1}{2} \\ 32\frac{1}{2} \\ 33 \\ \end{array}$	34	47 46 46 44 48 42 41 39 40 36 35 36 35 36 35 36	48 47 46 46 46 44 43 41 36 36 36 36 37 36	38½ 39½ 39½ 40	47 47 47 47 47 47 44 44 43 42 38 39 40 40	51 51 50 50 50 49 49 47 49 47 46 47	51 54 50 60 50 50 49 49 47 48 47 47	43 42 56 39 39 39 36 37	48 43 43 44			37 40 40 36 58 38 36 31 37 21 19 28 33 25 35	40 42 40 41 57 37 38 36 37 22 22 30 37 30 34	57 56 50 46 ¹ / ₂ 43 43 38 46 32 32 29 31 40 31 40	46 42 41 53 38 40 36 39 28 22 30 33 34 41	SW SE SW SW SE SW N NE N NE N NE SE	S W W SE S SW SW W W N N N SE S	1 2 1 6 1 3 1 2 2 1 1 1	233353423321121	$\begin{array}{c} 1^{\frac{1}{2}\frac{1}{12}} \\ 2^{\frac{1}{2}\frac{1}{2}} \\ 2^{\frac{1}{2}} \\ 2^{\frac{1}{2}\frac{1}{2}\frac{1}{12}} \\ 2^{\frac{1}{2}\frac{1}{2}\frac{1}{12}} \\ 2^{\frac{1}{2}\frac{1}{2}} \\ 1 \\ 1 \\ 1 \\ 1 \end{array}$	0 0 5 5 7 1 3 1 6 5 5 5 5 7	0 5 1 6 7 4 6 3 4 2 3 6 4 5 5	$\begin{bmatrix} 0 \\ 2^{\frac{1}{2}} \\ 3 \\ 5^{\frac{1}{2}} \\ 7 \\ 2^{\frac{1}{2}} \\ 4^{\frac{1}{2}} \\ 2 \\ 5 \\ 1^{\frac{1}{2}} \\ 3 \\ 6 \\ 4^{\frac{1}{2}} \\ 5 \\ 5 \\ \end{bmatrix}$	0.23
8.m.1	m	381	4	1	44	1 2	49	3						36	3				-		-				
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Observations.—1. Heavy frost. Smoky. 2. Light frost. Very smoky. Leaves falling fast. 3. Smoky, 4. Light frost. Smoky. Rain from 12 m. till 12 at night. Tem. rain 46°. 5. Rain from 12 at night all day. Tem. 54°. a. m. and 43° p. m. 6. Heavy frost. Smoky. 7. Heavy frost. Slight snow p. m. 8. Heavy frost. Ice froze 1-16 inch during night. 9. Smoky. A little snow 2 p. m. Snow fell 10 inches deep in Buffalo. 10 soil froze from 1 to 2 inches deep. 11. Soil froze from 2 to 3 inches. Snow commenced at 9 p. m. and continued through the next day. 13. Snow 4 inches deep. Melting fast. 14. Smoky. 15. Smoky. Snow and rain a. m. 16. Froze slightly during night. Smoky. 17. Smoky. 18. Froze slightly during night. 19. Froze slightly during night. Smoky. 20. Snow from 8 a. m. to 2 p. m. 21. Snow 2 inches deep. Smoky. 22. Smoky. 23. Smoky. 24. Rain p. m. 25. Smoky. 26. Snow 7 a. m. to 9 a. m. 28. Soil froze ½ inch deep during night. 29. Soil froze ½ inches during night. 30. Smoky. 31. Smoky.

Winds.—North 6½, East ½, South 8, West 6½, Northeast 1, Southeast 3. Northwest 1½, Southwest 4, Weather.—Clear 1½ days. More or less cloudy the rest of time. Rain on 4 days. Snow on 6 days. Warmest day 5th. Coldest day 10th. Highest temperature 58 degrees. Lowest temperature 19 degrees.

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н.		OBSE	RVA	TION	s on	r so:	ırs.				ATIO	NS						WEA	тне	R.	-				
DAY OF MONTH.	4 inc. below	Sui lace.	9 inc. below	surface.	2 feet below	sui lace.	4 feet below	adi idee	Horse	Chestinut.	Black	walnut.		Ai	r.		Direction of	Wind.		Force of Wind.		,	Degree of Cloudiness.		Inches rain.
	8 a.m	3 p.m	8 a.m	3 p.m	8 a. m	3 p, m	8 a.m	3 9. m	8 a.m	3 p. m	8 a.m	3 p.m	6 a. m	8 a.m	3 p.m	9 p. m	a.m	p. m	a.m	p.m	Mean	a. m		Mean	
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15	34 37 36 35 36 38 38 38 37 33 37 40	40 45 40 38 38 36 36 38 39 42 40 37 34 38 40	40 41 38 38 38 39 30 38 38 40 42 39 36 38 40	39 42 41 40 39 38 37 39 42 42 43 40 35 38 40	38 40	42 40 39 38 39 42 40 40 40 39 39 40	44 ¹ / ₂ 43 ¹ /	43 43 43 43 43 43 43 43 43 43 43 43 43 4	40 38 39 37 34 44 43 40 41 37 33 37 37 40 37 37 37 40 37 37 37 40 37 37 40 40 40 40 40 40 40 40 40 40 40 40 40	39 38 38 41 40 36 38 46 44 45 40 36 33 42 43				34 44 42 34 37 34 43 42 40 40 34 25 39 43	38 48 41 45 37 34 38 54 44 47 38 32 30 44 41		W S W NW NE N S S W S NW W W SE S	SW W		232222222222222222222222222222222222222	$\begin{bmatrix} 2^{\frac{1}{12}} \\ 2^{\frac{1}{2}} \\ 2 \\ 2 \\ 2 \\ 2^{\frac{1}{12}} \\ 2 \\ 2^{\frac{1}{12}} \\ 2 \\ 1^{\frac{1}{12}} \\ 1^{\frac{1}{12}} \\ 2 \\ 2 \\ 1^{\frac{1}{12}} \\ 2 \\ 1^{\frac{1}{12}} \\ 2 \\ 2 \\ 1^{\frac{1}{12}} \\ 2 \\ 2 \\ 1^{\frac{1}{12}} \\ 2 \\ 2 \\ 2 \\ 1^{\frac{1}{12}} \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 \\ 2 $	5 7 2 6 6 6 6 7 6 1 7 5 6 6 6 7 7	4 7 2 6 7 7 6 6 4 6 4 7 4 6 2 6 2	$\begin{array}{c} 4\frac{1}{2} \\ 7 \\ 2 \\ 6\frac{1}{2} \\ 6\frac{1}{2$	
I	nm. 3			39		40		$43\frac{1}{2}$	00	1)67				00	$39\frac{1}{2}$		GYY			- 6 7			0.1		
16 17 18 19 20 21 22 23 24 25 26 27 28 30 31	35 33 33 33 33 33 33 33 33 33 33 33 33 3		36 38 57 41 40 39 36 35 35 35 35 35 35 35 35	38 40 38 44 42 36 35 35 35 35 35 35 35 34 34		34 35 35 35 35	42 42 42 41 41 41 41 40	42 42 42 41 41 41 41 41 41	$\begin{array}{cccccccccccccccccccccccccccccccccccc$	33 29 26 28 33 21 22 27				28 41 40 42 ¹ / ₂ 36 26 13 10 10 35 29 13 31 28 32	37 46 47 62 34 22 18 15 28 38 30 29 35		SW N S N N N S S N W N N W W	N	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	3 1 2 2 2 3 4 2 5 2 2 2 2 2 2 2 2 2 2 1 1	21 1 1 2 2 2 2 1 1 2 2 2 1 1 1 2 2 2 1	5 3 4 5 7 1 6 6 1 5 1 7	6 2 4 5 7 7 1 7 6 3 7 1 	$ \begin{array}{cccccccccccccccccccccccccccccccccccc$	
3.n	n.m me.	$\frac{34}{35\frac{1}{2}}$	35 34	53	3		4	$\frac{1\frac{3}{4}}{2\frac{1}{2}}$						30)1						- 2				

Observations.—2. Rainy. Heavy showers a. m. 3. Smoky. 4. Smoky. 5. Rain during night and from 9 a.m. to 12 at night. 6. Rain early a. m. till 12 at night. 7. Rain early a. m. all day. 8. Rain carly a. m. P. m. pleasant. Smoky. 9. Smoky. 10. Rained all night and most of day. 11. Smoky. 12. Commenced snowing 2 p. m. and continued at intervals through the night. 13. Smoky. 14. Smoky. Rain from 8 a. m. to 10 a. m. 16. Soil froze \(\frac{1}{4}\) inch deep. 17. Rain during uight. Day pleasant. 18. Smoky. 19. Smoky. 20. Smoky. 21. Snow fell \(\frac{1}{2}\) linch deep last night. Commenced again 11 a. m. and continued through the day. 22. Snow 3 p. m. 12 inches deep. 23. Smoky, pleasant. 24. Hail and snow from 8 a. m. to 3 p. m. 25. Rain from 11 a. m. through the day. 26. Smoky. 27. Snow commenced falling 8 a. m. at 3. p. m. it had fallen 3 inches deep. 28. 8 inches of snow has fallen since 8 a. m. yesterday. 30, 4 inches of snow fell during night. 31. Pleasant. Thaws in sun.

OBSERVATIONS

ON THE TEMPERATURE ON THE SOIL, AT HOOSIC FALLS, RENSSELAER COUNTY, NEW-YORK.

By L. C. BALL, Esq.

These observations were made daily during the past summer. I have selected only the extremes of each half month as reported. Temperature of the soil 4 inches below the surface.

	FIRST	HALF.	SECOND	HALF.
	Air.	Soil.	Air.	Soil.
MAY.			l	
Highest,			91	78
Highest, Lowest,			54	52
Mean,			764	673
JUNE.			-	-
Highest,	85	80	101	90
Lowest,	58	55	72	72
Mean,	$72\frac{1}{3}$	653	853	801
July.	2	*	•	
Highest,	94	84	100	81
Lowest,	72	66	70	63
Mean,	781	76	80	75
August.	*			
Highest,	100	87	109	82
Lowest,	55	71	60	63
Mean,	891	781	803	721
September.	. 3	2	*	_
Highest,	86	89	72	64
Lowest,	51	49	50	50
Mean,	703	67	62	58₹

Note.—The observations were continued through October and a part of November, but being taken at the depth of one foot, and a change also being made as to the position of the air thermometer, they are omitted, as they do not correspond with those made in the preceding months.

OBSERVATIONS.

ON THE TEMPERATURE OF THE SOIL, MADE AT SCOTT, CORTLAND COUNTY, NEW-YORK. By Mr. C. B. SALISBURY.

OBSERVATION	NS AT 5 A. M. OR BEF	MARCH, 1848.	OBSERVATIONS AT 3 P. M.
DAY OF MONTH.	Temperature of air in shade. Tem. of naked soil 2 in. below surfaue. Tem. soil 2 feet below the surface. Tem. soil 4 feet below the surface.	Temperature dew, Temperature rain. Depth rain fallen sinee yesterday 3 p. m. Degrec of eloudines, Course of wind.	Temperature of air in shade. Tem. loose naked soil 2 inches below surface. Tem. soil 2 feet below the surface. Tem. soil 4 feet below the surface. Tem. rain. Depth rain fallen since this morning 5 a. m. Degree of cloudiness. Course of wind.
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30			32 32 31 33 31 5 NW 26 21 31 33 32 3 NW 14 13 32 33 32 2 NW 28 22 33 34 32 0 NW 48 34 34 35 33 0 S 51 40 2 32 33 32 3 S 20 20 32 33 4 NW 36 30 32 34 6 N 11 11 32 34 3 NW 3 1 1 32 34 3 NW 3 1 2 32 34 3 NW 3 1 2 32 34 3 NW 3 1 2 32 34 3 NW 3 1 3 3 3 3 NW 3 1 3 3 3 3 NW 3 2 3 3 3 3 NW 3 3 3 3 3 3 NW 4 4 3 3 3 3 NW 5 4 4 3 3 3 4 NW 5 5 3 3 3 3 4 NW 5 4 4 3 3 3 4 NW 5 4 4 3 3 3 4 NW 5 4 4 3 3 3 4 NW 5 5 3 3 3 3 3 4 NW 5 5 3 3 3 3 4 NW 5 5 3 3 3 3 3 4 NW 5 5 3 3 3 3 3 4 NW 5 5 3 3 3 3 3 3 3 5 5 5 3 3 3 3 3 6 5 6 5 3

REMARKS AT 5 A. M.—4. Snow 8. inches deep.. 5. Much drifted. 6. Snowing. 9. Fell 1 inch snow. 10. Snow fell 6 inches. 11. Snow fell 4 inches. 12. Snowing a little. 13. Snowing. 15. 2 inches snow fell. 18. Hazy. 19. Hazy. 20. Foggy. 21. Some foggy. Sap runs. 23. Snowing. 27. Foggy. 28. Foggy.

REMARKS AT 3 P. M.—3. Snowing. 4. 4 more inches of snow. 5. Ground is frozen about 8 inches, bare. 8. Smoky. Sap runs. 9. Snowing. 12. Raining. 13. Snowing a little. 14. Snowing a little. Sleighing. 19. Raining. Turned to snow. 20. Rain. Thunder and lightning at night. 21. Little rain. Sap runs. 26. Foggy. 27. Lowry. 31. Mud and snow mostly disappeared.

OBSERVATION	S AT 5 A	. M. or	вего	RE SUN	RISE.					01	BSER	VAT)	ions	ат З	P. 1	M.
DAY OF MONTH.	remperature of air in shade.	rem. 100se nakeu son z inches below surface. Tem of soil 9 feet below surface.	Tem of soil 4 feet below surface.	Tem. dew. Tem. rain.	Depth rain fallen since yesterday, 3 p. m.	Degree of cloudiness.	Course of wind.	Force of wind.	Tem. air in shade.	loose naked s below sur	Tem. soil 2 feet below surface.	Tem. soil 4 feet below surface.	ain.	Depth rain fallen since morning, 5 a. m.	Degree of cloudiness.	Course wind.
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 -23 24 25 26 27 28 29	35 39	35 36 37 38 39 39 39 39 40 40 39 43 42 40 42 40 43 40 42 40 40 40 br>40 40 40 40 40 40 40 40 40 40 40 40 40 40 40 40 40 4	36 38 37 38	336	3 in 12	4 N N S S N N N N N N N N N N N N N N N	IW IW SS IW W IW IW IW IW IW IW IW IW IW IW IW I	22442131311134322251111322112	42 40 49 40 38 44 56 66 65 66 65 66 39 54 41 52 37 50 58 66 69 66 69 66 66 66 66 66 66 66 66 66	48	35 36 37 37 39 39 39 39 40 42 42 42 42 42 42 42 42 42 42 42 42 42	36 38 38 38 38 38 38 38 39 40 40 40 41 41 41 41 42 42 42 42 42 42 42 42		1-16 in	123431113233543322122	NW S S NW NW NW NW NW NW NW S S NW NW NW NW S S NW NW NW NW S S NW NW NW NW NW NW NW NW NW NW NW NW NW

Remarks at 5 a. m.-1. Sprinkle of snow. 14. Snow, rain, hail. 20. Smoky. 21. Smoky. 22. Smoky. heavy frost. 24. Foggy on streams. 26. Hazy. 27. Heavy frost. 28. Smoky. Remarks at 3 p. m.-13. Some show. 17. Hazy. 18. Smoky. 22. Some rain. 23. Smoky. 28. Smoky.

				Μ.	AY	, 18	348.											
OBSERVATIONS	3 АТ 5 А	. м.,	OR BEF	ORE	SUNI	RISE.					0	BSEF	RVAT	ions	ат 3	Р.	м.	
DAY OF MONTH.		Tem. loose naked soil 4 inches below surface.	soil 4 surfac	Tem. dew.	Tem. rain.	Depth rain fallen since yesterday, 3 p. m.	Degree of cloudiness.	Course wind.	Force wind.	Tem. air in shade.	Tem. loose soil 4 inches below surface.	1em. soil 2 feet below surface.	Tem. soil 4 feet below surface.	Tem. rain.	Depth rain fallen since morning, 5 a. m.	Degree of cloudiness.	Course wind.	Fores wind
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30	36 40 38 40 55 51 54 40 44 48 53 54 56 57 53 40 50 60 60 60 60 60 60 60 60 60 60 60 60 60	43 44 44 54 55 54 55 55 56 66 58 66 66 68 66 66 68 66 66 66 66 66 66 66	51	50-h 51-h 46-h 54 56 60-h 552-h	60	1 1 2 3 3 8 9–16 1–16 	34413122436532151113333412113	S S S S S S S S S S S S S S S S S S S	233131122122311441122224112244	55 61 67 78 82 80 66 56 57 48 50 60 60 58 54 54 54 60 60 78 86 63 76 63 77 88 88 88 88 88	52 55 61 72 73 82 82 68 64 63 53 50 66 65 75 78 72 73 82 75 75 78 87 75 77 88 88 88 88 88 88 88 88 88 88 88 88	44 46 47 50 50 51 51 51 49 49 49 50 52 54 54 54 54 54 55 56 58	42 43 44 44 44 44 44 44 46 46 46 46 46 48 48 48 48 48 49 49 49 50 50 50 50 50 50 50 50 50 50 50 50 50	69 49 50 50 	7–16	53223292 :56332231143355332125	S NEW S W N N N N S S N N N N N S S N N N N	

Remarks at 5 a. m.—Smoky and hazy. 6. Wild plum in blossom. 9. Raining. 11. Raining still. 16. Raining. 20. Fruit trees in full bloom. 23. Raining. Foggy.

Remarks at 3 p. m.—1. Rain. 4. Hazy and smoky. 5. Some thunder. Sprinkle of rain. 6. Woods begin to look green. 7. Heavy thunder shower. Hail storm, breaking windows and doing much damage. 10. Raining. 11. Raining. 19. Heavy thunder shower. 21. Some rain and thunder. 22. Dense cold fog. Lowry. 31. Looking for a frost.

OBSERVATIONS	ат 5	A. M	., or	BEF	ORE	sun	RISE.					0	BSEI	RVAT	IONS A	т 3	P. 1	1.	
DAY OF MONTH.	Temperature of air in shade.	Tem. loose soil 4 inehes below surface.	Tem. soil 2 feet below surface.	Tem. soil 4 feet below surface.	Tem. dew.	Tem. rain.	Depth rain fallen sinee yesterday, 3 p. m.	Degree of eloudiness.	Course of wind.	Foree of wind.	Tem. air in shade.	Tem. loose soil 4 inelies below surface.	Tem. soil 2 feet below surface.	Tem. soil 4 feet below surface.	Tem. rain. Depth rain fallen sinee	rning, 5 a. m	Degree of cloudiness.	Course wind.	Foree wind.
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30	30 50 62 55 60 50 50 50 50 50 50 50 5	40 51 52 55 55 55 55 55 55 55	566 566 566 566 566 556 556 555 557 588 560 660 660 660 660 660 660 660 660	51 52 52 52 52 52 52 52 52 52 52 52 52 52	30 fr 54 h 46 h 52 h 52 59 56	62 38 49 54 68 63 56 	1.16 1.16 1.16	133244541311112112332244421135	NW S S NW	3231333321234321221221323521221	52 72 72 72 72 72 72 85 56 56 56 57 58 83 88 88 78 79 70 66 86 86 86 86 86 86 86	56 83 75 81 69 54 52 62 70 70 73 55 64 59 70 79 78 68 68 69 70 69 70 70 78 85 85 85 85 85 85 85 86 86 86 86 86 86 86 86 86 86 86 86 86	56 56 56 56 56 56 56 56 56 56 56 56 56 5	51 51 52 52 52 52 52 52 52 52 52 52 52 52 52	65	-16	1 4 2 3 5 4 4 1 1 1 2 1 1 2 2 3 3 3 3 3 2 3 1 1 1 1 1	NW S S N NW N	

REMARKS AT 5 A. M.—1. Black frost, injured corn slightly. 3, Foggy. 4. Foggy. 5. Raining. 7. Foggy. 11. Light dew. 14. Smoky. 16. Hazy. 17. Heavy dew. 22. Heavy dew. Hazy. 26. Heavy dew. 28. Light dew. 30. Light dew. REMARKS AT 3 P. M.—13. Smoky. 14. Rain. 15. Beautiful weather. 17. Some smoky. 23. Thunder shower. 28. Heavy rain. Thunder. 30. Some rain. Sprinkle.

OBSERVATION	S AT 5 A. N	1., OR BE	FORE S	UNRISE.				obs	ERVAT	IONS AT	З Р.	м.
DAY OF MONTH.	Temperature of air in shade. Tem, loose soil finches		Tem. dew.	Tem. rain. Depth rain fallen since yesterday 3 p. m.	Degree of cloudiness. Course of wind.	Force of wind.	Tem of air in shade.	Tem. loose soil 4 inches below surface.		Tem. rain. Depth rain fallen since morning 5 a. m.	Degree cloudiness.	Course wind.
1 2 3 4 5 6 7 8 9 40 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30		0 62 56 0 62 56 0 62 5 1 63 5 1 7 7 8 1 7 8 7 8 1 7 8 8 1 8 8 8 1 8 8 8 8 8 1 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8 8	5 50 h 5 66 h 7 7 56 h 7 7 56 h 7 7 55 h 8 55 h 8 8 8 8 8 8 8 8 8 8 8 8 8 8	61 1-16	3 NV 2 NV V 2 NV V V NV V V NV V V V NV V V V	V V V V V V V V V V V V V V V V V V V	73 65 68 48 69 80 76 63 90 76 82 80 70 75 74 84 84 84 74 75	62 66 67 66 66 67 66 66 67 66 66 67 68 67 68 67 68 68 68 68 68 68 68 68 68 68 68 68 68	32 54 56 56 52 56 52 57 52 57 52 57 52 57 52 56 51 56 52 56 52 56 53 58 53 58 53 58 53 58 53 58 53 58 53 58 53 58 53 58 63 58 63 58 63 58 63 58 63 58 63 58 63 58 63 58 63 58 63 58 63 58 63 58 63 58 63 58 63 <	65 3	343212225334330322 : : : : : : : : : : : : : : : : : :	NW SWNW NW SE SS SNWWNW NW N

REMARKS AT 5 A. M.—2. Light dew. 3. Continued raining all night. Still at it. 10. Lowry and foggy. 13. Foggy. 14. Foggy. 24. Raining.

REMARKS AT 3 P. M.—2. Commenced raining. Some thunder at evening. 4. Ceased raining p. m. 9. Lowry. 10. Thunder. 11. Some thunder. 12. Raining. 13. Some thunder. 16. Sprinkle rain. 24. Day showery. 31. Raining.

^{*} The blank space above was owing to loosing some of our notes.

		AUG	UST, 1848.	·
OBSERVATIO	NS AT 5 A.	M. OR BEFORE SU	NRISE.	observations at 3 p. m.
DAY OF MONTH.	Temperature of air in shade. Tem. of naked soil 2 in.	Tem Tem Tem	Depth rain fallen sinee yesterday 3 p. m. Degree of cloudines, Course of wind.	Force of wind. Temperature of air in shade. Tem. loose naked soil 2 inches below surface. Tem. soil 2 feet below the surface. Tem. soil 4 feet below the surface. Tem. rain. Tem. rain. Depth rain fallen since this morning 5 a. m. Degree of cloudiness.
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30	56 63 48 60 61 64 61 64 58 63 54 60 56 59 70 68 71 62 70 64 72 58 64 70 64 72 58 64 70 64 72 58 64 70 68 71 68 67 70 68 68 67 70 68 67 70 68 68 67 70 68 72 69 59 64 68 68 68 68 72 59 64 68 68 68 72 59 64 68 68 68 68 72 59 64 68 68 68 72 59 64 68 68 68 72 59 64 68 68 68 72	63 58 48 h		V

REMARKS AT 5 A. M.—2. Foggy in valley. 3. Smoky. 5. Smoky. 7. Clear and cool. 9. Hazy. 10. Smoky. 11. 12. Smoky. 13. Smoky. 15. Smoky. 18. Raining. 28. Raining. 29. Foggy and lowry. 30. Foggy.

REMARKS AT 3 P. M.—1. Smoky. 2. Smoky and hazy. 6. Hazy. 7. Smoky. 8. Very smoky. 10. Smoky. 11. Smoky. 12. Smoky. 14. Smoky. 15. Smoky. 16. Smoky. 17. Foggy. Raining commenced a. m. 19. Very changeable weather. 26. Sprinkle rain. Thunder.

OBSERVATION	8 AT 5	A. M.	OR BE	FORE	sun	RISE.					01	BSER	VAT	ons	ат З	Р.	м.	
, DAY OF MONTH.	Temperature of air in shade.		Tem of soil 9 feet below surface. Tem of soil 4 feet below	surface. Tem. dew.	Tem. rain.	Depth rain fallen sinee yesterday, 3 p. m.	Degree of cloudiness.	Course of wind.	Foree of wind.	Tem. air in shade.	Tem. loose naked soil 4 inches below surface.	Tem. soil 2 feet below surface.	Tem. soil 4 feet below surface.	Tem. rain.	Depth rain fallen since morning, 5 a. m.	Degree of cloudiness.	Course wind.	
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29	58 57 58 62 54 54 54 39 42 55 49 47 58 40 52 44 44 42 59 44 42 58 24 42 42	62 60 61 62 61 62 55 58 57 58 50 56 48 50 50 56 44 44 44 44 45 44 44 44 44	63 6663 6662 55662 556656 556 556 556 554 554 554 554 554	61 h 52 h 53 h 552 h 553 h 557 30 566	59	13 13 1-16	5663130005036006402403544355054	W NW N N N N N N N N N N N N N N N N N	1 1 2 1 1 2 2 2 2	63 67 72 82 66 63 65 72 61 70 62 58 48 48 48 55 58 70 56 52 38 49 52 59 44 59 59 44 59 59 59 59 59 59 59 59 59 59 59 59 59	666 644 656 662 600 655 645 550 550 550 550 550 550 550 550 550 5	63 62 62 62 62 62 62 62 62 62 60 60 58 58 56 56 56 56 56 54 54 54 53	59 59 59 59 59 59 59 59 59 59 59 59 59 5		1-16 	52202003606000632335253350236	NW NW NW NW NW NW NW SE NW	

REMARKS AT 5 A. M.—5. Smoky and foggy. 10. Frost. 13. Heavy frost. 18. Lowry. 21. Sprinkle of rain. 23. Hills white with snow. 27. Frost. 30. Raining and foggy.

REMARKS AT 3 P. M.—11. Thunder shower. 15. Raining. 22. Snow. 24. Sprinkle of rain. 29. Raining. 30. Raining.

		OCT	OBER	, 1848.								
OBSERVATIONS	AT 6 A. M., OR	BEFORE	SUNRISE.				OBSEF	VAT:	ions at 3	P. I	ı.	
DAY OF MONTH.	Temperature of air in shade. Tem. loose naked soil 4 inches below surface. Tem. soil 2 feet below surface.	Tem. soil 4 feet below surface. Tem. dew.	Tem. rain. Depth rain fallen sinee yesterday, 3 p. m.	Degree of cloudiness.	Foree wind.	Tem. air in shade.	Tem. loose soil 4 inches below surface. tem. soil 2 feet below surface.	Tem. soil 4 feet below surface.	Tem. rain. Depth rain fallen sinee morning, 5 a.m.	Degree of cloudiness.	Course wind.	Force wind.
1 2 3 4 5 6 7 8 9 10 11 12 13 14 15 16 17 18 19 20 21 22 23 24 25 26 27 28 29 30 31	54 53 56 53 50 53 50 53 36 53 36 53 34 53 32 54 30 54 32 54 32 54 32 54 32 54 32 54 33 53 34 52 34 52 34 52 34 52 34 52 34 52 34 52 34 52 34 52 35 51 51 51 53 51	54 54 54 54 54 54 54 54 54 55 52 52 52 52 52 53 53 53 53 53 54 54 54 54 54 55 56 57 58 59 50	54 7-16 53 1-16 52 1-16 1	6 S	2333212221221	56 52 54 64 60 50 54 56 44 58 36 48		54 54 54 54 54 54 54 54 54 54 54 54 54 5		$\begin{bmatrix} 0 & 3 & 4 & 4 & 1 \\ 4 & 4 & 1 & \ddots \\ 2 & \ddots & 0 & 0 & 0 & 0 & 2 & 2 & 2 & 5 & 2 & 6 & 4 & 4 & 3 & 3 & 1 & 7 & 2 & 3 & 3 & 3 & 7 & 4 & 3 & 3 & 1 & 7 & 2 & 3 & 3 & 3 & 7 & 4 & 3 & 3 & 3 $	S S S NW S N N N N N S S S S N N N N N N	1220223;12122134434322232255124

REMARKS AT 6 A. M.—1. Foggy. 2. Raining. Lowry. 7. Frost. 9. Hard frost. 11. Frost. 12. Frost. 13. Frost. 17. Snow and hail, soon disappears. 18. Raining. 27. Frost. 27. Heavy fog. Smoky atmosphere. 29. Rain. 30. Foggy. 31. Thunder.

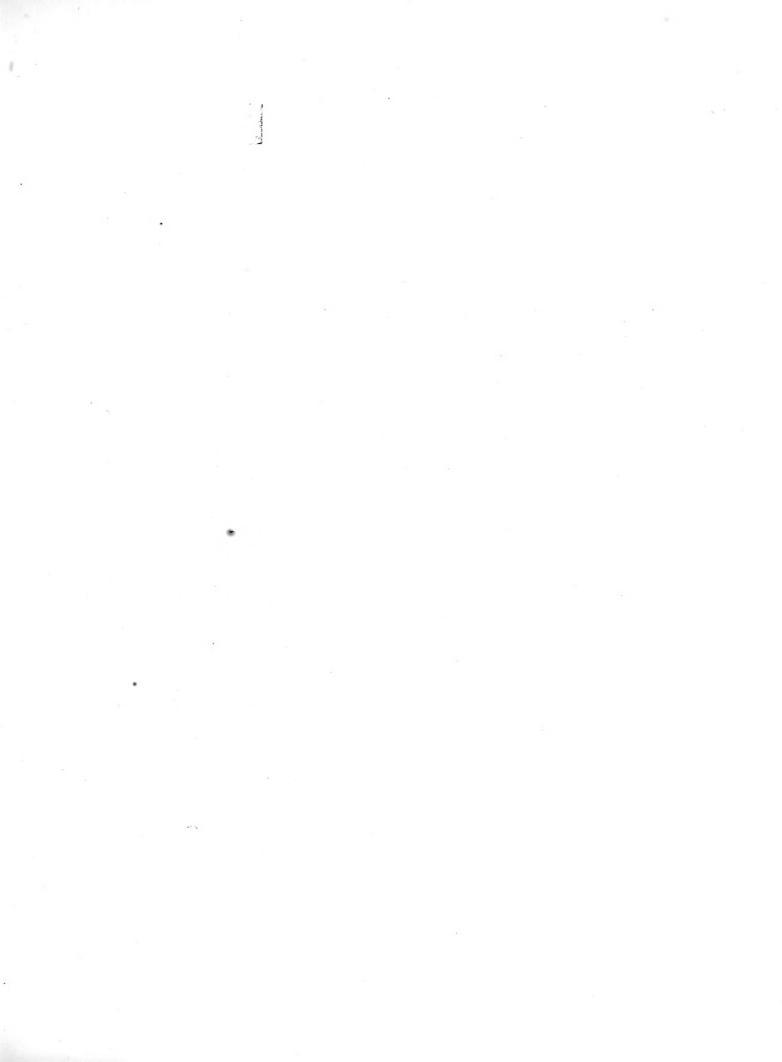
REMARKS AT 3 P. M.—1. Lowry. 2. Lowry. 4. Foggy. 19. Some rain. 24. Raining. 26. Sprinkle of rain. 28. Smoky. 29. Some rain. 30. Appearance of Rain.

TABLE

Showing the ranges of temperature of the air and soil at Cazenovia, Madison county, New-York, for December, 1848.

	1			1			1			1				1
	Tem	peratui	re air.	Tem.	2 feet urface.			4 feet urface.		Dire Wi		Wea	ther.]
DAY OF MONTH.	a. m.	p. m.	eve.	a. m.	p.m.	eve.	a. m	p.m.	eve.	a.m	p,m	a.m.	p.m.	G
9	32	33	33	41	40	40	40	40	40	NW	SE	Cl'dy.		
10	38	51	50	41	39	40	40	41	40	S	S		Fair.	
11	30	28	27	39	39	39	41	41	41	W	NW	C.	C.	
12	25	27	24	39	39	39	41	41	41	SW	SW	C.	C.	
13	23	32	31	38	38	38	41	41	41	SE	SE	C.	C.	
14	34	39	38	38	37	37	41	41	41	S	SE	F.	C.	
15	39	33	31	37	37	37	41	41	41	SW	W	R.	C.	
16	32	38	40	37	37	37	40	40	40	S	SE		C & R	
17	33	39	37	37	37	37	40	40	40	S	SW		F.	1.
18	33	44	44	37	37	37	40	40	40	sw	SE	F.	F.	1.
19	42	54	38	37	38	38	40	40	40	sw	sw	C.	C.	
20	27	31	28	38	38	38	40	40	40	sw	SW	C.	C.	
21	16	17	13	38	38	38	40	40	40		NW		F.	1.
22 23	8	18	20	38	38	38	40	40	40	SE	SE	F.	F.	1.
23	5	16	10	37	37	37	40	40	40	W	sw	C.	C.	
24	15	29	30	37	37	37	40	40	40	S	S	Sleet.		
25	32	39	32	36	36	36	40	40	40	S	S	R.	R.	ı.,
26	19	19	16	36	36	36	39	39	39	SW	w	C.	C.	
27	14	22	21	36	36	36	39	39	39	S	S	Snow	Snow.	
28	22	21	14	36	36	36	39	39	39	SW	SW	S.	S.	
29	21	25	23	36	36	36	39	39	39	S	S	C.	C.	١.,
30	24	29	23	36	35	35	39	39	39	SW	SW	C fog.	C.	١.,
31	23	24	23	35	35	35	39	39	39		SW		C.	l

Observations.—10. Rain commenced at 10 preceding night. 16. Rain at 11 p. m. 17. Aurora. 20. Light sonw. 22. Wind S. and SW. at sunset. 26. High Wind.



	3.1			
		(2)		
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1.4			- ÷	
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	. 1			

EXPLANATION OF THE PLATES.

It is necessary to state, in explanation of the irregularity in the numbering of the plates, that it was expected that this work would embrace several, which I afterwards found it necessary to arrange in another volume. Many of the plates were already printed before this arrangement was concluded. An error also was committed by the engraver in numbering the three plates of the illustration of the diseases or brand of the cereals. The owners of the volume are requested to make the proper correction with the pen.

PLATES I, II, III,

Which illustrate the diseases of the cereals, are referred to in the text as Pl. lv, lvi, lvii.

PLATES 1 b, 2 b, 3 b, 4 b, 5 b, 6 b,

Represent the most common varieties of potatoes which are cultivated in this vicinity. For descriptions, see the text with its references.

PLATE 4.

Marrow Squash, Vegetable Marrow or (as called in other places) Boston Marrow Squash. It has long been known and cultivated in the vicinity of Albany as the Vegetable Marrow, which is probably the earliest name under which it has been known.

PLATE 5.

Brazil Squash. It is a kind which was but recently introduced. It is a valuable variety, and worthy of cultivation. Color green: short or depressed, with the ends pressed inwards.

PLATE 6.

Custard Squash, is new in this vicinity. It is a fine kind, and ranks with the Vegetable Marrow in quality, though I am not ready to admit that it is superior. It is easily distinguished from other kinds by its strong longitudinal ribs, extending its entire length, and by their alternating with short and imperfect ones.

PLATE 7.

Egg-plant. Fruit thick and somewhat pyriform, purple: stem armed with spines.

PLATE 8.

Structure of wood as magnified about five diameters.

Fig. 1. Chestnut, with its bark, showing in the wood the annual layers which are distinguished by large pores arranged in three lines, and by faint whitish perpendicular bands which show the arrangement of the secondary pores. The bark, in this as in all trees, undergoes remarkable changes in the arrangement of the annual layers, becoming in many instances oblique and obscure. The wood is clearly different in structure from the oak (fig. 5).

[AGRICULTURL Rep. Vol. II — App.]

- Fig. 2. White Elm. The oblique and undulatory arrangement of the secondary pores of the wood is a feature characteristic of the elms. Primary pores in a single row. The wood of three years in contrast with which is the last annual layer of bark, which shows the perfect regularity of the medullary rays, which are continuous with those of the wood. Old bark corky and finely granular.
- Fig. 3. Bass-wood (Tilia americana). The structure of the bark is worthy of attention; the medullary rays unite in the form of a pointed brush or pencil.
- Fig. 4. Pawpaw (Asimina triloba). The circle of primary pores single, and more distinct than in the Bass-wood. The arrangement of the medullary rays of the bark is quite similar to the bass.
- Fig. 5. Chinquapin Oak. Two sections cut transversely, and magnified about five times. The striking difference of structure is manifest on comparing the oak with the chestnut (fig. 1). All the small figures of Plate 8 indicate the size of the sections.

PLATE 9.

- Fig. 1. Transverse section of a portion of two adjacent annual layers of the Poplar (Populus tremuloides). Secondary pores belong to three kinds: the large, which are divided by a thin diaphragm; the middle sized are of a pentangular form, and are irregular in their arrangement; the smallest contain dots or subordinate cells.
- Fig. 2. Section perpendicular to the medullary rays.
- Fig. 3. Section parallel to the medullary rays.
- Fig. 4. White Elder (Sambucus canadensis). Transverse section.
- Fig. 5. Section perpendicular to the medullary rays.
- Fig. 6. Section parallel to the medullary rays.
- Fig. 7. White-wood (Liriodendron tulipifera). Transverse section.
- Figs. 8 & 9. Correspond with the former.

PLATE 10.

- Fig. 1. Transverse section of the Walnut or Hickory. An interrupted single circle of large pores marks the commencement of growth of the wood. The interrupted belts indicate the direction of fine lines in the wood. The dark lines represent the medullary rays.
- Figs. 2 & 3. Sections perpendicular and parallel to the medullary rays.
- Fig. 4. White Oak (Quercus alba). Section transverse to the trunk shows, in the darker parallel lines, the structure of the medullary rays.
- Fig. 5. Shows the structure of the same ray, when cut by a line tangent to its line of growth.
- Fig. 6. Section parallel to the medullary ray, showing its structure as brought out by this cut.
- FIG. 7. BLACK OAK. The three sections correspond to those of the White Oak.

PLATE XI.

Coniferous woods. 1. Transverse section of White Cedar.

Figs. 4, 5, 6. Sections of the Red Cedar.

Figs. 7, 8, 9. Sections of the Norway Pine.

The coniferous woods or pines exhibit a remarkable similarity of structure, as revealed in these sections. Still there are differences worthy of note, and might be employed in distinguishing species from each other.

PLATE XII.

Figs. 1, 2, 3. Button-wood (*Platamus occidentalis*). Sections corresponding to those already noticed.

Fig. 4. Beech. Transverse section. The dark belt shows the singular structure of the medullary ray.

Figs. 5 & 6. Sections of the same wood.

Fig. 7. Buckthorn. The oblong spaces which are left nearly white, are the spots thickly implanted with secondary pores. The wood is easily distinguished from all others by its singular structure.

Fig. 8. CHERRY.

Fig. 9. Transverse section of the Plum.

PLATE 13.

- Fig. 1. Melia azedarach. Section magnified four diameters, containing both bark and wood.
- Fig. 2. Section of the potato, showing its structure in a healthy state. The hexagonal cells contain the starch granules. The spiral vessels are indicated by the three spirals upon the right; these vessels are always found in the vicinity of an eye. In a diseased potato, the cells are broken down more or less, but the starch granules remain unchanged in the midst of the decomposing fluids. These decomposing fluids contain the nitrogenous bodies, as albumen and casein, which are at first acted upon, and which results in their decomposition, as is evident from the exhalation of ammonia. Hence the nutritive powers of the diseased potato are first diminished, while the calorifient power is the last to pass away.
- Fig. 3. Dogwood (Cornus florida). Medullary rays are remarkably continuous in this wood, extending frequently from the centre to the circumference.
- Fig. 4. High Blackberry (Rubus villosus). Transverse section of the stem.
- Fig. 6. Raspberry. From comparison of the structure of the two stems, it would seem expedient to separate the High Blackberry from the Raspberry.
- Fig. 7. Transverse section of Black Ash.
- Fig. 8. Transverse section of White Birch.
- Fig. 9. Transverse section of the Black Birch (Betula excelsa). It appears, from the structure of the stems of the several species of birch, that they are more closely allied than many other species. So close is the resemblance, that it is extremely difficult to distinguish them: the resemblance, in fact, is greater than in two species of the Pine.

PLATE 14.

- Fig. 1. Transverse section of the Canada Balsam.
- Fig. 2. Section parallel to the medullary rays, showing the dotted tissue in a medullary ray.
- Fig. 3. Section perpendicular to the medullary rays.
- Fig. 4. Transverse section of the Tamarack.
- Fig. 5. Transverse section of Hemlock, magnified about five diameters.
- Fig. 6. Transverse section of Hemlock, greatly magnified.
- Fig. 7. Section of the White Elm parallel to the medullary rays, showing the spiral vessels of the wood.
- Fig. 8. Transverse section of the Taxus.
- Fig. 9. Transverse section of the White Elm.

PLATE 26.

Varieties of Maize, Early Tuscarora Corn. For description, see text under the head of Corn or or Maize.

PLATES 27, 28, 29.

Varieties of Wheat. For description, see the text.

PLATE 48.

Landreth's Early Pea. This pea, as its name implies, is early, and is valuable, inasmuch as it is equal in goodness to the Marrowfat. The pod shows that it bears well.

PLATE 49.

Fig. 1. Orange Carrot, showing the structure of its root by a tranverse section.

Fig. 2. a, longitudinal section of the Beet; b, transverse section. The yellowish white band communicates with the double row of fibres upon the outside.

PLATE 54.

BLACK TANTAIN OAT.

PLATE 57.

Two varieties of Barley. Fig. 3. Barley brand. Fig. 4. Spores of the barley brand, or smut, greatly magnified.

Plates illustrating the mean monthly ranges of temperature, and the semi-monthly mean for the year, have been compiled from registers kept at Albany for the year 1848. The register was kept by Mr. J. Salisbury. For a part of the year, a register has also been kept by Nathan Salisbury, Esq. of Scott, Cortland county. These observations I have tabulated also, and they appear on the plate of the same month with the Albany observations. In 1844, I kept a register of the temperature of the soil and air. The entire series of observations agree in the indications of the range of temperature for the periods at which they were made. Scott is about 1200 feet above Albany; Hoosic-falls, about 450 – 500.

The object in constructing these tables, was to present to the eye the facts which are so obscure to most minds when given merely in figures. Observations at Rensselaerville, Albany county, made by my friend Mr. H. A. Gallup, and also by Mr. Bannister of Cazenovia, agree very nearly with those made at Scott. These places differ only a few feet in height.

The inspection of these plates will convey a clear conception of the changes of temperature in this section of the State. This fact, however, would have appeared in a stronger light had I constructed them from the extremes of observed temperature for the day.

The earth, at the depth of four feet, it will be seen, presents for some months a uniform range, but little change occurring for many days in succession. The difference between two feet and four feet is quite remarkable, amounting to two or three degrees. It is interesting to observe how the ranges of temperature represented by the lines cut the arches formed by the air line at different periods of the year.

It may be proper to observe, in explanation of these plates, that on the left margin the degrees are placed, and the day of the month upon the upper. If it is wished to know what the mean temperature was for any day of any given month, it is found at the junction of the horizontal and perpendicular lines: the line of temperature touches at each intersection.

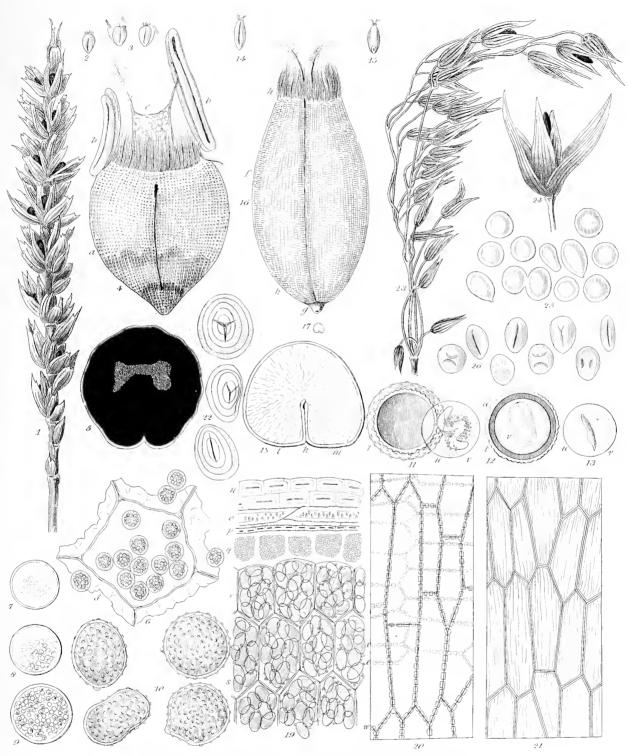


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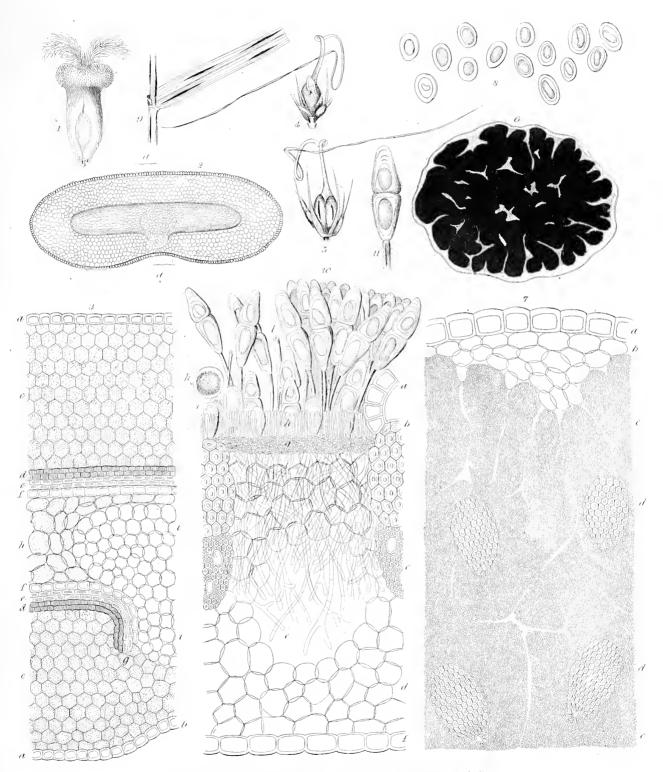


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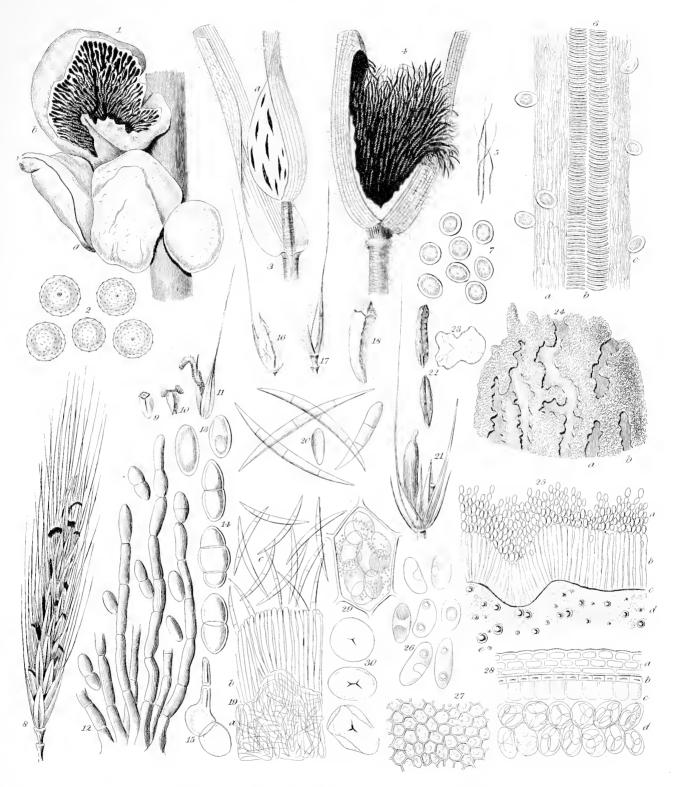
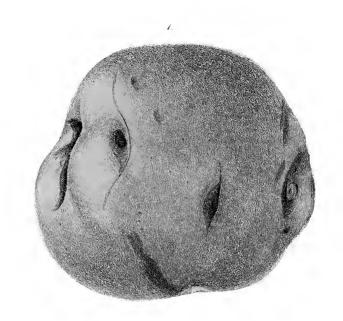
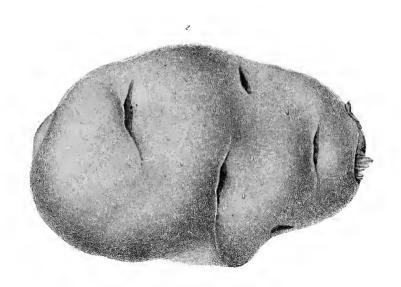


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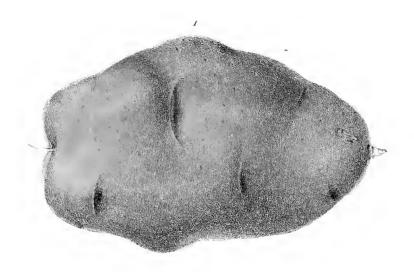
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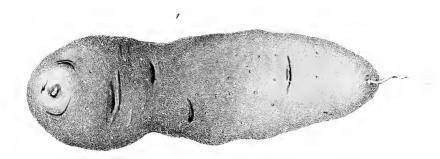
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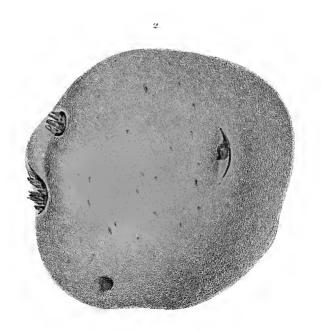
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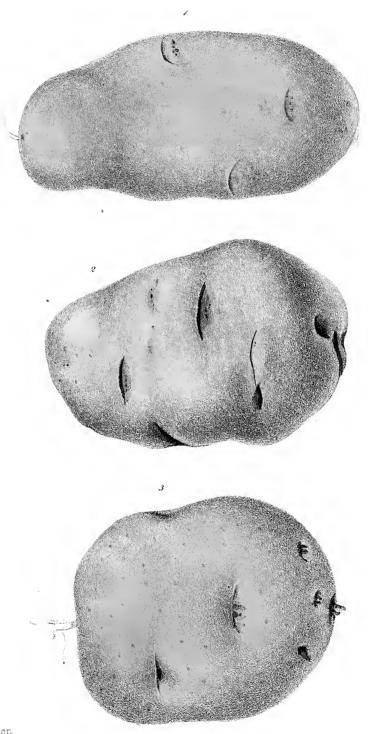


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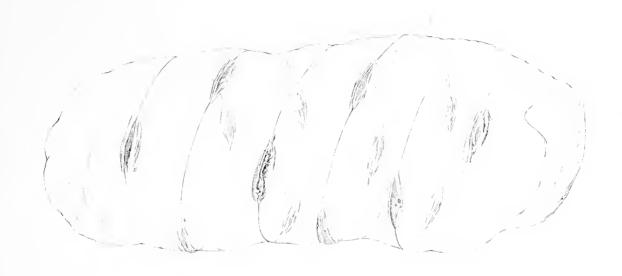


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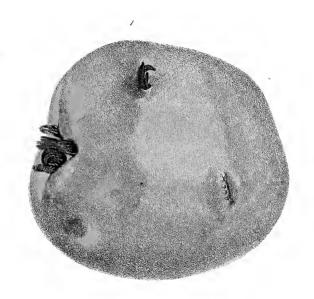
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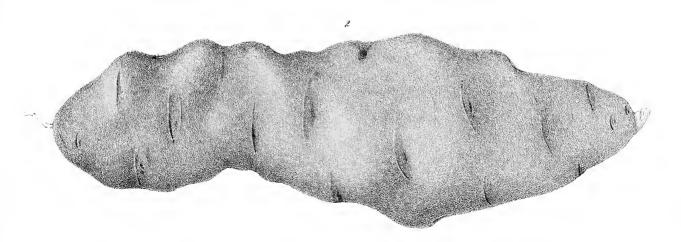




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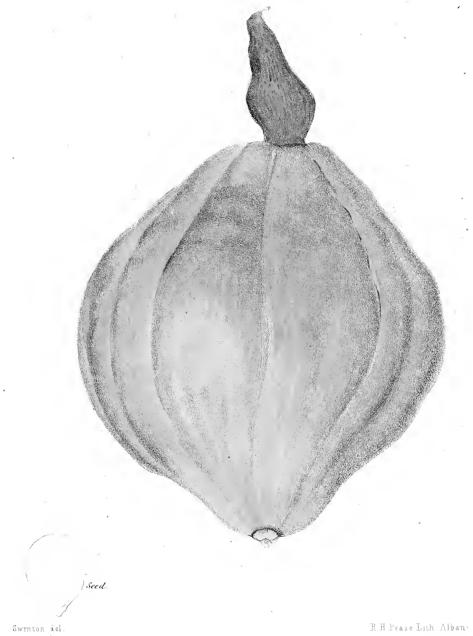


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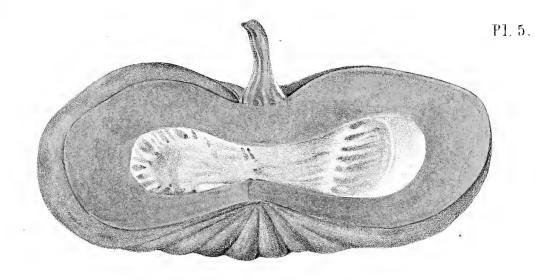
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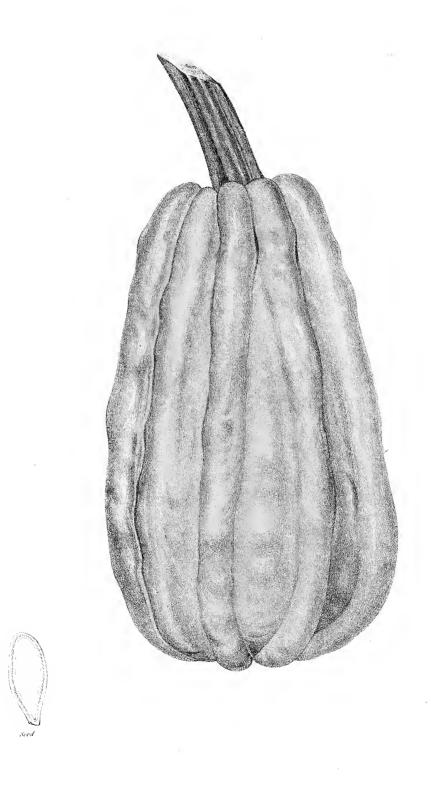
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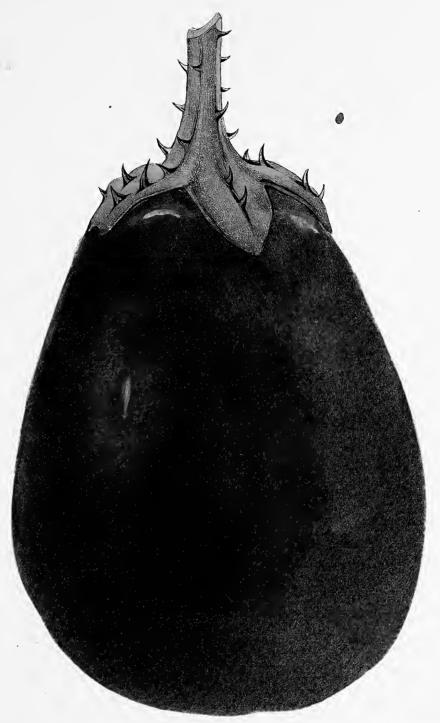


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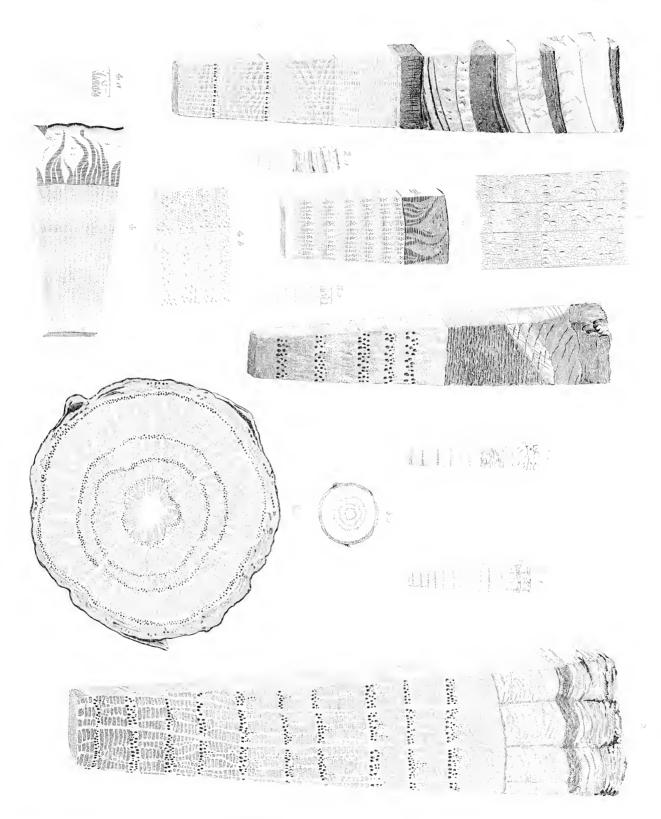


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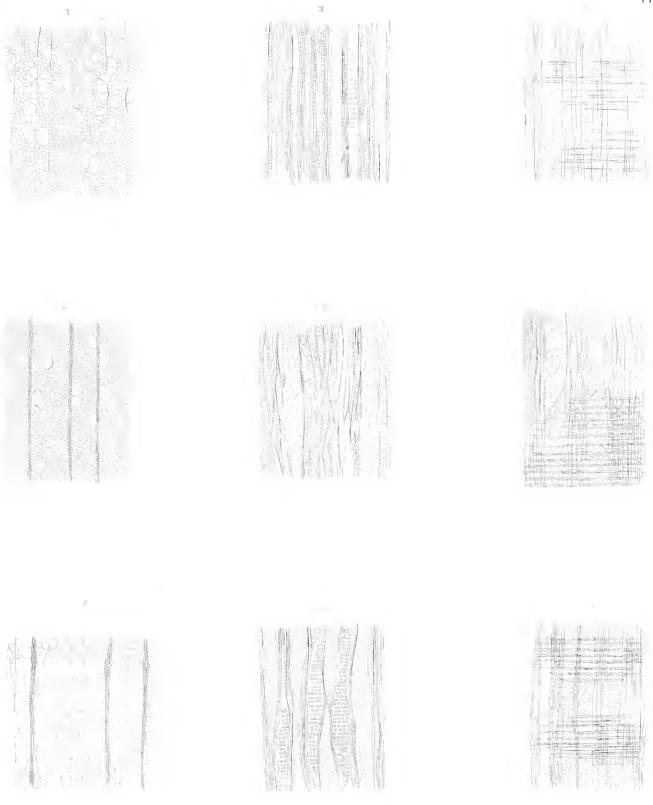
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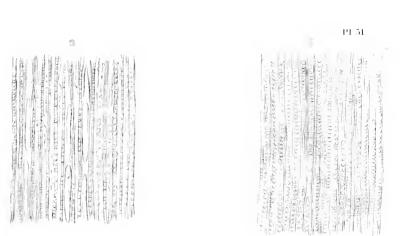


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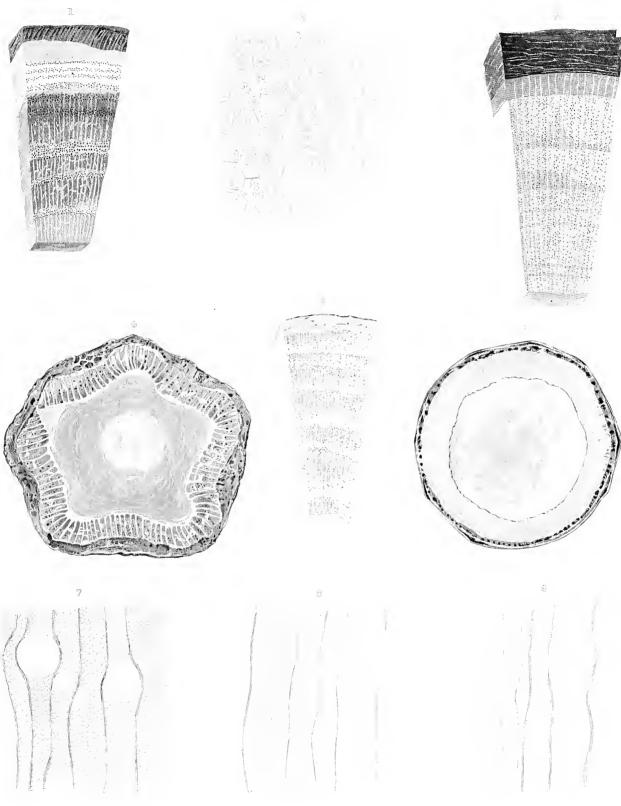
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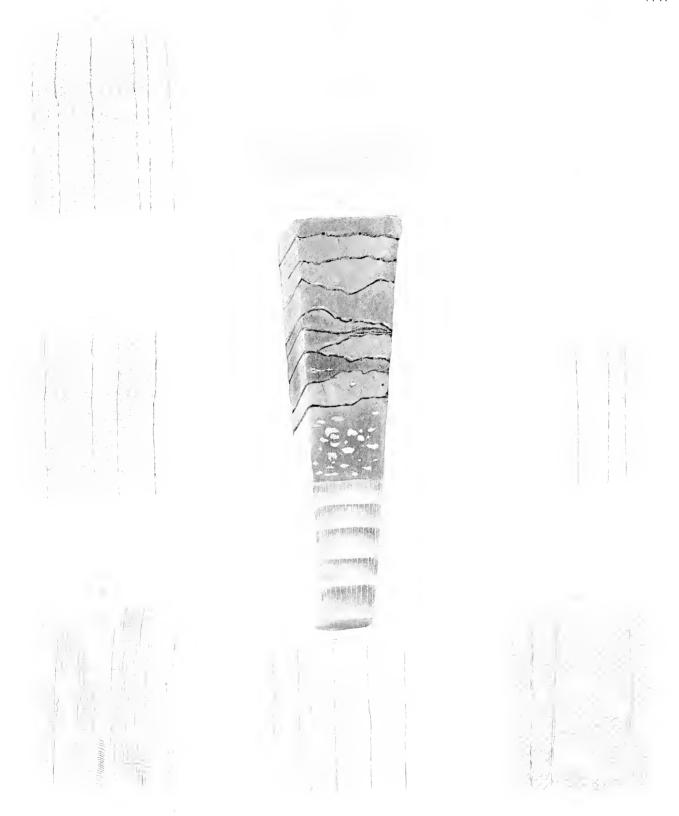




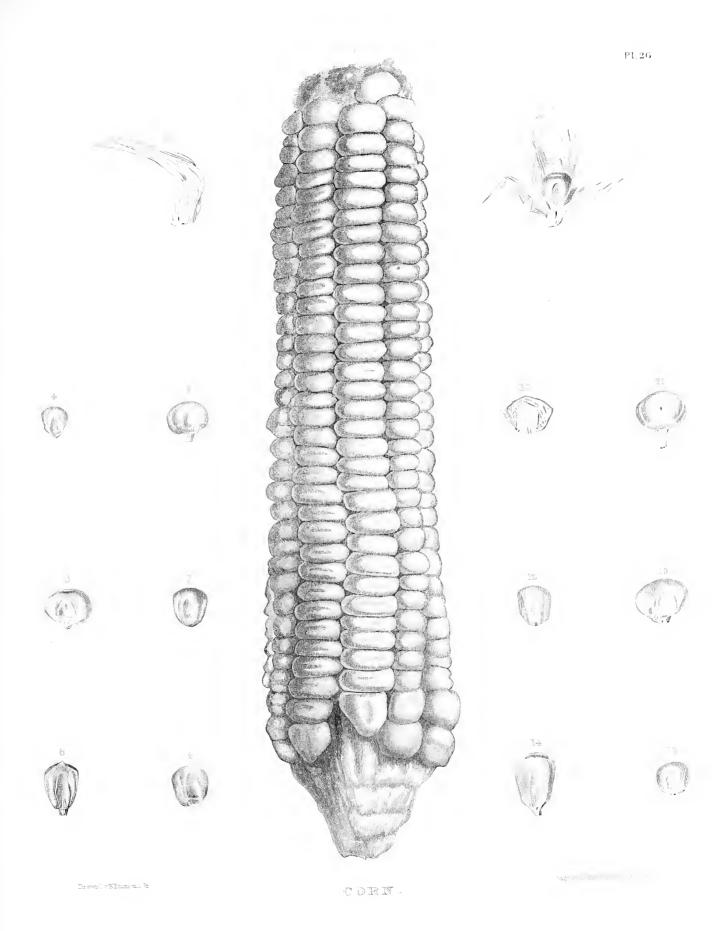
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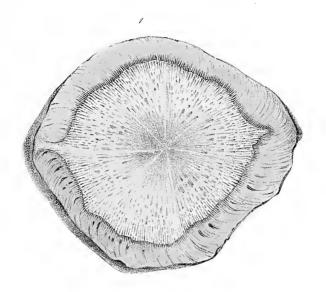


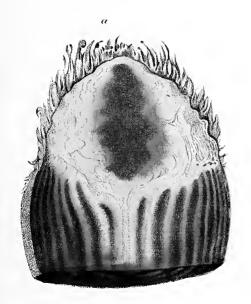
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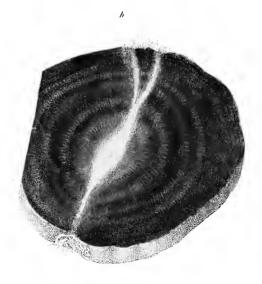
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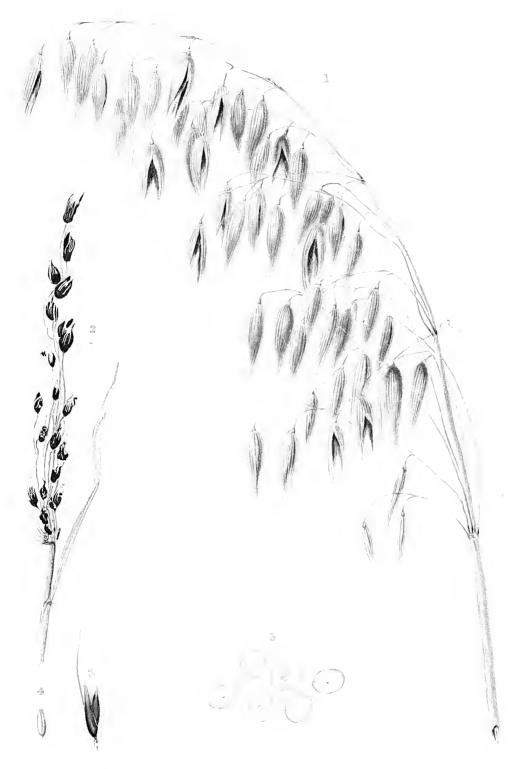
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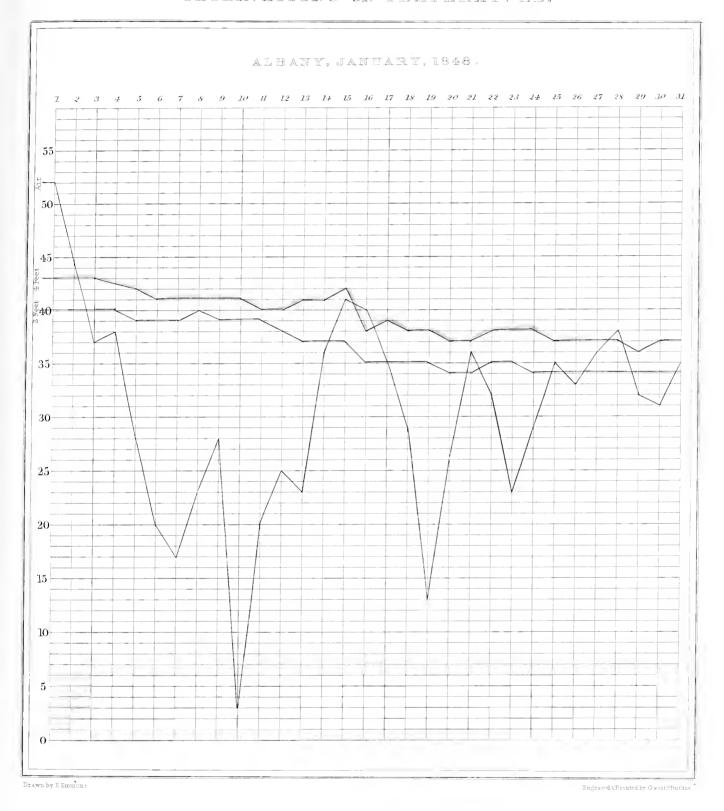
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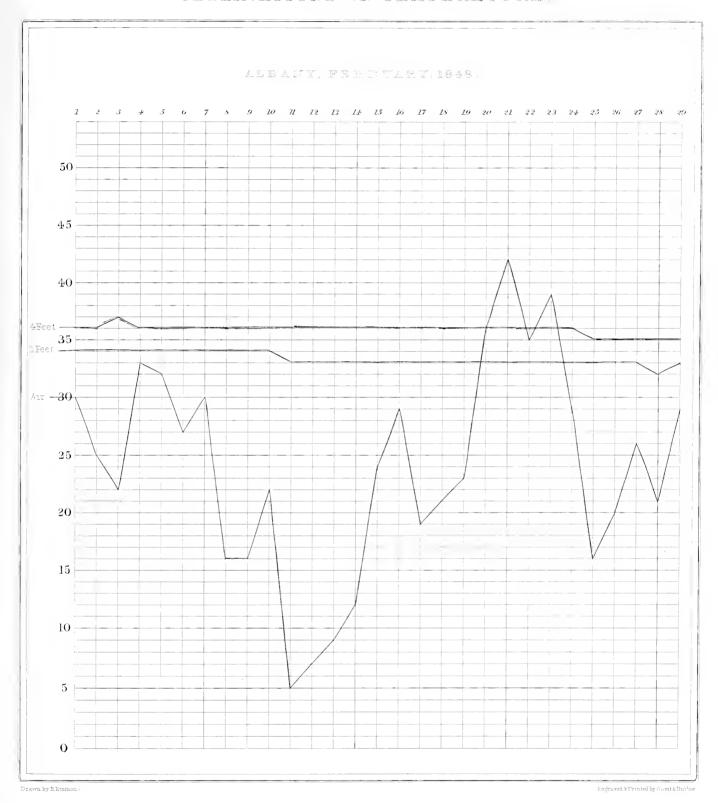
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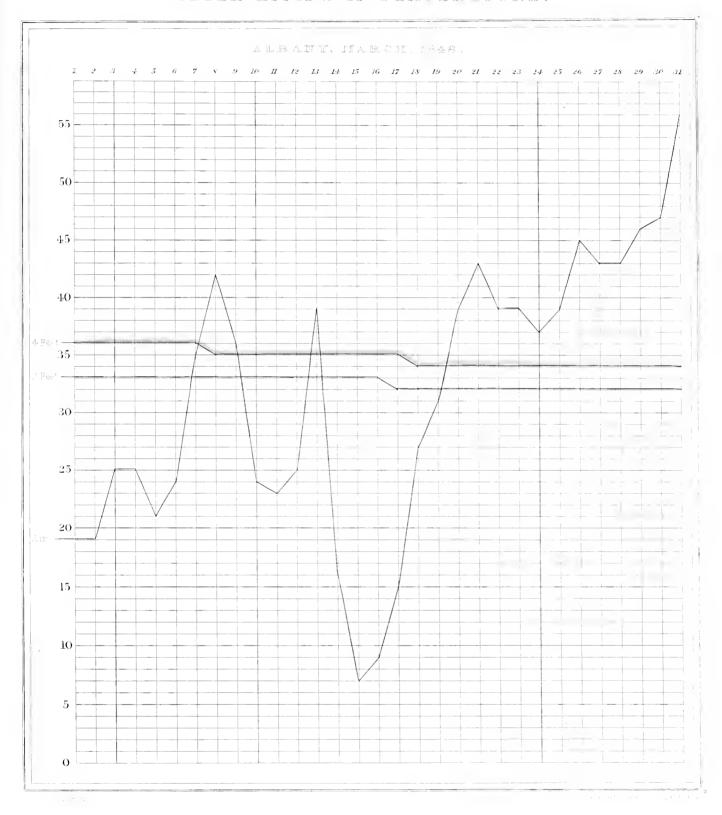
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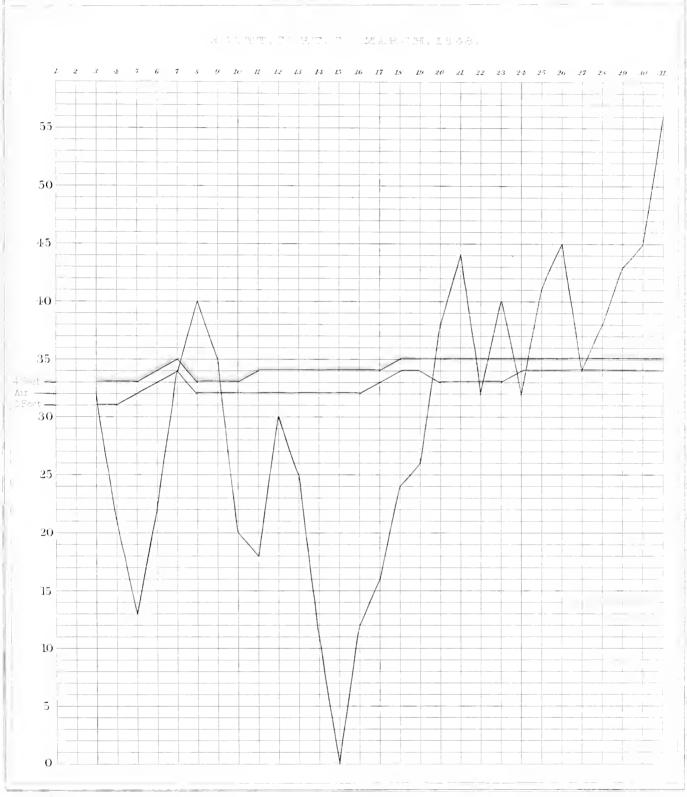


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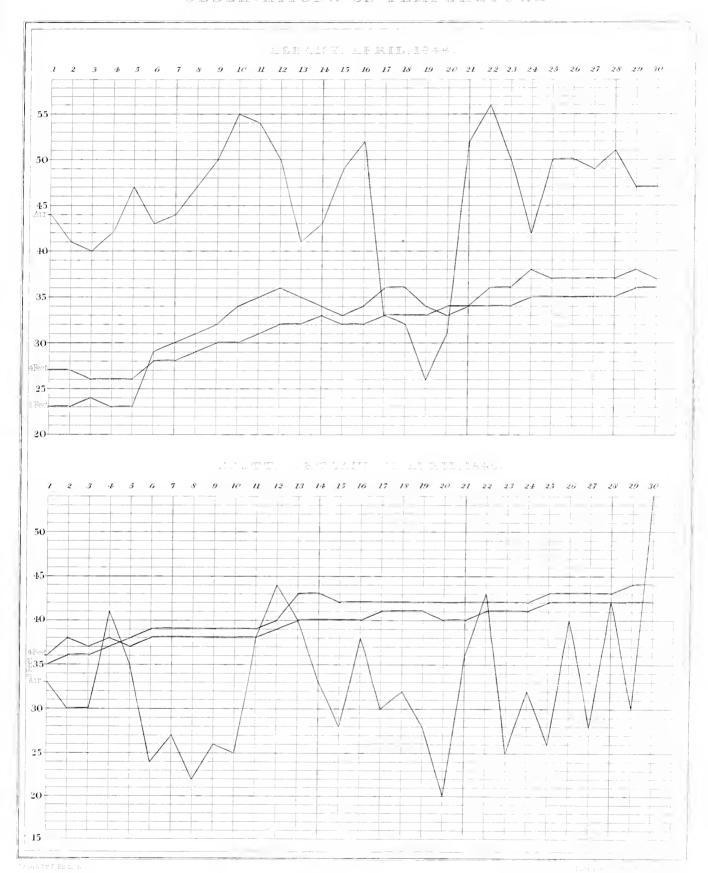


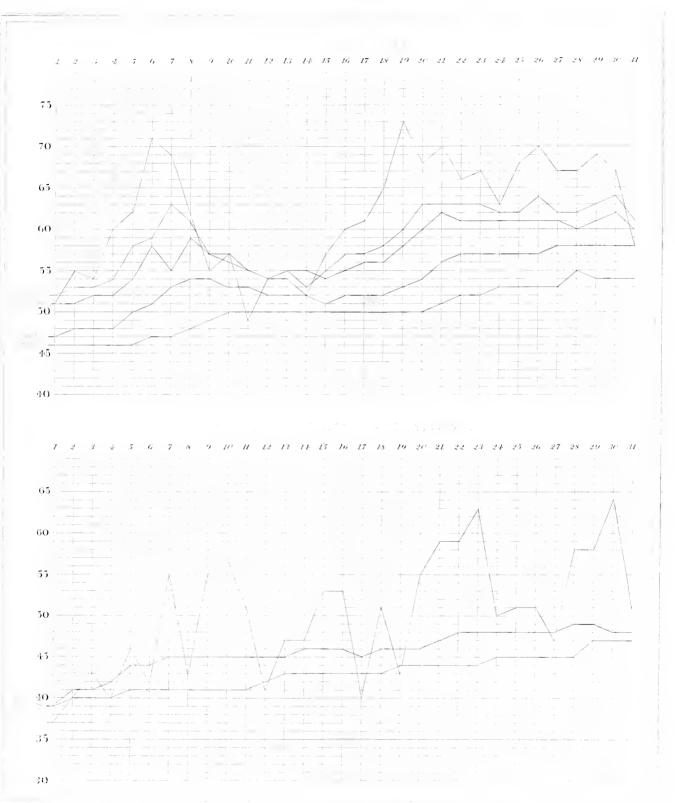
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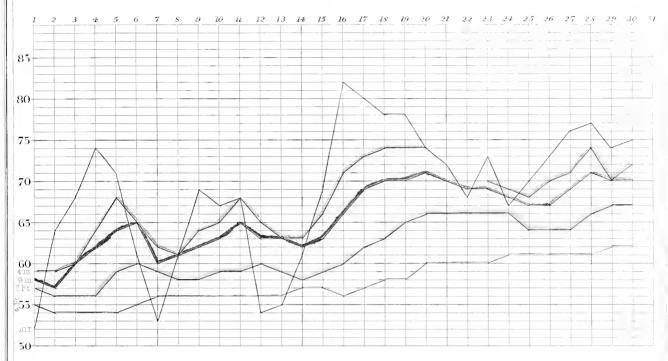
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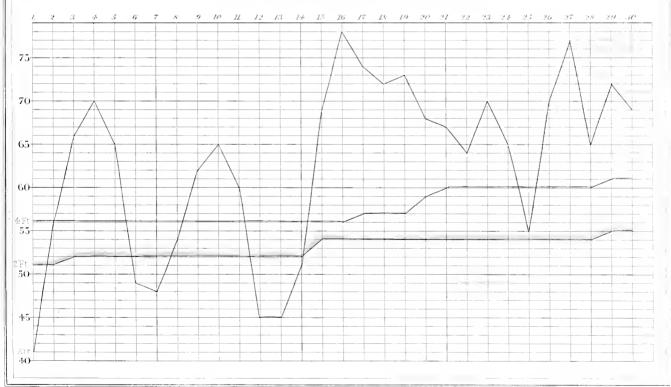


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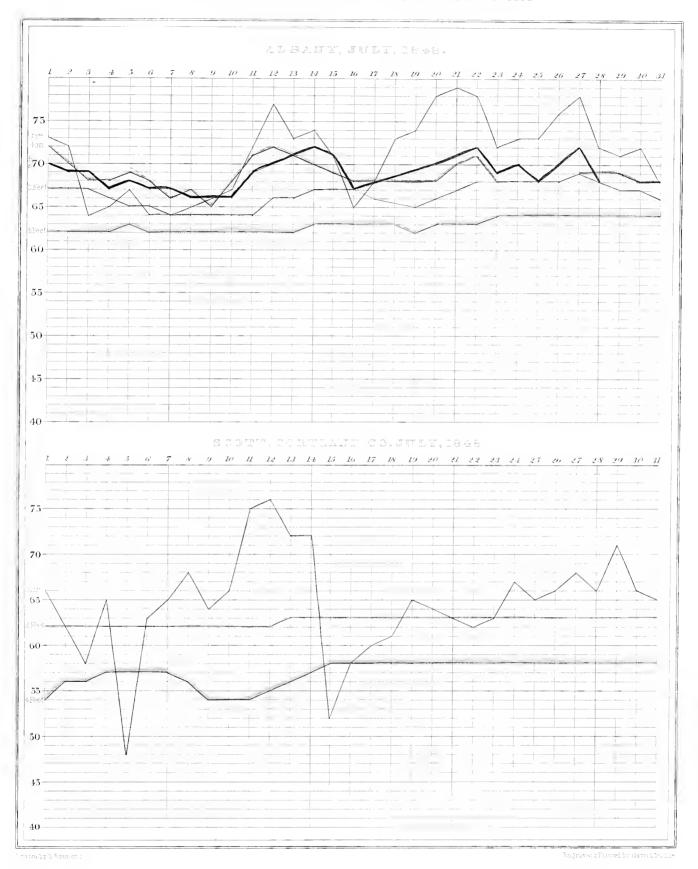




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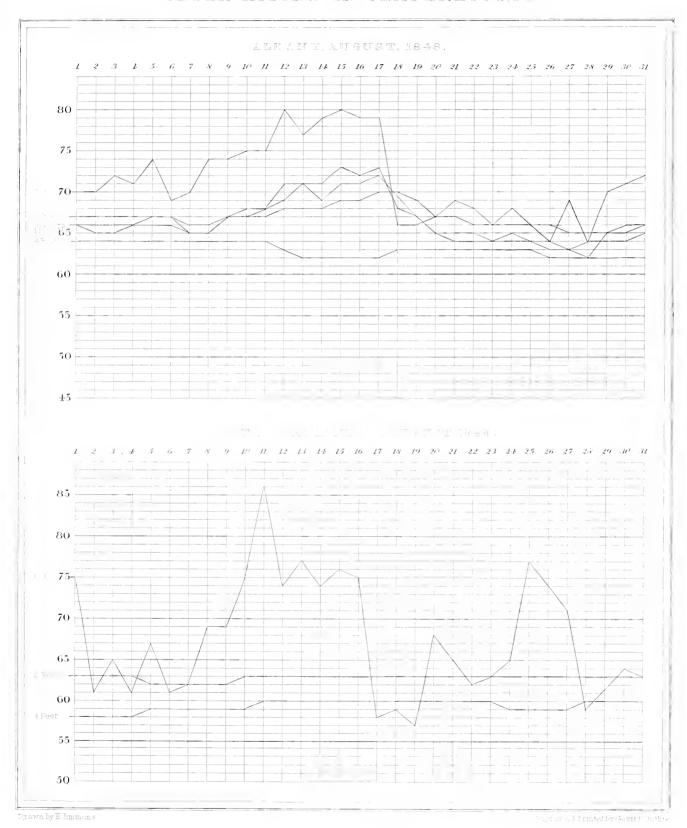


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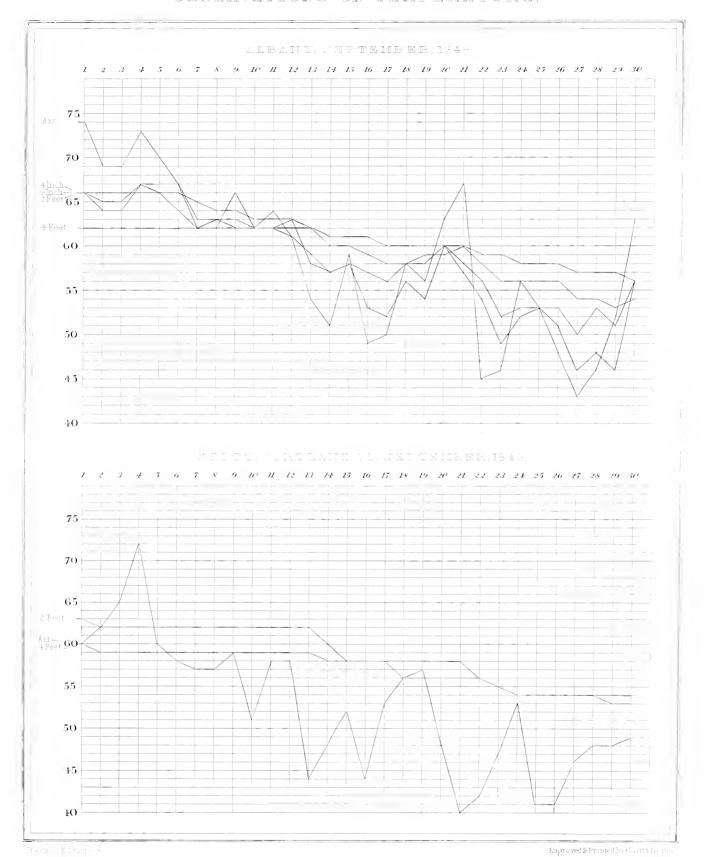
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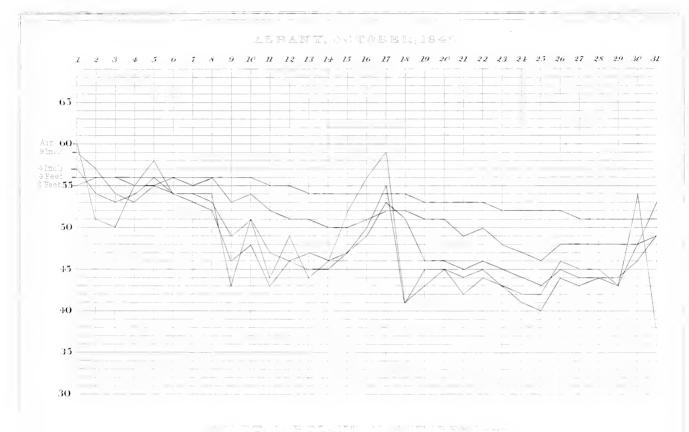


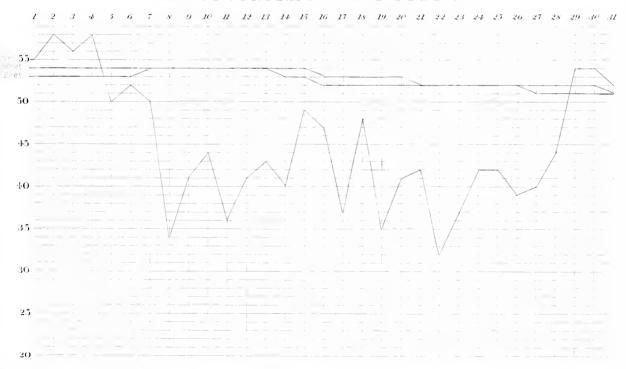
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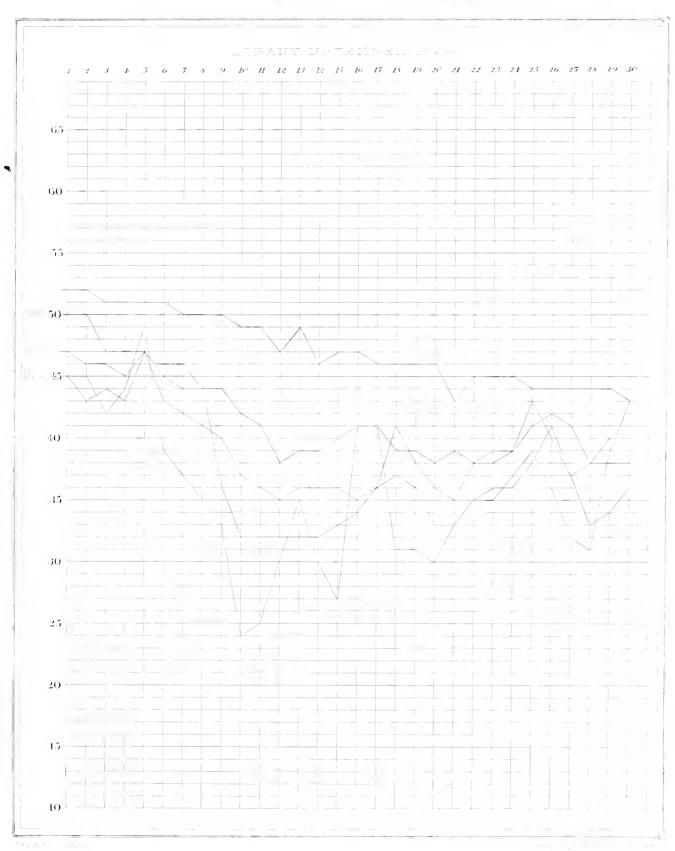


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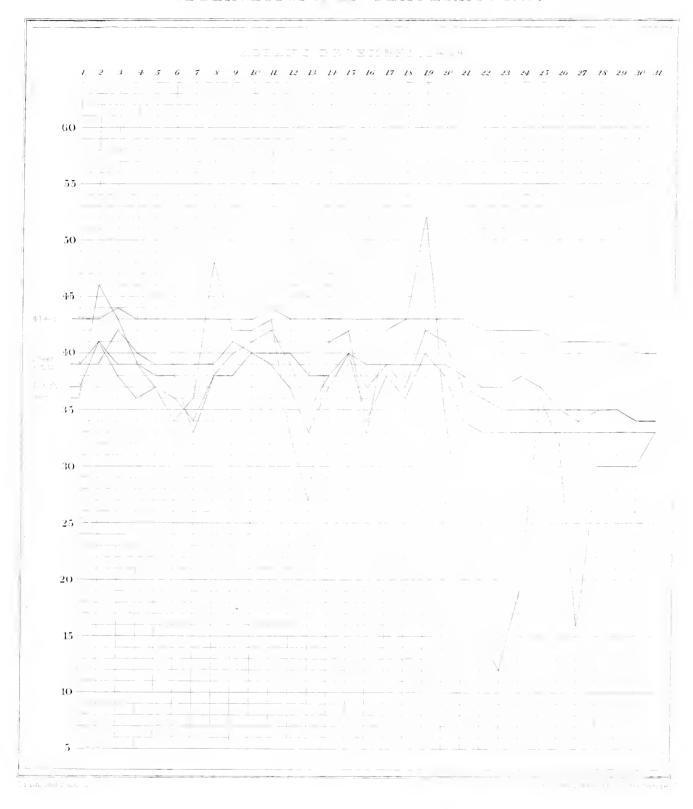




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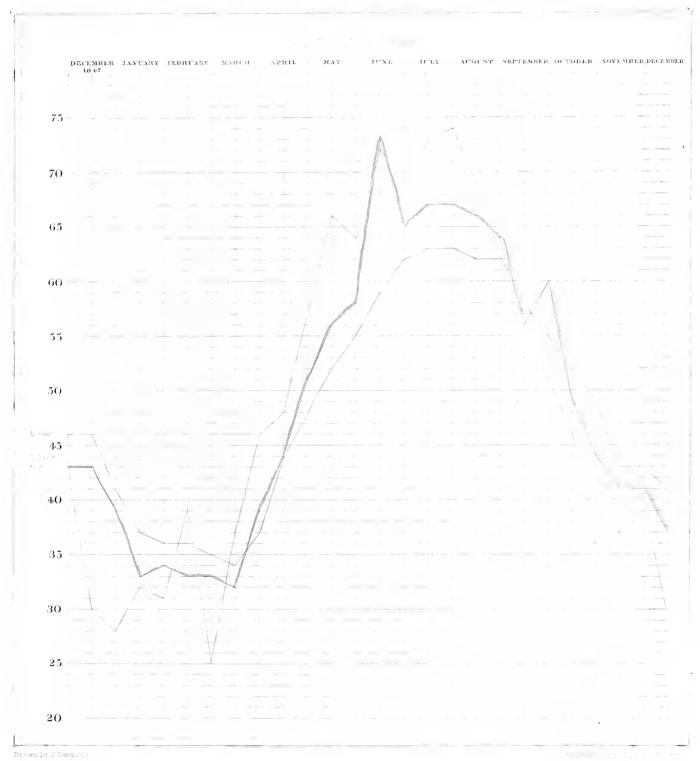


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